

**546. The Stereochemistry of Molecules Containing the C=C=N Group.
Part III.¹ The Crystal and Molecular Structure of N-Ethyl-2,2'-dimethylsulphonylvinylideneamine.**

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A three-dimensional structural analysis of a third compound containing the $\text{C}=\text{C}=\text{N}$ group is described. Crystals of *N*-ethyl-2,2'-dimethylsulphonylvinylideneamine are orthorhombic $P2_12_12_1$ with $a = 12.02$, $b = 5.85$, $c = 14.47 \text{ \AA}$. The bond angle at the nitrogen atom is 144° , in sharp contrast to the value of 180° found in two other vinylideneamines in which $\text{R} = \text{Me}$. The bond lengths and other bond angles exhibit no unusual features, except the $\text{C}=\text{N}$ bond which has the length of a normal $\text{C}\equiv\text{N}$ bond. It is suggested that the linearity of the $-\text{N}=$ system in *N*-methylvinylideneamines is due, at least partly, to hyperconjugation.

It has been shown in Parts I² and II¹ that the $-\text{N}=$ system in two *N*-methylvinylideneamines is linear. In order for this system to achieve linearity the lone pair of electrons on the nitrogen atom must be absorbed into the rest of the molecule. In Part II it was suggested that hyperconjugation or electronegativity, or both, might account for the transfer of the lone pair. In an attempt to assess the relative importance of these two effects the crystal structure of *N*-ethyl-2,2'-dimethylsulphonylvinylideneamine, $(\text{CH}_3\text{SO}_2)_2\text{C}=\text{C}=\text{NET}$, has been determined. Although some doubt has recently been cast on the importance of the part played by hyperconjugation in the ground state of molecules,^{3,4} the results of the analysis show that hyperconjugation must play some part in the bonding of *N*-methylvinylideneamines.

Experimental.—The synthesis of *N*-ethyl-2,2'-dimethylsulphonylvinylideneamine will be described separately.⁵

Molecular formula: $\text{C}_6\text{H}_{11}\text{O}_4\text{S}_2\text{N}$, $M = 225.29$. Orthorhombic, $a = 12.02 \pm 0.04 \text{ \AA}$, $b = 5.85 \pm 0.02 \text{ \AA}$, $c = 14.47 \pm 0.04 \text{ \AA}$. $U = 1017 \text{ \AA}^3$. $D_m = 1.44 \text{ g./c.c.}$ (by flotation), $Z = 4$, $D_c = 1.44 \text{ g./c.c.}$ $F(000) = 472$. Space group $P2_12_12_1$ (D_2^4 No. 41), origin at $1/4, 1/4, 1/4$ from the Int. Tab. origin.⁶ $\text{Cu}-K_\alpha$ radiation, $\mu = 44.3 \text{ cm}^{-1}$, single crystal oscillation and Weissenberg photographs about the a , b , and c axes. No correction was made for absorption.

The relative intensities of 1120 reflections were estimated by eye from a calibration strip and were placed on the same scale by equating $\sum|F_o|$ in common reciprocal rows. An approximate scale and temperature factor were obtained from $h0l$ intensities by the method of Howells, Phillips, and Rogers.⁷ A further 176 reflections, the intensities of which were too low to be observed, were given a value of one-half of the minimum observable intensity and are included in the R -values quoted.

The R -value used in this paper is the residual $\frac{\sum||F_o| - |F_c||}{\sum|F_o|}$.

Structure determination. The positions of the sulphur atoms were determined from the Patterson functions of the a - and b -axis projections. Fourier methods (in projection) were then used to obtain a complete set of co-ordinates. An attempt was made to refine these co-ordinates three-dimensionally by using 390 low-order planes. The least-squares method was used and the calculations were done on the Leeds University Pegasus computer. The programmes were devised by Dr. D. W. J. Cruickshank and Miss D. E. Pilling. The weights used were $w = 1$ throughout.

After four refinement cycles the residual fell slowly to 0.27, so the calculated phase angles were used to evaluate the electron density in three dimensions. This electron density showed that sites of the carbon atoms of the ethyl group (C5 and C6; see Fig. 1) were in error. New

¹ Part II, Bullough and Wheatley, *Acta Cryst.*, 1957, **10**, 233.

² Wheatley, *Acta Cryst.*, 1954, **7**, 68.

³ Rao, *Nature*, 1960, **187**, 913.

⁴ Dewar and Schmeising, *Tetrahedron*, 1959, **5**, 166.

⁵ Challenger, unpublished work.

⁶ "International Tables for X-Ray Crystallography," Vol. I, Kynoch Press, Birmingham, 1952.

⁷ Howells, Phillips, and Rogers, *Acta Cryst.*, 1950, **3**, 210.

positions were found for these atoms and the trial structure thus produced was eventually refined by using all 1296 intensities. The final value of the residual (after anisotropic refinement of the temperature factors) was 0.10.

Results.—The results of the analysis are summarized in the Tables and Figures.

TABLE 1. Final co-ordinates (\AA) for $(\text{CH}_3\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

	<i>x</i>	<i>y</i>	<i>z</i>		<i>x</i>	<i>y</i>	<i>z</i>
S 1	7.6002	1.2262	8.1286	C 1	7.1234	2.6999	8.9912
S 2	5.1042	0.9074	6.4295	C 2	4.9572	-0.8507	6.5980
O 1	9.0194	1.2757	7.8849	C 3	6.7959	1.2852	6.5881
O 2	4.7436	1.2630	5.0811	C 4	7.4642	1.5700	5.4431
O 3	7.0605	0.1090	8.8447	C 5	8.4458	2.6217	3.3902
O 4	4.4080	1.5310	7.5144	C 6	7.6254	3.9130	3.4620
N 1	8.0475	1.6813			4.4405		

FIG. 1. Numbering of the atoms and the bond lengths in $(\text{CH}_3\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

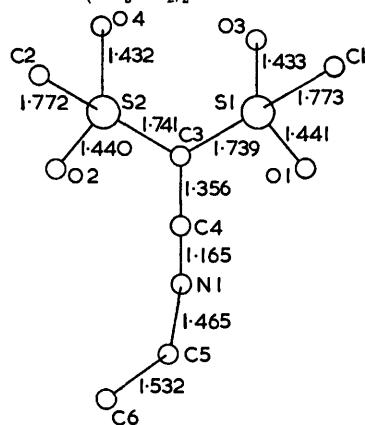


FIG. 2. The unit cell contents viewed down the *b*-axis, and some of the more important van der Waals distances.

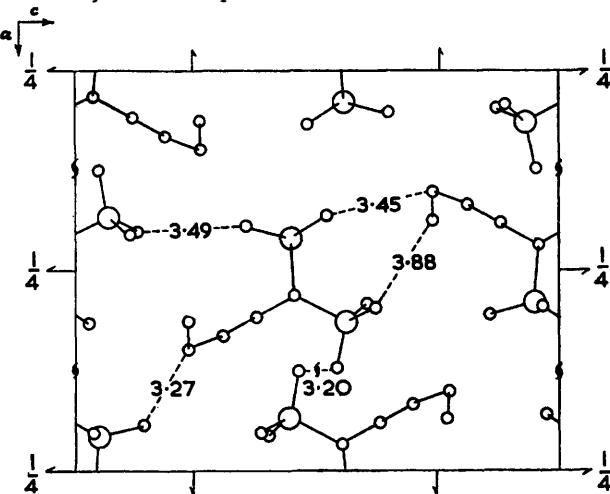


TABLE 2. U_{ij} (\AA^2) for $(\text{CH}_3\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

	U_{11}	U_{22}	U_{33}	$2U_{12}$	$2U_{23}$	$2U_{13}$
S 1	0.0375	0.0388	0.0404	0.0032	0.0075	-0.0037
S 2	0.0343	0.0698	0.0492	-0.0014	0.0164	-0.0085
O 1	0.0385	0.0602	0.0581	0.0089	0.0053	-0.0156
O 2	0.0563	0.1131	0.0659	-0.0247	0.0451	-0.0470
O 3	0.0670	0.0428	0.0466	-0.0111	0.0252	-0.0025
O 4	0.0437	0.0860	0.0678	0.0180	-0.0033	0.0201
N 1	0.0482	0.0734	0.0397	-0.0270	-0.0101	0.0064
C 1	0.0616	0.0485	0.0584	0.0052	-0.0118	-0.0042
C 2	0.0694	0.0567	0.0923	-0.0182	-0.0177	-0.0035
C 3	0.0466	0.0559	0.0423	-0.0069	0.0155	-0.0113
C 4	0.0435	0.0503	0.0511	0.0048	0.0078	-0.0069
C 5	0.0661	0.0800	0.0434	-0.0386	0.0160	0.0167
C 6	0.0691	0.0846	0.0821	-0.0161	0.0308	-0.0433

