

1366. *The Crystal Structure of Tris-(o-diphenylarsinophenyl)-arsineruthenium Dibromide*

By R. H. B. MAIS and H. M. POWELL

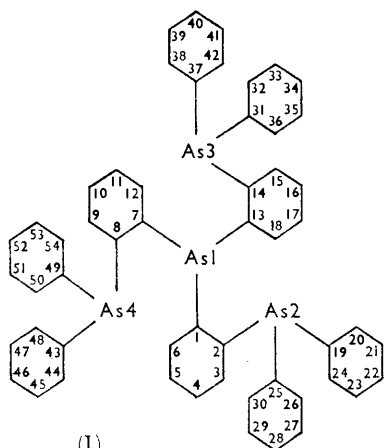
A three-dimensional crystal structure analysis shows that the ruthenium(II) atom in a six-co-ordinated complex has bonds in an arrangement not greatly distorted from the octahedral in spite of stereochemical difficulties to be expected with the quadridentate ligand that it contains. This is made possible by considerable distortions of bond angles in the ligand.

THE ligand tris-(*o*-diphenylarsinophenyl)arsine [QAS = (I)] reacts with nitrosoruthenium compounds to form complexes of the type RuQASX₂ (X = Cl, Br, I, CNS, or NO₃) which are monomeric, diamagnetic, and non-electrolytes in nitrobenzene solution. So far as is known all finite hexa-co-ordinated complexes, including those of Ru^{II} are octahedral, but it seemed unlikely that an octahedral configuration could be preserved in the RuQASX₂ series, in view of the stereochemical limitations of the quadridentate ligand QAS. Indeed, were the ruthenium octahedrally co-ordinated then either the bonds between the ruthenium and the ligand or the bonds within the ligand itself must be considerably distorted and yet

when X = Cl or Br the compound remains thermally stable at over 380°. To determine the environment of the ruthenium atoms in these compounds, the crystal structure of one member RuQASBr₂, has been examined.

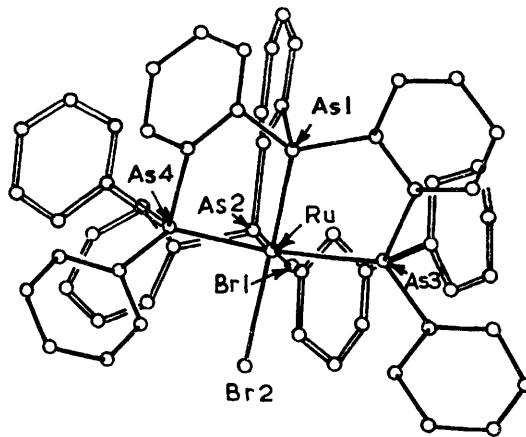
Crystal Data—RuQASBr₂, $M = 1251.5$; orthorhombic, $a = 31.53 \pm 0.08$, $b = 11.40 \pm 0.10$, $c = 13.03 \pm 0.02 \text{ \AA}$; $U = 4684.2 \text{ \AA}^3$; $D_m = 1.79$ (by flotation); $Z = 4$, $D_c = 1.76$; Space group $Pna2_1$ (C_{2v}^9 No. 33); Cu- $K\alpha$ radiation, single crystal photographs.

Structure Determination.—Over 2500 independent values of $F^2(hkl)$ were obtained from intensity measurements made from photographs taken by the equi-inclination Weissenberg method. Corrections were made for the Lorenz and polarisation factors but not for absorption or extinction. A three-dimensional Patterson map was computed from values of $F^2(hkl)$ sharpened to point atoms at rest.



(1)

The numbers represent the number of the carbon atom referred to in the Tables.



The space groups $Pna2_1$ and $Pnam$ are both compatible with the absent spectra, but since there are only four molecules in the unit cell $Pnam$ would require the molecule to have a mirror plane of symmetry passing through the ruthenium atom. The position of the vectors could not be reconciled both with this requirement and with the known chemical formula. A trial structure based on $Pna2_1$ of four octahedra whose positions were determined from the vector map gave an R value of 0.38. Application of shifts, obtained from a difference map, to the positional and isotropic thermal parameters of the heavy atoms reduced R to 0.29. Two rounds of least-squares refinement on the positional and isotropic thermal parameters of the heavy atoms further reduced R to 0.20. The least-squares refinement employed the block diagonal approximation and a weighting scheme such that $1/w = 1 + [(F_o - b)/a]^2$ where $a = 70$ and $b = 30$.

