# Low-frequency Single-crystal Vibrational Spectra of Bis(thiourea)dichlorocadmium and Tetrakis(thiourea)dichlorocadmium

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Raman spectra are reported for single crystals of the title compounds together with single-crystal i.r. reflectance data for the tetrakis-complex. The bis-complex is discussed on the basis of a factor-group analysis, but for the more complex unit cell of the tetrakis-complex a molecular model provides an adequate basis for assignment.

CO-ORDINATION complexes such as the title compounds yield complicated vibrational spectra, particularly in the low-frequency region. Assignment of such spectra is generally attempted on the basis of qualitative evidence, typically changes induced by variation of halogen and of metal atom. Although by no means proven, these assignments are commonly invoked by others interpreting the spectra of still more elaborate co-ordination compounds. In this paper we attempt to strengthen the assignment of the spectra of two thiourea complexes using single-crystal techniques. Two advantages stem from this approach. (i) Close-lying bands of different symmetry species are readily revealed in contrast to the rather broad composite bands commonly found in the spectra of polycrystalline samples. (ii) Knowledge of symmetry species considerably assists assignment, as

shown below. Since all of the internal modes of thiourea are found above  $400 \text{ cm}^{-1}$  this study is restricted to lower frequencies.

## EXPERIMENTAL

 $CdCl_2(tu)_2$  (tu = thiourea) was prepared by mixing stoicheiometric amounts of  $CdCl_2, 2\frac{1}{2}H_2O$  and thiourea in aqueous ethanol. Crystals were grown by evaporation of the solution over several days. They showed well-developed 001 and 110 faces (Figure), although in some the 010 face was slightly developed; grinding and polishing was necessary to increase the size of this face.  $CdCl_2(tu)_4$  was formed from stoicheiometric quantities of the reactants in absolute ethanol. Very slow evaporation of the solution over a period of weeks yielded crystals with well-developed 100 and 010 faces.

Crystals were oriented by X-ray diffraction techniques.

Raman spectra were obtained with a Coderg PH1 instrument using 632.8 and 488.0 nm excitation and a 'Cryocirc' cold cell. I.r. spectra were obtained using Beckman-RIIC FS-620 and FS-720 interferometers.



## RESULTS AND DISCUSSION

 $CdCl_2(tu)_2$ .—This complex crystallises in the orthorhombic system with the symmetry of space-group  $Pm2_1n$  ( $C_{2v}^r$ ), z = 2.1 Considering the molecules as  $CdCl_2(SC)_2$  the factor-group analysis in Table 1 results.

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	Fac	tor-gro	up anal	ysis of	CdCl <sub>2</sub> (S	$SC)_2$	
C?,	$N_{T}$	Α	Т	R	Ni	Activit	у
$A_1$	12	1	1	1	9	z; xx,	yy,zz
$A_{2}^{\dagger}$	9	0	1	<b>2</b>	6	xy	
$B_1^-$	9	1	0	<b>2</b>	6	x; xz	
$B_2$	12	1	1	1	9	y; yz	
A =	= Acoust	ic mod	les; T	= trans	slatory	lattice	modes
_			-	37 .		1 1	-

R = rotatory lattice modes;  $N_i$  = internal molecular modes excluding  $\nu$ (C=S) but including torsions.

v(C-S) modes occur well above the range of our study and must be subtracted but we must add the lowenergy torsional modes of thiourea (about the Cd-S bonds) which are excluded by the above approximation. Molecular symmetry is  $C_{2v}$ , with the molecules occupying sites of  $C_s$  symmetry: correlation of molecular and unitcell labels is shown in Table 2, and approximate descriptions of molecular modes in Table 3.

TABLE 2 Correlation table for  $CdCl_2(SC)_2$ 



The crystals showed pronounced dependence of spectrum upon orientation as evidenced by the intensities listed in Table 4. The observed Raman lines lie in three groups: (a) 170–250 cm<sup>-1</sup>, (b) 125–150 cm<sup>-1</sup>, (c) <110 cm<sup>-1</sup>. Since stretching modes must come higher in frequency than their associated bending modes  $\nu$ (Cd-Cl) will be in the highest group, a conclusion strengthened by much qualitative evidence on halogeno-complexes. Similarly  $\nu$ (Cd-S) will be higher than  $\delta$ (SCdS) and almost certainly above  $\delta$ - and  $\pi$ -(CdSC) modes. However, due to the small energy separations

