

**238.** *Synthetical Experiments in the Chromone Group. Part XII.*  
*Synthesis of 7-Hydroxyisoflavone and of  $\alpha$ - and  $\beta$ -Naphthaisoflavone.*

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IN the absence of a solvent and under suitable conditions the reaction between ethyl formate, sodium, and an *o*-hydroxyphenyl benzyl ketone (Joshi and Venkataraman, this vol., p. 513) is a general one, giving an *isoflavone* \* in excellent yield as the only product.

The action of benzyl chloride on 2 : 4-dihydroxyphenyl benzyl ketone led exclusively to the 4-*benzyl* ether, from which 7-*benzyloxyisoflavone* was prepared. Debenzylation of

\* The m. p.'s of *isoflavone* itself recorded in Part XI (*loc. cit.*) should be 132° and 138° instead of 142° and 148° respectively.

this gave 7-hydroxyisoflavone, which was methylated by methyl iodide and potassium carbonate, yielding 7-methoxyisoflavone. The reverse process was carried out, 7-methoxyisoflavone being prepared from 2-hydroxy-4-methoxyphenyl benzyl ketone and demethylated to give 7-hydroxyisoflavone.

3-Phenyl-1:4- $\alpha$ -naphthapyrone and 2-phenyl-1:4- $\beta$ -naphthapyrone (Badhwar and Venkataraman, J., 1932, 2420) also have been prepared by the general method.

## EXPERIMENTAL.

2-Hydroxy-4-benzyloxyphenyl Benzyl Ketone.—A mixture of 2:4-dihydroxyphenyl benzyl ketone (20 g.), anhydrous potassium carbonate (36 g.), benzyl chloride (66 g.), and acetone (200 c.c.) was boiled for 8 hours; the solvent was then recovered, and the residue diluted and distilled with steam to remove the excess of benzyl chloride. The 4-benzyl ether crystallised from alcohol in colourless plates (17 g.), m. p. 111° (Found: C, 79.0; H, 5.5.  $C_{21}H_{18}O_3$  requires C, 79.2; H, 5.6%), which gave a blood-red colour with alcoholic ferric chloride.

7-Benzyloxyisoflavone.—A clear solution of the preceding ether (10 g.) in dry redistilled ethyl formate (150 c.c.), cooled in ice, was added during 1 hour to sodium dust (6 g.), the mixture being kept in ice-salt. After 12 hours, the orange-brown semi-solid mass was treated with ice, and the yellow precipitate was washed and crystallised successively from alcohol, glacial acetic acid, alcohol-acetic acid, and alcohol. The colourless rhombohedral plates (7 g.) obtained, m. p. 171° (Found: C, 80.6; H, 5.0.  $C_{22}H_{16}O_3$  requires C, 80.5; H, 4.9%), gave no coloration with alcoholic ferric chloride and were insoluble in aqueous alkali. They were coloured yellow by sulphuric acid, and the colourless solution exhibited a sky-blue fluorescence.

7-Hydroxyisoflavone.—7-Benzyloxyisoflavone (6 g.) was dissolved in glacial acetic acid (170 c.c.) on the water-bath, concentrated hydrochloric acid (80 c.c.) added in two lots during 1 hour, and the mixture boiled for a few minutes. Dilution with water precipitated a crystalline substance which, recrystallised twice from alcohol, formed thick opaque leaflets (3.4 g.), m. p. 215° (Found: C, 75.5; H, 4.2.  $C_{15}H_{10}O_3$  requires C, 75.6; H, 4.2%). The colourless solution of 7-hydroxyisoflavone in sulphuric acid exhibited a sky-blue fluorescence. The colourless alcoholic solution gave no colour with ferric chloride or sodium amalgam and a pale yellow colour with magnesium and hydrochloric acid. The acetyl derivative crystallised from alcohol in fine lustrous needles, m. p. 139° (Found: C, 73.0; H, 4.2.  $C_{17}H_{12}O_4$  requires C, 72.8; H, 4.2%).

7-Benzyloxyisoflavone was also debenzylated when its solution in sulphuric acid was kept for 3 hours and poured into water.

7-Methoxyisoflavone.—(1) To a boiling solution of 7-hydroxyisoflavone (0.1 g.) in acetone (3 c.c.) and methyl iodide (1 c.c.), anhydrous potassium carbonate (0.5 g.) was added in small portions. After being heated for 5 hours, the mixture was poured into water; the precipitate obtained crystallised from dilute alcohol in colourless plates (0.06 g.), m. p. 156°. (2) A vigorous reaction ensued when a solution of 2-hydroxy-4-methoxyphenyl benzyl ketone (Baker and Robinson, J., 1929, 160) (2 g.) in ethyl formate (50 c.c.) was added to sodium dust (1 g.) cooled in ice-salt. After 12 hours, ice and ether were added, the mixture was shaken, and the ether-ester layer separated, washed, and distilled. The pale brown, oily residue solidified in contact with alcohol and then crystallised from alcohol in colourless, six-sided, irregular plates (1.2 g.), m. p. 156° alone or mixed with the above specimen (Found: C, 76.0; H, 4.7. Calc. for  $C_{16}H_{12}O_3$ : C, 76.1; H, 4.8%). The colourless solution in sulphuric acid exhibited a bright blue fluorescence. When the substance (0.1 g.) was heated with hydriodic acid (*d* 1.7; 2 c.c.) and acetic anhydride (2 c.c.) for 2 hours and poured into sodium bisulphite solution, the product obtained crystallised in leaflets and plates, m. p. (and mixed m. p. with 7-hydroxyisoflavone prepared from the benzyl ether) 215°.

3-Phenyl-1:4- $\alpha$ -naphthapyrone.—The interaction of 2-phenylacetyl-1-naphthol (Cheema and Venkataraman, J., 1932, 920) (10 g.), ethyl formate (120 g.), and sodium (5 g.) produced in 15 minutes a deep orange-yellow solid. Ice was added after 12 hours and the pale yellow solid was separated and crystallised from alcohol, forming a felted mass of colourless needles (3.5 g.), m. p. 187° (Found: C, 83.6; H, 4.3.  $C_{19}H_{12}O_2$  requires C, 83.8; H, 4.4%). The substance was coloured yellow by sulphuric acid and the very pale yellow solution had an intense bluish-green fluorescence; when it was poured into water, the naphthapyrone was recovered unchanged. The substance was also unaffected by boiling acetic anhydride and gave no colour with alcoholic ferric chloride.

*2-Phenyl-1 : 4-β-naphthopyrone*.—The reaction between 1-phenylacetyl-2-naphthol (Chadha, Mahal, and Venkataraman, J., 1933, 1462) (1.6 g.), ethyl formate (50 c.c.), and sodium (1 g.) was somewhat violent and strong cooling in ice-salt and very gradual addition of the ethyl formate solution were essential. The addition of ice after 12 hours gave a sticky solid which, after two crystallisations from warm dilute acetic acid and one from aqueous alcohol, formed very pale cream-coloured clusters of curved needles (0.5 g.), m. p. 129—130°, not depressed by admixture with the naphthopyrone prepared by Badhwar and Venkataraman (*loc. cit.*). The present specimen, however, did not contain difficultly removable water of crystallisation (Found in material dried at 110—115°: C, 83.6; H, 4.7. Calc. for  $C_{19}H_{12}O_2$ : C, 83.8; H, 4.5%). A final crystallisation from alcohol gave long, colourless, prismatic needles, m. p. 133°. The colourless solution in sulphuric acid exhibited a bright, pure blue fluorescence.

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