At this stage a second difference synthesis showed the position of all 54 carbon atoms and two more rounds of least-squares refinement on the positional and isotropic thermal parameters of all except the hydrogen atoms reduced R to 0.087.

DISCUSSION

The purpose of the structure determination was to discover the environment of the heavy atoms and not to examine the detail of the phenyl groups. In the case of the carbon atoms the standard deviations are naturally higher than those in the positional parameters of the heavy atoms, but even so the bond angles and distances within the phenyl groups are not unreasonable.

Hartley and Venanzi¹ reported that a molecular model showed that considerable

¹ Hartley and Venanzi, *J.*, 1962, 182.

distortion of the bond angles and distances would be necessary to form a complex in which the co-ordination of the ruthenium atom could be considered as approaching the octahedral. Nevertheless the co-ordinates of the heavy atoms establish that the bonds from ruthenium retain an octahedral configuration despite the restraints and limitations imposed by the ligand.

The trigonal symmetry² of the trigonal-bipyramidal ion [PtQASI]⁺ enables the bond angles at the apical arsenic atom As₁ to remain close to the tetrahedral, but it is not obvious how in the octahedral system the average deviation of the bonds at As₁ from the ideal tetrahedral can be as little as 2·5°, no greater than the average deviation (3·0°) in [PtQASI]⁺.

If the bonds from ruthenium were exactly octahedrally orientated and if the ruthenium atom were coplanar with each of the three bridging benzene rings, as is the platinum atom in [PtQASI]⁺, then the bond angles at As₁ would not be as close to the tetrahedral as they are found to be and overcrowding would occur among the free phenyl groups.

The actual orientation found differs from the ideal octahedral in that the bonds from ruthenium to As₃ and As₄ are bent 7° towards Br₁ and 7° towards As₁ and the bond to As₂ is bent 3° towards As₁. The benzene ring forming the bridge between As₁ and As₂ is found to be coplanar with Ru, As₁, As₂, Br₁, and Br₂. The plane containing these atoms forms an approximate mirror plane of symmetry within the molecule. Because the bonds from As₁ are almost exactly tetrahedrally orientated the benzene rings forming the bridges from As₁ to As₃ and As₄ though they are coplanar with the arsenic atoms to which they are attached are not coplanar with ruthenium (see Figure). The potential overcrowding mentioned above is avoided partly by the effects of the distortions necessary to provide tetrahedral bonds at As₁ and partly by the distortions of the bond angles at As₂, As₃, and As₄ which involve also the non-bridging phenyl groups. The deviations in the angles at As₂, As₃, and As₄ from the tetrahedral can be seen in Table 2. The largest deviations are found when bonds to non-bridging phenyls are involved. The occurrence of such deviations, which are geometrically impossible in the bridges, makes possible the retention of the octahedral form.

The bond from ruthenium to As₁ is short compared with those to As₂, As₃, and As₄. In the [PtQASI]⁺ ion the bond to As₁ from platinum is also short (2·31 Å) compared with those to As₂ (2·49 Å), As₃ (2·45 Å), and As₄ (2·43 Å). The shortening might be due to restrictions imposed by the ligand which in order to reduce the strain at As₁ requires that the angles Ru-As₁-C be slightly greater than the ideal tetrahedral.

The general conclusion from this structure is that the ruthenium bond angles are distorted from the octahedral about as much as they might be by the bridging requirements of a bidentate ligand. What seemed a possible way of preventing octahedral co-ordination fails because the valency angles at the arsenic atoms are not all sufficiently restricted by bridging.

EXPERIMENTAL

Preparation.—Samples were supplied by Dr. L. M. Venanzi, and had been recrystallised from tetrahydrofurfuryl alcohol. The yellow-orange crystals were about 0·25 mm. long.

X-Ray Photography.—The unit-cell dimensions were measured from zero layer Weissenberg photographs taken about the *b* and *c* axes. The intensities were estimated visually by the multiple film technique from equi-inclination Weissenberg photographs taken about the *b* axis. The layer scale factors were obtained by comparison of F²(*h**k*0) values with those observed on a [001] zero layer Weissenberg photograph.

