521. Chromones and Flavones. Part III. Synthesis of Some Cyano-chromones and -flavones.

By M. V. Shah and Suresh Sethna.

Cyano-chromones and -flavones have been synthesised by the Rosenmund-von Braun reaction from the iodo-chromones and -flavones and by Kostanecki-Robinson acylation of 3- and 5-cyano-2-hydroxy-4-methoxy-acetophenone. The action of alkali and sulphuric acid on the cyano-compounds has been studied.

DA RE et al.^{2,3} synthesised 6- and 8-cyano-2,3-dimethylchromone and 4'-cyanoflavone. No other work on the synthesis of cyano-chromones and -flavones has been reported. In the present work two iodochromones 4 (Ia and b) and an iodoflavone (IIa) were fused with cuprous cyanide (Rosenmund-von Braun reaction), affording the corresponding cyanochromones (Ic and d) and -flavone (IIc) in good yield. These were demethylated to the corresponding hydroxy-compounds.

In further work the cyano-ketones (IIIc and d) were synthesised by reaction of 3- and 5-iodo-2,4-dimethoxyacetophenone with cuprous cyanide and subsequent partial demethylation with aluminium chloride and then subjected to Kostanecki-Robinson acetylation and benzoylation. The 3-acetyl-8- and -6-cyanochromone and the benzoylflavones obtained were identical with those obtained by the Rosenmund-von Braun method. 3-Acetyl-8-cyano-7-methoxy-2-methylchromone was deacetylated to the cyanochromone (Ic) by hot alkali, but the 6-cyano-isomer gave a mixture of the cyano-ketone (IIId) and

¹ Part II, Shah and Sethna, J., 1960, 3899.

² Da Re, Farmaco (Pavia), Edn. Sci., 1956, 11, 678.

³ Da Re, Colleoni, and Verlicchi, Ann. Chim. (Italy), 1958, 48, 762.

⁴ Shah and Sethna, J., 1959, 2676.

n reactions.
Braun
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Rose
from
Products
TABLE 1.

Required (%)	Н	5.4	5.4	4.2	4.2	73.6 4.0 5.0		62.8 4.7 7.3	4.7	3.5	3.5	1
, Я							t above.ª					
	Z	9.9	6.7	8.9	6.3	5.3)emethylation products from compounds mentioned in column 4	7.2	7.5	7.2	7.2	
	Н	5.5	5.9	4.3	4.1	3.4		5.1	4.8	3.5	3.7	
	ပ	65.0	64.5	0.79	67.5	73.5		63.0	62.3	65.2	65.9	c c
	M. p.	131°	192	262	249	235		169°	180	368	340	1
	(%)	57	67	63	58	54						
	Product	3-Cyano-	5-Cyano-	ľ,	Id	$_{ m IIc}$		j.		-2-methylchromone	none	,
	temp.	$165 - 170^{\circ}$	185 - 190	235 - 240	245 - 250	220 - 225	T					
	Substance	Me ether of IIIa	Me ether of IIIb	Ia	Ib	IIa		IIIc	IIId	7-Hydroxy-8-cyano-2-methylch	7-Hydroxy-6-cyano-2-methylch	
	No.	1	2	က	4	5		1	2	_. .	4	

Nos. 3, 4, and 5 dissolve in sodium hydrogen carbonate solution with effervescence. No. 5 shows intense violet fluorescence with sulphuric acid.

Table 2. Kostanecki-Robinson "Acetylation" (A) or "Benzoylation" (B)

(%	Z	5.4	3.7	5.4	3.7
Required (%	Н	4.3	4.0	4.3	4.0
	ပ	65.4	75.6	65.4	75.6
	Formula	$C_{14}H_{11}O_4N$	C24H15O4N	C''H''O''N	$C_{24}H_{15}O_4N$
_	z	5.8	3.6	5.3	3.8
Found (%)	Н	4.0	3.8 8.8	4.4	4.3
Н	ပ	65.3	0.92	65.7	75.0
	М. р.	230°	258	220	281
Yield	(%)	33	20	33	57
	1ct *	ano-7-methoxy-2-methylchromone	oxyflavone	cy-2-methylchromone	oxyflavone
	Produ	3-Acetyl-8-cyano-7-meth	3-Benzoyl-8-cyano-7-meth	3-Acetyl-6-cyano-7-methox	3-Benzoyl-6-cyano-7-meth
	Reaction Produ	A 3-Acetyl-8-cyano-7-meth	B 3-Benzoyl-8-cyano-7-meth	A 3-Acetyl-6-cyano-7-methox	B 3-Benzoyl-6-cyano-7-meth

* Obtained also by heating the corresponding 3-acyl-6- or -8-iodo-chromone or -flavone (see Part II) with cuprous cyanide for 10 min, for chromones at 220° and for 8- and 6-iodoffavone at 250° and 285° respectively.

