# Crystal Structure of Sodium Tetratungstate, $\mathbf{N a}_{2} \mathbf{W}_{4} \mathbf{O}_{13}$ 

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The structure of $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ has been determined by Patterson and Fourier methods from 426 independent reflections measured photographically. The basic units, consisting of four edge-shared octahedra, are linked along two crystallographic directions by corner sharing to form octahedral layers, which are held together by sodium atoms. The W-O linkage is similar to that in $\mathrm{K}_{2} \mathrm{Mo}_{4} \mathrm{O}_{13}$, but is new for tungstates. Crystals are triclinic, space group $P \bar{T}$. with $Z=1$, in a unit cell of dimensions: $a=11 \cdot 163(5), b=3 \cdot 894(1), c=8 \cdot 255(3) \AA, \alpha=90 \cdot 60(2)$. $\beta=131 \cdot 36(2), \gamma=79 \cdot 70(2)^{\circ}$. The structure was refined by least squares to $R 0 \cdot 117$.

Caillet ${ }^{1}$ and later Bouillaud ${ }^{2}$ reported the preparation of $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ and presented powder data. Crystals for the present investigation were obtained by a similar procedure. They were tabular and showed an excellent cleavage parallel to ( 100 ) plane.

## EXPERIMENTAL

Powder data (Table 1) were obtained by use of a Guiniertype focusing camera internally calibrated with silicon. The lines were indexed according to my own method ${ }^{3}$ and lattice parameters refined by use of a program based on least squares. The discrepancies between the present data and those observed by the earlier workers may be attributed to the inability to resolve many reflections, if cameras other than Guinier-type are used.

Crystal Data.- $\mathrm{Na}_{2} \mathrm{O}_{13} \mathrm{~W}_{4}$, Triclinic, $M=0.000(0), a=$ $11 \cdot 163(5), b=3 \cdot 894(1), c=8 \cdot 255(3) \AA, \alpha=90 \cdot 60(2), \beta=$ $131 \cdot 36(2), \quad \gamma=79 \cdot 70(2)^{\circ}, \quad U=262 \cdot 2(3) \quad \AA^{3}, \quad D_{\mathrm{c}}=6 \cdot 27$, $Z=1, D_{\mathrm{m}}=6 \cdot 49, F(000)=550$. Space group $P \overline{1}$. Mo$K_{\alpha}$ radiation, $\lambda=0.7107 \AA ; \mu\left(\right.$ Mo- $\left.K_{\alpha}\right)=439.9 \mathrm{~cm}^{-1}$.

The structure was derived from three-dimensional Weissenberg data by Mo- $K_{\alpha}$ radiation. The intensities of the 426 independent reflections used for determining the structure were measured photometrically. An approximate correction for absorption effects of the prismatic crystal was made with the values given in ref. 4. Intensities were then corrected for Lorentz and polarisation factors. A Patterson

* See Notice to Authors No. 7 in J.C.S. Dalton, 1973, Index issue (items less than 10 pp . are supplied as full-size copies).

1 P. Caillet, Compt. rend., 1963, 256, 1986.
${ }_{2}$ Y. Bouillaud, Bull. Soc. France Mineral. Crist., 1968, 91, 292.
projection then revealed the positions of the tungsten atoms. From the positions of the Patterson peaks and their relative heights, it could be concluded that an inversion centre exists between the four tungsten atoms. A three-dimensional difference Fourier synthesis using the tungsten parameters yielded the parameters of the thirteen oxygen atoms. A subsequent difference-Fourier calculated with tungsten and oxygen atoms indicated the positions of sodium atoms. All parameters together with individual temperature factors were then refined by full-matrix least-squares (a modified version of the Busing-Levy ORFLS program) to a final $R$ of $0 \cdot 117$. Final observed and calculated structure factors are listed in Supplementary Publication No. SUP 21082 (5 pp., 1 microfiche).* Atom co-ordinates are listed in Table 2.