TABLE 3. Standard deviations (\AA) of the co-ordinates and σ_{rms} for $(\text{CH}_3\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

	<i>x</i>	<i>y</i>	<i>z</i>	σ_{rms}		<i>x</i>	<i>y</i>	<i>z</i>	σ_{rms}
S 1	0.0023	0.0025	0.0025	0.0024	C 1	0.0121	0.0117	0.0117	0.0118
S 2	0.0025	0.0032	0.0029	0.0029	C 2	0.0129	0.0129	0.0143	0.0134
O 1	0.0064	0.0079	0.0070	0.0071	C 3	0.0096	0.0113	0.0095	0.0102
O 2	0.0079	0.0113	0.0081	0.0092	C 4	0.0098	0.0111	0.0101	0.0103
O 3	0.0081	0.0076	0.0071	0.0076	C 5	0.0125	0.0143	0.0111	0.0127
O 4	0.0071	0.0098	0.0078	0.0083	C 6	0.0132	0.0158	0.0136	0.0142
N 1	0.0084	0.0110	0.0080	0.0093					

TABLE 4. Bond lengths and bond angles, and their standard deviations in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

Bond	Length (\AA)	σ (\AA)	Angle	Size	(σ)	Angle	Size	(σ)
S1-O1	1.441	0.0075	S1-C3-S2	121° 32'	0° 36'	O2-S2-C2	108° 15'	0° 35'
S2-O2	1.440	0.0097	S1-C3-C4	121° 49'	0° 48'	O2-S2-C3	105° 58'	0° 31'
S1-O3	1.433	0.0080	S2-C3-C4	116° 36'	0° 47'	O3-S1-C1	107° 41'	0° 31'
S2-O4	1.432	0.0088	N1-C4-C3	173° 21'	1° 08'	O3-S1-C3	107° 10'	0° 30'
S1-C1	1.773	0.0121	N1-C5-C6	111° 13'	1° 02'	O4-S2-C2	108° 38'	0° 34'
S2-C2	1.772	0.0137	O1-S1-O3	118° 51'	0° 27'	O4-S2-C3	108° 00'	0° 30'
S1-C3	1.739	0.0104	O1-S1-C1	108° 35'	0° 30'	C1-S1-C3	106° 10'	0° 31'
S2-C3	1.741	0.0106	O1-S1-C3	107° 44'	0° 28'	C2-S2-C3	106° 42'	0° 34'
N1-C4	1.165	0.0139	O2-S2-O4	118° 41'	0° 32'	C4-N1-C5	144° 31'	1° 03'
N1-C5	1.465	0.0157						
C3-C4	1.356	0.0145						
C5-C6	1.532	0.0191						

TABLE 5. Standard deviations (\AA^2) of the U_{ij} for $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

	U_{11}	U_{22}	U_{33}	$2U_{12}$	$2U_{23}$	$2U_{13}$
S 1	0.0010	0.0011	0.0011	0.0019	0.0019	0.0018
S 2	0.0011	0.0017	0.0013	0.0025	0.0028	0.0021
O 1	0.0032	0.0044	0.0042	0.0070	0.0081	0.0066
O 2	0.0046	0.0079	0.0049	0.0120	0.0117	0.0084
O 3	0.0046	0.0039	0.0039	0.0079	0.0070	0.0074
O 4	0.0038	0.0062	0.0048	0.0093	0.0104	0.0075
N 1	0.0045	0.0064	0.0043	0.0104	0.0098	0.0077
C 1	0.0066	0.0060	0.0065	0.0121	0.0115	0.0118
C 2	0.0074	0.0071	0.0093	0.0134	0.0150	0.0148
C 3	0.0050	0.0059	0.0050	0.0101	0.0102	0.0088
C 4	0.0049	0.0057	0.0053	0.0103	0.0104	0.0090
C 5	0.0072	0.0096	0.0059	0.0151	0.0136	0.0114
C 6	0.0078	0.0100	0.0088	0.0160	0.0176	0.0146

TABLE 6. Minimum van der Waals contacts (\AA) in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$.

O-O	C-O	C-C	O-N	C-N
3.2	3.3	3.9	3.4	3.6

TABLE 7. List of $|F_o|$, $|F_c|$, A_c , and B_c (for N-ethyl-2,2'-dimethylsulphonylvinylidene-amine) on ten times the absolute scale.

h	k	l	$ F_o $	$ F_c $	A_c	B_c	h	k	l	$ F_o $	$ F_c $	A_c	B_c							
0	0	2	1058	1138	1138	0	2	0	12	61	55	-55	0	4	0	14	81	80	-80	0
0	0	4	151	129	129	0	2	0	13	302	284	-284	0	4	0	15	92	73	-73	0
0	0	6	475	449	-449	0	2	0	14	25	3	3	0	4	0	16	139	130	-130	0
0	0	8	400	349	-349	0	2	0	15	107	108	-108	0	4	0	17	16	1	-1	0
0	0	10	503	502	-502	0	2	0	16	21	19	19	0	4	0	18	31	55	-55	0
0	0	12	305	279	-279	0	2	0	17	26	32	-32	0	5	0	1	151	139	0	-139
0	0	14	26	3	3	0	2	0	18	47	41	41	0	5	0	2	55	47	0	-47
0	0	16	226	197	197	0	3	0	1	341	315	0	-315	5	0	3	118	105	0	105
0	0	18	116	116	116	0	3	0	2	149	141	0	141	5	0	4	29	33	0	-33
1	0	1	412	458	0	458	3	0	3	530	485	0	-485	5	0	5	179	159	0	159
1	0	2	170	207	0	207	3	0	4	595	634	0	634	5	0	6	127	134	0	-134
1	0	3	180	158	0	-158	3	0	5	447	438	0	-438	5	0	7	389	368	0	368
1	0	4	63	30	0	30	3	0	6	396	389	0	389	5	0	8	183	173	0	-173
1	0	5	382	345	0	-345	3	0	7	20	39	0	39	5	0	9	308	297	0	297
1	0	6	289	293	0	293	3	0	8	6	14	0	14	5	0	10	26	21	0	-21
1	0	7	235	218	0	-218	3	0	9	355	365	0	365	5	0	11	26	34	0	34
1	0	8	323	304	0	304	3	0	10	34	27	0	-27	5	0	12	159	153	0	153
1	0	9	123	122	0	-122	3	0	11	135	102	0	102	5	0	13	26	38	0	-38
1	0	10	67	72	0	-72	3	0	12	213	191	0	-191	5	0	14	25	4	0	-4
1	0	11	25	16	0	16	3	0	13	26	23	0	-23	5	0	15	160	154	0	-154
1	0	12	93	83	0	-83	3	0	14	26	2	0	-2	5	0	16	46	44	0	-44
1	0	13	166	151	0	-151	3	0	15	29	15	0	15	5	0	17	94	93	0	-93
1	0	14	106	88	0	-88	3	0	16	46	51	0	-51	6	0	1	173	151	-151	0
1	0	15	96	102	0	+102	3	0	17	47	45	0	45	6	0	1	173	151	-151	0
1	0	16	22	24	0	-24	3	0	18	48	46	0	46	6	0	2	517	508	-508	0
1	0	17	47	42	0	42	4	0	0	1340	1397	-1397	0	6	0	3	215	194	-194	0
1	0	18	56	65	0	-65	4	0	1	705	678	678	0	6	0	4	166	176	176	0
2	0	0	179	164	0	4	0	2	340	335	-335	0	6	0	5	116	111	-111	0	
2	0	1	165	203	0	4	0	3	280	250	250	0	6	0	6	165	145	145	0	
2	0	2	439	397	0	4	0	4	225	208	208	0	6	0	7	78	64	-64	0	
2	0	3	1293	1337	0	4	0	5	112	150	150	0	6	0	8	91	83	83	0	
2	0	4	379	352	-352	0	4	0	6	129	126	126	0	6	0	9	75	66	66	0
2	0	5	222	211	0	4	0	7	21	39	39	0	6	0	10	260	247	247	0	
2	0	6	79	55	0	4	0	8	89	57	57	0	6	0	11	170	152	152	0	
2	0	7	111	118	0	4	0	9	8	20	20	0	6	0	12	173	148	148	0	
2	0	8	20	29	0	4	0	10	151	142	142	0	6	0	13	105	104	104	0	
2	0	9	103	97	0	-97	4	0	11	26	7	-7	0	6	0	14	68	58	58	0
2	0	10	48	69	0	4	0	12	87	65	65	0	6	0	15	21	27	27	0	
2	0	11	314	310	-310	0	4	0	13	56	62	-62	0	6	0	16	18	11	-11	0

