## **951.** 1:7:10- and 1:6:10-Trimethylphenanthrene.

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The compounds named in the title have been synthesised by standard methods.

During degradative studies concerning the constitution of the fungal diterpene rosolo-lactone, 1:7:10- and 1:6:10-trimethylphenanthrene have been synthesised as reference compounds, by standard methods.

Condensation of 1:8-dimethylnaphthalene with methylsuccinic anhydride by the Friedel–Crafts method furnished a mixture of acids (I; R = H, R' = Me, and vice versa). Reduction of the keto-acid (I; R = H, R' = Me) by the Huang-Minlon  $^2$  modification of the Wolff–Kishner method gave  $\gamma$ -(4:5-dimethyl-1-naphthyl)- $\alpha$ -methylbutyric acid (II; R = H, R' = Me), the acid chloride of which was cyclised to 1:2:3:4-tetrahydro-2:8:9-trimethyl-1-oxophenanthrene (III; R = H, R' = Me). Reduction of this cyclic ketone by a second application of the Huang-Minlon process yielded 1:2:3:4-tetrahydro-2:8:9-trimethylphenanthrene (IV; R = H, R' = Me) which furnished 1:7:10-trimethylphenanthrene on dehydrogenation.

Similarly 1:6:10-trimethylphenanthrene was prepared from the acid (I; R = Me, R' = H) by the stages (II), (III), and (IV; R = Me, R' = H).

Reduction of  $\beta$ -(4:5-dimethyl-1-naphthoyl)- $\alpha$ -methylpropionic acid (I; R = H, R' = Me) by the Clemmensen method furnished a product which, in agreement with the analytical data and infrared absorption at 1783 cm.<sup>-1</sup>, is formulated as the pinacol-di- $\gamma$ -lactone (VI) (cf. Huang-Minlon<sup>2</sup>).

In accordance with general principles the condensation of 1:8-dimethylnaphthalene with methylsuccinic anhydride could furnish either the 2- or the 4-naphthoylpropionic acid. Since the ultraviolet absorption spectra show conclusively that the final tricyclic hydrocarbons are phenanthrenes and not anthracenes, it follows that the condensation

<sup>&</sup>lt;sup>1</sup> Harris, Robertson, and Whalley, J., 1958, 1807; and unpublished work from these laboratories.

<sup>&</sup>lt;sup>2</sup> Huang-Minlon, J. Amer. Chem. Soc., 1946, 68, 2487.

furnishes the 4-naphthoylpropionic acids (I), as expected. Exclusive production of 4-substituted naphthalenes has been observed in analogous Friedel-Crafts condensations between succinic anhydride and 1-methylnaphthalene 3 and acenaphthene.4 The orientation of the two trimethylphenanthrenes and hence of the cognate derivatives follows from the structures of the isomeric cyclic ketones (III; R = H, R' = Me and vice versa). The comparatively slow rate of reduction of one ketone (IIIA) in the Wolff-Kishner reaction together with its reluctance to form a 2:4-dinitrophenylhydrazone, in contrast to the isomer (B) which furnishes a derivative immediately and is more rapidly reduced by the Wolff-Kishner method, indicate steric hindrance in the isomer (A) which may thus be formulated as 1:2:3:4-tetrahydro-2:8:9-trimethyl-1-oxophenanthrene (III; R = H, R' = Me). Rates of formation of ketonic derivatives have been used <sup>5</sup> to distinguish between 2(or 8)-methyl- and 2:8-dimethyl-1-tetralones. Collateral evidence for the assignments is that the cyclic ketone formulated as (III; R = H, R' = Me) does not furnish a piperonylidene derivative, whilst the isomer (III; R = Me, R' = H) does.

Attempts to oxidise the two trimethylphenanthrenes to dimethyl-2: 2'-diphenic acids with hydrogen peroxide—acetic acid gave intractable products, although 1:7-dimethylphenanthrene readily furnished 3:4'-dimethyl-2:2'-diphenic acid (V), the expected oxidation product from 1:7:10-trimethylphenanthrene.

Improvements in the synthesis of 9-methoxy-1: 7-dimethylphenanthrene are recorded below.

## EXPERIMENTAL

1: 8-Dimethylnaphthalene.—Finely divided naphthalic anhydride (10 g.) was reduced when extracted from a Soxhlet thimble by a refluxing suspension of lithium aluminium hydride (5 g.) in ether (500 ml.) during 100 hr. Excess of reagent was destroyed by ethyl acetate, and the reaction complex decomposed with 2N-sulphuric acid (400 ml.). Evaporation of the washed ethereal extract furnished 1:8-di(hydroxymethyl)naphthalene which separated from benzene in needles (6 g.), m. p. 158°. Ghilardi and Kalopissis 6 record m. p. 154-155°.