## DISCUSSION

The structure consists of infinite chains of shared $\mathrm{WO}_{6}$ octahedra (Figure 1) running parallel to the $c$ axis, the repeating unit consisting of only four edge-shared octahedra. Thus the structure is similar to that of $\mathrm{K}_{2} \mathrm{Mo}_{4} \mathrm{O}_{13},{ }^{5}$ but not identical because the latter contains eight edge-shared octahedra within the basic unit. Thus this type of linkage can be considered new, especially for the tungstates. The octahedra, marked $A$ and $B$ in Figure 1 are related to those marked $\mathrm{A}^{\prime}$ and $\mathrm{B}^{\prime}$, through a centre of inversion. All four tungsten atoms lie almost in the plane (120). Each basic unit of four octahedra is
${ }^{3}$ K. Viswanathan, Amer. Mineral., 1968, 53, 2047.
${ }^{4}$ K. Sagel, 'Tabellen zur Röntgenstrukturanalyse,' ch. 8, Band VIII, Seite 85, Springer Verlag, Berlin, 1958.
${ }^{5}$ M. Gatehouse and P. Leverett, J. Chem. Soc. (A), 1971, 2107.
linked with identical units through common corners, parallel to $b$ and $c$ axes, thus giving rise to an octahedral layer parallel to the (100) plane. Such rigid octahedral layers are held together by the comparatively weak