TABLE 7. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	
6	0	17	11	6	6	0	13	0	1	46	36	0	-36	3	1	10	60	39	-36	-16	
7	0	1	156	127	0	-127	13	0	3	27	25	0	-25	3	1	12	75	84	-77	-34	
7	0	2	268	262	0	-262	13	0	4	148	140	0	140	3	1	13	154	155	-49	147	
7	0	3	60	68	0	-68	13	0	5	55	54	0	54	3	1	14	89	84	-44	72	
7	0	4	591	607	0	-607	13	0	6	168	167	0	167	3	1	15	46	42	-18	38	
7	0	5	230	234	0	-234	13	0	7	55	47	0	47	3	1	16	83	101	-49	89	
7	0	6	535	562	0	-562	13	0	8	38	36	0	36	3	1	17	16	31	-23	-21	
7	0	7	115	97	0	-97	13	0	9	103	94	0	94	3	1	18	11	18	9	13	
7	0	8	227	221	0	-221	14	0	0	142	128	128	0	4	1	0	80	68	0	-68	
7	0	9	71	49	0	-49	14	0	0	142	128	128	0	4	1	1	163	180	-179	20	
7	0	10	176	164	0	-164	14	0	1	45	25	25	0	4	1	2	366	337	-141	-306	
7	0	11	36	56	0	-56	14	0	2	129	118	118	0	4	1	3	147	147	146	-20	
7	0	12	80	215	0	-215	14	0	3	8	2	2	0	4	1	4	380	348	348	19	
7	0	13	24	4	0	-4	14	0	4	66	55	55	0	4	1	4	36	27	-23	-14	
7	0	14	129	111	0	-111	14	0	5	7	15	-15	0	4	1	6	213	181	180	-18	
7	0	15	19	17	0	-17	14	0	6	92	85	-85	0	4	1	7	235	216	199	-84	
7	0	16	38	39	0	-39	14	0	7	9	19	-19	0	4	1	8	196	183	-33	180	
8	0	0	212	192	192	0	15	0	1	98	90	0	-90	4	1	9	245	214	203	-69	
8	0	1	240	215	-215	0	15	0	2	6	19	0	-19	4	1	10	193	183	-131	128	
8	0	2	136	135	135	0	15	0	3	53	51	0	-51	4	1	11	25	27	5		
8	0	3	159	174	-174	0	15	0	4	4	11	0	11	4	1	12	85	54	-40	36	
8	0	4	45	16	-16	0	1	1	266	264	0	-264	4	1	14	59	40	-5	-40		
8	0	5	9	26	26	0	0	1	1	266	264	0	-264	4	1	15	87	77	-77	4	
8	0	6	11	5	-5	0	0	1	2	190	241	241	0	4	1	16	47	47	30	-36	
8	0	7	83	69	-69	0	0	1	3	448	354	0	354	4	1	16	47	47	30	-36	
8	0	8	63	63	63	0	0	1	4	635	652	0	-652	0	4	1	17	36	41	-40	6
8	0	9	39	34	-34	0	0	1	5	40	11	0	-11	5	1	6	76	67	46	-49	
8	0	10	78	45	-45	0	0	1	6	276	270	0	-270	5	1	0	611	589	0	589	
8	0	11	72	53	-53	0	0	1	7	305	283	0	283	5	1	1	297	285	-281	-47	
8	0	12	24	22	22	0	0	1	8	21	19	-19	0	5	1	2	462	431	74	425	
8	0	13	21	20	20	0	0	1	9	6	27	0	27	5	1	3	102	101	-89	-49	
8	0	14	60	51	51	0	0	1	10	48	23	23	0	5	1	4	205	206	102	-179	
8	0	15	15	29	29	0	0	1	11	34	55	0	-55	5	1	5	120	112	92	63	
9	0	1	374	344	0	-344	0	1	13	94	77	0	-77	5	1	7	175	170	95	-140	
9	0	2	128	113	0	-113	0	1	14	35	48	-48	0	5	1	8	125	117	3	-117	
9	0	3	180	182	0	-182	0	1	15	25	6	0	-6	5	1	9	153	144	138	-43	
9	0	4	193	182	0	-182	0	1	16	22	21	-21	0	5	1	10	218	209	12	-209	
9	0	5	72	62	0	-62	0	1	17	19	5	0	-5	5	1	11	182	155	18	-154	
9	0	6	225	230	0	-230	0	1	18	47	51	-51	0	5	1	12	92	80	-19	-77	
9	0	7	312	301	0	-301	0	1	12	26	24	24	0	5	1	13	56	58	39	-44	
9	0	8	45	12	0	-12	1	1	0	1352	1437	0	-1437	5	1	4	65	60	-7	60	
9	0	9	343	336	0	-336	1	1	1	523	559	534	-167	5	1	15	21	27	-26	9	
9	0	10	80	70	0	-70	1	1	2	694	744	274	-692	5	1	16	65	63	-24	58	
9	0	11	58	168	0	-168	1	1	3	529	540	157	-516	5	1	17	14	17	-17	2	
9	0	12	22	33	0	-33	1	1	4	389	338	306	143	5	1	13	56	58	39	-44	
9	0	13	20	31	0	-31	1	1	5	443	431	-111	-416	6	1	0	95	119	0	-119	
9	0	14	68	65	0	-65	1	1	6	98	107	96	45	6	1	1	192	211	-175	-118	
9	0	15	72	68	0	-68	1	1	7	248	237	-216	-98	6	1	2	152	141	-129	-57	
10	0	0	58	57	0	-57	1	1	9	146	137	-135	-21	6	1	4	262	258	-249	-65	
10	0	1	29	41	-41	0	0	1	10	272	269	-85	255	6	1	5	279	272	211	-171	
10	0	2	149	149	-149	0	0	1	11	307	304	-100	287	6	1	6	134	109	-58	92	
10	0	3	76	65	-65	0	0	1	12	139	132	0	132	6	1	7	348	335	304	-142	
10	0	4	139	137	-137	0	0	1	13	188	175	-1	175	6	1	8	119	120	-97	70	
10	0	5	59	47	-47	0	0	1	14	114	106	-82	-67	6	1	9	162	139	138	-13	
10	0	6	11	32	32	0	0	1	15	71	60	4	60	6	1	10	91	85	72	46	
10	0	7	56	32	32	0	0	1	16	114	114	-2	-113	6	1	11	135	120	90	79	
10	0	8	154	138	138	0	0	1	17	51	61	56	26	6	1	12	160	145	145	-9	
10	0	9	122	113	-113	0	0	1	18	69	68	-15	-66	6	1	13	82	92	-36	85	
10	0	10	41	37	37	0	0	1	19	142	134	3	124	7	1	6	95	88	-64	-60	
10	0	11	48	49	-49	0	2	1	0	437	387	0	-387	6	1	15	99	93	-88	30	
10	0	12	19	15	15	0	2	1	1	366	341	231	251	6	1	16	32	39	31	-24	
10	0	13	16	29	-29	0	2	1	2	322	332	306	129	6	1	17	65	97	-97	-2	
10	0	14	11	17	-17	0	2	1	3	129	151	-145	40	2	1	4	514	535	506	-173	
11	0	1	83	74	0	-74	2	1	5	315	331	-181	277	7	1	1	319	291	-175	232	
11	0	2	133	136	0	-136	2	1	6	339	305	304	15	7	1	2	161	157	-101	-121	
11	0	3	52	62	0	-62	2	1	7	392	370	-354	107	7	1	3	148	137	-36	133	
11	0	4	187	188	0	-188	2	1	8	205	191	187	38	7	1	4	101	102	-98	28	
11	0	5	111	97	0	-97	2	1	9	95	78	-60	51	7	1	5	134	136	19	134	
11	0	6	279	254	0	-254	2	1	10	133	124	3	124	7	1	6	95	88	-64	-60	
11	0	7	94	86	0	-86	2	1	11	88	81	-29	-76	7	1	7	168	170	72	154	
11	0	8	96	78	0	-78	2	1	12	136	130	-128	24	7	1	8	25	15	-15	-3	
11	0	9	61	66	0	-66	2	1	13	92	77	-53	-56	7	1	9	54	38	34	-17	
11	0	10	99	99	0	-99	2	1	14	48	59	-58	-10	7	1	10	88	78	-10	-78	
11	0	11	26	14	0	-14	2	1	15	94	95	-90	-31	7	1	11	119	102	43	-92	
11	0	12	118	103	0	-103	2	1	16	63	64	-62	-14	7	1	12	25	39	30	-25	
12	0	0	56	46	46	0	2	1	18	12	18	-18	4	7	1	14	65	57	0	-57	
12	0	1	14	14	-14	0	3	1	0	702	747	0	747	7	1	16	41	43	26	-35	
12	0	2	53	41	41	0	3	1	1	356	317	-35	-315	2	1	2	161	155	151	-34	
12	0	3	78	84	-84	0	3	1	1	37	427	355	238	8	1</						

