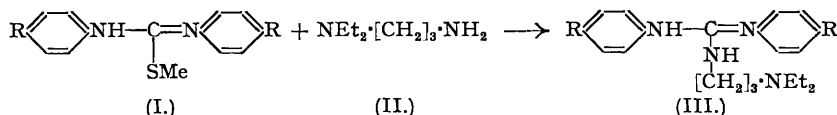


146. Some Derivatives of Diphenylguanidine.

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Certain diaryl-(γ -diethylaminopropyl)guanidines have been synthesised with a view to pharmacological tests.

SEVERAL lines of investigation suggested that diphenylguanidine derivatives containing a second basic centre in the molecule might have interesting pharmacological properties; three such compounds have therefore been synthesised by condensing an appropriate diphenylmethylisothiourea derivative (I) with γ -diethylaminopropylamine (II) to form a guanidine derivative (III) by loss of methylthiol. The necessary isothioureas were prepared by the method of Deck and Dains (*J. Amer. Chem. Soc.*, 1933, **55**, 4986) from the corresponding thiourea and methyl sulphate. 4:4'-Dichloro- (I; R = Cl) and 4:4'-dimethoxy-diphenylisothioureas (I; R = OMe) have not previously been described. Condensation with (II) proceeded smoothly to give excellent yields of the guanidine (III; R = H, Cl or OMe), but only their oxalates could be obtained crystalline, except that diphenyl-(γ -diethylaminopropyl)guanidine (III; R = H) gave also a crystalline picrolonate. For pharmacological test the hydrochlorides were prepared; they were intractable gums which, however, gave satisfactory analyses.



EXPERIMENTAL.

(All compounds were dried in a high vacuum at 60° for analysis.)

N,N'-Diphenyl-N-(γ -diethylaminopropyl)guanidine.—Diphenylmethylisothiourea (9.68 g.) (Deck and Dains, *loc. cit.*) and γ -diethylaminopropylamine (5.20 g.) were dissolved in xylene (30 c.c.) and boiled for 15 hours under reflux. The guanidine oxalate, precipitated in theoretical yield by addition of excess oxalic acid in ethyl acetate, crystallised from alcohol in colourless felted needles, m. p. 99° (Found: C, 55.1; H, 6.5; N, 10.9. $\text{C}_{20}\text{H}_{28}\text{N}_4 \cdot 2\text{C}_2\text{H}_2\text{O}_4 \cdot \text{H}_2\text{O}$ requires C, 55.2; H, 6.5; N, 10.7%). The picrolonate crystallised from acetone after a month and was recrystallised with seeding from alcohol, forming orange rhombs, m. p. 111° with gas-evolution (Found: C, 55.2; H, 5.4; N, 19.1. $\text{C}_{20}\text{H}_{28}\text{N}_4 \cdot 2\text{C}_{10}\text{H}_8\text{O}_5\text{N}_4 \cdot \text{H}_2\text{O}$ requires C, 55.2; H, 5.3; N, 19.3%). Other derivatives could not be obtained crystalline.

Dianisylmethylisothiourea.—Prepared in 88% yield by the method of Deck and Dains (*loc. cit.*), the isothiourea crystallised from benzene-light petroleum in large polyhedra, m. p. 84–85° (Found: C, 63.1; H, 5.8; S, 10.1. $\text{C}_{18}\text{H}_{18}\text{O}_2\text{N}_2\text{S}$ requires C, 63.5; H, 6.0; S, 10.6%).

Bis-(*p*-chlorophenyl)methylisothiourea.—Prepared similarly in 89% yield, the isothiourea crystallised from ether-light petroleum in colourless needles, m. p. 133° (Found: C, 53.8; H, 3.9. $\text{C}_{14}\text{H}_{12}\text{N}_2\text{Cl}_2\text{S}$ requires C, 54.0; H, 3.9%).

N,N'-Dianisyl-N-(γ -diethylaminopropyl)guanidine.—Prepared in the same way as the diphenylguanidine and in almost theoretical yield, this guanidine gave an oxalate which crystallised only with extreme difficulty and showed dimorphism; rounded rhombs, m. p. 100–105°, separated first from alcohol, and these gradually changed into pale yellow prisms, m. p. 150° with gas-evolution (Found: C, 50.6; H, 7.0; N, 9.3. $\text{C}_{22}\text{H}_{32}\text{O}_2\text{N}_4 \cdot 2\text{C}_2\text{H}_2\text{O}_4 \cdot 3\text{H}_2\text{O}$ requires C, 50.5; H, 6.8; N, 9.1%). No other derivative could be obtained crystalline; the hydrochloride was an extremely hygroscopic gum (Found: C, 56.9; H, 8.0; N, 12.2. $\text{C}_{22}\text{H}_{32}\text{O}_2\text{N}_4 \cdot 2\text{HCl} \cdot \frac{1}{2}\text{H}_2\text{O}$ requires C, 56.7; H, 7.5; N, 12.0%).

N,N'-Bis-(*p*-chlorophenyl)-N-(γ -diethylaminopropyl)guanidine.—Prepared similarly, the guanidine oxalate crystallised with great reluctance from alcohol in feathery, cream-coloured needles, m. p. 166–168° with gas-evolution (Found: C, 49.4; H, 5.7; N, 9.4. $\text{C}_{20}\text{H}_{28}\text{N}_4\text{Cl}_2 \cdot 2\text{C}_2\text{H}_2\text{O}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$ requires C, 49.5; H, 5.3; N, 9.6%). No other derivative could be obtained crystalline; the hydrochloride was an exceptionally hygroscopic gum (Found: C, 50.9; H, 6.6; N, 11.7. $\text{C}_{20}\text{H}_{28}\text{N}_4\text{Cl}_2 \cdot 2\text{HCl} \cdot \frac{1}{2}\text{H}_2\text{O}$ requires C, 51.0; H, 6.1; N, 11.9%).

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