| Table 1 |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Powder pattern of $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ obtained on a Guinier focusing-type camera with $\mathrm{Cu}-K_{\alpha_{1}}$ radiation |  |  |  |  |  |
| $h$ | $k$ | l | $d_{0}$ | $d_{\text {c }}$ | $I_{\text {o }}$ |
| 1 | 0 | -1 | $8 \cdot 203$ | $8 \cdot 205$ | 18 |
| 0 | 0 | 1 | 6.129 | $6 \cdot 130$ | 30 |
| $\because$ | 0 | -1 | $5 \cdot 499$ | $5 \cdot 491$ | 28 |
| $\because$ | 0 | -2 | $4 \cdot 108$ | $4 \cdot 103$ | 40 |
| 2 | 0 | 0 | 4.073 | $4 \cdot 080$ | 70 |
| 1 | $1)$ | -2 | 3.843 | 3.838 | 23 |
| 1 | 1 | 0 | 3.792 | $3 \cdot 784$ | 100 |
| $1)$ | 1 | 0 | $3 \cdot 792$ | $3 \cdot 790$ |  |
| 3 | 0 | $-2$ | $3 \cdot 513$ | 3.514 | 80 |
| 0 | $-1$ | $-1$ | $3 \cdot 459$ | $3 \cdot 457$ | 22 |
| 0 | 1) | 2 | $3 \cdot 064$ | 3.065 | 42 |
| 1 | 1 | 1 | $3 \cdot 005$ | 3.006 | 9 |
| 2 | - 1 | -1 | $2 \cdot 889$ | $2 \cdot 889$ | 20 |
| 3 | 1 | $-2$ | $2 \cdot 762$ | $2 \cdot 769$ | 35 |
| 3 | 0 | -3 | 2.736 | 2.735 | 32 |
| $\because$ | - 1 | $-2$ | 2.736 | $2 \cdot 736$ |  |
| : | 0 | 0 | 2.716 | 2.719 | 25 |
| 2 | ) | $-3$ | 2.684 | $2 \cdot 683$ | 6 |
| 1 | 1 | -2 | $2 \cdot 601$ | $2 \cdot 601$ | 22 |
| 1 | --1 | $-2$ | $2 \cdot 573$ | 2.574 | 5 |
| 4 | ) | -3 | $2 \cdot 512$ | 2.512 | 3 |
| 2 | -- 1 | 0 | $2 \cdot 503$ | $2 \cdot 505$ | 11 |
| 3 | 1 | 0 | $2 \cdot 495$ | 2.497 | 18 |
| 2 | 1 | 1 | $2 \cdot 439$ | $2 \cdot 442$ | 13 |
| 1 | 0 | $-3$ | $2 \cdot 396$ | $2 \cdot 396$ | 9 |
| 3 | $\cdots 1$ | $-1$ | $2 \cdot 336$ | $2 \cdot 337$ | 7 |
| 3 | 1 | - 3 | $2 \cdot 258$ | $2 \cdot 256$ | 3 |
| $\stackrel{1}{2}$ | $\cdots 1$ | $-3$ | $2 \cdot 232$ | $2 \cdot 233$ | 14 |
| 4 | 1 | $-3$ | $2 \cdot 201$ | $2 \cdot 200$ | 3 |
| 3 | - 1 | -3 | $2 \cdot 182$ | $2 \cdot 182$ | 10 |
| 5 | 0 | -3 | $2 \cdot 164$ | $2 \cdot 164$ | 24 |
| 5 | 0 | -2 | $2 \cdot 164$ | $2 \cdot 160$ |  |
| 0 | 0 | 3 | 2.042 | 2.043 | 22 |
| 3 | -1 | 0 | 2.001 | 2.003 | 3 |
| 4 | $\cdots 1$ | $-3$ | 2.001 | 2.001 |  |
| 3 | 1 | 1 | 1.979 | 1.981 | 5 |
| 1 | $\underline{2}$ | 0 | 1.948 | 1.947 | 22 |
| 1 | 1 | $-3$ | 1.937 | 1.937 | 4 |
| 0 | $\cdots$ | -1 | 1.891 | 1.890 | 4 |
| 1 | $\because$ | -1 | 1.861 | 1.861 | 5 |
| 2 | 1 | $\stackrel{\square}{2}$ | 1.861 | 1.861 |  |
| 1 | 2 | 1 | 1.850 | 1.850 | 16 |
| 6 | 0 | $-3$ | 1.831 | 1.830 | 23 |
| 3 | 1 | $-4$ | 1.782 | 1.782 | 6 |
| 4 | - 1 | $-4$ | 1.775 | 1.778 | 12 |
| 6 | 1 | -3 | 1.775 | 1.777 |  |
| 3 | 2 | -2 | 1.766 | 1.766 | 6 |
| $\stackrel{ }{2}$ | -- 1 | -4 | 1.766 | 1.766 |  |
| \% | $\underline{\square}$ | 0 | 1.753 | 1.755 | 9 |
| 1 | $\cdots$ | -2 | 1.753 | 1.752 |  |
| 6 | 1 | -2 | 1.753 | 1.751 |  |
| 2 | 2 | -2 | 1.743 | $1 \cdot 744$ | 4 |
| 1 | 6 | -4 | 1.731 | 1.731 | 2 |
| $\underline{9}$ | 2 | 1 | 1.731 | 1.729 |  |
| 4 | 2 | - 2 | 1.706 | 1.708 | 16 |
| 2 | $-2$ | $-1$ | 1.699 | $1 \cdot 701$ | 17 |
| 0 | --1 | 3 | 1.699 | 1.698 |  |
| 1 | 1 | 3 | 1.684 | $1 \cdot 685$ | 9 |
| 6 | 1 | -4 | $1 \cdot 677$ | 1.679 | 4 |
| 1 | 2 | $-2$ | 1.649 | 1.650 | 15 |
| 1 | - 1 | -4 | $1 \cdot 649$ | 1.649 |  |

$\mathrm{Na}-\mathrm{O}$ bonds, thus accounting for the excellent cleavage parallel to (100). The alkali-metal atom has only five oxygens as near neighbours ( $\mathrm{Na}-\mathrm{O}$ distances varying between $2 \cdot 16$ and $2 \cdot 65 \AA$, Table 3 ), the co-ordinating polyhedron approaching the shape of a tetragonal pyramid.

