

372. Synthetic Antimalarials. Part XLIII. Some Dithiobiurets and 1 : 2 : 4-Triazoles related to "Paludrine."

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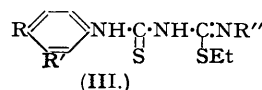
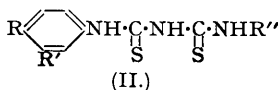
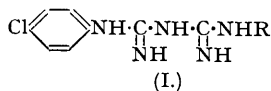
1-*p*-Chlorophenyl-5-isopropylidithiobiuret and a number of related substances of type (II) have been prepared by reaction of the corresponding 1-aryl-4-ethyl-5-alkyl-4-isodithiobiurets (III) with sodium hydrogen sulphide, and examined for antimalarial activity. Unlike the analogous diguanides described in Part X (Curd and Rose, *J.*, 1946, 729), none exhibited any activity.

Using the 5-isopropyl compound as an example, it has been shown that the primary ethylation product of the 1-aryl-5-alkyldithiobiurets, like that of the 1-aryldithiobiurets (Johnson, *Amer. Chem. J.*, 1903, **30**, 167) is the 2-*S*-ethyl derivative (VI). Similarly, the 4-*S*-ethyl compounds (III) from which the normal dithiobiurets of type (II) were derived, and which were prepared by condensation of aryl isothiocyanates with *S*-ethyl-*N*-alkylisothioureas, were further ethylated to give 2 : 4-di-*S*-ethyl derivatives (VII). By the action of hydrazine on this type of compound a number of 3-arylamino-5-alkylamino-1 : 2 : 4-triazoles (VIII) were prepared, but despite their obvious structural similarity to the diguanides (I) they were without antimalarial activity. This is discussed.

1-*p*-Chlorophenyl-5-isopropylidithiobiuret and its 2- and 4-*S*-ethyl derivatives were all converted into *N*¹-*p*-chlorophenyl-*N*⁵-isopropylidiguanide by reaction with ammonia and mercuric oxide. The suggestion is made that such reactions proceed *via* intermediate carbodi-imide stages and must accordingly be stepwise.

The investigation also included some miscellaneous work on monothiobiurets, such as the preparation of 1-*p*-chlorophenyl-5-isopropyl-4-thiobiuret (X; R = Pr¹), and the corresponding 5-ethyl compound. Neither showed any antimalarial activity.

THE preceding paper described the synthesis, and examination for antimalarial activity, of a number of guanylurea and biuret analogues of the active *N*¹-aryl-*N*⁵-alkyldiguanides such as "Paludrine" (I; R = Pr¹). In the present investigation this exploration of compounds related to the diguanides has been extended, in the first place to the analogous dithiobiurets (II). When our investigation was complete it came to our notice that a number of 1-aryldithiobiurets had been examined under the American wartime antimalarial programme (Wiselogle, "Survey of Antimalarial Drugs," 1941—1945, p. 656) without the disclosure of any marked antimalarial activity. More recently, the preparation of the exact dithiobiuret analogue (II; R = Cl, R' = H, R'' = Pr¹) of "Paludrine" has been reported (Fullhart, *Iowa State College J. Sci.*, 1947, **22**, 27) but no detailed account has yet appeared and no results of antimalarial tests have so far been published. Dr. Fullhart has, however, kindly sent us details of his work. From this it appears that isopropyl isothiocyanate was condensed with *N*-*p*-chloro-



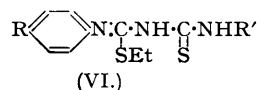
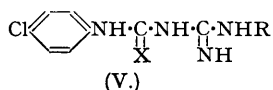
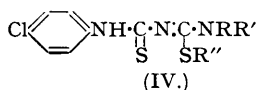
phenyl-*S*-methylisothioureas in aqueous alcoholic solution to give 1-*p*-chlorophenyl-2-methyl-5-isopropyl-2-isodithiobiuret which, without purification, was thiohydrolysed with sodium hydrogen sulphide to give (II; R = Cl, R' = H, R'' = Pr¹). One of our first approaches to this compound was by the same method, but since, in one experiment, we failed to effect the condensation of isopropyl isothiocyanate with *N*-*p*-chlorophenyl-*S*-methylisothioureas in ethereal solution, we turned our attention to the condensation of *p*-chlorophenyl isothiocyanate with *S*-ethyl-*N*-isopropylisothioureas. This was successfully effected in ether, but more conveniently in aqueous-alcoholic solution, to give 1-*p*-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret (III; R = Cl, R' = H, R'' = Pr¹), which was thiohydrolysed by boiling with alcoholic sodium hydrogen sulphide, saturated with hydrogen sulphide, to give 1-*p*-chlorophenyl-5-isopropylidithiobiuret.

Using similar procedures, the following dithiobiurets of type (II) were made: 1-*p*-chlorophenyl-5-methyl-, -5-ethyl-, -5-*n*-propyl-, -5-*n*-butyl-, 1-*m*-chlorophenyl-5-isopropyl-, 1-*p*-bromophenyl-5-isopropyl-, 1-*p*-iodophenyl-5-isopropyl-, and 1-3' : 4'-dichlorophenyl-5-isopropyl-.

That the synthesis had indeed taken the expected course was proved by treating not only 1-*p*-chlorophenyl-5-isopropylidithiobiuret, but also the 4-*S*-ethyl compound (III; R = Cl, R' = H, R'' = Pr¹) from which it was derived, with alcoholic ammonia in presence of mercuric oxide to give, in both cases, *N*¹-*p*-chlorophenyl-*N*⁵-isopropylidiguanide. In a similar manner, 1-*p*-chlorophenyl-5-methylidithiobiuret afforded *N*¹-*p*-chlorophenyl-*N*⁵-methylidiguanide (I; R = Me).

In the conversion of 1-*p*-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret into N^1 -*p*-chlorophenyl- N^5 -isopropylidiguamide it seemed possible that the desulphurising agent was unnecessary for the replacement of the S-ethyl group by NH_2 . This was supported by the observation that 1-*p*-chlorophenyl-4-methyl-4-isodithiobiuret (IV; $R = R' = H$, $R'' = Me$) reacted with isopropylamine, without the intervention of any desulphurising agent, to give *N-p*-chlorophenyl- N' -isopropylguanylthiourea (V; $X = S$, $R = Pr^i$) identical with material prepared by condensation of *p*-chlorophenyl isothiocyanate with isopropylguanidine (see Part XXIX, *J.*, 1948, 1636). Analogously (IV; $R = R' = H$, $R'' = Me$) shaken with aqueous alcoholic methylamine gave *N-p*-chlorophenyl- N' -methylguanylthiourea (V; $R = Me$), and reacted with methanolic ammonia to give *N-p*-chlorophenyl- N' -guanylthiourea (Part XXXVII, this vol., p. 475).

