Additions and Corrections

Synthesis and Reactions of Osmium(III) Chloro Carboxylates. *X*-Ray Crystal Structure of Tetra(μ-n-butyrato)-dichlorodiosmium(III) (*Os–Os*) (1983, 2109)

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The electrochemical results described on pp. 2111 (two lines from the bottom of the left-hand column) and 2112 (first three lines) have been reinterpreted as follows.

Recent re-examination of the electrochemical data reveals that the facile one-electron reversible step at +0.57 and +0.53 V, exhibited by the propionate and butyrate respectively, is a *reduction* not an oxidation process. Hence, this electrochemical evidence, correctly interpreted, establishes that the formally mixed valence $[Os_2^{II,III}(O_2CR)_4Cl_2]^-$ anions have been produced and *not* the $[Os_2^{III,IV}(O_2CR)_4Cl_2]^+$ cations as originally suggested. Preliminary controlled-potential electrogeneration studies at potentials *more negative* than the peak potential reveal that these anions are stable at ambient temperature.

The assignments of all the other electrochemical data given in this paper are correct.