## TABLE 3

Approximate mode descriptions for molecular  $CdCl_2(SC)_2, C_{2v}$ 

<i>a</i> <sub>1</sub>	$\nu_1 \\ \nu_2 \\ \nu_3 \\ \nu_4 \\ \nu_4$	$\nu(S-C)$ $\nu(Cd-S)$ $\nu(Cd-Cl)$ $\delta(CdSC)$	$b_1$	$   \nu_9  $ $   \nu_{10}  $ $   \nu_{11}  $ $   \nu_{12}   $	$\begin{array}{l} \nu(S-C) \\ \nu(Cd-S) \\ \delta(CdSC) \\ \rho_w(CdCl_2) \end{array}$
	V 5		h		-(Cd-CI)
	<i>v</i> <sub>6</sub>	0(3003)	0	ν <sub>13</sub> ν <sub>14</sub>	$\pi(Cd=Cl)$ $\pi(CdSC)$
$a_2$	$\nu_7$	$\pi(CdSC)$		$\nu_{15}$	$\rho_r(CdCl_2)$
	$\nu_8$	$CdCl_2$ $CdS_2$ torsion			
	Tors	ions of thiourea about S	S-C bond,	$a_2 +$	<i>b</i> <sub>2</sub> .

TABLE 4

Raman frequencies (cm<sup>-1</sup>), peak heights (arbitrary units), and assignments for single-crystal CdCl<sub>2</sub>(tu)<sub>2</sub> at ambient temperature

	Factor-					
	group					Molecular
	assignment	;				assignment
	Č7,	z(xx)y	z(yx)y	z(xz)y	z(yz)y	C 20
<b>28</b>	$\vec{B}_{\bullet}$	4			2	<u>۱</u>
38	$A_{1}$					Lattice
44	$B_1$			9		modes
<b>58</b>	$B_{2}$	3		4	170	J
69	$A_2$		74	4		$\nu_8$
<b>72</b>	$A_1$	18				<u></u>
75	$B_2$				6	<b>β</b> γ <sub>6</sub>
<b>78</b>	$B_1$			31		$\nu_{12}$
84	$A_{1} + B_{2}$	12			47	$\nu_5$
91	$A_2$		12			Ligand
						<pre>&gt; torsions?</pre>
93	$B_1$			4		$V_{16 17}$
101	$B_2$				14	$\nu_{15}$
106	$A_2$	0	12		-	$\nu_7$
125	$B_2$				6	$\nu_{14}$
137	$B_1$			15		$\nu_{11}$
148	$A_1$	21			0	v4
172	$B_2$		0	0	42	} V10
182	$(B_2)^{a}$				•	j -13
206	$A_1$	6	0	-	0	Ya + Ya
224	$A_1$	86	2	7	8	) 2173
234	$A_2$		I	0		} V10
244	$B_1$			3		j <b>10</b>
	<sup>a</sup> From	i.r. spec	trum, sy	mmetry	y assume	ed.

between bands there must be substantial mixing of internal co-ordinates; our descriptions, Table 3, must therefore be regarded as a guide only.

The highest band observed is of  $B_1$  symmetry; it must, therefore, originate (Table 2) in a molecular mode of either  $a_2$  or  $b_1$  symmetry. Therefore (Table 3) it is to be associated with  $v_7(a_2)$ ,  $v_{10}$ , or  $v_{11}(b_1)$ ; we regard  $v_{10}$ , v(Cd-S), as the most probable description. If the region 170–250 cm<sup>-1</sup> is associated with v(Cd-S) and v(Cd-Cl) modes we would expect to find  $3A_1 + 1A_2 +$ 

<sup>1</sup> M. Nardelli, L. Cavalca, and A. Braibanti, *Gazzetta*, 1957, 87, 137.