Table 2
Co-ordinates of the atoms, with standard deviations in parentheses

| Atom | $x$ | $y$ | $z$ | $\beta / \AA^{2}$ |
| :---: | :---: | :---: | :---: | :---: |
| W(1) | $0 \cdot 1938(5)$ | 0.8382(2) | $0 \cdot 2132(5)$ | 0-12(03) |
| W(2) | $0 \cdot 1744$ (5) | $0 \cdot 9774(3)$ | $0 \cdot 8054(3)$ | $0 \cdot 11(04)$ |
| $\mathrm{O}(1)$ | $0 \cdot 0$ | $0 \cdot 0$ | 0.5 | $2 \cdot 14(80)$ |
| $\mathrm{O}(2)$ | $0 \cdot 663(6)$ | 0.160(7) | $0 \cdot 195(5)$ | $1 \cdot 77(52)$ |
| $\mathrm{O}(3)$ | $0 \cdot 684(7)$ | $0 \cdot 166$ (5) | $0 \cdot 892(7)$ | 1.43(71) |
| $\mathrm{O}(4)$ | $0 \cdot 995$ (5) | $0 \cdot 005(4)$ | $0 \cdot 157(8)$ | $1 \cdot 10(74)$ |
| $\mathrm{O}(5)$ | $0 \cdot 346$ (3) | $0 \cdot 855(5)$ | $0 \cdot 485(3)$ | $0 \cdot 13(52)$ |
| $\mathrm{O}(6)$ | $0 \cdot 198(4)$ | $0 \cdot 389$ (7) | $0 \cdot 223$ (3) | 0.43 (65) |
| O (7) | $0 \cdot 148(7)$ | $0 \cdot 422(5)$ | $0 \cdot 827(6)$ | 0.93(46) |
| Na | $0 \cdot 430$ (8) | $0 \cdot 285$ (4) | 0.736(6) | 0.82(55) |



Figure 1 Projection of the structure on the (120) plane; projection axis approximately $b$. The plane, defined by $b$ and $2 \overline{1} 0$, is exactly centred. Hence the relationship between the octahedral chains, shaded and unshaded, is given by the vector $\left[\frac{b}{2}+\frac{[2 \overline{1} 0]}{2}\right]$. Sodium atoms (solid circles) are at height $0 \cdot 5$, others are at 0 . The basis oxygen atoms have been numbered (see Table 3)

Table 3
Distances $(\AA)$ in $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$
(a) W(1) octahedron

(b) W(2) octahedron

| $\mathrm{W}(2)-\mathrm{O}(1)$ | 1.90 | $\mathrm{~W}(2)-\mathrm{O}\left(9^{\prime}\right)$ | 2.10 |
| :--- | :---: | ---: | :--- |
| $\mathrm{~W}(2)-\mathrm{O}\left(2^{\prime}\right)$ | 1.80 | $\mathrm{~W}(2)-\mathrm{O}(7)$ | 2.26 |
| $\mathrm{~W}(2)-\mathrm{O}\left(3^{\prime}\right)$ | 1.88 | $\mathrm{~W}(2)-\mathrm{O}\left(7^{\mathrm{I}}\right)$ | 1.73 |
|  | $\mathrm{O}-\mathrm{O}$ | $2.57-2.98$, mean 2.72 |  |

(c) Na polyhedron

| $\mathrm{Na}-\mathrm{O}\left(2^{\prime}\right)$ | 2.44 | $\mathrm{Na}-\mathrm{O}(5)$ | $\mathbf{2 . 4 5}$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Na}-\mathrm{O}\left(2^{\prime} \mathrm{I}\right)$ | 2.41 | $\mathrm{Na}-\mathrm{O}\left(5^{\mathrm{I}}\right)$ | $\mathbf{2 . 6 5}$ |
| $\mathrm{Na}-\mathrm{O}(3)$ | 2.16 |  |  |

(d) W distances
$\left.\begin{array}{ll}W(1)-W(2) & \mathbf{3 \cdot 2 7 2} \\ W(1)-W(1) & \mathbf{3 \cdot 2 7 4}\end{array}\right\}$ edge linked

Primed and unprimed atoms are related by a centre of inversion; Roman numeral superscript I denotes atoms related by translation along $b$.

The tungsten octahedra are considerably distorted and the observed range in the $\mathrm{W}-\mathrm{O}$ distances of $1.71-2 \cdot 28 \AA$ is typical of tungstates.