Concentration of the benzene mother-liquors furnished a product which on chromatography from benzene on aluminium oxide gave 8-hydroxymethyl-1-naphthoic lactone (ca. 0.5 g.) which separated from benzene-light petroleum (b. p. 80-100°) in plates, m. p. 157° (Found: C, 78·4; H, 4.5.  $C_{12}H_8O_2$  requires C, 78.3; H, 4.4%),  $v_{max}$ . 1727 cm.<sup>-1</sup> (lactone). The mixed m. p. with naphthoic anhydride or 1:8-di(hydroxymethyl)naphthalene was depressed. This lactone was insoluble in cold 2n-aqueous sodium hydroxide but dissolved in hot aqueous-alcoholic 2nsodium hydroxide to furnish a solution which remained clear on dilution and from which the parent lactone was precipitated by acid. Reduction of this lactone with lithium aluminium

- Haworth and Mavin, J., 1932, 2720.
  Fieser and Peters, J. Amer. Chem. Soc., 1932, 54, 4347.
  Cocker, Fateen, and Lipman, J., 1951, 926.
  Ghilardi and Kalopissis, Bull. Soc. chim. France, 1952, 217.

hydride gave 1:8-di(hydroxymethyl)naphthalene, m. p. 62°, by the method of Ghilardi and Kalopissis,<sup>6</sup> who record no yield.

1:7:10-Trimethylphenanthrene.—Aluminium chloride (8.3 g.), dissolved in nitrobenzene (15 ml.), was added slowly to a stirred solution of 1:8-dimethylnaphthalene (5 g.) and methylsuccinic anhydride (3.7 g.) in nitrobenzene (15 ml.) at 0°, and 5 days later the mixture was decomposed by ice and excess of hydrochloric acid and extracted with ether. Distillation of the ether, followed by removal of the nitrobenzene in a current of steam, furnished a resinous, acid which was esterified by boiling 10% alcoholic sulphuric acid for 4 hr. Isolated in the usual manner followed by distillation, the mixed esters (5.3 g.) had b. p. 185—195°/0.2 mm., and were saponified by refluxing them for 1 hr. with potassium hydroxide (1 g.) in water (50 ml.) and alcohol (50 ml.). Isolation with ether from the cooled, acidified hydrolysate, followed by purification from benzene-light petroleum (b. p.  $80-100^{\circ}$ ), gave  $\gamma$ - $(4:5-dimethyl-1-naphthyl)-\alpha$ methyl-y-oxobutyric acid (2.7 g.) as plates, m. p. 139-141°, unchanged by further crystallisation or by chromatography on silica-Celite (Found: C, 75.3; H, 6.8. C<sub>17</sub>H<sub>18</sub>O<sub>3</sub> requires C, 75.5; H, 6.7%). The mother-liquor from this acid slowly deposited a semi-crystalline acid (1.5 g.) which on repeated purification from benzene-light petroleum (b. p. 60-80°) gave  $\gamma$ -(4:5-dimethyl-1-naphthyl)-β-methyl-γ-oxobutyric acid in pale yellow needles, m. p. 137—138° (Found: C, 75.6; H, 6.8%). Occasionally the acids separated in the reverse order during purification from benzene-light petroleum (b. p. 60-80°).

Fractional distillation of the methylated mixture did not achieve a more satisfactory separation of these two acids since hydrolysis of each fraction furnished a crystalline acid, m. p. 139—141°, together with the semi-crystalline butyric acid. The mixed m. p. of these acids was undepressed.

A mixture of the above  $\alpha$ -methylpropionic acid (2 g.), 100% hydrazine hydrate (0.6 ml.), potassium hydroxide (1.3 g.), and diethylene glycol (25 ml.) was refluxed at 170—180° for 2 hr., then excess of hydrazine and water were removed by distillation until the reaction temperature, which was maintained for a further 3 hr., was 220°. Isolated with ether from the cooled, acidified reaction mixture, the product was purified by chromatography (silica) from chloroform—light petroleum (b. p. 60—80°) (1:10), to give  $\gamma$ -(4:5-dimethylnaphthyl)- $\alpha$ -methylbutyric acid which separated from light petroleum (b. p. 80—100°) in prisms (1.2 g.), m. p. 92° (Found: C, 79.5; H, 7.8.  $C_{17}H_{20}O_{2}$  requires C, 79.7; H, 7.9%).

