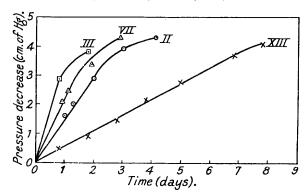
## 158. Surface Reaction between Acetylene and Iodine.

By CECIL P. ELLIS.

According to Berthelot (Ann. Chim. Phys., 1866, 9, 428), iodine and acetylene do not combine at the ordinary temperature even in direct sunlight. This is not correct. If iodine is sublimed on to the sides of a flask, which is then filled with acetylene and kept for two or three days at 25°, most of the iodine crystals are converted into colourless elastic needles of acetylene di-iodide, m. p. 73°, b. p. 192° (corr.).



The rate of the reaction depends to some extent on the nature of the surface used, but not to so marked a degree as in the reaction between ethylene and iodine (see Mooney and Reid, J., 1931, 2597). The rate of reaction is decreased by a film of paraffin wax on the sides of the flask, but is accelerated by a film of alcohol. It is significant that the usual method of preparation of the di-iodide consists in passing acetylene over iodine moistened with alcohol (Sabanejeff, *Annalen*, 1875, 178, 109). The speed of the reaction is slightly greater with sublimed than with powdered iodine.

A 500-c.c. glass flask was used. Before each experiment the flask was immersed in chromic acid, washed, and dried in a current of hot air. The paraffin-wax surface was obtained as described by Mooney and Reid (loc. cit.).

The iodine was weighed and introduced into the flask. The air in the flask was then displaced by acetylene, purified by passage through a solution of copper sulphate containing sulphuric acid and dried over anhydrous calcium chloride. The flask was connected to a mercury manometer, and was completely immersed in a large tank of water. As the temperature in this country varies very little, this arrangement sufficed to keep the temperature of the flask constant within 1° or 2°.

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Typical results are given below, and are illustrated in the accompanying figure. In each case 1 g. of iodine was used, and the surface area was  $300 \text{ cm.}^2$ ; dp/dt gives the rate of change of pressure (in cm. of Hg) with time (in days).

Expt. No.	Iodine surface.	Other surface.	dp/dt.
XIII	Powder	Paraffin	0.55
II	,,	Glass	1.4
VII	Sublimed	,,	$2 \cdot 1$
III	Powder	Alcohol	3.6

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