

List of Physico-chemical Symbols adopted by the Chemical Society.

[See J.C.S., 1921, 119, 502—512.]

1. *Mathematical Symbols.*

	Usual symbol.	Alternative symbol.
Base of natural (Napierian) logarithms ...	e	
Diameter	d	
Radius	r	
Ratio of circumference to diameter	π	
Summation.....	Σ	
Variation	δ	
Total differential	d	
Partial differential	∂	

2. *Universal Constants.*

Acceleration due to gravity.....	g	
Mechanical equivalent of heat	J	
Avogadro's constant [number of molecules in 1 gram-molecule (mole)]	N	
Gas constant per mole	R	
Faraday's constant (number of coulombs per gram-equivalent of an ion)	F	
Charge on an electron	e	

3. *General Physics and Chemistry.*

Length	l	
Height.....	h	
Mass	m	
Time	t	
Volume	v, V	
Density (mass per unit volume)	d	D
Pressure	p, P	
Concentration	c, C	
Mole fraction	x	
Critical constants : pressure, volume, temperature (centigrade), temperature (absolute), density	$\left\{ \begin{array}{l} p_c, v_c \\ t_c, T_c \end{array} \right.$	
Reduced quantities : pressure, volume, temperature, density		$\left\{ \begin{array}{l} p_r, v_r \\ t_r, T_r, d \end{array} \right.$
van der Waals's constants	a, b	
Fluidity	ϕ	
Viscosity	η	
Surface tension	γ	σ
Diffusion coefficient	Δ	
Atomic weight	A	
Molecular weight	M	
Velocity coefficient of reaction	k	
Equilibrium constant	$K, (K_s, K_p)$	
van 't Hoff coefficient	i	
Degree of dissociation (electrolytic, thermal etc.)	α	

4. Heat and Thermodynamics.

	Usual symbol.	Alternative symbol.
Temperature (centigrade)	t	θ
Temperature (absolute)	T	
Critical temperature	t_c, T_c	
Reduced temperature	t_r, T_r	
Critical solution temperature	t_{cs}, T_{cs}	
Quantity of heat	Q	
Entropy	S	
Specific heat	c	
Specific heat at constant pressure	c_p	
Specific heat at constant volume	c_v	
Ratio of specific heats, $c_p : c_v$	γ	
Molecular heat	C	
Molecular heat at constant pressure	C_p	
Molecular heat at constant volume	C_v	
Latent heat per gram	l	
Latent heat per mole	L	
Maximum work (diminution of free energy)	A	

5. Optics.

Wave-length of light	λ	
Refractive index	n	n_r
Specific refractive power (Gladstone and Dale)	$r_G, [r_G]_\lambda^c$	
Specific refractive power (Lorentz and Lorenz)	$r_L, [r_L]_\lambda^f$	
Molecular refractive power	$\left\{ \begin{array}{l} R_G, R_L \\ [R_G]_\lambda^c, [R_L]_\lambda^f \end{array} \right.$	
Angle of optical rotation	α	
Specific rotatory power	$[\alpha]$	
Molecular rotatory power	$M[\alpha]$	
Specific magnetic rotation	$[\omega]$	
Molecular magnetic rotation	$M[\omega]$	

6. Electricity and Magnetism.

Quantity of electricity	Q	
Current intensity	I	
Resistance	R	W
Electromotive force	E	
Electrode potential, or discharge potential of an ion	E	ϵ
Electrode potential referred to the normal hydrogen or normal calomel electrode respectively, the potential of which is taken as zero	E_h, E_c	ϵ_h, ϵ
Normal potential, <i>i.e.</i> , the electrode potential referred to the normal hydrogen or normal calomel electrode respectively, when the solution is molecular-normal in respect of all participating substances and ions of variable concentration	${}_0E_h, {}_0E_c$	${}^0\epsilon_h, {}^0\epsilon$
Dielectric constant	ϵ	
Conductivity (specific conductance)	κ	
Equivalent conductivity	Λ	
Equivalent conductivity at different dilutions—volumes in litres containing 1 gram-equivalent	$\Lambda_{10}, \Lambda_{100}, \Lambda_{200}$	

6. *Electricity and Magnetism*—(continued).

	Usual symbol.	Alternative symbol.
Equivalent conductivity of cation and of anion	Λ_k, Λ_a	
Equivalent conductivity of specified ions...	Λ_K, Λ_{Cl}	
Molecular conductivity	μ	
Velocity of cation and of anion in cm./sec. when the potential gradient is 1 volt per cm.	U_k, U_a	
Transport number of cation and of anion ...	n_k, n_a	
Magnetic permeability	μ	
Magnetic susceptibility	κ	

List of Symbols, Arranged Alphabetically.