Evidence has been adduced to support the belief that the conversion of *N*-aryl- N' -alkylguanylthioureas and *N*-arylguanyl- N' -alkylthioureas and their respective S-alkyl derivatives into N^1 -aryl- N^5 -alkylidiguamides, described in Parts XXIX and XXX (*J.*, 1948, 1636, 1645), proceeded *via* a carbodi-imide stage, and if the same mechanism operated in the analogous conversion of dithiobiurets of types (II) and (III) into diguanides the reaction must necessarily have been stepwise. Some support for this mechanism is provided by the Indian patent 37,045 which claims the conversion of 1-substituted dithiobiurets into dicyandiamides by reaction with ammonia in presence of a desulphurising agent, and by our failure to convert into diguanides isodithiobiurets of the type (IV; R and $R' = \text{alkyl}$) in which the necessary hydrogen atoms for carbodi-imide formation are not available. The compounds studied were 1-*p*-chlorophenyl-5 : 5-dimethyl-4-ethyl-4-isodithiobiuret (IV; $R = R' = Me$, $R'' = Et$) and 1-*p*-chlorophenyl-5-methyl-4-ethyl-5-isopropyl-4-isodithiobiuret (IV; $R = Me$, $R' = Pr^i$, $R'' = Et$) which were prepared by the action of *p*-chlorophenyl isothiocyanate on *NN*-dimethyl-S-ethylisothiourea



and *N*-methyl-S-ethyl-*N*-isopropylisothiourea respectively, with a view to the preparation of 1-*p*-chlorophenyl-5 : 5-dimethyl- and 1-*p*-chlorophenyl-5-methyl-5-isopropyl-dithiobiuret. However, attempts to effect the thiohydrolysis of (IV; $R = R' = Me$, $R'' = Et$) and (IV; $R = Me$, $R' = Pr^i$, $R'' = Et$) were unsuccessful, apparently because of breakdown of the molecule, since neither starting material nor dithiobiuret could be isolated.

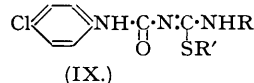
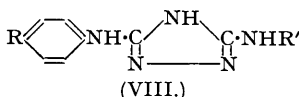
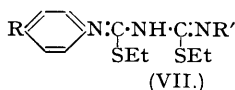
This was unfortunate because we were particularly anxious, as a means of throwing further light on the reaction mechanism discussed above, to study the action of ammonia and a desulphurising agent on 1-*p*-chlorophenyl-5 : 5-dimethyl- and 1-*p*-chlorophenyl-5-methyl-5-isopropyl-dithiobiuret. It seemed possible that, in contrast to the corresponding 4-isodithiobiurets, replacement of the 2-sulphur atom might occur to give the corresponding *N-p*-chlorophenylguanyl- N' -dimethyl- and N' -methyl- N' -isopropyl-thioureas, but that the reactions would not proceed further to give diguanides because of the absence, in either case, of a hydrogen atom on the nitrogen atom carrying the alkyl groups.

The methods used by Fullhart and ourselves for the preparation of 1-aryl-5-monoalkyldithiobiurets were based on the earlier syntheses of 1-aryl- and 1 : 5-diaryl-dithiobiurets by the condensation of aryl isothiocyanates with S-alkylisothioureas and *N*-aryl-S-alkylisothioureas, respectively (cf. Johnson, *Amer. Chem. J.*, 1903, **30**, 167), followed by thiohydrolysis of the resulting 1-aryl- and 1 : 5-diaryl-4-alkyl-4-isodithiobiurets (cf. Underwood and Dains, *Kansas Univ. Sci. Bull.*, 1937, **24**, 5). The reaction of arylamines with perthiocyanic acid (xanthan hydride) to give 1-aryldithiobiurets (for references see Part XXX, *loc. cit.*) cannot be modified to give the 5-alkyl compounds.

Previously, abortive attempts had been made to prepare 1-*p*-chlorophenyl-5-isopropyl-dithiobiuret by the direct interaction of *p*-chlorophenyl isothiocyanate with isopropylthiourea, but no condensation occurred when the reactants were heated together in boiling xylene for $\frac{3}{4}$ hour, in pyridine at 120°, or on fusion together at 170°. Likewise, the reaction of isopropyl isothiocyanate with *p*-chlorophenylthiourea could not be effected.

Tursini (*Ber.*, 1884, **17**, 584) observed that 1-phenyldithiobiuret was alkylated with ethyl iodide in alcohol in presence of ammonia, and Johnson (*loc. cit.*) concluded that the product was 1-phenyl-2-ethyl-2-isodithiobiuret (VI; $R = R' = H$) since it was not identical with the 4-S-ethyl compound (III; $R = R' = R'' = H$) prepared by interaction of phenyl isothiocyanate with S-ethylisothiourea. Further proof has now been provided that alkylation of 1-aryldithiobiurets occurs first in the 2-position by the observation that 1-*p*-chlorophenyl-5-isopropylidithio-

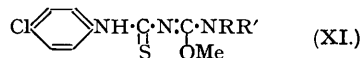
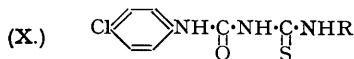
biuret on treatment with ethyl iodide under Tursini's conditions must have given 1-*p*-chlorophenyl-2-ethyl-5-isopropyl-2-isodithiobiuret (VI; R = Cl, R' = Prⁱ), since it was not only different from the 1-*p*-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret described above, but also gave N¹-*p*-chlorophenyl-N⁵-isopropylidiguanide on treatment with alcoholic ammonia in presence of mercuric oxide.