The acid (I; R = H; R' = Me) (0.5 g.) was esterified with diazomethane, and the oily ester reduced by refluxing it for 36 hr. with 25% hydrochloric acid (7 ml.), toluene (3 ml.), zinc amalgam (1 g.), and acetic acid (0.5 ml.): additional quantities (1.5 ml.) of concentrated hydrochloric acid were added at six-hourly intervals. Isolated with ether, the di- $\gamma$ -lactone (VI) of 4:5-bis-(4:5-dimethyl-1-naphthyl)-4:5-dihydroxy-2:7-dimethyloctane-1:8-dioic acid separated from benzene-light petroleum (b. p. 60— $80^{\circ}$ ) in needles (0.05 g.), m. p.  $282^{\circ}$  [Found: C,  $80\cdot3$ ; H,  $7\cdot0\%$ ; M (Rast), 467.  $C_{34}H_{34}O_4$  requires C,  $80\cdot6$ ; H,  $6\cdot8\%$ ; M, 507]. This substance was insoluble in 2N-sodium hydrogen carbonate but readily dissolved in warm aqueous-alcoholic sodium hydroxide to furnish a solution which remained clear on dilution and from which acid precipitated the parent lactone.

A solution of  $\gamma$ -(4:5-dimethyl-1-naphthyl)- $\alpha$ -methylbutyryl chloride [prepared from the acid (1 g.) and phosphorus pentachloride (0.9 g.)] in tetrachloroethane (10 ml.) was added slowly to a stirred suspension of aluminium chloride (0.7 g.) in tetrachloroethane (10 ml.) at 0°. 24 Hr. later the mixture was decomposed by ice and excess of 2N-hydrochloric acid, the tetrachloroethane removed by distillation with steam, the residue extracted with ether, and the extract washed with 2N-hydrochloric acid, water, 2N-sodium hydrogen carbonate, dried, and evaporated to yield a semi-crystalline solid which was purified by chromatography from light petroleum (b. p. 60–80°) on aluminium oxide. Purification of the eluate from light petroleum (b. p. 40–60°) furnished 1:2:3:4-tetrahydro-2:8:9-trimethyl-1-oxophenanthrene in prisms (0.6 g.), m. p. 84°,  $\lambda_{max}$ . 216, 256, 281, 291, 351, (log  $\epsilon$  5·34, 5·74, 4·83, 4·90, 4·29 respectively) (Found: C, 85·4; H, 7·9.  $C_{17}H_{18}O$  requires C, 85·7; H, 7·6%).

This ketone in alcohol did not give a derivative with alcoholic 2:4-dinitrophenylhydrazine sulphate during 24 hr. at room temperature. However the derivative separated when the mixture was heated on the steam-bath for 20 min. Purification from benzene-ethanol gave the 2:4-dinitrophenylhydrazone in scarlet needles, m. p. 242 (Found: C,  $66\cdot3$ ; H,  $5\cdot3$ ; N,  $13\cdot2$ .  $C_{23}H_{22}O_4N_4$  requires C,  $66\cdot0$ ; H,  $5\cdot3$ ; N,  $13\cdot4\%$ ). When a solution of the ketone  $(0\cdot05 \text{ g.})$  and piperonaldehyde  $(0\cdot03 \text{ g.})$  in alcohol (1 ml.) containing 1% alcoholic sodium ethoxide (1 ml.)

was heated on the steam-bath during 3 hr., no piperonylidene derivative was obtained but unchanged ketone (0.04 g.), m. p. and mixed m. p.  $82-83^{\circ}$ , was recovered.

Reduction of the foregoing ketone (0·7 g.) with hydrazine hydrate (0·25 ml.), potassium hydroxide (0·4 g.), and diethylene glycol (12 ml.) was carried out at  $140^{\circ}$  for 2 hr. and then at  $200^{\circ}$  for 4 hr. after removal of excess of hydrazine. Isolated with ether the product was separated by chromatography from light petroleum (b. p.  $60-80^{\circ}$ ) on activated aluminium oxide into unchanged starting material (0·3 g.) and 1:2:3:4-tetrahydro-2:8:9-trimethyl-phenanthrene (0·3 g.), b. p.  $170-180^{\circ}/0·1$  mm., m. p.  $38-40^{\circ}$  (Found: C, 90·6; H, 9·0.  $C_{17}H_{20}$  requires C, 91·0; H, 9·0%).