Calculations.—The main programmes used were written by J. S. Rollett (structure factor calculations and least-squares refinement), O. S. Mills (Fourier summations), G. A. Mair (standard deviations by procedures of Cruickshank³ and Darlow⁴). Final structure factors are

² Mair, Powell, and Venanzi, *Proc. Chem. Soc.*, 1961, 170.

³ Ahmed and Cruickshank, *Acta Cryst.*, 1953, **6**, 385.

⁴ Darlow, *Acta Cryst.*, 1960, **13**, 683.

TABLE 1

Structure factors in absolute units with (90 - alpha) in degrees, given in every sixth column

Table with 15 columns: k, h, l, F_o, F_c, k, h, l, F_o, F_c, k, h, l, F_o, F_c, k, h, l, F_o, F_c. It lists structure factor values for various Miller indices (hkl) and includes a 60-degree phase shift column. The table is organized into groups of rows, with indices increasing from 0 to 34.

TABLE I (Continued)

<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C	<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C	<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C	<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> ₀	<i>F</i> _C					
1	21	32	94					10	43	42	93				8	79	83	349	4	43	42	143		
2	86	84	265					11	111	111	270				9	33	32	349	5	43	52	314		
3	87	88	30		4	5	0	1	185	179	360				10	96	99	229	6	38	39	160		
4	76	71	282		2	211	209	212							11	21	24	314	7	33	47	288		
5	79	77	111		3	129	126	100							0	40	41	270	0	21	28	270		
6	62	62	259		4	97	95	279							2	27	30	351	3	60	65	26		
7	75	75	85		5	187	174	14							3	40	41	90	4	22	21	307		
9	34	41	66		6	186	175	204							4	42	33	5	5	76	88	130		
11	39	54	121		7	121	121	54							5	66	72	67	6	31	35	246		
3	26	1	22	22	8	120	115	177							6	27	28	2	7	44	47	72		
3	22	20	152		9	71	72	76							8	56	53	48	8	21	17	284		
5	22	29	118		10	57	57	207							9	22	40	47	9	35	42	44		
3	27	1	147	152	11	76	74	356							11	21	29	2	10	33	39	286		
2	62	55	268		12	37	48	177							0	53	59	270	11	47	54	118		
3	82	76	195		13	23	49	51							1	58	57	120	12	19	26	258		
4	22	29	307		4	6	0	28	29	90					2	101	102	223	4	24	0	67	62	270
5	90	93	181		1	131	137	337							3	106	103	284	1	56	56	8		
6	57	60	29		2	104	97	174							4	80	82	285	2	21	28	239		
7	41	47	213		3	45	46	15							5	168	173	246	3	65	67	5		
8	28	33	308		4	40	31	215							6	79	71	188	4	22	22	243		
3	28	2	22	22	5	65	64	0							7	55	55	269	6	62	58	224		
3	29	0	142	151	6	84	80	159							8	64	63	232	4	25	0	65	69	90
2	87	88	109		7	59	53	288							9	88	88	254	1	72	72	182		
3	31	35	155		8	51	46	185							10	58	57	284	2	31	38	296		
4	74	80	93		4	7	0	338	350	270					11	51	56	237	4	58	60	345		
5	21	27	45		1	199	208	152							12	27	31	196	5	58	58	173		
6	73	77	100		2	153	149	270							14	19	28	256	6	72	79	25		
8	25	32	171		3	140	132	143							0	29	26	90	7	30	24	199		
9	23	20	86		4	155	153	205							1	50	53	206	8	35	44	115		
3	31	1	29	95	5	138	130	146							2	38	31	104	10	39	38	10		
3	41	38	23		6	170	161	279							3	18	19	271	11	25	29	172		