TABLE 7. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	<i>A</i> _c	<i>B</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	<i>A</i> _c	<i>B</i> _c	<i>h</i>	<i>k</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _c	<i>A</i> _c	<i>B</i> _c	
8	1	8	160	138	-128	-52	15	1	2	108	133	18	-132	4	2	15	25	60	-25	54	
8	1	9	163	140	-139	11								4	2	16	21	88	82	33	
8	1	10	155	162	128	-99	0	2	0	1199	1183	-1183	0	5	2	0	202	221	-221	0	
8	1	11	98	96	-64	71	0	2	1	457	497	0	-497	5	2	1	185	176	-152	88	
8	1	12	127	110	109	-15	0	2	2	671	681	-681	0	5	2	2	313	274	-118	248	
8	1	13	80	68	13	67	0	2	3	252	239	0	-239	5	2	2	189	187	10	-187	
8	1	14	109	114	114	13	0	2	4	352	325	-325	0	5	2	3	189	187	10	-187	
8	1	15	46	48	28	38	0	2	5	149	147	0	-147	5	2	4	58	66	-64	14	
9	1	0	32	23	0	-23	0	2	7	288	278	0	-278	5	2	6	356	327	172	-278	
9	1	1	75	56	-29	-49	0	2	8	475	458	458	0	5	2	7	210	193	-38	-189	
9	1	2	85	55	-55	-1	0	2	9	356	333	0	333	5	2	8	138	131	124	42	
9	1	3	52	51	-32	-40	0	2	10	343	336	336	0	5	2	9	108	98	-35	-91	
9	1	4	133	129	-66	110	0	2	11	183	178	0	178	5	2	10	190	170	133	106	
9	1	5	161	160	22	-158	0	2	12	133	105	105	0	5	2	11	31	46	46	-3	
9	1	6	11	21	-21	-4	0	2	13	28	33	0	-33	5	2	12	29	38	-13	36	
9	1	7	38	21	6	-20	0	2	14	87	74	-74	0	5	2	13	28	27	25	10	
9	1	8	103	90	-14	-89	0	2	15	80	55	0	-55	5	2	14	73	62	-48	-38	
9	1	9	69	48	14	46	0	2	16	103	91	-91	0	5	2	15	116	98	22	96	
9	1	10	35	26	19	19	0	2	17	87	90	0	-90	5	2	16	52	66	-58	-33	
9	1	11	92	75	16	74	1	2	0	55	39	39	0	6	2	0	11	8	8	0	
9	1	12	21	20	-5	-19	1	2	1	207	212	104	-185	6	2	1	214	218	-10	218	
9	1	13	38	46	9	45	1	2	2	587	512	195	-473	6	2	2	180	170	145	89	
9	1	14	31	43	18	-39	1	2	3	115	110	83	-72	6	2	3	179	190	186	37	
9	1	15	26	24	-9	22	1	2	4	147	148	100	-109	6	2	4	139	135	133	23	
10	1	0	145	127	0	127	1	2	6	243	241	-62	233	6	2	6	51	58	14	56	
10	1	1	192	184	7	1	2	7	350	352	63	346	6	2	7	196	188	143	-122		
10	1	2	149	153	99	117	1	2	8	31	25	-22	-11	6	2	8	85	95	-95	-9	
10	1	3	86	91	81	42	1	2	9	216	211	54	204	6	2	9	181	163	-103	-126	
10	1	4	115	105	101	29	1	2	10	160	159	-38	-155	6	2	10	189	182	-181	9	
10	1	5	158	155	-155	5	1	2	11	109	108	-71	82	6	2	11	183	155	-139	-69	
10	1	6	67	56	40	-39	1	2	12	63	66	1	-66	6	2	12	31	61	-31	-53	
10	1	7	207	192	-186	51	1	2	13	52	39	-39	4	6	2	13	122	105	-103	-20	
10	1	8	143	143	73	-123	1	2	14	35	52	-2	52	6	2	14	26	5	-5	2	
10	1	9	169	162	-162	8	1	2	15	56	66	-28	-60	6	2	15	22	23	-12	20	
10	1	10	72	72	-24	-67	1	2	16	49	58	3	58	6	2	16	18	41	-40	-10	
10	1	11	89	87	-87	-3	1	2	17	39	56	-22	-51	7	2	0	95	94	-94	0	
10	1	12	45	48	-41	-25	2	2	0	143	150	150	0	7	2	1	78	70	-70	3	
10	1	13	36	43	38	-21	2	2	1	187	165	133	-97	7	2	2	282	284	-42	280	
10	1	14	16	34	-19	28	2	2	2	353	370	88	-359	7	2	3	353	347	-304	166	
11	1	0	98	95	0	95	2	2	3	119	102	-101	7	7	2	4	329	355	-85	345	
11	1	1	38	31	30	7	2	2	4	281	242	-95	-222	7	2	5	215	197	-186	-64	
11	1	2	165	163	-55	154	2	2	5	303	281	-279	31	7	2	6	208	236	43	232	
11	1	3	98	92	12	92	2	2	6	206	233	21	-222	7	2	7	216	194	-163	-96	
11	1	4	58	58	-54	19	2	2	7	210	210	-200	66	7	2	8	93	82	52	64	
11	1	5	100	97	-2	97	2	2	8	55	39	-12	-37	7	2	9	133	109	56	-93	
11	1	6	78	80	-54	54	2	2	9	141	128	89	92	7	2	10	300	317	-317	0	
11	1	7	13	25	2	25	2	2	10	142	139	82	112	8	2	0	300	317	-317	0	
11	1	8	46	43	2	-43	2	2	11	336	347	347	11	8	2	1	32	49	18	46	
11	1	9	31	30	-1	30	2	2	12	153	145	-32	141	8	2	2	125	137	-135	23	
11	1	10	20	29	4	-29	2	2	13	159	161	160	13	8	2	3	118	131	-129	24	
11	1	11	19	3	-3	0	2	2	14	91	105	-14	104	8	2	4	63	64	11	63	
11	1	12	14	26	-1	2	2	15	22	39	-12	-37	8	2	5	9	9	9	-2	-2	
12	1	0	13	5	0	5	3	2	0	73	116	-116	0	8	2	9	22	22	22	4	
12	1	1	47	45	-8	45	3	2	1	400	367	98	354								
12	1	2	95	98	96	16	3	2	1	2	154	145	-145	6	9	2	0	308	375	275	0
12	1	3	63	60	26	54	3	2	2	154	145	-145	6	9	2	1	270	262	-18	-262	
12	1	4	120	126	126	7	3	2	3	207	185	166	-82	9	2	2	145	127	127	6	
12	1	5	91	92	15	91	3	2	4	479	460	122	-444	9	2	2	145	127	127	-25	
12	1	6	160	167	166	-16	3	2	5	169	145	145	5	9	2	3	116	110	-107	-25	
12	1	7	39	35	19	29	3	2	6	246	240	-25	-238	9	2	4	103	110	60	93	
12	1	8	27	25	15	-20	3	2	7	189	170	98	139	9	2	5	220	206	-126	162	
12	1	9	42	36	34	13	3	2	8	167	147	117	-88	9	2	6	227	215	-113	183	
12	1	10	86	85	-85	-3	3	2	9	127	118	-67	97	9	2	7	142	123	-30	120	
12	1	11	40	58	28	-51	3	2	10	120	120	78	91	9	2	8	143	134	-133	14	
13	1	0	47	48	0	-48	3	2	12	119	109	47	-149	9	2	9	147	138	6	138	
13	1	1	83	75	67	33	3	2	13	87	94	-76	-55	10	2	0	201	195	195	0	
13	1	2	56	56	-10	-56	3	2	14	26	27	-13	24	10	2	1	93	85	-60	61	
13	1	3	56	50	36	35	3	2	15	81	78	-39	-67	10	2	2	108	107	104	26	
13	1	4	45	54	-29	-46	3	2	16	20	18	-17	5	10	2	3	103	100	-76	65	
13	1	5	60	72	-21	69	3	2	17	15	11	8	-8	10	2	4	56	62	57	25	
13	1	6	33	33	1	33	3	2	11	39	64	-47	-44	9	2	9	147	138	6	138	
13	1	7	27	31	-31	1	4	2	0	594	518	518	0	10	2	6	91	88	-73	49	
13	1	8	74	74	-1	74	4	2	1	129	104	50	91	10	2	7	32	24	-11	-21	
13	1	9	78	91	-38	-82	4	2	2	183	157	117	-104	10	2	8	42	40	-39	9	
14	1	0	34	36	0	-36	4	2	4	168	173	87	-149	11	2	0	91	91	91	0	
14	1	1	81	80	-80	3	4	2	5	132	115	33	-110	11	2	1	39	34	34	-5	
14	1	2	42	40	0	-40	4	2	6	100	79	-27	74	11	2	1	39	34	34	-5	
14	1	3	55	49	-29	40	4	2	7	235	223	-204	-90	11	2	2	183	181	78	-163	
14	1	4	59	63	-3	4	2	8	114	115	-										