 $1B_1 + 3B_2$  modes since v(Cd-S) spans the representation  $A_1 + A_2 + B_1 + B_2$  and v(Cd-Cl)  $2A_1 + 2B_2$ . A sole  $A_2$  mode is, in fact, observed at 234 cm<sup>-1</sup>; its proximity to the sole  $B_1$  strengthens our supposition that both are associated with v(Cd-S) motions. Certainly, from symmetry considerations, they are not due to v(Cd-Cl); unless  $\delta(CdSC)$  is higher than  $\nu(Cd-Cl)$  (an improbable event) the above argument and assignment is valid. Two of the expected three  $A_1$  modes are found and must clearly be described as  $\nu(Cd-S) + \nu(Cd-Cl)$ ,  $\nu_{2,3,13}$ . Correlation interaction yields a  $B_2$  mode at 172 cm<sup>-1</sup>, the separation of ca. 30 cm<sup>-1</sup> from the  $A_1$  components being reasonable for metal-halogen motion. In addition there is an intense band at 182 cm<sup>-1</sup> in the i.r. spectrum,<sup>2</sup> not found in the Raman experiments; this must be of  $A_1$  or  $B_2$  symmetry the latter being the more probable. The association of the 172 cm<sup>-1</sup> line with a Cd-Cl mode is supported by its non-appearance in Raman spectra of polycrystalline samples of  $CdBr_2(tu)_2$  and  $CdI_2(tu)_2$ , Table 5.

## TABLE 5

Raman spectra (cm<sup>-1</sup>) of powder samples of CdX<sub>2</sub>(tu)<sub>2</sub>

X =	Cl	$\mathbf{Br}$	I		
	243w	234w	<b>232</b> wm	l	u(Cd-S)
	224s	) 214s	<b>21</b> 5wm	ſ	v(Cu-5)
	205w	> 160s	136vs		$\nu$ (Cd-X)
	173wm				,
	147wm	144m		1	N/0.10(0)
	138wm			Ĵ	o(CaSC)
	106vw	108w	100 wm		
		<b>94</b> wm			
		87m	81ms		
	73s	71w	64w		
	58s	58s	54vs		
		53vs	37vs		
		<b>38</b> m	$28 \mathrm{wm}$		
			23ms		

Factor-group analysis <sup>3</sup> predicts  $6B_1$  internal modes for two molecules  $CdCl_2(SC)_2$ . One of these is  $\nu(S=C)$  and is above the frequency range discussed but there is a torsional mode which will be of low frequency and probably also of low intensity which must be added, keeping the total at six. There are no translational modes in  $B_1$ and the two rotatory modes allowed cannot be high in frequency. Coupling between the two molecules in the unit cell requires  $B_1$  modes to be accompanied by an  $A_2$ component and vice-versa. There are  $A_2/B_1$  pairs at 234/244 and 91/93 cm<sup>-1</sup>; it is permissible to assume that the  $A_2$  modes at 105 and 69 cm<sup>-1</sup> and the  $B_1$  modes at 137 and 78 cm<sup>-1</sup> have vanishingly weak  $B_1$  and  $A_2$  components respectively. Thus in the region above 60 cm<sup>-1</sup>, the region in which external modes are unlikely for this molecular crystal, it appears that all six of the internal mode pairs have been located.

Why should some of the components be missing? In this crystal an  $a_2$  molecular mode yields  $A_2 + B_1$ factor-group components. Thus, if we assume that  $a_2$  molecular modes are responsible for the  $A_2$  unit-cell

modes with missing  $B_1$  components we find that these must be assigned as  $v_7$  and  $v_8$  respectively. Similarly the lone  $B_1$  modes at 137 and 78 cm<sup>-1</sup> may be labelled  $v_{11}$  and  $v_{12}$  respectively. In any case, its position as the second highest  $B_1$  mode observed makes the 137 cm<sup>-1</sup> line a strong candidate for  $v_{11}$ . The  $A_2/B_1$  pair ca. 92 cm<sup>-1</sup> could be one of torsional modes, or  $\nu_{12}$ 

Similar arguments lead to the assignments shown for the  $A_1, B_2$  unit-cell modes.

A partial assignment of the Raman spectra of the powders  $CdBr_2(tu)_2$  and  $CdI_2(tu)_2$  follows by comparison with the crystal assignment. The highest feature, assigned as  $v_{10}(vCd-S)$  is roughly constant in frequency. The most striking change evident when looking at the three powder spectra together is that the most intense band moves steadily to lower frequency with change to heavier halogen. Bands ca. 140 cm<sup>-1</sup> are also present throughout [due to  $\delta$ (CdSC)] but in the iodide the intense  $\nu$ (Cd–I) band overlays them.