4	61	68	355		7	197	192	93							4	18	19	271	12	15	26	7		
5	68	76	6		8	55	49	356							6	52	52	7	4	26	0	79	77	270
6	19	16	6		9	29	30	158							9	22	34	348	1	44	41	185		
7	39	48	29		10	43	45	307							10	48	44	290	2	62	58	272		
8	22	27	30		11	22	29	94							11	21	26	22	4	31	34	239		
3	33	0	71	78	12	21	20	261							12	19	28	237	5	38	42	183		
3	38	43	268		13	28	41	42							0	24	23	90	0	114	119	90		
4	32	36	277		4	8	0	131	131	270					1	80	80	55	1	22	23	287		
5	56	58	256		1	87	87	19							2	35	40	250	2	58	62	95		
6	33	47	271		2	91	85	259							3	18	31	203	3	31	39	220		
7	21	23	328		3	18	31	10							4	91	92	170	4	52	50	82		
8	22	20	312		4	115	107	235							5	34	43	269	5	39	36	313		
3	35	4	42	54	6	48	39	259							6	29	44	122	6	59	62	94		
5	21	26	180		7	18	25	71							7	56	57	118	7	40	50	260		
2	65	64	102		8	28	36	91							8	75	77	191	4	28	0	45	45	90
3	68	71	238		9	39	31	178							9	76	73	180	1	48	49	179		
8	27	24	318		10	138	138	90							10	61	67	141	3	38	34	165		
10	71	76	270		1	282	102	183							11	41	48	171	4	30	38	160		
4	1	37	56	261	2	212	213	41							12	18	19	119	5	21	25	129		
3	197	192	341		3	168	162	178							13	27	41	137	6	20	24	133		
4	12	8	303		4	165	159	122							0	62	61	90	8	26	30	188		
5	136	123	191		5	213	208	176							1	26	24	84	4	29	1	87	95	12
6	101	94	68		6	98	97	25							2	26	17	44	1	22	34	139		
7	84	78	226		7	56	47	173							3	27	27	17	3	37	43	11		
9	81	81	208		8	28	37	191							5	29	34	156	5	65	74	349		
10	107	106	77		9	21	24	244							7	48	47	121	6	20	26	197		
11	96	100	175		10	31	37	113							8	38	46	281	7	41	45	15		
12	72	78	66		11	22	28	155							9	31	36	189	4	30	3	21	28	97
13	50	57	195		12	30	39	311							10	30	39	302	5	34	37	72		
14	26	31	101		4	77	75	270							11	34	35	170	7	18	23	84		
15	18	16	176		1	144	145	147							0	74	77	270	4	31	0	97	106	270
4	2	1	26	14	2	44	45	257							1	32	41	330	2	46	50	297		
3	131	117	10		3	76	77	213							2	71	70	310	3	20	27	320		
4	65	59	343		4	64	62	220							3	75	78	159	4	44	49	257		
5	69	65	177		5	57	66	216							4	53	57	194	5	27	32	252		
6	77	76	8		6	56	45	206							5	36	39	344	6	47	59	270		
7	55	48	166		7	19	18	53							6	21	23	30	8	15	15	346		
8	69	54	21		9	21	24	250							8	38	47	77	4	32	6	24	19	335
9	20	33	196		10	31	36	143							9	31	40	106	4	33	1	47	55	188
11	58	70	168		0	273	282	90							10	72	73	108	3	18	27	177		
1	86	94	90		1	109	110	315							11	34	39	347	4	43	54	176		
1	96	99	311		2	205	211	67							12	46	54	71	5	41	49	180		
2	146	145	205		3	145	140	29							0	19	10	15	7	19	27	188		
3	79	65	240		4	176	171	91							2	48	50	15	4	18	18	90		
4	31	33	185		5	114	114	62							3	20	24	248	2	25	24	79		
6	132	113	323		6	125	124	62							4	35	40	51	4	34	42	90		
7	177	171	114		7	27	41	309							5	30	26	267	1	17	24	261		
7	152	140	274		8	77	81	40							6	65	62	34	2	16	23	89		
8	123	115	180		9	33	32	57							7	31	27	224	3	21	25	100		
9	54	52	336		10	22	30	123							8	22	31	28	5	24	34	90		
10	86																							

TABLE I (Continued)