TABLE 7. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	F_o	F_c	A_c	B_c	<i>h</i>	<i>k</i>	<i>l</i>	F_o	F_c	A_c	B_c	<i>h</i>	<i>k</i>	<i>l</i>	F_o	F_c	A_c	B_c	
12	2	0	73	68	68	0	3	3	9	138	136	-84	-106	9	3	0	47	49	0	49	
12	2	1	49	48	17	-45	3	3	10	107	101	98	25	9	3	1	60	60	20	57	
12	2	2	76	73	47	56	3	3	11	96	113	-42	-105	9	3	2	55	58	-13	56	
12	2	3	95	79	69	-38	3	3	12	131	146	145	17	9	3	3	165	171	53	162	
12	2	4	107	103	-65	79	3	3	13	42	41	40	-9	9	3	4	53	64	-25	-58	
12	2	5	39	39	39	7	3	3	14	88	106	94	-48	9	3	5	108	119	-18	118	
12	2	6	73	77	-42	64	3	3	15	45	48	26	41	9	3	6	44	42	36	-21	
12	2	7	11	4	-4	2	3	3	16	45	63	5	-63	9	3	7	58	51	-6	51	
12	2	8	33	36	36	-5									9	3	8	11	21	21	-2
12	2	9	67	52	-44	28	4	3	0	334	313	0	313	9	3	9	53	44	-2	-44	
13	2	0	39	41	-41	0	4	3	1	79	70	14	-68	9	3	10	12	33	17	-28	
13	2	1	86	72	31	65	4	3	3	174	174	-165	54	9	3	12	9	19	-18	4	
13	2	2	16	26	-6	-25	4	3	4	166	167	-167	-1	9	3	13	28	73	-10	-73	
13	2	3	89	82	81	-10	4	3	5	79	65	3	-65								
13	2	4	63	68	2	-67	4	3	6	236	236	-193	-137	10	3	0	203	220	0	-220	
13	2	5	69	115	85	-78	4	3	7	46	60	2	60	10	3	1	67	75	-30	-68	
13	2	6	63	62	6	-62	4	3	8	103	107	30	-102	10	3	2	185	195	-95	-170	
13	2	7	49	54	38	-38	4	3	9	91	76	48	59	10	3	3	32	42	17	-39	
13	2	8	26	28	23	-17	4	3	10	167	168	107	-130	10	3	5	47	64	-61	-18	
14	2	0	68	50	-50	0	4	3	12	94	75	70	-25	10	3	6	88	85	-30	79	
14	2	1	100	89	28	-84	4	3	13	26	46	0	46	10	3	7	53	47	32	-35	
14	2	2	78	71	-70	9	4	3	14	12	16	-16	3	10	3	8	140	163	-17	162	
14	2	3	51	50	30	-40	4	3	15	9	20	7	19	10	3	9	28	17	12	-11	
14	2	4	33	37	-28	24	4	3	16	7	32	-11	30	10	3	10	56	79	-12	78	
14	2	5	31	33	25	21								10	3	11	8	27	21	18	
14	2	6	28	38	37	-8	5	3	0	194	185	0	-185	11	3	0	113	125	0	-125	
0	3	1	11	5	0	-5	5	3	1	263	259	-10	11	3	1	3	38	40	13	38	
0	3	2	96	98	-98	0	5	3	2	115	99	42	-90	11	3	2	94	108	30	-103	
0	3	3	178	180	0	-180	5	3	3	159	170	92	-143	11	3	3	9	9	-8	4	
0	3	4	415	415	415	0	5	3	4	81	64	-43	-48	11	3	4	59	58	58	7	
0	3	5	53	37	0	-37	5	3	5	103	100	33	-94	11	3	5	34	37	-21	31	
0	3	6	314	317	317	0	5	3	6	39	54	-48	25	11	3	6	86	91	75	53	
0	3	7	15	47	0	-47	5	3	7	160	159	-158	12	11	3	7	45	55	-64	10	
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0	3	9	79	73	0	-73	5	3	9	207	207	-199	55	11	3	9	7	19	-1	-19	
0	3	10	170	169	-169	0	5	3	10	91	92	-26	88	11	3	10	8	41	-23	34	
0	3	11	14	8	0	-8	5	3	11	156	165	-119	114								
9	3	12	114	98	-98	0	5	3	12	14	15	1	12	3	0	11	7	0	-7		
0	3	13	14	17	0	-17	5	3	13	71	80	15	79	12	3	1	76	75	56	-49	
0	3	14	13	11	-11	0	5	3	14	12	28	-14	-24	12	3	2	21	30	-28	10	
0	3	15	12	16	0	16	5	3	15	38	45	40	21	12	3	3	106	126	14	-126	
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0	3	17	35	58	0	58	6	3	0	122	100	0	100	12	3	5	96	115	-34	-109	
1	3	0	316	324	0	324	6	3	2	216	215	61	206	12	3	7	40	51	-38	-35	
1	3	1	426	394	-393	32	6	3	3	212	214	2	214	12	3	8	47	3	0	-3	
1	3	2	320	344	-287	190	6	3	4	49	32	32	-3								
1	3	3	235	239	-224	83	6	3	5	201	218	-131	174	13	3	0	9	44	0	-44	
1	3	4	249	247	-240	58	6	3	6	113	109	-88	65	13	3	1	71	76	-71	-27	
1	3	5	236	223	22	222	6	3	7	159	155	-100	118	13	3	2	33	32	25	-19	
1	3	6	274	270	-267	38	6	3	8	170	169	21	-168	13	3	3	35	48	-47	-10	
1	3	7	216	222	218	45	6	3	9	88	84	-80	-25	13	3	4	52	69	60	33	
1	3	8	142	150	-106	106	6	3	10	65	70	43	-55	13	3	5	5	19	1	-19	
1	3	9	310	309	-2	309	6	3	11	14	53	-5	-53	13	3	6	6	15	11	10	
1	3	10	88	97	59	-77	6	3	12	13	21	-12	-18								
1	3	11	218	232	-154	6	3	13	61	73	7	-72	14	3	0	51	85	0	5		
1	3	12	128	115	112	-25	6	3	14	42	50	0	50	14	3	1	36	49	47	-13	
1	3	13	40	62	-20	59	6	3	15	58	84	59	-59	14	3	2	45	63	38	50	
1	3	14	113	93	91	22								14	3	3	28	43	0	-43	
1	3	15	54	59	-59	-2	7	3	0	269	283	0	283								
1	3	16	9	24	17	17	7	3	1	108	110	97	-51	0	4	0	342	340	340	0	
1	3	17	32	62	-60	-16	7	3	2	138	147	103	105	0	4	1	347	337	0	337	
2	3	0	259	247	0	247	7	3	4	134	145	145	-6	0	4	3	209	197	0	197	
2	3	1	182	182	-121	-136	7	3	5	94	101	43	-92	0	4	4	99	118	-118	0	
2	3	2	133	129	-107	-72	7	3	6	119	114	77	-84	0	4	5	108	118	0	-118	
2	3	3	279	267	-30	-266	7	3	7	120	115	-8	-115	0	4	6	9	4	4	0	
2	3	4	167	169	-164	-39	7	3	8	86	98	91	-36	0	4	7	170	167	0	-167	
2	3	5	214	212	71	-199	7	3	9	88	85	-78	32	0	4	8	78	72	0	-290	
2	3	6	34	29	-25	14	7	3	10	34	38	-9	-37	0	4	9	297	290	0	-290	
2	3	7	342	337	-298	-158	7	3	11	49	71	-39	59	0	4	10	28	34	34	0	
2	3	8	99	103	-99	31	7	3	12	41	57	-57	-7	0	4	11	141	126	0	-126	
2	3	9	183	170	-2	7	3	13	45	53	-45	27	0	4	12	26	40	-40	0		
2	3	10	54	38	-38	8	7	3	14	39	57	-33	46	0	4	13	24	3	0	3	
2	3	11	41	54	47	26								0	4	14	20	21	-21	0	
2	3	12	14	37	-3	-37	8	3	0	154	157	0	-157	0	4	15	56	62	0	62	
2	3	13	59	68	-14	67	8	3	1	159	154	-60	142								
2	3	14	53	63	30	-56	8	3	2	163	180	93	-154	1	4	1	179	182	-182	0	
2	3	15	82	89	-71	54	8	3	3	242	238	52	232	1	4	1	269	257	-16	257	
2	3	16	9	26	16	-20	8	3	4	146	151	149	18	1	4	2	192	204	-117	-167	
3	3	0	345	341	0	-341	8	3	6	189	203	197	49	1	4	4	252	273	-110	-250	
3	3	1	142	116	69	-93	8	3	7	89	82	2	82	1	4	5	87	98	-19	-96	
3	3	2	303	272	-214	-167	8	3	8	92	86	38	77	1	4	6	148	164	127	-104	
3	3	3	120	118	63</																