Johnson (*loc. cit.*) has shown that 1 : 5-diaryl-4-alkyl-4-isodithiobiurets are further alkylated with methyl iodide to give 2 : 4-di-S-alkyl derivatives, *i.e.*, 1 : 5-diaryl-2 : 4-dialkyl-2 : 4-diisodithiobiurets. Analogously, further alkylation of the 1-aryl-5-alkyl-4-S-ethylisodithiobiurets of type (III) with ethyl iodide in alcoholic solution in the cold led to compounds of the type (VII), from which we were able to prepare a number of 3-arylamino-5-alkylamino-1 : 2 : 4-triazoles of type (VIII) by reaction with hydrazine (*cf.* the preparation of 3-amino-5-arylamino-1 : 2 : 4-triazoles from 1-aryl-2 : 4-dialkyl-2 : 4-diisodithiobiurets; Underwood and Dains, *loc. cit.*). In general, the intermediate 1-aryl-2 : 4 : 5-trialkyl-2 : 4-diisodithiobiurets were obtained as oils, or low-melting solids, which were difficult to crystallise, although in one case, that of 1-*p*-chlorophenyl-5-isopropyl-2 : 4-diethyl-2 : 4-diisodithiobiuret, the compound was isolated as its *hydriodide*. Usually, therefore, the crude compounds of type (VII) were caused to react directly with hydrazine in alcoholic solution. The compounds of type (VIII) prepared included 3-*p*-chloroanilino-5-isopropylamino-, -5-methylamino-, -5-ethylamino-, and -5-*n*-butylamino-1 : 2 : 4-triazoles and 3-*p*-bromoanilino-5-isopropylamino-1 : 2 : 4-triazole, but none showed any antimalarial activity when tested against *P. gallinaceum* in chicks, despite their apparent close similarity to the diguanides.

In the course of the investigations described in this series of papers we have continually attempted to relate antimalarial activity with chemical structure. Thus in Part XII (*J.*, 1947, 154) the suggestion was made that in the diguanides (I), and in the earlier pyrimidine types from which they were evolved, the possession of antimalarial activity was to be associated with the contribution to a resonance hybrid of the polarised form in which a conjugated path could be traced through alternate carbon and nitrogen atoms from the anilino-nitrogen to the nitrogen atom carrying the terminal alkyl- or dialkylaminoalkyl-group. On this basis the triazoles of type (VIII) might have been expected to exhibit activity.

Another suggestion which has been made (Curd and Rose, *Nature*, 1946, 158, 707) to explain the antimalarial activity of the diguanide (I; R = Prⁱ) was that it might be capable of interfering with a porphyrin-containing enzyme system essential to the malaria parasites, because of a similarity between the copper complex of the drug (containing two molecules of diguanide to one atom of copper, and assuming the symmetrical disposition of the diguanide molecules in a planar structure) and the metal protoporphyrins. Manifestly the triazoles (VIII) cannot form similarly constituted copper complexes, nor can they, through hydrogen-bond resonance such as is possible in the case of the diguanides (*cf.* Gage, Part XXXIV, this vol., p. 221), give a structure simulating a portion of the metal protoporphyrins.



The investigation was terminated with some miscellaneous experiments on monothiobiurets. It has been recorded in the preceding paper that *p*-chlorophenyl isocyanate reacted with *ON*-dialkylisoureas to give 1-*p*-chlorophenyl-4 : 5-dialkyl-4-isobiurets which were hydrolysed to 1-*p*-chlorophenyl-5-alkylbiurets. Analogously, the same isocyanate reacted with *NS*-dialkylisothioureas, for instance with *NS*-diethylisothiourea and with *S*-ethyl-*N*-isopropylisothiourea, to give respectively 1-*p*-chlorophenyl-4 : 5-diethyl-4-isothiobiuret (IX; R = R' = Et) and 1-*p*-chlorophenyl-4-ethyl-5-isopropyl-4-isothiobiuret (IX; R = Prⁱ, R' = Et), which on treatment with a boiling alcoholic solution of sodium hydrogen sulphide afforded the corresponding monothiobiurets, 1-*p*-chlorophenyl-5-ethyl- (X; R = Et) and -5-isopropyl-4-thiobiuret (X; R = Prⁱ). Neither of these compounds exhibited any activity when tested against *P. gallinaceum* in chicks. Their constitution followed from the method of synthesis, but additional proof was provided by reaction of (X; R = Prⁱ) with ammonia in presence of mercuric oxide whereby it was converted into *N*-*p*-chlorophenyl-*N'*-isopropylguanylylurea (V; X = O,

R = Prⁱ), identical with the compound made by condensation of *p*-chlorophenyl isocyanate with isopropylguanidine (preceding paper).

It has been reported (Lakra and Dains, *J. Amer. Chem. Soc.*, 1929, **51**, 220) that 1-phenyl-4-thiobiuret was converted into 1-phenylbiuret by the action of lead monoxide in alcohol, but when 1-*p*-chlorophenyl-5-isopropyl-4-thiobiuret was treated with mercuric oxide in methanol it was converted into 1-*p*-chlorophenyl-4-methyl-5-isopropyl-4-isobiuret described in the preceding paper. To account for this it is suggested that desulphurisation to give a carbodiimide first occurs, which then adds on the elements of methanol.

Just as phenyl isocyanate was condensed with *N-p*-chlorophenyl-*O*-methylisourea to give 1-phenyl-5-*p*-chlorophenyl-4-methyl-4-isobiuret (Part XLII, *loc. cit.*), *p*-chlorophenyl isothiocyanate was condensed with the same compound to give 1 : 5-*di-p*-chlorophenyl-4-methyl-4-iso-2-thiobiuret (XI; R = H, R' = C₆H₄Cl-*p*) and with the appropriate *O*-methyl-*N*-alkyl- or *NN*-dialkyl-isourea to give 1-*p*-chlorophenyl-4 : 5-dimethyl-5-isopropyl-, -4-methyl-5 : 5-*di-n*-butyl-, and -4-methyl-5-*n*-butyl-4-iso-2-thiobiuret (XI; R = H, R' = Buⁿ). None of these compounds was hydrolysed to the corresponding monothiobiuret in view of the inactivity of the compounds of type (X), but it was also shown that *p*-chlorophenyl isocyanate condensed with *N-p*-chlorophenyl-*S*-methylisothiurea to give 1 : 5-*di-p*-chlorophenyl-4-methyl-4-iso-4-thiobiuret (IX; R = C₆H₄Cl-*p*, R' = Me). Fullhart (*loc. cit.*) has reported the analogous condensation of phenyl isocyanate with *N-p*-chlorophenyl-*S*-methylisothiurea and thiohydrolysis of the resulting 1-phenyl-5-*p*-chlorophenyl-4-methyl-4-iso-4-thiobiuret to give 1-phenyl-5-*p*-chlorophenyl-4-thiobiuret.