 $CdCl_2(tu)_4$ .—This compound is isostructural with the cobalt analogue for which the full three-dimensional

	,	TABLE 6				
Fac	ctor-group	analysis o	of CdCl <sub>2</sub> (S	SC)4		
$C_{4h}^4$	$A_{g}$	$B_{g}$	$E_{g}$	$A_{ m u}$	$B_{\mathbf{u}}$	$E_{\mathbf{u}}$
$N_T$	15	15	15	18	18	18
Α				1		1
Т				2	3	2
R	3	3	3			
$N_i$	12	12	12	15	15	15
$\mathbf{Activity} ig \{ \mathbf{I.r.} \\ \mathbf{Ramar} ig \}$	xx + yy;	xx — yy;	(xz,yz)	Z		<i>x</i> , <i>y</i>
	<b>Z</b> Z	xy				

structure has been determined.<sup>4</sup> It has the symmetry of the tetragonal group  $P4_2/n$  (C<sup>4</sup><sub>4h</sub>), z = 4. With four molecular units in the primitive cell, factor-group analysis predicts a very large number of optically active modes for this crystal, Table 6. The physical process which generates this multiplicity of modes is correlation interaction between the four symmetry-related molecules. Proceeding via the site symmetry,  $C_i$ , we see (Table 7)

## TABLE 7 Correlation Table for CdCl<sub>2</sub>(SC)<sub>4</sub>



that, for example, each molecular  $a_g$  mode (assuming idealised  $C_{4h}$  symmetry, see below) will yield a vector <sup>4</sup> J. E. O'Connor and E. L. Amma, Inorg. Chem., 1969, 8, 2367.

<sup>&</sup>lt;sup>2</sup> D. M. Adams and J. B. Cornell, J. Chem. Soc. (A), 1967, 884. <sup>3</sup> D. M. Adams and D. C. Newton, 'Tables for Factor Group and Point Group Analysis,' Beckman-RIIC Limited, Croydon, 1970.

 $A_g + B_g + E_g$ . Interaction of this type is clearly evidenced by the correlation splittings observed in such crystals as  $Cr(CO)_3(\pi-C_6H_6)^5$  and  $K_2CrO_4.^6$  However, these are crystals in which the bonds which are associated with the most strongly perturbed vibrations are oriented in such a way that physical interaction between molecular units is strong. In this thiourea complex the molecules are 'insulated' from each other by the bulky ligands. If we consider the region below 400 cm<sup>-1</sup> (because all of the ligand modes are higher than this), we might reasonably expect to interpret the crystal spectrum in terms of that due to a single molecule, plus lattice modes. We note in particular that the nonbonded Cl · · · Cl distances are so large that no correlation interaction can be physically significant. The numbers of bands found in each  $C_{4h}$  symmetry species is well below that required by factor-group analysis but close to what is required for one molecule plus lattice modes.

Molecular symmetry is only  $C_i$  if we include the ligands. The modes of the  $CdS_4Cl_2$  kernel might reasonably be considered in terms of a point group with a  $C_4$  axis and, since this is approximately parallel to the crystal c axis, the symmetry species can be correlated directly with those of the factor group  $(C_{4h})$ . Combining a  $C_4$  axis with the known centre of symmetry requires  $C_{4h}$  to be used as the approximation to molecular symmetry, thus making a one-to-one correspondence between molecular and crystal symmetry species. Deviations from this model are to be expected for modes involving Cd-S-C bending and torsions of thiourea with respect to the rest of the molecule; a few weak bands might therefore appear in addition to those predicted by the molecular  $C_{4h}$  model (Table 8) due to lifting of degeneracies.

## TABLE 8

Approximate normal modes for  $CdCl_2(SC)_4$  with idealised  $C_{4b}$  symmetry

				2	
ag	$ $	$\nu(C-S)$ $\nu(Cd-S)$ $\nu(Cd-Cl)$ $\delta(CdSC)$	a <sub>u</sub>	$     \frac{     \nu_{11}}{     \nu_{12}}     \nu_{13}   $	$\nu$ (Cd–Cl) $\delta$ (CdSC) $\delta$ (SCdCl)
	-4	0(0000)	b.,	v	δ(CdSC)
$b_{g}$	$\nu_5$	$\nu(C-S)$	υu	$\nu_{15}^{\nu_{14}}$	$\delta(SCdCl)$
	$\nu_6$	$\nu(Cd-S)$			
	$\nu_7$	$\delta(CdSC)$	$e_{\mathbf{u}}$	$\nu_{16}$	$\nu(C-S)$
	Vs	δ(SCdS)		$\nu_{17}$	$\nu$ (Cd $-$ S)
	Ū	. ,		V19	$\delta(CdSC)$
$e_{\mathbf{g}}$	$\nu_9$	$\delta$ (CdSC)		$\nu_{19}^{13}$	δ(SCdS)
	$\nu_{10}$	δ(SCdCl)		$\nu_{20}$	δ(SCdCl)
	Torsions of	thioures about C-	S bond	$ls a \perp b$	$h \perp e_{\mu}$

Torsions of thiourea about C-S bonds,  $a_g + b_g + e_u$ .