Symbol.	Name of quantity.
<i>A</i>	Atomic weight; maximum work.
<i>a</i>	Van der Waals's constant.
<i>b</i>	Van der Waals's constant.
<i>C</i>	Concentration; molecular heat.
<i>c</i>	Concentration; specific heat.
<i>C_p, C_v</i>	Molecular heat at constant pressure, and at constant volume.
<i>c_p, c_v</i>	Specific heat at constant pressure, and at constant volume.
<i>D</i>	Alternative symbol for density.
<i>d</i>	Diameter; total differential; density.
<i>d_c</i>	Critical density.
<i>d</i>	Reduced density.
<i>E</i>	Electromotive force; electrode potential.
<i>e</i>	Base of Napierian logarithms; charge on an electron.
<i>E_n, E_c</i>	Electrode potential referred to the normal hydrogen or the normal calomel electrode, respectively, the potential of which is taken as zero.
<i>°E_n, °E_c</i>	Normal potential, that is, the electrode potential referred to the normal hydrogen or the normal calomel electrode respectively, when the solution is molecular-normal in respect of all participating substances and ions of variable concentration.
<i>F</i>	Faraday's constant (number of coulombs per gram-equivalent of an ion).
<i>g</i>	Acceleration due to gravity.
<i>h</i>	Height.
<i>I</i>	Current.
<i>i</i>	Van 't Hoff's coefficient.
<i>J</i>	Mechanical equivalent of heat.
<i>K</i>	Equilibrium constant.
<i>K_n, K_p</i>	Equilibrium constant, when molar concentrations and partial pressures respectively are employed.
<i>k</i>	Velocity coefficient of reaction.
<i>L</i>	Latent heat per mole.
<i>l</i>	Length; latent heat per gram.
<i>M</i>	Molecular weight.
<i>M_[α]</i>	Molecular rotatory power.
<i>M_[ω]</i>	Molecular magnetic rotatory power.
<i>m</i>	Mass.
<i>N</i>	Avogadro's constant (Loschmidt's number) or number of molecules in 1 gram-molecule.
<i>n</i>	Refractive index.

List of Symbols, Arranged Alphabetically—(continued).

Symbol.	Name of quantity.
n_{\pm}, n_{θ}	Transport number of cation and of anion.
n_r	Refractive index (alternative symbol).
P	Pressure.
p	Pressure.
p_0, p_r	Critical pressure: reduced pressure.
Q	Quantity of heat; quantity of electricity.
R	Gas constant per mole; electrical resistance.
R_G, R_L	Molecular refractive power, according to Gladstone and Dale, and to Lorentz and Lorenz respectively.
r	Radius.
r_G, r_L	Specific refractive power according to Gladstone and Dale, and to Lorentz and Lorenz respectively.
S	Entropy.
T	Absolute temperature.
T_a	Critical temperature (on the absolute scale).
T_r	Reduced temperature (absolute).
$T_{s,}$	Critical solution temperature (absolute).
t	Time; temperature (centigrade).
t_c	Critical temperature (centigrade).
$t_{s,}$	Critical solution temperature (centigrade).
t_r	Reduced temperature (centigrade).
U_{\pm}, U_a	Velocity of kation and of anion in cm./sec. when the potential gradient is 1 volt per cm.
V	Volume.
v	Volume.
v_0, v_r	Critical volume: reduced volume.
W	Electrical resistance (alternative symbol).
x	Mole fraction.
α	Degree of dissociation (electrolytic, thermal, etc.); angle of optical rotation.
$[\alpha]$	Specific rotatory power.
γ	Surface tension; ratio of specific heats.
Δ	Diffusion coefficient.
δ	Variation.
∂	Partial differential.
ϵ	Electrode potential (alternative symbol); dielectric constant.
ϵ_h, ϵ_r	Electrode potential referred to the normal hydrogen or the normal calomel electrode respectively, the potential of which is taken as zero (alternative symbols).
ϵ^h, ϵ^e	Normal potential, that is, the electrode potential referred to the normal hydrogen or the normal calomel electrode respectively, when the solution is molecular-normal in respect of all participating substances and ions of variable concentration (alternative symbols).
η	Viscosity.
θ	Temperature (centigrade), (alternative symbol).
κ	Specific conductance (conductivity); magnetic susceptibility.
Λ	Equivalent conductivity.
$\Lambda_{10}, \Lambda_{\theta}, \Lambda_{\infty}$	Equivalent conductivity at different dilutions (volumes in litres containing 1 gram-equivalent).
Λ_k, Λ_a	Equivalent conductivity of cation and of anion.
λ	Wave-length of light.
μ	Molecular conductivity; magnetic permeability.
π	Ratio of circumference to diameter.
Σ	Summation.
σ	Surface tension (alternative symbol).
ϕ	Fluidity.
$[\omega]$	Specific magnetic rotation.