

306. β -isoPropylnaphthacene.

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THE simplest derivative of 1:2-benzanthracene in which cancer-producing activity is shown to a marked degree is 6-isopropyl-1:2-benzanthracene (Cook, J., 1932, 456). In order to make a comparison between the ring systems of 1:2- and 2:3-benzanthracene, the analogous 6-isopropyl-2:3-benzanthracene (β -isopropylnaphthacene) was synthesised. This hydrocarbon has been applied in benzene solution to the skin of a series of 10 mice, all of which were still alive after 1 year, and 4 of them after 2 years. No tumours were obtained. For these animal experiments I am indebted to Professor E. L. Kennaway, F.R.S. The method used for the synthesis of 6-isopropyl-2:3-benzanthracene was exactly analogous to that described for the preparation of 6- and 7-isopropyl-1:2-benzanthracenes (Cook, *loc. cit.*).

2-Cuminoyl-3-naphthoic Acid—Anhydrous aluminium chloride (18 g) was slowly added, with agitation, to a suspension of naphthalene-2:3-dicarboxylic anhydride (kindly presented by Imperial Chemical Industries, Ltd.), (12 g.) in cumene (40 c.c.). The resulting amber syrup was kept at room temperature for 6 hours with occasional shaking. The keto-acid was isolated in the usual way and recrystallised from benzene (yield, 8 g.). After recrystallisation from acetic acid, *2-cuminoyl-3-naphthoic acid* formed colourless needles, m. p. 218—219°, giving a carmine solution in concentrated sulphuric acid (Found : C, 79.1; H, 5.65. $C_{21}H_{18}O_3$ requires C, 79.2; H, 5.7%).

2-p-isoPropylbenzyl-3-naphthoic Acid.—A solution of the keto-acid (6 g.) in *N*-potassium hydroxide solution (300 c.c.) was boiled with zinc dust (15 g.) for 24 hours and filtered hot. On cooling, the filtrate gave a colourless precipitate, which was collected, dissolved in water, and the solution acidified. The resulting *2-p-iso-propylbenzyl-3-naphthoic acid* was recrystallised from aqueous alcohol (yield, 4.6 g.) and then benzene, forming colourless needles, m. p. 183°, which gave an orange-red solution in concentrated sulphuric acid (Found : C, 82.8; H, 6.7. $C_{21}H_{20}O_2$ requires C, 82.85; H, 6.6%).

6-isoPropyl-2:3-benz-10-anthrone.—A mixture of the isopropylbenzyl-naphthoic acid (2.5 g.) and anhydrous zinc chloride (7.5 g.) was stirred at 185—190° for $\frac{1}{2}$ hour. After cooling, the brittle yellow mass was powdered, extracted with water and then dilute sodium carbonate solution, and the residual solid recrystallised from alcohol. Recrystallisation of the resulting *6-iso-propyl-2:3-benz-10-anthrone* (1.7 g.) from acetic acid gave almost colourless needles, m. p. 180°, which, like 2:3-benz-9-anthrone (Fieser, *J. Amer. Chem. Soc.*, 1931, 53, 2331), gave a purple solution in alcoholic alkali, rapidly decolorised by air (Found : C, 88.0; H, 6.4. $C_{21}H_{18}O$ requires C, 88.1; H, 6.3%). This anthrone was also obtained by dehydration of the benzyl-naphthoic acid with ice-cold concentrated sulphuric acid (2 hours).

6-isoPropyl-2:3-benzanthraquinone was obtained by oxidising the anthrone with sodium dichromate in boiling acetic acid. It crystallised from alcohol in small yellow needles, m. p. 131—132°, and gave a purple solution in concentrated sulphuric acid (Found : C, 83.9; H, 5.4. $C_{21}H_{16}O_2$ requires C, 84.0; H, 5.4%).

6-isoPropyl-2:3-benzanthracene.—The anthrone (1 g.) was heated on the water-bath for 15 minutes with zinc dust (2 g.), alcohol (50 c.c.), and 25% alcoholic sodium hydroxide (5 c.c.). The yellow solution was filtered from zinc, and treated with concentrated hydrochloric acid (10 c.c.) in order to dehydrate the dihydroanthranol. *6-isoPropyl-2:3-benzanthracene* then crystallised, and was recrystallised from xylene, forming golden-orange leaflets (0.4 g.), m. p. 273—274°, very sparingly soluble in the usual media (Found : C, 93.2; H, 6.6. $C_{21}H_{18}$ requires C, 93.3; H, 6.7%).

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