TABLE 3 Products of hydrolysis by sulfhuric acid

	(%) pe	H		3.7	3.7	4.3	{	4.4	4.7	4.1	68.1 3.6		68.1 3.6	;
	Required (%)	ပ	1	0.09	0.09	61.5	1	69.1	Ż	68.9	68.1		$68 \cdot 1$:
		Formula	1	$\mathrm{C_{11}H_8O_5}$	$C_{11}H_{8}O_{k}$	$C_{12}H_{10}\tilde{O}_{2}$	• • •	C,'H,'O,N	• •	$C_{17}H_{12}O_{5}$	$C_{16}^{-}H_{10}^{-}O_{5}^{-}$		67.8 3.6 C ₁₆ H ₁₀ O ₅	;
	%	Η		4.0	4.0	4.4		4.6	N, 4·9 68·7 4·6	4.6	3.7		3.6	
	Found (%)	ပ	1	59.8	60.5	61.3		69.1	ż	2.89	67.9		8.79	
		M. p.	249°	259	302	245	240	296		216	242		311	,
TABLE 3. I TOUNCES OF NYMOVYSES OF SURPRINT MIN.	Yield	(%)		36		36		37		18	46		50, 45 311	
		Product *	7-Hydroxy-2-methylchromone	7-Hydroxy-2-methylchromone-8-carboxylic acid	7-Hydroxy-2-methylchromone-6-carboxylic acid	7-Methoxy-2-methylchromone-6-carboxylic acid	7-Hydroxyflavone	(i) 8-Carbamoyl-7-methoxyflavone b and		(ii) 7-Methoxyflavone-8-carboxylic acid	7-Hydroxyflavone-8-carboxylic acid		7-Hydroxyflavone-6-carboxylic acid	
	4 SO	(% by wt.)	7	95 4	70	χĊ	20	20			50		70	;
		Parent	Ic and 3-Ac of Ic	8-Cyano-7-hydroxy-	Id and 3-Ac of Id	Id	IIc and 3-Bz of IIc	IIc			8-Cyano-7-hydroxy-	flavone	IId and 3-Bz of IId	
		No.	_	61	က	4	ū	9			7		œ	•

* Nos. 2, 4, 6 (ii), and 7 are soluble in sodium hydrogen carbonate solution with effervescence whereas Nos. 3 and 8 form a sodium salt with sodium hydrogen carbonate. Nos. 2, 3, 7, and 8 give a reddish-violet colour with alcoholic ferric chloride and melt with effervescence. Nos. 2, 3, 6 (ii), 7, and 8 were decarboxylated when heated with quinoline and copper powder, and the products were compared with authentic specimens.

* Heated on a steam-bath for 3 hr. * Yielded 7-hydroxyflavone-8-carboxylic acid when heated in acetic acid with hydriodic acid.

the acid (IVd). 3-Benzoyl-8- and -6-cyano-7-methoxyflavone, on treatment with alkali, gave the acids (IVc and d) and small quantities of the β-diketones (Vc and d) identical with those synthesised by the Baker-Venkataraman transformation of the O-benzoyl derivatives of the cyano-ketones (IIIc and d). The β-diketones cyclised to the cyanoflavones (IIc and d) in concentrated sulphuric acid. When the cyano-chromones and -flavones were treated with 70% sulphuric acid, demethylation, deacylation (in the case of 3-acyl compounds), and hydrolysis of the cyano-group to carboxyl occurred. In the cases (Ic) and (IIc) the cyano-group was eliminated. When 50% sulphuric acid was used no demethylation was observed.

EXPERIMENTAL

Rosenmund—von Braun Synthesis (see Table 1).—The iodo-derivative (0.01 mole) was stirred with cuprous cyanide (0.02 mole) for 10 min. at the temperature specified in Table 1. The mixture was powdered and extracted with acetone or alcohol; the product recrystallised from alcohol in needles. Addition of copper sulphate (0.001 mole) or the cyano-derivative (trace) from a previous run to the reaction mixture improved the yield.⁵

Kostanecki-Robinson "Acetylations" (see Table 2).—The cyano-ketone (2 g.) was heated with freshly fused sodium acetate (5 g.) and acetic anhydride (25 ml.) at 170—180° for 8 hr. The mixture was then added to cold water; the product crystallised from acetic acid (charcoal) in needles.

Kostanecki-Robinson "Benzoylations" (see Table 2).—The cyano-ketone (1 g.) was heated with freshly fused sodium benzoate (3 g.) and benzoic anhydride (10 g.) at 170—180° for 8 hr. The mixture was then treated repeatedly with hot water and sodium hydrogen carbonate, and the residue crystallised from acetic acid (charcoal), forming needles.

