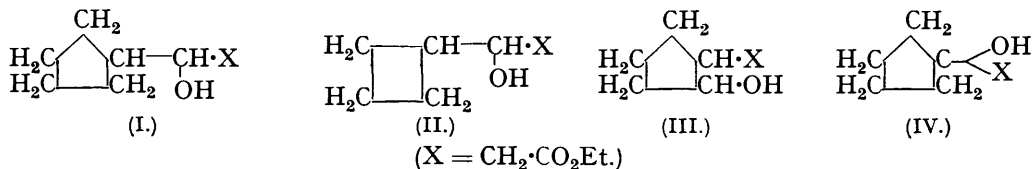


97.—*The Reformatsky Reaction with Compounds of the Ethylene Oxide Type.*

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IN a previous paper (J., 1932, 1778), we investigated the course of the Reformatsky reaction on *cyclohexene oxide*, using ethyl bromoacetate, and stated that possibly the oxide first isomerised to give *cyclopentane aldehyde*, which then reacted to give ethyl β -hydroxy- β -*cyclopentylpropionate* (I).



This has now been found to be the case. *cyclopentane aldehyde* was prepared according to the method of Bedos (*Compt. rend.*, 1929, 189, 255), who, however, gives no experimental details. The Reformatsky reaction with the aldehyde gave (I) as an oil which readily formed a hydrazide, m. p. 154°, identical with that obtained from *cyclohexene oxide*.

It was clearly of interest to repeat the work, *cyclopentene oxide* being used. If this reaction were analogous to the previous one, the product would be the *cyclobutane* derivative (II), but it seemed more probable, however, that the reaction would either give (III) or that the *cyclopentene oxide* would first isomerise in the course of the reaction to *cyclopentanone* and then give ultimately ethyl *cyclopentan-1-ol-1-ylacetate* (IV). The latter alternative has been found to be correct, since when either *cyclopentene oxide* or *cyclopentanone* is treated with zinc and bromoacetic ester the resulting ester (IV) gives in both cases the same crystalline hydrazide, m. p. 143°.

cyclopentane Aldehyde.—12 G. of *cyclohexene oxide* were added to magnesium bromide etherate prepared from Mg (3 g.), Br (20 g.), and Et₂O (200 c.c.) according to the method of Menshutkin (*Z. anorg. Chem.*, 1906, 49, 34). It was necessary to cool the reaction flask. The whole was then kept over-night, the Et₂O removed, and to the cooled residue H₂O was cautiously added. The dark-coloured upper layer was extracted with Et₂O and to the ethereal solution sat. NaHSO₃ aq. was added. After 12 hr., the bisulphite compound was collected, added to K₂CO₃ aq. and, after 2 hr., steam-distilled. The *cyclopentane aldehyde* (2.5 g., b. p. 135°) was obtained from the distillate in the usual way and gave a cryst. semicarbazone, m. p. 123° (Found: C, 54.3; H, 8.3. Calc. for C₇H₁₂ON₃: C, 54.2; H, 8.4%).

Ethyl β-Hydroxy-β-cyclopentylpropionate (I).—*cyclopentane aldehyde* (2 g.), ethyl bromoacetate (4.5 g.), and Zn wool (2 g.) in dry C₆H₆ (15 c.c.) were heated on the water-bath for 3 hr. The usual method of working up (cf. J., 1932, 1782) gave the ester (I) (3.25 g.), b. p. 117°/14 mm. (Found: C, 64.5; H, 9.2. C₁₀H₁₈O₃ requires C, 64.5; H, 9.7%). The ester, mixed with hydrazine hydrate (90—95%), gave a *hydrazide*, needles from H₂O, m. p. 154° alone or mixed with the hydrazide of the product of the Reformatsky reaction on *cyclohexene oxide* (Found: C, 56.2; H, 9.15; N, 16.25. C₈H₁₆O₂N₂ requires C, 55.8; H, 9.3; N, 16.3%).

The Action of Zinc and Ethyl Bromoacetate on cyclopentene Oxide.—The oxide (2 g.), made in the same way as *cyclohexene oxide* ("Organic Syntheses," Vol. 5, p. 35), bromoacetic ester (4.5 g.), and Zn wool (2 g.) in C₆H₆ (15 c.c.) were heated for 3 hr. on the water-bath. The usual method of working up yielded a sweet-smelling ester (1 g.), b. p. 108°/18 mm. (Found: C, 63.25; H, 9.1. Calc. for C₉H₁₆O₃: C, 62.8; H, 9.3%), which gave a *hydrazide*, shining platelets from EtOAc, m. p. 143° (Found: C, 52.8; H, 8.8. C₇H₁₄O₂N₂ requires C, 53.2; H, 8.9%).

Ethyl cyclopentan-1-ol-1-ylacetate (IV).—The reaction was carried out as just described, except that *cyclopentanone* (2 g.) was used instead of *cyclopentene oxide*. The usual method of working up gave ethyl *cyclopentan-1-ol-1-ylacetate*, b. p. 108°/18 mm., which with hydrazine hydrate gave a *hydrazide*, m. p. 143°, identical with that obtained above.