EXPERIMENTAL.

1-*p*-Chlorophenyl-4-ethyl-5-isopropyl-4-isothiobiuret (III; R = Cl, R' = H, R'' = Prⁱ).—*S*-Ethyl-*N*-isopropylisothiurea hydrobromide (22.8 g.) (Part XXXVII, *loc. cit.*) was dissolved in a solution of potassium hydroxide (5.6 g.) in water (50 c.c.), and a solution of *p*-chlorophenyl isothiocyanate (16.8 g.) in alcohol (50 c.c.) added. The mixture was stirred for 20 hours. The oil first formed gradually solidified. The product was then collected, dried, and crystallised from light petroleum (b. p. 100—120°) to give 1-*p*-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret (6281) as colourless prisms, m. p. 122—124° (Found : C, 49.8; H, 6.1; N, 13.3; S, 20.8. C₁₃H₁₈N₃ClS₂ requires C, 49.4; H, 5.7; N, 13.3; S, 20.3%).

1-*p*-Chlorophenyl-5-methyl-4-ethyl-4-isodithiobiuret.—*N*-Methylthiurea (16.5 g.) was dissolved in alcohol (100 c.c.), ethyl bromide (22 g.) added, and the solution boiled under reflux for 20 hours. The reaction mixture was evaporated to dryness under reduced pressure and the residue stirred with ethyl acetate. The resulting crystalline *N*-methyl *S*-ethylisothiurea hydrobromide was collected and dried in a vacuum (yield, 31.35 g.). Because of its hygroscopic nature it was used directly, without further purification or analysis, for reaction with *p*-chlorophenyl isothiocyanate, as described above, to give 1-*p*-chlorophenyl-5-methyl-4-ethyl-4-isodithiobiuret as colourless flat prisms, m. p. 104—105° (Found : C, 46.0; H, 5.0; N, 14.7. C₁₁H₁₄N₃ClS₂ requires C, 45.9; H, 4.9; N, 14.6%) (6355).

1-*p*-Chlorophenyl-4 : 5-diethyl-4-isodithiobiuret.—Prepared similarly from *NS*-diethylisothiurea hydrobromide (Part XXXVII, *loc. cit.*) and *p*-chlorophenyl isothiocyanate, this biuret crystallised from light petroleum (b. p. 80—100°) as colourless prisms, m. p. 110—112° (Found : C, 47.7; H, 5.4; N, 13.7; S, 22.0. C₁₂H₁₆N₃ClS₂ requires C, 47.8; H, 5.3; N, 13.9; S, 21.3%) (6356).

1-*p*-Chlorophenyl-4-ethyl-5-*n*-butyl-4-isothiobiuret.—*N-n*-Butylthiurea was treated with ethyl bromide as described above for *N*-methylthiurea. The resulting crude *S*-ethyl-*N-n*-butylisothiurea hydrobromide (12.1 g.) was dissolved in a solution of potassium hydroxide (2.8 g.) in water (50 c.c.), and a solution of *p*-chlorophenyl isothiocyanate (8.5 g.) in alcohol (50 c.c.) added. The mixture was stirred at room temperature for 4 hours during which the oil first precipitated gradually solidified. The product was collected, washed with water, and dried in a vacuum. Crystallisation from light petroleum (b. p. 80—100°) gave 1-*p*-chlorophenyl-4-ethyl-5-*n*-butyl-4-isodithiobiuret as colourless rhombs, m. p. 77—79° (Found : C, 51.2; H, 5.9; N, 12.3. C₁₄H₂₀N₃ClS₂ requires C, 51.0; H, 6.1; N, 12.7%) (6569).

1-*m*-Chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, similarly prepared from *m*-chlorophenyl isothiocyanate (Hofmann, *Ber.*, 1880, **13**, 14) and *S*-ethyl-*N*-isopropylisothiurea hydrobromide, crystallised from light petroleum (b. p. 60—80°) as colourless prisms, m. p. 80—81° (Found : C, 48.9; H, 5.5; N, 13.8. C₁₃H₁₈N₃ClS₂ requires C, 49.4; H, 5.7; N, 13.3%) (6459).

1-*p*-Bromophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, prepared analogously but from *p*-bromophenyl isothiocyanate, crystallised from light petroleum (b. p. 100—120°) as colourless elongated prisms, m. p. 129—130° (Found : C, 43.2; H, 5.0; N, 11.6. C₁₃H₁₈N₃BrS₂ requires C, 43.3; H, 5.0; N, 11.7%).

1-*p*-Iodophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, prepared similarly from *p*-iodophenyl isothiocyanate (see Part XXIX, *loc. cit.*) and *S*-ethyl-*N*-isopropylisothiurea hydrobromide, crystallised from light petroleum (b. p. 100—120°) as colourless parallelepipeds, m. p. 124—125° (Found : C, 38.7; H, 5.0; N, 10.5. C₁₃H₁₈N₃I₂ requires C, 38.3; H, 4.5; N, 10.3%) (6492).

1-3' : 4'-Dichlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, prepared in a similar manner from 3 : 4-dichlorophenyl isothiocyanate (Dyson, George, and Hunter, *J.*, 1926, 3041), separated from light petroleum (b. p. 80—100°) as colourless laminae, m. p. 104° (Found : C, 44.7; H, 4.9; N, 11.7. C₁₃H₁₇N₃Cl₂S₂ requires C, 44.55; H, 4.9; N, 12.0%).

1 : 5-*Di-p*-chlorophenyl-4-methyl-4-isodithiobiuret (IV; R = H, R' = C₆H₄Cl-*p*, R'' = Me).—*N-p*-Chlorophenyl-*S*-methylisothiurea (6.68 g.) (Part XXV, Crowther, Curd, and Rose, *J.*, 1948, 586) was dissolved in dry ether (50 c.c.), and a solution of *p*-chlorophenyl isothiocyanate (5.65 g.) in dry ether

(50 c.c.) added. After standing for 3 days the solvent was evaporated and the residue crystallised first from chlorobenzene and then from 2-ethoxyethanol to give the product as colourless flat prisms, m. p. 158—160° (Found : C, 48.5; H, 3.6; N, 11.6; S, 17.8. Calc. for $C_{15}H_{13}N_3Cl_2S_2$: C, 48.6; H, 3.5; N, 11.3; S, 17.3%) (6254) (Fullhart, *loc. cit.*, gives m. p. 154—155°).