<i>k</i>	<i>h</i>	<i>l</i>	<i>F_O</i>	<i>F_C</i>	<i>k</i>	<i>h</i>	<i>l</i>	<i>F_O</i>	<i>F_C</i>	<i>k</i>	<i>h</i>	<i>l</i>	<i>F_O</i>	<i>F_C</i>	<i>k</i>	<i>h</i>	<i>l</i>	<i>F_O</i>	<i>F_C</i>		
		11	36	31	332			2	76	81	206			7	36	29	306	5	45	50	143
5	2	1	18	23	120			3	21	19	20			8	47	47	101	6	16	19	112
		2	97	94	89			4	54	50	263			10	27	30	130	7	25	28	149
		3	41	40	355			5	88	77	3			11	24	22	332	5	18	19	64
		4	105	96	113			6	76	72	197			1	34	42	248	6	26	27	326
		5	24	19	335			7	61	58	322			3	73	74	331	5	30	31	90
		6	163	145	85			8	64	62	206			4	58	55	333	1	25	30	230
		8	68	60	213			9	62	61	14			5	47	41	230	2	30	32	86
		9	20	19	79			10	30	33	294			6	21	27	277	3	17	24	144
		10	62	63	91			1	86	78	281			7	36	40	258	4	16	36	61
5	3	0	55	61	270			2	15	22	128			8	29	27	303	5	15	19	87
		1	44	45	360			3	22	15	77			9	28	35	308	6	30	35	208
		2	16	14	291			4	40	39	75			10	32	39	288	5	13	14	23
		3	39	35	168			5	60	54	53			11	23	27	156	6	12	17	23
		4	80	73	331			6	26	22	141			1	45	41	354	5	30	31	90
		5	92	86	330			7	75	71	289			2	81	71	254	3	15	22	107
		6	27	32	302			8	20	30	102			3	68	61	353	3	20	23	19
		7	30	21	39			9	51	50	27			4	66	57	276	4	29	37	15
		9	20	24	121			11	54	51	358			5	21	17	27	5	16	19	1
		10	29	29	308			1	79	74	106			6	81	81	254	5	19	26	306
5	4	1	54	50	349			2	128	122	252			7	21	31	340	5	13	18	158
		2	82	78	230			3	65	63	165			8	20	8	138	6	27	22	34
		3	65	50	222			4	93	76	293			9	19	28	340	6	20	23	94
		4	87	82	346			5	18	25	160			1	50	41	90	4	183	181	97
		5	155	146	3			6	62	62	233			2	21	29	212	6	233	222	83
		6	32	30	111			7	56	57	83			5	47	44	221	8	64	53	228
		7	58	50	340			8	29	39	222			6	21	23	88	1	42	51	358
		8	33	31	311			9	21	27	179			7	21	25	209	2	13	7	110
		9	35	37	39			10	30	27	294			8	25	34	233	4	18	21	217
		10	36	34	45			1	57	54	175			9	32	38	211	5	42	39	292
5	5	0	55	52	350			2	57	54	175			11	33	39	135	6	16	16	216
		1	76	75	126			3	66	60	16			1	31	27	270	6	2	263	339
		2	58	48	60			4	99	86	330			2	93	87	171	2	62	62	336
		3	78	71	248			5	55	45	57			2	72	70	271	3	148	142	13
		4	29	31	169			6	51	46	215			3	21	18	215	4	107	107	7
		5	75	66	230			8	21	38	274			4	36	45	236	5	228	224	3
		6	86	80	256			9	36	29	43			5	73	74	168	6	90	82	13
		7	18	14	125			10	75	74	291			6	29	28	224	7	76	66	8
		8	47	40	345			11	28	29	56			7	20	22	184	8	72	73	346
		10	30	32	256			1	64	64	90			8	34	40	217	9	54	46	132
		11	30	31	138			2	44	41	315			9	26	20	275	10	72	68	11
5	6	0	160	163	270			1	127	129	159			10	16	17	194	11	55	45	318
		1	24	22	124			3	122	124	175			11	20	27	164	13	25	32	167
		2	15	25	219			4	40	33	122			1	39	40	3	2	35	28	42
		3	55	52	203			5	70	66	193			2	21	35	327	3	26	19	353
		4	88	74	297			6	28	23	58			3	35	40	338	4	33	34	197
		5	88	73	296			7	41	51	85			4	20	21	246	4	33	35	222
		6	109	105	239			8	30	28	167			5	27	30	139	5	23	30	249
		9	54	55	251			10	29	20	138			6	24	25	203	6	224	272	270
		10	59	54	277			1	80	86	120			7	31	39	94	1	12	20	2
		11	36	39	308			2	48	36	304			8	22	24	275	2	92	88	266
5	7	0	31	28	90			3	18	32	230			9	73	68	90	3	135	136	256
		1	41	42	153			4	76	69	321			1	36	37	165	4	117	114	250
		2	59	59	35			5	76	80	216			3	66	64	144	5	184	189	278
		3	63	54	346			6	66	54	357			4	21	24	129	6	157	145	260
		4	69	65	134			7	62	53	134			5	41	45	113	7	119	118	281
		5	72	58	184			8	59	60	289			6	53	54	91	8	39	40	345
		6	29	29	45			9	47	48	207			7	33	33	122	9	104	99	259
		7	63	53	180			10	54	49	341			8	18	26	184	10	72	62	288
		8	47	49	21			11	33	43	162			9	29	32	106	11	71	69	294
		9	20	29	230			1	83	89	90			10	14	22	103	12	28	34	235
		10	47	47	126			2	17	19	247			1	42	46	270	6	30	36	270
		11	66	59	178			3	119	122	67			2	21	28	211	1	19	25	131
5	8	0	41	47	270			4	31	34	135			3	21	28	211	4	44	44	214
		1	97	102	118			4	92	85	120			4	21	20	190	5	15	16	172
		2	67	64	357			8	21	23	22			5	40	42	346	7	18	22	238
		3	55	44	208			9	36	37	86			6	48	51	205	10	21	29	127
		4	117	109	159			11	27	31	22			7	32	32	2	6	14	26	270
		5	78	73	184			1	91	92	90			8	24	32	125	1	50	48	121
		6	70	61	235			1	18	14	107			9	23	18	12	2	57	47	135
		7	69	73	85			2	58	53	73			10	18	21	161	3	58	50	301
		9	71	58	216			3	92	88	197			0	42	41	90	4	178	176	169
		10	21	29	161																