This structure bears no resemblance to the tunnel
structure of the analogous tungstate, $\mathrm{K}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$, which contains $\mathrm{WO}_{6}$ octahedra linked by corners to form sixmembered rings. ${ }^{6}$ Considering that both are tetratungstates, this change of structure is radical, and must be attributed to the environment of the sodium. The bigger potassium ion requires a larger number of neighbours and the rigid single chain of $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ cannot be distorted to any extent. However, it should be theoretically possible for the band structure of $\mathrm{K}_{2} \mathrm{Mo}_{4} \mathrm{O}_{13}$ to accommodate the bigger potassium ions and to produce an analogous $\mathrm{K}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$. Incidentally, the sodium atom in $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ has almost the same co-ordination as one of the sodium atoms in $\mathrm{Na}_{2} \mathrm{~W}_{2} \mathrm{O}_{7}$ or $\mathrm{Na}_{2} \mathrm{Mo}_{2} \mathrm{O}_{7}{ }^{7,8}$

A comparison of the structures of the bimolybdates of sodium and potassium ${ }^{9}$ reveals that replacement of the alkali-metal atoms does cause a change in the nature of the bond between the molybdenum polyhedra, although the chain structure remains.

A comparison of the structure of $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ with that of $\mathrm{Li}_{2} \mathrm{~W}_{2} \mathrm{O}_{7}$ (ref. 10) is interesting, since both contain four edge-shared octahedra in the fundamental unit (Figure 2). In the former the three shared edges within the unit lie almost on a single plane (120) and the basic units are linked in two directions by sharing corners to form octahedral layers, giving rise to a tetratungstate polyanion, $\left(\mathrm{W}_{4} \mathrm{O}_{13}\right)^{2-}$. On the other hand, the three shared edges do not lie on a plane in $\mathrm{Li}_{2} \mathrm{~W}_{2} \mathrm{O}_{7}$ and the basic units are

[^0]linked by sharing edges to form octahedral double chains, giving rise to a polyanion, $\left(\mathrm{W}_{4} \mathrm{O}_{14}\right)^{4-}$.

A study of the structures of the different tungstates suggests that the $\mathrm{W}-\mathrm{O}$ co-ordination is determined by the tungsten : oxygen ratio; thus in bronzes, $\mathrm{M}_{x} \mathrm{WO}_{3}$, with a $\mathrm{W}: \mathrm{O}$ ratio of $1: 3$, in the parent oxide, $\mathrm{WO}_{3}$, and in the tetratungstate $(4: 13)$, the co-ordination is exclusively

(a)

(b)

Figure 2 (a) Band of $W$ octahedra with the composition $\left(W_{4} \mathrm{O}_{14}\right)$ in the structure of $\mathrm{Li}_{2} \mathrm{~W}_{2} \mathrm{O}_{7}$. (b) Idealised octahedral chain of composition $\left(\mathrm{W}_{4} \mathrm{O}_{13}\right)$ in the $\mathrm{Na}_{2} \mathrm{~W}_{4} \mathrm{O}_{13}$ structure
octahedral, whereas in the ditungstate, $\mathrm{Na}_{2} \mathrm{~W}_{2} \mathrm{O}_{7}$, both tetrahedral and octahedral co-ordinations are observed. At the extreme ratio of $1: 4$, as in $\mathrm{Na}_{2} \mathrm{WO}_{4}$, only tetrahedral co-ordination has been observed. The molybdates of potassium behave similarly. ${ }^{5}$
[4/400 Received, 28th February, 1974]
${ }^{9}$ S. A. Magarill and R. F. Klevtsova, Soviet Phys. Cryst., 1972, 16, 645.
${ }^{10}$ S. A. Magarill, R. F. Klevtsova, and V. V. Bakakin, Soviet Phys. Cryst., 1973, 18, 166.


[^0]:    ${ }^{6}$ M. Seleborg, Chem. Comm., 1967, 1126.
    ${ }^{7}$ M. Seleborg, Acta Chem. Scand., 1967, 21, 499.
    ${ }^{8}$ I. Lindquist, Acta Chem. Scand., 1950, 4, 1066.