 $B_{\rm g}$  Species.—One  $B_{\rm g}$  mode v(Cd-S), v<sub>6</sub>, is expected well above the others followed by v<sub>7</sub>,  $\delta$ (CdSC) and v<sub>8</sub>,  $\delta$ (SCdS); Raman lines at 209, 147, and 87 cm<sup>-1</sup> are accordingly assigned to them (Table 9). The lines at 60 and 52 cm<sup>-1</sup> are associated with rotatory modes which must certainly be combined with torsional modes of thiourea. A very weak line at 107 cm<sup>-1</sup> is in a region which appears to be characteristic of  $\delta$ (CdSC) modes and is presumed to be

<sup>6</sup> H. J. Buttery, G. Keeling, S. F. A. Kettle, I. Paul, and P. J. Stamper, *J. Chem. Soc.* (A), 1969, 2077; D. M. Adams and A. Squire, *ibid.*, 1970, 814. evidence of the symmetry approximations discussed above.

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Raman frequencies (cm<sup>-1</sup>), intensities (arbitrary units) and assignments for single-crystal CdCl<sub>2</sub>(tu)<sub>4</sub> at ambient temperature

Frequency and

assig	nment					Mode
(	C44	x(zz)y	x(yx)y	x(zx)y	x(yz)y	description
41 42	$E_{g}$	94		13	15	Molecular
49 52	$E_{g}$	01	5	12	14	rotatory modes
60 60	$B_{g}^{a}$		36			mixed
62 69	$E_{g} A_{''}$	4		4	4	thiourea
72	$E_{\mathbf{g}}^{\mathbf{s}}$	10		38	39	torsional modes
87	$B_{g}^{Ag}$	10	56		J	ν <sub>s</sub>
91 107	$A_{g} B_{\sigma}$	4	2			δ
109	$A_{g}$	100			-	$\nu_4$
$112 \\ 129$	$E_{g}$ $A_{a}$	4		8	9	$\delta^{\nu_{10}}$
137	$E_{\mathbf{g}}^{\mathbf{s}}$		0	9	10	ν <sub>9</sub>
147 153	$B_{g}$ $A_{\sigma}$	5	6			ν <sub>7</sub> ν
208	$A_{g}^{\bullet}$	10	,			$\nu_2$
209	Bg		1			$\nu_{10}$

δ = Modes not predicted on idealised C<sub>4ħ</sub> molecular symmetry, due to δ(CdSC) motions.

 $E_g$  Species.—The two highest  $E_g$  modes are of the  $\delta$ (CdSC) type and are, therefore, associated with the 137 and 112 cm<sup>-1</sup> lines. There is a clear gap between 112 cm<sup>-1</sup> and a series of low-lying  $E_g$  lines beginning at 72 cm<sup>-1</sup> and due to torsions and rotatory lattice modes.

 $A_g$  Species.—From their frequency the 208 and 153 cm<sup>-1</sup> lines can be assigned to  $v_2$ , v(Cd-S) and  $v_3$ , v(Cd-Cl).  $v_3$  Can hardly be lower than this as even this frequency represents the motion of a largely ionic-bonded chlorine. In addition this symmetry species shows weak bands (129 and 91 cm<sup>-1</sup>) of a frequency compatible with  $\delta(CdSC)$ 



motion but which are not expected on the idealised  $C_{4h}$  model.

A<sub>u</sub> Species.—The three internal modes can be assigned
D. M. Adams, M. A. Hooper, and M. H. Lloyd, J. Chem. Soc. (A), 1971, 946.

as shown in Table 10 with a fair degree of confidence. The intense antisymmetric Cl–Cd–Cl mode at 180 cm<sup>-1</sup> is above the corresponding symmetric stretch,  $\nu_3$ , and the other modes lie in regions consistent with the Raman assignment.

 $E_{u}$  Species.—In this species, apart from  $v_{17}$ , v(Cd-S) we

expect a group of  $\delta$ (CdSC) modes (including splitting due to departure from C<sub>4h</sub>),  $\nu_{19}$ ,  $\delta$ (SCdS) and several lattice modes. Assignments are shown in Table 10.

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