TABLE 1 (Continued)

<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c	<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c	<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c	<i>k</i>	<i>h</i>	<i>l</i>	<i>F</i> _o	<i>F</i> _c						
7	26	9	26	34	81	8	6	5	76	69	177	8	13	4	89	81	161	3	38	33	330				
		0	35	46	270			0	89	97	90			0	64	63	90	4	27	31	283				
		3	20	27	282			1	71	78	328			1	73	71	146	5	38	34	222				
		6	44	51	286			2	17	18	155			2	62	62	48	6	27	34	234				
		7	29	29	317			4	56	52	357			3	60	51	179	8	20	0	66	70	90		
		8	22	35	9			5	84	73	7			4	44	53	128	2	38	51	53				
		9	18	33	314			0	26	42	71			5	64	61	198	3	47	60	188				
7	28	3	27	46	343	8	7	0	78	78	90			6	27	38	40	4	47	46	75				
		5	25	34	171			1	64	65	272			8	14	0	33	53	270	6	47	58	470		
8	0	2	36	33	242			2	18	24	182			1	41	52	270	8	21	2	27	42	259		
		4	76	62	245			3	20	30	69			3	25	31	75	4	38	32	211				
		6	125	117	251			4	38	47	133			5	38	47	45	8	22	1	55	63	350		
8	1	1	53	66	352			6	76	69	103			8	15	0	65	58	90	2	61	63	81		
		4	28	35	294			0	47	48	270			1	60	61	323	4	38	35	108				
		5	73	63	332			1	65	61	42			2	86	84	150	5	53	59	339				
		6	24	36	228			2	19	22	241			4	59	57	111	8	23	3	38	46	165		
8	2	0	18	24	270			3	29	40	262			5	38	35	15	8	24	0	54	61	270		
		1	88	108	176			4	74	70	324			6	38	43	34	1	38	42	355				
		4	20	33	213			5	42	50	300			8	16	1	25	35	211	3	53	53	343		
		5	142	136	167			0	27	37	270			2	44	51	201	6	34	44	260				
8	3	0	94	130	270			1	33	42	345			4	38	37	309	8	25	5	24	30	30		
		1	39	46	107			2	28	33	188			5	27	31	104	8	26	0	36	38	270		
		3	35	36	317			3	37	38	66			6	27	40	229	1	36	33	198				
		4	28	34	236			4	46	48	320			8	17	0	26	32	270	2	25	39	265		
		6	73	66	243			5	73	67	40			1	68	61	341	5	32	35	198				
8	4	0	105	136	90			1	64	63	116			3	46	50	350	8	27	0	25	38	270		
		1	63	65	218			2	21	23	350			4	27	22	350	5	22	27	125				
		2	22	29	99			3	54	46	243			5	38	42	335	8	28	0	24	27	90		
		4	70	69	95			4	33	44	190			8	18	1	59	60	152	5	21	20	114		
		6	92	84	62			5	70	64	254			3	27	34	168	8	29	1	32	40	184		
8	5	0	20	32	90			0	51	42	270			4	27	29	334	5	27	37	169				
		1	67	72	189			1	82	74	128			5	39	42	194	8	31	0	44	52	90		
		2	44	42	294			2	31	38	252			8	19	0	27	33	270	2	19	28	86		
		4	47	43	157			3	40	44	186														