A mixture of this tetrahydrophenanthrene (1·2 g.), and 30% palladium—charcoal (0·1 g.) was heated under reflux at 300° for 2 hr. and then at 350° for  $\frac{1}{2}$  hr. The cooled mixture was extracted with benzene, and the crude product purified by chromatography from the same solvent on activated aluminium oxide, to give 1:7:10-trimethylphenanthrene which separated from light petroleum (b. p. 40—60°) in plates (1 g.), m. p. 93°,  $\lambda_{\text{max}}$ . 224, 258, 280, 290, 302, 318, 327, 334, 351 (log  $\epsilon$  4·28, 4·74, 4·07, 3·97, 4·05, 2·56, 2·40, 2·52, and 2·23 respectively) (Found: C, 92·0; H, 7·6.  $C_{17}H_{17}$  requires C, 92·7; H, 7·3%). The picrate separated from alcohol in orange needles, m. p. 149° (Found: C, 61·7; H, 4·1; N, 9·3.  $C_{23}H_{19}O_7N_3$  requires C, 61·5; H, 4·3; N, 9·4%).

1:6:10-Trimethylphenanthrene.—The Wolff-Kishner reduction of  $\gamma$ -(4:5-dimethyl-1-naphthyl)-β-methyl- $\gamma$ -oxobutyric acid (1·3 g.) by the method used for the isomer furnished  $\gamma$ -(4:5-dimethyl-1-naphthyl)-β-methylbutyric acid which separated from light petroleum (b. p. 80—100°) in prisms (0·4 g.), m. p. 87°. The mixed m. p. with the isomeric acid was undepressed.

Cyclisation of the acid chloride (0·4 g.) of this acid in tetrachloroethane (5 ml.) containing aluminium chloride (0·3 g.) at 0° furnished 1:2:3:4-tetrahydro-3:8:9-trimethyl-1-oxophenan-threne which separated from light petroleum (b. p. 40—60°) in needles (0·2 g.), m. p. 94°,  $\lambda_{\text{max}}$ . 218, 260, 286, 295, 306, 360 (log  $\epsilon$  5·30, 5·68, 4·85, 4·92, 4·77, and 4·51 respectively) (Found: C, 85·9; H, 7·5%). The mixed m. p. with the isomeric 1:2:3:4-tetrahydro-2:8:9-trimethyl-oxophenanthrene was depressed. When a solution of alcoholic 2:4-dinitrophenyl-hydrazine sulphate was added to a cold solution in alcohol of this ketone the 2:4-dinitrophenylhydrazone separated rapidly and was purified from benzene-ethyl acetate, forming red needles, m. p. 305° (decomp.) (Found: C, 65·7; H, 5·1; N, 13·2%). Prepared from this ketone (0·1 g.) and piperonaldehyde (0·1 g.) in alcohol (3 ml.) containing 1% alcoholic sodium ethoxide (1 ml.) on the steam-bath during 3 hr., the difficultly purifiable piperonylidene derivative separated from benzene-acetone in pale yellow needles, m. p. 226°; insufficient material was obtained for analysis. No unchanged ketone was recovered.

Reduction of this ketone (1.9 g.) with 100% hydrazine hydrate (0.75 ml.), potassium hydroxide (1.35 g.), and diethylene glycol (25 ml.) proceeded completely during 2 hr. at  $140^{\circ}$  followed by 4 hr. at  $200^{\circ}$ , to furnish 1:2:3:4-tetrahydro-3:8:9-trimethylphenanthrene in needles (1.6 g.), m. p. 63° [from light petroleum (b. p. 40— $60^{\circ}$ )] (Found: C, 90.8; H, 8.9%).

Dehydrogenation of this tetrahydrophenanthrene (1·1 g.) with 30% palladium—charcoal (0·1 g.) as before furnished 1:6:10-trimethylphenanthrene, needles (1 g.) [from methanol or light petroleum (b. p. 40—60°)], m. p. 83°,  $\lambda_{\text{max}}$ . 216, 260, 283, 292, 305, 325, 340, 357 (log  $\epsilon$ , 4·52, 4·82, 4·15, 4·04. 4·07, 2·68, 2·87, and 2·70 respectively) (Found: C, 92·4; H, 7·1%). The mixed m. p. with the isomeric phenanthrene was depressed. The *picrate* separated from alcohol in orange needles, m. p. 140° (Found: C, 61·6; H, 4·1; N, 9·0%) (the mixed m. p. with the isomeric picrate was depressed).