Alkaline Hydrolysis.—(a) 3-Acetyl-8-cyano-7-methoxy-2-methylchromone was heated with 2% alcoholic sodium carbonate solution for $\frac{1}{4}$ hr. The product obtained on acidification was extracted first with sodium hydrogen carbonate solution and then with sodium hydroxide, and 3-cyano-2-hydroxy-4-methoxybenzoic acid (IVc), m. p. 234—235° (effervescence) (Found: C, 55·9; H, 3·4; N, 7·8. $C_0H_7O_4N$ requires C, 56·0; H, 3·7; N, 7·3%), and 3-cyanopeonol (see Table 1) were isolated. The residue was the cyanochromone (Ic).

- (b) 3-Acetyl-6-cyano-7-methoxy-2-methylchromone on similar treatment with 5% alcoholic sodium carbonate solution for 2 hr. gave 5-cyano-2-hydroxy-4-methoxybenzoic acid (IVd), m. p. 232—233° (effervescence) (Found: C, 56·0; H, 3·7; N, 7·5%), and 5-cyanopeonol (see Table 1).
- (c) 3-Benzoyl-8-cyano-7-methoxyflavone was refluxed with 10% alcoholic potassium hydroxide for 2 hr. The product obtained on acidification was treated with aqueous sodium hydrogen carbonate, and the cyano-phenolic acid (IVc) separated. The undissolved product, 3-cyano-2-hydroxy-4-methoxydibenzoylmethane (Vc), crystallised from alcohol in needles, m. p. 230° (Found: C, 69·4; H, 4·5; N, 4·8. $C_{17}H_{13}O_4N$ requires C, 69·1; H, 4·4; N, 4·7%). On treatment with concentrated sulphuric acid it yielded the cyano-flavone (IIc). This β -diketone was also synthesised from 2-benzoyloxy-3-cyano-4-methoxyacetophenone [prepared from the ketone (IIIc) by reaction with benzoyl chloride and pyridine], m. p. 182° (from alcohol) (Found: C, 69·4; H, 4·2; N, 5·1. $C_{17}H_{13}O_4N$ requires C, 69·1; H, 4·4; N, 4·7%). This ketone (2 g.) in benzene (30 ml.) was refluxed with pulverised sodium (0·3 g.) for 4 hr. and the precipitated sodium salt was decomposed with acetic acid to afford the β -diketone.
- (d) 3-Benzoyl-6-cyano-7-methoxyflavone, on similar treatment with alcoholic potassium hydroxide, gave the acid (IVd) and 5-cyano-2-hydroxy-4-methoxydibenzoylmethane (Vd), m. p. 204° (Found: C, 69·1; H, 4·5; N, 4·3%). This was also synthesised from 2-benzoyloxy-5-cyano-4-methoxyacetophenone [prepared from the ketone (IIId)], m. p. 195—196° (Found: C, 69·5; H, 4·4; N, 4·2%). This ketone, when refluxed in toluene with sodium for 4 hr. and worked up as above, yielded the β -diketone.

6-Cyano-7-methoxyflavone (IId).—The above β -diketone was dissolved in concentrated sulphuric acid and kept for 4 hr. The product, obtained by pouring the mixture over ice, crystallised in needles, m. p. 226—227°, from alcohol (Found: C, 73·1; H, 4·3; N, 5·1. $C_{17}H_{11}O_3N$ requires C, 73·6; H, 4·0; N, 5·0%).

Sulphuric Acid Hydrolysis (see Table 3).—The cyano-compound (0.5 g.) was gently refluxed

⁵ Cf. Koelsch and Whitney, J. Org. Chem., 1941, 6, 795.

with sulphuric acid (25 ml.), of the specified strength, for 3 hr. The product obtained by pouring the mixture on ice was treated with aqueous sodium hydrogen carbonate to separate the acid. Both the acid and the residue, if any, were crystallised from dilute alcohol.

Demethylations (see Table 1).—The cyano-ketones were partially demethylated with aluminium chloride in the dry state or by heating them in dry nitrobenzene solution on a steam-bath for 3 hr. The product obtained after steam-distillation of nitrobenzene was purified by extraction with alkali. It crystallised from alcohol in needles. The cyano-chromones and -flavones were demethylated by heating their acetic acid solutions with hydriodic acid at 120-125° for 3 hr. The mixture was poured into sodium hydrogen sulphite solution and the product was extracted with alkali. It crystallised as needles from acetic acid. In the case of (IId), 7-hydroxy-2-methylchromone-6-carboxylic acid was obtained by use of hydriodic acid; hence it was demethylated by aluminium chloride at 120° (3 hr.).

In general, demethylations with aluminium chloride were smooth, whereas those with hydriodic acid gave undesirable coloured products especially at reaction temperatures above 130°.

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