1-*p*-Chlorophenyl-5-isopropylthiobiuret (II; R = Cl, R' = H, R'' = Pr).—Sodium (0.92 g.) was dissolved in alcohol (25 c.c.), and hydrogen sulphide passed into the solution until the gain in weight was 1.36 g. This solution was added to a solution of 1-*p*-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret (6.3 g.) in alcohol (50 c.c.), and the mixture boiled under reflux for 1 hour while hydrogen sulphide passed through it. The hot reaction mixture was then carbon-treated, filtered, and acidified with acetic acid. On cooling, the product crystallised and was collected and recrystallised from light petroleum (b. p. 100—120°); colourless, long, flat prisms, m. p. 148° (Found : C, 45.8; H, 5.0; N, 14.4; S, 22.4. Calc. for $C_{11}H_{14}N_3ClS_2$: C, 45.9; H, 4.9; N, 14.6; S, 22.3%) (6337) (Fullhart, *loc. cit.*, quotes m. p. 135—137°).

1-*p*-Chlorophenyl-5-methylthiobiuret, prepared similarly by thiohydrolysis of 1-*p*-chlorophenyl-5-methyl-4-ethyl-4-isodithiobiuret, crystallised from toluene as colourless plates, m. p. 164—166° (Found : C, 42.0; H, 4.1; N, 16.1; S, 25.0. $C_9H_{10}N_3ClS_2$ requires C, 41.6; H, 3.9; N, 16.2; S, 24.7%) (6406).

1-*p*-Chlorophenyl-5-ethylthiobiuret, similarly prepared from 1-*p*-chlorophenyl-4:5-diethyl-4-isodithiobiuret, crystallised from benzene as colourless minute prisms, m. p. 158—160° (Found : C, 43.8; H, 4.5; N, 15.4. $C_{10}H_{12}N_3ClS_2$ requires C, 43.85; H, 4.4; N, 15.3%) (6587).

1-*p*-Chlorophenyl-5-*n*-propylthiobiuret.—*N*-*n*-Propylthiourea (18.9 g.) (Hecht, *Ber.*, 1890, **23**, 283) in alcohol (200 c.c.) was heated under reflux with ethyl bromide (20 g.) overnight, and the reaction mixture evaporated to dryness under reduced pressure. Without further purification the resulting *S*-ethyl-*N*-*n*-propylthiourea hydrobromide was treated with potassium hydroxide to liberate the base, which was brought into reaction with *p*-chlorophenyl isothiocyanate in aqueous-alcoholic solution, as described above in similar cases, to give 1-*p*-chlorophenyl-4-ethyl-5-*n*-propyl-4-isodithiobiuret. Complete purification of this substance was difficult and so the crude material was thiohydrolysed in the usual way to give 1-*p*-chlorophenyl-5-*n*-propylthiobiuret, which crystallised from light petroleum (b. p. 100—120°) as a colourless microcrystalline powder, m. p. 140—141° (Found : C, 46.3; H, 4.5; N, 14.7. $C_{11}H_{14}N_3ClS_2$ requires C, 45.9; H, 4.9; N, 14.6%) (6813).

1-*p*-Chlorophenyl-5-*n*-butylthiobiuret, prepared as described for the corresponding 5-isopropyl compound but from 1-*p*-chlorophenyl-4-ethyl-5-*n*-butyl-4-isodithiobiuret, crystallised from light petroleum (b. p. 100—120°) as colourless needles, m. p. 132—133° (Found : C, 47.7; H, 5.2; N, 13.8. $C_{12}H_{16}N_3ClS_2$ requires C, 47.8; H, 5.3; N, 13.9%) (6589).

1-*m*-Chlorophenyl-5-isopropylthiobiuret, similarly prepared from 1-*m*-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, crystallised from benzene-light petroleum (b. p. 60—80°) as colourless laminae, m. p. 128—129° (Found : C, 46.0; H, 5.5; N, 15.0. $C_{11}H_{14}N_3ClS_2$ requires C, 45.9; H, 4.9; N, 14.6%) (6491).

1-*p*-Bromophenyl-5-isopropylthiobiuret, prepared, as described for 6337, from the corresponding 4-*S*-ethyl compound, separated from benzene-light petroleum (b. p. 60—80°) as colourless flat prisms, m. p. 150—152° (Found : C, 39.8; H, 4.3; N, 12.4. $C_{11}H_{14}N_3BrS_2$ requires C, 39.8; H, 4.2; N, 12.6%) (6493).

1-*p*-Iodophenyl-5-isopropylthiobiuret, prepared similarly from 1-*p*-iodophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, crystallised from benzene-light petroleum (b. p. 60—80°) as colourless elongated prisms, m. p. 136—138° (Found : C, 34.9; H, 3.9; N, 11.1. $C_{11}H_{14}N_3IS_2$ requires C, 34.8; H, 3.7; N, 11.1%) (6568).

1-3':4'-Dichlorophenyl-5-isopropylthiobiuret, prepared in an analogous manner from 1-3':4'-dichlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, crystallised from benzene as colourless felted needles, m. p. 158° (Found : C, 41.4; H, 4.1; N, 12.8. $C_{11}H_{13}N_3Cl_2S_2$ requires C, 41.0; H, 4.1; N, 13.0%).

Reaction of 1-p-Chlorophenyl-5-isopropylthiobiuret with Ammonia and Mercuric Oxide.—1-*p*-Chlorophenyl-5-isopropylthiobiuret (2.85 g.), mercuric oxide (8.68 g.), and saturated alcoholic ammonia (100 c.c.) were stirred together at room temperature for 20 hours. The reaction mixture was then filtered and evaporated to dryness under diminished pressure. The residue was dissolved in acetone, and the solution made acid to litmus with acetic acid. The crystalline material precipitated was collected and crystallised from alcohol-acetone to give *N*¹-*p*-chlorophenyl-*N*⁵-isopropylthiobiuret acetate, m. p. and mixed m. p. 184—185°.