TABLE 2

RuQASBr₂Some distances in Å and standard deviations in Å × 10³ shown in parentheses*Within the bridging groups*

Ru-As1	2.308(5)	As3-C14	1.974(30)	C7-C8	1.438(57)
Ru-As2	2.398(5)	As3-C31	1.945(40)	C8-C9	1.438(56)
Ru-As3	2.465(5)	As3-C37	1.901(40)	C9-C10	1.532(57)
Ru-As4	2.472(5)	As4-C8	1.947(42)	C10-C11	1.369(65)
Ru-Br1	2.610(5)	As4-C43	1.938(40)	C11-C12	1.369(55)
Ru-Br2	2.615(5)	As4-C49	1.991(40)	C12-C7	1.413(47)
As1-C1	1.940(40)	C1-C2	1.400(47)	C13-C14	1.398(51)
As1-C7	1.960(29)	C2-C3	1.441(52)	C14-C15	1.419(51)
As1-C13	1.947(39)	C3-C4	1.489(57)	C15-C16	1.417(52)
As2-C2	1.915(36)	C4-C5	1.365(51)	C16-C17	1.458(58)
As2-C19	2.049(38)	C5-C6	1.420(59)	C17-C18	1.369(57)
As2-C25	1.966(39)	C6-C1	1.448(60)	C18-C13	1.459(51)

In the phenyl groups

C-C Average 1.41; average deviation 0.04; maximum deviation 0.10 Å

Some angles in degrees and standard deviations in degrees × 10 shown in parentheses

As1-Ru-As2	86.7(2)	Ru-As2-C2	106.1(10)	As2-C2-C1	121.5(28)
As1-Ru-As3	82.7(2)	Ru-As2-C19	120.3(27)	As3-C14-C13	117.8(24)
As1-Ru-As4	82.7(2)	Ru-As2-C25	120.7(11)	As4-C8-C7	121.9(28)
As2-Ru-As3	96.6(2)	C2-As2-C19	108.9(25)	C6-C1-C2	123.2(36)
As2-Ru-As4	97.1(2)	C2-As2-C25	150.5(15)	C1-C2-C3	116.4(33)
As3-Ru-As4	159.3(2)	C19-As2-C25	94.2(36)	C2-C3-C4	120.9(30)
Br1-Ru-As1	88.1(2)	Ru-As3-C14	103.5(9)	C3-C4-C5	118.0(37)
Br1-Ru-As2	174.8(2)	Ru-As3-C31	123.8(11)	C4-C5-C6	122.7(37)
Br1-Ru-As3	82.5(2)	Ru-As3-C37	123.9(12)	C5-C6-C1	117.3(35)
Br1-Ru-As4	82.5(2)	C14-As3-C31	97.2(16)	C12-C7-C8	121.7(31)
Br1-Ru-Br2	92.4(2)	C14-As3-C37	104.9(15)	C7-C8-C9	120.1(35)
Br2-Ru-As1	178.9(2)	C31-As3-C37	99.2(16)	C8-C9-C10	115.3(36)
Br2-Ru-As2	92.8(2)	Ru-As4-C8	102.6(13)	C9-C10-C11	119.2(38)
Br2-Ru-As3	98.4(2)	Ru-As4-C43	123.9(12)	C10-C11-C12	124.4(38)
Br2-Ru-As4	96.4(2)	Ru-As4-C49	122.3(11)	C11-C12-C7	118.6(36)
Ru-As1-C1	110.3(11)	C8-As4-C43	107.2(17)	C18-C13-C14	119.7(34)
Ru-As1-C7	114.0(10)	C8-As4-C49	96.6(17)	C13-C14-C15	121.5(30)
Ru-As1-C13	111.5(12)	C43-As4-C49	100.2(16)	C14-C15-C16	118.4(34)
C1-As1-C7	104.0(15)	As1-C1-C2	115.3(28)	C15-C16-C17	120.4(35)
C1-As1-C13	109.5(17)	As1-C7-C8	109.2(24)	C16-C17-C18	120.2(35)
C7-As1-C13	107.1(13)	As1-C13-C14	114.5(26)	C17-C18-C13	119.7(35)

C-C-C in phenyl groups; average 119.9°; average deviation 3.9°; maximum deviation 10°.

