Corrigendum

Rotational Isomerism in 1,1′-Dinitrobicyclopentyl, 1,1′-Dinitrobicyclohexyl and 1,1′-Dinitrobicycloheptyl

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Page 2035: *Delete* Table 5 and *replace* with the following.

Table 5 Comparison between 1,1′-dinitro- and 1,1′-dicyano-bicycloalkanes

Compound	ΔE^+/kJ mol^{-1}		Torsion angle of *gauche* rotamer/°		% *gauche* population		Barrier height/kJ mol^{-1} between *gauche* and *trans* forms
	Experimental[d]	AM1	Experimental	AM1	Experimental	AM1	AM1
1,1′-Dinitrobicyclopentyl[a]	−2.25	−1.69	67	58	83	80	21.82
1,1′-Dinitrobicyclohexyl[b]	2.59	1.44	90	59	42	53	32.68
1,1′-Dinitrobicycloheptyl[b]	5.99	5.82	64	57	16	16	49.92
	Experimental	MMP2	Experimental	MMP2	Experimental	MMP2	MMP2
1,1′-Dicyanobicyclopentyl[c]	9.28	10.71	85	61	4.5	2.5	22.34
1,1′-Dicyanobicyclohexyl[c]	6.60	11.17	86	60	11	2	27.48
1,1′-Dicyanobicycloheptyl[c]	3.05	3.51	67	52	38	33	30.25

[a] AM1 calculations with full geometry optimization. [b] AM1 calculations using fixed geometries. [c] MMP2 calculations with full geometry optimization. [d] Experimental values are based on the Lennard-Jones–Pike method of calculation using data from measurements on CCl_4 solutions. ΔE^+ is defined as $E(gauche) - E(trans)$.

Non-linear Brønsted Correlations: Evidence for a Levelling Off in the Reactivity of Oximate Ions in Aqueous Solution

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Page 157: *Delete* eqn. (13) and *replace* with

$$k_{app} = \frac{1}{[PNPA]_0}\left(\frac{d[PNP^-]}{dt}\right)_0 = \frac{k_0 + \dfrac{k_1B + k_2C}{A}[Ox]_0}{1 + 10^{7.05 - pH}} \quad (13)$$

Page 158: *Delete* eqn. (17) and *replace* with

$$k_{app} = \frac{1}{[PNPA]_0}\left(\frac{d[PNP^-]}{dt}\right) = k_0 + k_2[Ox]_0 \quad (17)$$