TABLE 7. (Continued.)

<i>k</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	<i>k</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	<i>k</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>			
1	4	13	21	47	-24	41	7	4	8	73	69	-63	-28	2	5	5	129	129	-113	63			
1	4	14	19	44	-18	40	7	4	9	11	23	-22	5	2	5	6	112	119	24	-116			
1	4	15	14	35	6	35	8	4	0	182	176	176	0	2	5	8	120	116	-59	73			
2	4	0	114	109	-109	0	8	4	1	66	60	-5	59	2	5	9	69	62	-59	19			
2	4	1	31	31	24	21	8	4	2	143	151	150	-5	2	5	10	44	62	21	-58			
2	4	2	281	278	-83	265	8	4	3	58	44	44	3	2	5	11	20	39	-38	-9			
2	4	3	54	43	20	-38	8	4	4	8	13	-13	-2	2	5	12	18	3	1	-2			
2	4	4	174	183	-77	166	8	4	5	99	108	107	-15	2	5	13	14	39	-33	-21			
2	4	5	58	87	-82	27	8	4	6	63	59	-55	22	2	5	8	120	116	9	-115			
2	4	6	179	173	27	171	8	4	7	89	87	60	-64	3	5	0	140	145	0	145			
2	4	7	121	141	-136	-38	8	4	8	72	75	-73	-19	3	5	1	100	83	-83	-10			
2	4	8	111	106	104	20	8	4	9	54	41	-4	-41	3	5	2	101	99	97	17			
2	4	9	69	56	-55	-10	8	4	10	55	59	-55	22	3	5	3	74	73	-56	47			
2	4	10	80	88	62	-63	9	4	0	176	171	-171	0	3	5	4	56	60	58	12			
2	4	11	25	39	-32	-23	9	4	1	51	45	0	-45	3	5	5	121	123	61	106			
2	4	12	119	125	6	-125	9	4	2	159	129	-129	4	3	5	6	49	36	-11	-34			
2	4	13	20	9	-3	8	9	4	3	105	102	101	-17	3	5	7	115	118	77	90			
2	4	14	78	78	-15	-77	9	4	4	34	30	-29	-7	3	5	8	11	16	-15	-5			
2	4	15	13	14	14	1	9	4	5	66	62	62	-6	3	5	9	91	80	79	8			
3	4	0	28	37	37	0	9	4	7	32	26	17	-19	3	5	11	21	44	33	-29			
3	4	1	267	255	-55	249	9	4	8	115	117	117	0	3	5	12	18	23	-23	-1			
3	4	2	29	27	-19	19	9	4	9	8	18	-6	-17	3	5	13	13	44	-37	-24			
3	4	3	181	189	-188	18	137	-18	136	10	4	0	48	34	-34	0	4	5	0	18	21	0	21
3	4	4	128	183	-94	-157	10	4	1	95	70	45	-54	4	5	1	141	139	114	80			
3	4	5	174	183	-94	-157	10	4	2	88	84	-73	41	4	5	2	44	42	37	-21			
3	4	6	87	81	31	75	10	4	3	91	89	70	-55	4	5	3	116	105	22	103			
3	4	7	165	159	-97	-126	10	4	3	91	89	70	-55	4	5	4	52	56	55	-9			
3	4	8	33	28	21	20	10	4	4	80	81	-54	61	4	5	4	50	100	101	-35	94		
3	4	9	93	89	71	-53	10	4	5	125	118	118	10	4	5	5	100	101	35	35			
3	4	10	27	28	-16	-23	10	4	6	8	7	-1	7	4	5	6	33	35	35	-3			
3	4	11	82	73	40	-61	10	4	7	78	82	68	46	4	5	7	31	36	-34	-11			
3	4	12	25	37	-23	-29	10	4	8	54	57	57	-4	4	5	8	9	16	-13	9			
3	4	13	74	86	80	-31	10	4	9	60	49	-36	32	4	5	9	69	55	-11	-54			
3	4	14	19	27	-27	4	11	4	0	56	59	-59	0	4	5	11	44	40	-21	34			
3	4	15	42	53	43	30	11	4	1	92	71	-28	-65	4	5	12	56	17	-16	-4			
4	4	0	182	176	-176	0	11	4	2	8	13	-9	-9	6	5	0	41	48	0	48			
4	4	1	138	125	17	-124	11	4	3	100	105	-98	-37	5	5	0	41	48	0	48			
4	4	2	49	48	-48	4	11	4	4	16	20	-20	4	5	5	1	113	100	-100	-5			
4	4	3	132	117	-111	-37	11	4	5	75	72	-71	-10	5	5	2	89	85	-83	19			
4	4	4	121	122	110	53	11	4	6	51	47	35	32	5	5	3	62	63	-9	63			
4	4	5	182	197	-180	80	11	4	7	54	56	-41	38	5	5	4	105	101	-74	-69			
4	4	6	29	30	17	-24	11	4	8	29	37	36	7	5	5	5	87	77	20	74			
4	4	7	165	162	-116	113	12	4	6	45	50	29	-41	5	5	6	78	65	-63	17			
4	4	8	52	59	-38	45	12	4	0	71	69	-69	0	5	5	7	83	74	62	41			
4	4	9	145	131	-5	131	12	4	1	9	2	-2	0	5	5	8	69	71	1	71			
4	4	10	29	33	-25	-21	12	4	2	61	65	-61	-21	5	5	9	79	64	54	-34			
4	4	11	96	94	79	51	12	4	3	27	27	25	10	5	5	10	72	58	54	21			
4	4	12	26	18	4	-17	12	4	4	51	51	-7	-51	5	5	11	20	65	21	-61			
4	4	13	71	84	70	-47	12	4	5	5	12	3	12	5	5	12	15	53	48	-22			
4	4	14	19	29	-27	-11	12	4	6	45	50	29	-41	6	5	0	116	95	0	-95			
5	4	0	131	122	122	0	13	4	0	67	65	65	0	6	5	1	34	31	4	-31			
5	4	1	107	105	0	105	13	4	1	6	15	-10	-10	6	5	2	136	126	73	-103			
5	4	2	175	169	103	133	13	4	2	42	46	16	-43	6	5	3	29	37	36	5			
5	4	3	107	101	-15	100	13	4	2	42	46	16	-43	6	5	4	53	54	51	-19			
5	4	4	143	157	16	156	0	5	1	140	120	0	-120	6	5	5	39	52	44	-28			
5	4	5	28	36	-25	25	0	5	2	143	128	-128	0	6	5	6	61	61	8	61			
5	4	6	158	150	-114	99	0	5	3	167	158	0	-158	6	5	7	55	49	10	-48			
5	4	7	51	55	52	18	0	5	4	153	149	-149	0	6	5	8	102	98	5	98			
5	4	8	95	91	-91	-8	0	5	5	141	132	0	-132	6	5	9	25	14	-14	3			
5	4	9	22	19	14	-14	0	5	6	95	89	-89	0	6	5	10	22	43	-8	43			
5	4	10	98	88	-81	-36	0	5	7	13	4	0	-4	6	5	11	18	23	4	23			
5	4	11	26	38	15	-35	0	5	8	11	18	18	0	7	5	0	99	89	0	-89			
5	4	12	24	16	14	-7	0	5	9	59	66	0	66	7	5	1	59	51	39	32			
5	4	13	21	33	13	-31	0	5	10	26	13	13	0	50	7	5	3	40	41	34	23		
5	4	14	16	58	43	-39	0	5	11	32	50	0	50	7	5	4	89	90	-87	-23			
6	4	0	56	47	47	0	0	5	13	52	54	0	54	7	5	5	151	30	-28	12			
6	4	1	142	130	-67	-112	1	5	0	102	113	0	-113	7	5	6	51	46	-27	37			
6	4	2	146	140	52	-130	1	5	1	160	152	149	26	7	5	7	63	71	-70	7			
6	4	3	86	83	-83	-7	1	5	0	102	113	0	-113	8	5	1	73	61	-41	-46			
6	4	4	151	151	83	-126	1	5	1	160	152	149	26	7	5	2	34	43	22	37			
6	4	5	120	110	-102	-43	1	5	2	163	160	160	13	7	5	9	69	56	-56	-2			
6	4	6	73	58	-23	-53	1	5	3	89	83	56	-62	7	5	7	62	56	-56	-2			
6	4	7	53	51	-27	43	1	5	4	126	128	120	45	8	5	0	12	13	0	13			
6	4	8	121	120	-116	-29	1	5	5	52	65	-57	31	8	5	1	73	61	-41	-46			
6	4	9	86	69	39	56	1	5	6	93	97	96	-13	8	5	2	34	43	22	37			
6	4	10	91	62	-49	38	1	5	7	67	84	-83	8	5	5	3	138	121	4	-121			
6	4	11	26	91	78	48	1	5	8	93	93	17	-92	8	5	4	7	4	-2	4			
6	4	12	24	53	16	51	1	5	9	94	90	-88	17	8	5	5	91	125	23	-123			
6	4	13	20	53	46	27	1	5	10	24	39	-29	-26	8	5	6	27	34	-34	-3			
7	4	0	88	91	91	0	1	5	12	19	46	-41	21	8	5								