TABLE 3
RuQASBr₂

Atomic co-ordinates in fractions of the unit cell lengths, *b*, *a*, and *c* and isotropic thermal factors in Å⁻²

	Y	X	Z	B		Y	X	Z	B
Ru	0.17175	0.12016	0.00075	3.612	C25	0.3350	0.2004	-0.1634	4.245
Br1	0.02127	0.05987	0.03928	4.606	C26	0.2254	0.2143	-0.2056	9.166
As1	0.31962	0.07205	0.03372	3.556	C27	0.2355	0.2429	-0.2884	9.655
As2	0.32378	0.17107	-0.03054	3.756	C28	0.3344	0.2541	-0.3448	5.363
Br2	0.00548	0.17469	-0.04045	4.587	C29	0.4313	0.2381	-0.3023	9.490
As3	0.17119	0.12685	0.18927	3.525	C30	0.4422	0.2092	-0.2172	8.916
As4	0.17533	0.08630	-0.17039	3.860	C31	0.2809	0.1620	0.2656	4.782
C1	0.4717	0.0983	0.0128	5.002	C32	0.3960	0.1473	0.2693	5.172
C2	0.4688	0.1411	-0.0151	3.365	C33	0.4818	0.1741	0.3259	5.047
C3	0.5798	0.1630	-0.0163	5.072	C34	0.4440	0.2131	0.3691	6.528
C4	0.6920	0.1392	-0.0063	7.237	C35	0.3283	0.2269	0.3618	6.050
C5	0.6863	0.0970	0.0164	5.478	C36	0.2462	0.2013	0.3034	6.555
C6	0.5789	0.0758	0.0369	7.388	C37	0.0339	0.1366	0.2697	3.892
C7	0.3208	0.0232	-0.0591	3.915	C38	0.0427	0.1428	0.3768	5.832
C8	0.2376	0.0302	-0.1404	5.316	C39	-0.0694	0.1532	0.4318	6.497
C9	0.2105	-0.0034	-0.2112	5.214	C40	-0.1694	0.1560	0.3806	7.544
C10	0.2874	-0.0431	-0.2021	7.063	C41	-0.1803	0.1474	0.2715	5.441
C11	0.3624	-0.0468	-0.1206	5.876	C42	-0.0736	0.1390	0.2121	4.798
C12	0.3842	-0.0152	-0.0510	5.257	C43	0.0378	0.0785	-0.2556	5.046
C13	0.3100	0.0500	0.1730	5.261	C44	-0.0660	0.0904	-0.2152	6.875
C14	0.2305	0.0713	0.2363	3.764	C45	-0.1653	0.0823	-0.2721	5.823
C15	0.2049	0.0568	0.3369	5.093	C46	-0.1701	0.0683	-0.3705	7.540
C16	0.2635	0.0202	0.3734	5.670	C47	-0.0599	0.0550	-0.4153	5.600
C17	0.3464	-0.0023	0.3080	5.942	C48	0.0426	0.0623	-0.3556	6.033
C18	0.3680	0.0117	0.2104	5.016	C49	0.2931	0.1017	-0.2769	5.573
C19	0.3313	0.2259	0.0536	6.770	C50	0.4008	0.0864	-0.2635	6.868
C20	0.2269	0.2474	0.0520	6.112	C51	0.4850	0.1008	-0.3423	8.192
C21	0.2305	0.2867	0.1119	8.380	C52	0.4606	0.1239	-0.4243	6.469
C22	0.3363	0.2986	0.1583	6.681	C53	0.3406	0.1389	-0.4344	6.757
C23	0.4306	0.2752	0.1622	5.911	C54	0.2588	0.1291	-0.3604	5.930
C24	0.4382	0.2356	0.1047	5.071					

given in Table 1, positional and thermal parameters in Table 3 and bond angles and distances together with the estimated standard deviations in Table 2.

We thank Dr. L. M. Venanzi for material, the D.S.I.R. for a maintenance grant (to R. H. B. M.), and the Director and staff of the Oxford University Computing Laboratory for facilities.

CHEMICAL CRYSTALLOGRAPHY LABORATORY, OXFORD.

[Received, January 9th, 1964.]