Reaction of 1-p-Chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret with Ammonia and Mercuric Oxide.—1-*p*-Chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret (3.15 g.), mercuric oxide (8.68 g.), and saturated alcoholic ammonia (50 c.c.) were stirred together at 30—35° for 20 hours, and the mixture filtered and evaporated to dryness under reduced pressure. The residue was extracted with warm 2*N*-hydrochloric acid, the extract carbon-treated, filtered, and made just alkaline to brilliant-yellow with ammonia. The precipitated solid was collected and crystallised from water to give *N*¹-*p*-chlorophenyl-*N*⁵-isopropylthiobiuret hydrochloride, m. p. 243—244°, undepressed in admixture with an authentic specimen.

Reaction of 1-p-Chlorophenyl-5-methylthiobiuret with Ammonia and Mercuric Oxide.—The dithiobiuret (2.6 g.), mercuric oxide (8.68 g.), and saturated alcoholic ammonia (100 c.c.) were stirred at 30—35° for 20 hours and the mixture worked up as in the preceding experiment to give *N*¹-*p*-chlorophenyl-*N*⁵-methylthiobiuret hydrochloride, m. p. 227—228° either alone or in admixture with authentic material.

Reaction of 1-p-Chlorophenyl-4-methyl-4-isodithiobiuret with isoPropylamine.—1-*p*-Chlorophenyl-4-methyl-4-isodithiobiuret (1 g.) (Part XXX, *loc. cit.*) and isopropylamine (5 c.c.) were heated together at 60° for 2 hours. The mixture was then poured into water and kept until the product solidified. Collected and crystallised from benzene, it formed colourless small flat prisms, m. p. 143°, undepressed in admixture with *N*-*p*-chlorophenyl-*N*⁵-isopropylthioureia prepared as described in Part XXIX (*loc. cit.*).

Reaction of 1-p-Chlorophenyl-4-methyl-4-isodithiobiuret with Methylamine.—The dithiobiuret (6 g.), methylamine (20 c.c. of 21% aqueous solution), and alcohol (10 c.c.) were shaken together overnight at room temperature. The starting material gradually dissolved and was replaced by a granular product.

This was collected, washed with water, dried, and crystallised from ethyl acetate-light petroleum (b. p. 60—80°) to give *N-p-chlorophenyl-N'-methylguanylthiourea* as colourless rhombic crystals, m. p. 130—132° (Found: C, 44·3; H, 4·3; S, 13·2. $C_9H_{11}N_4ClS$ requires C, 44·5; H, 4·6; S, 13·2%).

Reaction of 1-p-Chlorophenyl-4-methyl-4-isodithiobiuret with Ammonia.—1-*p-Chlorophenyl-4-methyl-4-isodithiobiuret* (2 g.) and saturated methanolic ammonia (20 c.c.) were heated at 50—60° for 10 hours. The mixture was then filtered and the filtrate poured into water. The precipitated crystalline material was collected, dried, and crystallised first from benzene-alcohol and then from xylene-butanol to give *N-p-chlorophenyl-N'-guanylthiourea*, m. p. 195°, either alone or in admixture with material prepared as in Part XXXVII (*loc. cit.*).

1-*p-Chlorophenyl-5:5-dimethyl-4-ethyl-4-isodithiobiuret* (IV; R = R' = Me, R'' = Et).—*NN*-Dimethylthiourea (5·2 g.) (Wallach, *Ber.*, 1899, **32**, 1874), alcohol (150 c.c.), and ethyl bromide (12 g.) were allowed to stand together for 3 days and the solution then evaporated to dryness under reduced pressure. The residual oil readily solidified and consisted substantially of *NN-dimethyl-S-ethylisothiourea hydrobromide*. A sample crystallised from alcohol-ethyl acetate had m. p. 142—143° (Found: C, 27·6; H, 6·0; S, 15·1. $C_5H_{12}N_2S$, HBr requires C, 28·1; H, 6·1; S, 15·0%). 6·89 G. of this were added to a solution of potassium hydroxide (1·68 g.) in water (50 c.c.), followed by a solution of *p*-chlorophenyl isothiocyanate (5·05 g.) in alcohol (50 c.c.). The precipitated product was collected and crystallised, first from alcohol and then from benzene, to give 1-*p-chlorophenyl-5:5-dimethyl-4-ethyl-4-isodithiobiuret* as colourless prisms, m. p. 172—174° (Found: C, 47·8; H, 4·8. $C_{12}H_{16}N_3ClS_2$ requires C, 47·8; H, 5·3%).

N-Methyl-N-isopropylthiourea.—Hydrogen sulphide was passed into a solution of methylisopropylcyanamide (20 g.) (Part XXXIII, Ainley, Curd, and Rose, this vol., p. 98) in alcohol (50 c.c.) saturated with ammonia. When precipitation had visibly ceased, the reaction mixture was kept for 3 days and then evaporated to dryness under reduced pressure. The residual solid crystallised from water to give *N-methyl-N-isopropylthiourea* as colourless leaves, m. p. 100—102° (Found: C, 44·9; H, 8·9; N, 20·7. $C_5H_{12}N_2S$ requires C, 45·4; H, 9·2; N, 21·2%).

Ethyl bromide (6·0 g.) was added to a solution of *N-methyl-N-isopropylthiourea* (6·6 g.) in dry alcohol (50 c.c.), and the mixture kept overnight. After evaporation under reduced pressure, the residue was stirred with ethyl acetate and filtered to give *N-methyl-S-ethyl-N-isopropylisothiourea hydrobromide*, m. p. 130—132° (not analysed).

1-*p-Chlorophenyl-5-methyl-4-ethyl-5-isopropyl-4-isodithiobiuret* (IV; R = Me, R' = Pr^t, R'' = Et).—Prepared from *p*-chlorophenyl isothiocyanate and *N-methyl-S-ethyl-N-isopropylisothiourea hydrobromide*, as described above in similar cases, this *biuret* crystallised from light petroleum (b. p. 100—120°) as colourless prisms, m. p. 126° (Found: C, 51·2; H, 6·1; N, 12·7. $C_{14}H_{20}N_3ClS_2$ requires C, 51·0; H, 6·1; N, 12·7%).

1-*p-Chlorophenyl-2-ethyl-5-isopropyl-2-isodithiobiuret* (VI; R = Cl, R' = Pr^t).—1-*p-Chlorophenyl-5-isopropylidithiobiuret* (8·61 g.), ethyl iodide (5·15 g.), alcohol (150 c.c.), and ammonia (3 c.c.) were mixed and kept for 3 days. Evaporation of the solution under diminished pressure and crystallisation of the residue from alcohol gave the *product*, which was recrystallised from light petroleum (b. p. 60—80°); clusters of colourless prisms, m. p. 82—84° (Found: C, 49·3; H, 5·6; N, 13·7. $C_{13}H_{18}N_3ClS$ requires C, 49·4; H, 5·7; N, 13·3%).