TABLE 7. (Continued.)

<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>	<i>h</i>	<i>k</i>	<i>l</i>	<i>F_o</i>	<i>F_c</i>	<i>A_c</i>	<i>B_c</i>
9	5	5	42	89	-5	-88	2	6	9	28	25	21	-14	7	6	0	60	51	-51	0
9	5	6	78	75	70	26	2	6	10	14	41	-15	-38	8	6	1	85	78	-30	-73
9	5	7	54	59	-26	-53	3	6	0	80	70	70	0	8	6	2	61	49	-42	-26
9	5	8	18	18	17	-5	3	6	1	81	74	59	-45	8	6	3	47	49	-35	-34
10	5	0	65	73	0	73	3	6	2	46	54	51	16	8	6	3	47	49	-35	-34
10	5	1	66	55	-50	22	3	6	3	69	75	67	-32	8	6	4	29	26	9	-24
10	5	2	56	49	10	48	3	6	4	20	20	19	-6	8	6	5	31	36	-31	18
10	5	3	32	36	-23	28	3	6	5	74	78	77	-11	9	6	0	72	70	70	0
10	5	4	35	34	23	26	3	6	6	55	65	-64	-13	9	6	0	72	70	70	0
10	5	5	5	34	-11	32	3	6	7	32	45	31	33	9	6	1	6	25	2	25
10	5	6	33	37	25	-28	3	6	8	65	67	-67	-2	9	6	2	51	48	42	-23
10	5	7	18	36	15	32	3	6	9	60	55	-26	48	3	6	10	14	34	-33	6
11	5	1	60	49	-49	-2	36	4	6	0	16	27	-27	0	7	2	44	40	-40	0
11	5	2	42	38	12	-33	4	6	1	92	72	32	64	0	7	3	102	93	0	93
11	5	3	45	43	-27	-33	4	6	2	47	35	-2	35	0	7	4	6	2	-2	0
11	5	4	19	27	16	22	4	6	2	47	35	-2	35	0	7	5	98	103	0	103
0	6	0	56	49	49	0	4	6	4	49	45	-20	40	0	7	1	53	49	0	49
0	6	1	142	128	0	-128	4	6	5	66	66	57	-33	1	7	0	40	38	0	-38
0	6	2	20	28	0	4	6	6	41	49	25	42	1	7	1	19	23	-7	-22	
0	6	3	94	80	0	-80	4	6	7	53	76	19	-73	1	7	2	44	43	-40	-17
0	6	4	8	5	5	0	4	6	8	7	18	15	10	1	7	3	48	40	-20	-34
0	6	5	20	27	0	27	4	6	9	68	70	6	-69	1	7	4	45	40	-36	18
0	6	6	44	46	-46	0	4	6	9	68	70	6	-69	1	7	5	5	6	0	-6
0	6	7	92	91	0	91	5	6	0	15	18	-18	0	1	7	6	32	37	-33	16
0	6	8	33	30	-30	0	5	6	1	79	47	2	-47	1	7	6	32	37	-33	16
0	6	9	103	92	0	92	5	6	2	36	36	-35	6	2	7	0	83	91	0	-91
0	6	10	18	18	-18	0	5	6	3	81	82	65	-50	2	7	1	22	24	20	-14
0	6	11	39	59	0	59	5	6	4	20	31	-4	-31	2	7	2	93	96	-10	-95
1	6	0	94	90	90	0	5	6	6	63	71	38	-60	2	7	3	6	7	-5	-5
1	6	1	29	55	-43	-34	5	6	7	59	13	3	13	2	7	4	34	38	1	-38
1	6	2	133	120	0	5	6	8	44	48	36	-31	2	7	5	12	13	-12	-6	
1	6	3	71	73	-72	-14	5	6	9	52	46	-26	38	2	7	6	40	58	17	56
1	6	4	54	52	19	48	5	6	5	42	41	36	-19	2	7	3	6	7	-5	-5
1	6	5	59	62	-62	2	6	6	0	62	65	-65	0	3	7	0	34	4	0	4
1	6	6	85	95	-61	72	6	6	1	16	10	10	0	3	7	1	62	60	45	-40
1	6	7	86	24	-5	24	6	6	2	94	99	-48	86	3	7	2	7	10	9	5
1	6	8	94	93	-87	31	6	6	3	7	9	-8	-5	3	7	3	61	57	31	-48
1	6	9	55	57	43	38	6	6	4	94	100	7	100	3	7	4	15	20	20	4
1	6	10	15	66	-60	-27	6	6	5	41	41	31	28	3	7	5	32	30	-25	-15
1	6	11	9	46	41	22	6	6	6	73	70	21	66	2	7	0	27	28	0	-28
2	6	0	60	64	64	0	6	6	8	22	29	27	11	4	7	1	29	26	-4	-25
2	6	1	52	46	38	26	5	6	0	62	66	66	66	4	7	2	45	34	30	-17
2	6	2	87	101	34	-95	7	6	0	52	51	-51	0	4	7	3	66	70	-26	-65
2	6	3	91	93	90	21	7	6	1	32	30	-29	6	4	7	4	26	26	12	-23
2	6	4	86	80	4	-80	7	6	2	31	31	-26	-18	4	7	5	33	56	-17	-54
2	6	5	36	38	37	-9	7	6	3	80	69	-69	8	2	7	0	26	27	0	27
2	6	6	66	55	-23	-51	7	6	4	40	36	-28	-22	5	7	0	26	27	0	27
2	6	7	27	35	29	-21	7	6	5	58	52	-52	2	5	7	1	7	2	-2	0
2	6	8	26	25	-24	-7	7	6	6	33	39	26	-29	5	7	2	59	71	69	19

Table 2 gives the anisotropic thermal parameters (in Å²). These are the components of Cruickshank's anisotropic thermal motion tensor.⁸

Table 3 gives the standard deviations and σ_{rms} for the atomic co-ordinates. The values were found from the standard least-squares formulæ.⁹

For Table 4 the deviations were calculated from the formulæ given by Cruickshank and Robertson.¹⁰

Table 5 gives the standard deviations of the anisotropic thermal parameters.