3:4'-Dimethyl-2:2'-diphenic Acid.—Hydrogen peroxide (1.7 ml. of 30%) was added gradually to a solution of 1:7-dimethylphenanthrene (0.5 g.) in acetic acid (6 ml.) at 80°. After 3 hr. at this temperature the acetic acid was removed by distillation, and a solution in benzene of the residual oil was washed with 2N-sodium hydrogen carbonate. Acidification of the alkaline extract furnished a solid which, after purification by chromatography from chloroform on silica, gave 3:4'-dimethyl-2:2'-diphenic acid which separated from benzene-light petroleum (b. p. 60—80°) in needles (0.2 g.), m. p. 205° (decomp.) (Found: C, 71·1; H, 5·3.  $C_{16}H_{14}O_4$  requires C, 71·1; H, 5·2%).

1:7-Dimethyl-9-methoxyphenanthrene.—The following improvements were effected in the synthesis recorded by Ruzicka and Waldmann.

Dehydrogenation of 7-methyl-1-tetralone  $^7$  (10·1 g.) with 30% palladium-charcoal (0·5 g.)

<sup>7</sup> Ruzicka and Waldmann, Helv. Chim. Acta, 1932. 15, 907.

was effected at 320° for 3 hr., then the cooled reaction mixture was extracted with ether and the extract washed with 2N-sodium hydroxide. Acidification of this furnished 7-methyl-1-naphthol which separated from light petroleum (b. p. 80—100°) in needles (6·1 g.), m. p. 111°. Ruzicka and Waldmann 7 record m. p. 110—111°. The neutral fraction from the dehydrogenation gave 2-methylnaphthalene (3 g.), m. p. and mixed m. p. 34°.

Reduction of  $\gamma$ -(4-methoxy-6-methyl-1-naphthyl)- $\gamma$ -oxobutyric acid  $^7$  (5 g.) with 100% hydrazine hydrate (1·4 g.), potassium hydroxide (3·1 g.), and diethylene glycol (60 ml.) was achieved under reflux for 1 hr. at 180—190° and then for 4 hr. at 205—225°. The cooled mixture was diluted with water and extracted with ether. The ethereal extract was washed with 2N-sodium hydrogen carbonate, and the washings were acidified. Methylation of the dried precipitate with methyl sulphate-potassium carbonate in boiling acetone during 6 hr. furnished methyl  $\gamma$ -(4-methoxy-6-methyl-1-naphthyl)butyrate which was purified by chromatography from benzene, followed by elution with benzene-light petroleum (b. p. 60—80°) on neutralised aluminium oxide. Saponification of this ester gave  $\gamma$ -(4-methoxy-6-methyl-1-naphthyl)butyric acid (2·6 g.) in needles, m. p. 144° (from methanol). Ruzicka and Waldmann  $^7$  record m. p. 142°.

Cyclisation of this acid followed by interaction of the resultant ketone with methylmagnesium iodide according to the method of Ruzicka and Waldmann <sup>7</sup> furnished 1:2:3:4-tetrahydro-1-hydroxy-9-methoxy-1:7-dimethylphenanthrene which separated from light petroleum (b. p.  $60-80^{\circ}$ ) in needles, m. p.  $111^{\circ}$ , with infrared absorption at  $3393 \, \mathrm{cm}^{-1}$  (Found: C,  $79\cdot1$ ; H,  $7\cdot7$ .  $C_{17}H_{20}O_{2}$  requires C,  $79\cdot7$ ; H,  $7\cdot9\%$ ).

9-Methoxy-1: 7-dimethylphenanthrene, purified by chromatography from light petroleum (b. p. 40—60°) on activated aluminium oxide, separated from the same solvent in plates, m. p. 131° (Found: C, 85·9; H, 7·0; OMe, 12·8. Calc. for  $C_{16}H_{13}$ ·OMe: C, 86·4; H, 6·8; OMe, 13·1%). Ruzicka and Waldmann 7 record m. p. 126—127°.

The analyses were by Mr. A. S. Inglis, M.Sc., and his associates of this Department. The ultraviolet spectra were determined in alcohol with a Unicam S.P. 500 spectrophotometer, and the infrared spectral data were obtained in Nujol by means of a Grubb-Parsons S3 spectrophotometer. One of us (J. A. C.) is indebted to D.S.I.R. for a Maintenance Grant.

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