Reaction of 1-p-Chlorophenyl-2-ethyl-5-isopropyl-2-isodithiobiuret with Ammonia and Mercuric Oxide.—The preceding compound (2·55 g.), mercuric oxide (7·1 g.), and saturated alcoholic ammonia (50 c.c.) were stirred together at 30—35° for 20 hours. The mixture was then filtered and evaporated to dryness under diminished pressure. The residue was extracted with hot 2*N*-hydrochloric acid, and the extract filtered from a little insoluble matter and made faintly alkaline to brilliant-yellow with ammonia. Addition of a little salt induced separation of the product, which was collected, crystallised from water, and identified as *N¹-p-chlorophenyl-N²-isopropylidiguamide hydrochloride*, m. p. and mixed m. p. 243—244°.

1-*p-Chlorophenyl-2:4-diethyl-5-isopropyl-2:4-diisodithiobiuret* (VII; R = Cl, R' = Pr^t).—A solution of 1-*p-chlorophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret* (6·3 g.) in alcohol (75 c.c.) was treated with ethyl iodide (6·24 g.) and the mixture kept for 4 days. It was then warmed until homogeneous, more ethyl iodide (6·24 g.) and alcohol (25 c.c.) added, and the whole set aside for a further 24 hours and then evaporated to dryness under diminished pressure. The residue solidified on standing overnight and was crystallised from alcohol-ether to give 1-*p-chlorophenyl-2:4-diethyl-5-isopropyl-2:4-diisodithiobiuret hydriodide* as colourless prisms, m. p. 134—136° (Found: C, 37·9; H, 4·5; N, 9·4. $C_{15}H_{22}N_3ClS$, HI requires C, 38·2; H, 4·9; N, 8·9%) (6595).

The corresponding base liberated from the hydriodide by stirring with sodium carbonate solution and isolated by extraction with ether formed a low-melting (*ca.* 40°) solid.

3-*p-Chloroanilino-5-isopropylamino-1:2:4-triazole* (VIII; R = Cl, R' = Pr^t).—The above base (2·75 g.) was dissolved in alcohol (20 c.c.), hydrazine hydrate (2·2 c.c.) added, and the mixture heated under reflux for 4 hours. The *triazole* which separated on cooling and dilution with water was collected and crystallised from benzene; colourless laminae, m. p. 162—164° (Found: C, 52·5; H, 5·5; N, 27·4. $C_{11}H_{14}N_5Cl$ requires C, 52·5; H, 5·6; N, 27·8%) (6488).

3-*p-Chloroanilino-5-methylamino-1:2:4-triazole.*—1-*p-Chlorophenyl-5-methyl-4-ethyl-4-isodithiobiuret* was treated with ethyl iodide in alcohol as described above to give 1-*p-chlorophenyl-5-methyl-2:4-diethyl-2:4-diisodithiobiuret* as an oil. This oil (8·58 g.) was boiled under reflux for 4 hours with alcohol (50 c.c.) containing hydrazine hydrate (7·9 c.c. of 35%). The product, which separated on cooling, was collected and crystallised from aqueous 2-ethoxyethanol to give the *triazole* as colourless elongated prisms, m. p. 261—263° (Found: C, 48·4; H, 4·6; N, 30·9. $C_9H_{10}N_5Cl$ requires C, 48·3; H, 4·5; N, 31·3%) (6442).

3-*p-Chloroanilino-5-ethylamino-1:2:4-triazole*, prepared similarly from 1-*p-chlorophenyl-4:5-diethyl-4-isodithiobiuret* by *S*-ethylation followed by reaction with hydrazine, crystallised from aqueous 2-ethoxyethanol as colourless elongated flat prisms, m. p. 228—230° (Found: C, 50·5; H, 4·7; N, 29·2. $C_{10}H_{12}N_5Cl$ requires C, 50·5; H, 5·1; N, 29·5%) (6588).

3-*p-Chloroanilino-5-n-butylamino-1:2:4-triazole*, similarly prepared from 1-*p-chlorophenyl-4-ethyl-*

5-*n*-butyl-4-isodithiobiuret, separated from alcohol in the form of elongated flat prisms, m. p. 196—198° (Found: C, 54.1; H, 5.7; N, 26.7. $C_{12}H_{16}N_2Cl$ requires C, 54.2; H, 6.1; N, 26.4%) (6614).

3-*p*-Bromoanilino-5-isopropylamino-1:2:4-triazole, likewise synthesised from 1-*p*-bromophenyl-4-ethyl-5-isopropyl-4-isodithiobiuret, crystallised from chlorobenzene as colourless prisms, m. p. 154—156° (Found: C, 45.0; H, 4.7; N, 23.4. $C_{11}H_{14}N_5Br$ requires C, 44.6; H, 4.8; N, 23.6%) (6626).

1-*p*-Chlorophenyl-4:5-diethyl-4-isothiobiuret (IX; R = R' = Et).—A solution of potassium hydroxide (3.7 g.) in water (20 c.c.) was added to a solution of *N*S-diethylisothioureia hydrobromide (15.16 g.) in water (80 c.c.), and the mixture extracted with ether. To the dried ethereal solution a solution of *p*-chlorophenyl isocyanate (9.2 g.) in ether (50 c.c.) was added. When the ensuing reaction was complete the solvent was evaporated off, and the residue crystallised from light petroleum (b. p. 80—100°) to give the *thiobiuret* as colourless elongated prisms, m. p. 122—124° (Found: C, 50.2; H, 5.4; N, 15.3. $C_{12}H_{16}ON_3ClS$ requires C, 50.4; H, 5.6; N, 14.7%) (6427).

The 4-ethyl-5-isopropyl analogue, prepared similarly from *p*-chlorophenyl isocyanate and *S*-ethyl-*N*-isopropylisothioureia hydrobromide, separated from light petroleum (b. p. 80—100°) as colourless thin prisms, m. p. 132° (Found: C, 51.5; H, 5.8; N, 13.9. $C_{13}H_{18}ON_3ClS$ requires C, 52.1; H, 6.0; N, 14.0%) (6408).