Table 6 gives some of the smaller Van der Waals contacts.

Table 7 gives |*F_o*|, |*F_c*|, *A_c*, and *B_c*.

Discussion.—In Table 8 the bond lengths obtained in the present analysis are compared with standard lengths given in "Interatomic Distances."¹¹ It will be seen that the agreement is everywhere good except for the C=N bond, which is not significantly different from a carbon-nitrogen triple bond. The same feature was observed in the two previous studies on vinylideneamines.^{1,2}

The angle C3=C4=N1 differs significantly from 180° ($\Delta/\sigma = 6.0$) but the direction of displacement of the nitrogen atom is perpendicular to the plane S1-C4-S2. Such a

⁸ Cruickshank, *Acta Cryst.*, 1956, **9**, 747.

⁹ Whittaker and Robinson, "Calculus of Observations," Chapter IX, Blackie and Sons, London, 1940.

¹⁰ Cruickshank and Robertson, *Acta Cryst.*, 1953, **6**, 698.

¹¹ Interatomic Distances (L. E. Sutton), *Chem. Soc. Special Publ.* No. 11, London, 1958.

distortion preserves the *mm* symmetry of the π -bond in C4=N1, and thus might occur particularly easily.

In Table 9 the present molecular dimensions are compared with the analogous ones obtained by Wheatley for $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{NMe}$. Some differences in bond angles occur but these, with the exception of C4=N1-C5, are small and are probably attributable to

TABLE 8. Bond lengths (\AA) and angles in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$ compared with standard values taken from "Interatomic Distances."¹⁰

Bond	Length in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{NET}$	Length in standard	Standard source	Angle	Size in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$	Size in standard
S1-O1	1.441	1.43	Sulphones	S1-C3-S2	121.5°	120.0°
S2-O2	1.440	1.43	Sulphones	S1-C3-C4	121.8	120.0
S1-O3	1.433	1.43	Sulphones	S2-C3-C4	116.6	120.0
S2-O4	1.432	1.43	Sulphones	N1-C4-C3	173.4	180.0
S1-C1	1.773	1.80	S-methyl	N1-C5-C6	111.2	109.5
S2-C2	1.772	1.80	S-methyl	O1-S1-O3	118.8	109.5
S1-C3	1.739	1.73	S-C heterocyclic	O1-S1-C1	108.6	109.5
S2-C3	1.741	1.73	S-C heterocyclic	O1-S1-C3	107.7	109.5
N1-C4	1.165	—	Not enough data	O2-S2-O4	118.7	109.5
N1-C5	1.465	1.475	N-C normal	O2-S2-C2	108.2	109.5
C3-C4	1.356	1.337	C-C normal	O2-S2-C3	106.0	109.5
C5-C6	1.532	1.541	C-C normal	O3-S1-C1	107.7	109.5
				O3-S1-C3	107.2	109.5
				O4-S2-C2	108.6	109.5
				O4-S2-C3	108.0	109.5
				C1-S1-C3	106.2	109.5
				C2-S2-C3	106.7	109.5
				C4-N1-C5	144.5	120.0

TABLE 9. Comparison of the bond lengths (\AA) and angles in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{C}_2\text{H}_5$ with corresponding values in $(\text{CH}_3\cdot\text{SO}_2)_2\text{C}=\text{C}=\text{N}\cdot\text{CH}_3$.

Bond	Length in NET compound	Length in NMe compound	Δ/σ	Angle	Size in in NET compound	Size in in NMe compound	Δ/σ
S1-O1	1.441	1.431	1.4	S1-C3-S2	121.5°	122.6°	0.7
S2-O2	1.440	1.431	0.9	S1-C3-C4	121.8	118.7	3.9
S1-O3	1.433	1.435	0.3	S2-C3-C4	116.6	118.7	2.6
S2-O4	1.432	1.435	0.3	N1-C4-C3	173.4	180.0	6.0
S1-C1	1.773	1.770	0.3	N1-C5-C6	111.2	—	—
S2-C2	1.772	1.770	0.1	O1-S1-O3	118.8	118.4	0.8
S1-C3	1.739	1.726	1.3	O1-S1-C1	108.6	108.7	0.2
S2-C3	1.741	1.726	1.4	O1-S1-C3	107.7	107.3	0.8
N1-C4	1.165	1.154	0.8	O2-S2-O4	118.7	118.4	0.6
N1-C5	1.465	1.426	2.4	O2-S2-C2	108.2	108.7	0.8
C3-C4	1.356	1.342	0.9	O2-S2-C3	106.0	107.3	2.6
C5-C6	1.532	—	—	O3-S1-C1	107.7	106.7	2.0
				O3-S1-C3	107.2	108.3	2.2
				O4-S2-C2	108.6	106.7	3.2
Distance				O4-S2-C3	108.0	108.3	0.6
S1-S2	3.03	3.03	—	C1-S1-C3	106.2	106.7	1.0
				C2-S2-C3	106.7	106.7	—
				C4-N1-C5	144.5	180.0	31.0

packing requirements. The lengths of the C3=C4, C4=N1, and N1-C5 bonds are all found to be slightly, though not significantly, greater than those found by Wheatley (Δ/σ values are 0.9, 0.8, and 2.4 respectively). It is interesting that the N1-C5 bond is significantly longer than was found by Bullough and Wheatley for $\text{CH}_3\cdot\text{SO}_2\cdot\text{C}(\text{SO}_2\cdot\text{Ph})=\text{C}=\text{NMe}$ ($\Delta/\sigma = 5.2$).

There is no significant difference between the length of the N1-C5 bond found here (1.465 \AA) and the standard single bond value (1.475 \AA). Thus the length of the N1-C5 bond is consistent with the absence of hyperconjugation.

The most noticeable feature of the present analysis is the value found for the angle C4=N1-C5. The angle between the valencies of the nitrogen atom is no longer 180°,

nor has the angle dropped to 120° , but an intermediate value of 144° is adopted. These values indicate that the methyl group, attached to the nitrogen atom in *N*-methylvinylideneamines, brings about complete delocalisation of the lone pair of electrons on the nitrogen, while a similarly substituted ethyl group cannot do so. By analogy, a carbanion of the type $\text{RR}'\text{C}=\text{C}=\bar{\text{C}}-\text{Me}$ ought to have a linear $\text{C}=\text{C}-\text{Me}$ angle.

In view of these results it seems reasonable to classify the *N*-methylvinylideneamines investigated^{1,2} as examples of hyperconjugation in molecules in the ground state. However, it cannot be assumed that hyperconjugation is the only factor affecting linearity. In the *N*-ethylvinylideneamine the shortness of the $\text{N}1=\text{C}4$ bond, and the intermediate value adopted by the $\text{C}4=\text{N}1-\text{C}5$ angle could be ascribed either to the reduced hyperconjugation of the ethyl group or to the electron-attracting power of the sulphonyl groups. A more thorough examination of the effects of electronegativity is required to establish the relative importance of hyperconjugation and electronegativity.

I thank Dr. P. J. Wheatley for suggesting the problem and for many helpful discussions, and Dr. J. H. Robertson for his kindly encouragement.

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[Received, January 12th, 1961.]