1-*p*-Chlorophenyl-5-ethyl-4-*thiobiuret* (X; R = Et).—Sodium (0.9 g.) was dissolved in alcohol (50 c.c.), and hydrogen sulphide passed through the mixture until the gain in weight was 1.4 g. The resulting solution was added to a suspension of (IX; R = R' = Et) (6 g.) in alcohol (50 c.c.), and the mixture boiled under reflux for 1 hour with the passage of hydrogen sulphide. The resulting solution was carbon-treated, filtered, and acidified with acetic acid. Addition of water precipitated the *thiobiuret*, which was collected, washed with water, dried, and crystallised from chlorobenzene; colourless flat needles, m. p. 184° (efferv.) (Found, in material dried in a vacuum at 80°: C, 46.3; H, 4.8; N, 16.3. $C_{10}H_{12}ON_3ClS$ requires C, 46.6; H, 4.7; N, 16.3%) (6949).

1-*p*-Chlorophenyl-5-isopropyl-4-*thiobiuret*, similarly prepared from (IX; R = Pr, R' = Et (by thiohydrolysis, separated from benzene as colourless flat needles, m. p. 176—178° (Found: N, 15.5. $C_{11}H_{14}ON_3ClS$ requires N, 15.5%) (8607).

Reaction of 1-*p*-Chlorophenyl-5-isopropyl-4-*thiobiuret* with Ammonia and Mercuric Oxide.—The preceding compound (1.36 g.), mercuric oxide (4.3 g.), and saturated alcoholic ammonia (100 c.c.) were stirred together at 30—35° for 2 hours. After standing overnight, the mixture was filtered and evaporated to dryness. The solid residue crystallised from benzene to give *N-p*-chlorophenyl-*N'*-isopropylguanyleurea, m. p. 130—132° undepressed in admixture with material prepared as described in the preceding paper.

Reaction of 1-*p*-Chlorophenyl-5-isopropyl-4-*thiobiuret* with Methanol and Mercuric Oxide.—A mixture of (X; R = Pr) (1.2 g.), mercuric oxide (2.1 g.), and methanol (100 c.c.) was boiled under reflux for 1 hour, then filtered hot, and the filtrate evaporated to dryness. The residue crystallised from aqueous methanol to give 1-*p*-chlorophenyl-4-methyl-5-isopropyl-4-*isobiuret*, m. p. 82—84° undepressed in admixture with authentic material (Part XLII).

1:5-Di-*p*-chlorophenyl-4-methyl-4-iso-2-*thiobiuret* (XI; R = H, R' = C_6H_4Cl-p).—*p*-Chlorophenyl isothiocyanate (5.1 g.), dissolved in ether (25 c.c.), was added to a solution of *N-p*-chlorophenyl-*O*-methylisoureia (5.5 g.) in ether (25 c.c.), and the whole kept for 3 days; light petroleum (b. p. 40—60°) was then added, and the precipitated solid filtered off. The mother-liquors were evaporated to dryness, leaving a solid residue. The combined solids were crystallised from benzene—light petroleum (b. p. 60—80°) to give the *thiobiuret* as colourless prisms, m. p. 143—145° (Found: C, 51.1; H, 3.6; N, 12.0. $C_{15}H_{13}ON_3Cl_2S$ requires C, 50.85; H, 3.7; N, 11.9%) (6441).

1-*p*-Chlorophenyl-4:5-dimethyl-5-isopropyl-4-iso-2-*thiobiuret*.—*p*-Chlorophenyl isothiocyanate (5.1 g.), *ON*-dimethyl-*N*-isopropylisoureia (3.9 g.), and water (25 c.c.) were mixed and warmed on the steam-bath for 10 minutes. After standing overnight, the solid product was collected, washed with water, and dried. Crystallisation from benzene gave the *thiobiuret* as minute colourless prisms, m. p. 147° (Found: C, 52.3; H, 6.1; N, 14.4. $C_{13}H_{18}ON_3ClS$ requires C, 52.1; H, 6.0; N, 14.0%) (6374).

O-Methyl-*NN*-di-*n*-butylisoureia.—Sodium (8.5 g.) was dissolved in dry methanol (125 c.c.), di-*n*-butylcyanamide (52 g.) added, and the mixture stirred at 50—60° for 2 hours. After cooling and dilution with water, hydrochloric acid was added to render the solution just acid, and the mixture extracted with ether. The aqueous layer was made alkaline with sodium hydroxide, and the precipitated product extracted with benzene. Evaporation of the dried (K_2CO_3) benzene extract then gave *O*-methyl-*NN*-di-*n*-butylisoureia as a colourless oil, b. p. 122°/18 mm. (Found: N, 15.2. $C_{10}H_{22}ON_2$ requires N, 15.0%).

1-*p*-Chlorophenyl-4-methyl-5:5-di-*n*-butyl-4-iso-2-*thiobiuret* (XI; R = R' = Buⁿ), prepared as described for 6374 from *p*-chlorophenyl isothiocyanate and *O*-methyl-*NN*-di-*n*-butylisoureia, crystallised from alcohol as colourless parallelepipeds, m. p. 124—126° (Found: C, 57.4; H, 7.3; N, 12.1. $C_{17}H_{26}ON_3ClS$ requires C, 57.4; H, 7.4; N, 11.8%) (6314).

The *monobutyl* analogue, prepared similarly from *p*-chlorophenyl isothiocyanate and *O*-methyl-*N*-*n*-butylisoureia, crystallised from light petroleum (b. p. 80—100°) as colourless prisms, m. p. 94—96° (Found: C, 52.4; H, 6.1; N, 14.0. $C_{13}H_{18}ON_3ClS$ requires C, 52.1; H, 6.0; N, 14.0%) (6357).

1:5-Di-*p*-chlorophenyl-4-methyl-4-*isothiobiuret* (IX; R = C_6H_4Cl-p , R' = Me).—*p*-Chlorophenyl isocyanate (7.67 g.), dissolved in ether (25 c.c.), was added to a solution of *N-p*-chlorophenyl-*S*-methylisothioureia (10.02 g.) in ether (20 c.c.). Heat was evolved, and on standing a crystalline product separated. Collected and crystallised from benzene, this *thiobiuret* formed colourless thin prisms, m. p. 139—140° (Found: C, 50.7; H, 3.6; N, 11.8. $C_{15}H_{13}ON_3Cl_2S$ requires C, 50.85; H, 3.7; N, 11.9%) (6409).