Polyhedral Thiaborane Chemistry: a Comparative Nuclear Magnetic Resonance Study of *arachno*-[6-SB₉H₁₂]⁻ and *nido*-6-SB₉H₁₁

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A nuclear magnetic resonance study of $nido-6-SB_9H_{11}$ and $arachno-[6-SB_9H_{12}]^-$ using ¹¹B, ¹¹B-{¹¹B}, [¹¹B-¹¹B]-COSY, and [¹H-¹H]-COSY techniques has enabled unequivocal assignment of the ¹¹B and ¹H spectra, and has thereby shown that the basic known behavioural patterns of their binary borane analogues ($arachno-[B_{10}H_{14}]^{2-}$ and $nido-B_{10}H_{14}$) also hold for the 6-thiadecaborane system. Interestingly, the effects on cluster nuclear shielding upon replacement of the decaborane B(6) centres with sulphur are very similar for both systems in spite of the fundamentally different *arachno versus nido* electronic structures. A long-term hydrolysis product of [6-SB_9H_{12}]^- is confirmed as [*endo*-9-(OH)-*arachno*-6-SB_9H_{11}]^-.

The factors that influence nuclear magnetic shielding in polyhedral boron compounds are not understood, and few reliable structure-shielding correlations that are useful in the general case have emerged.¹⁻⁴ An exception to this generalization is in the $C_{2\nu}$ open ten-vertex cluster type (Figure 1), in which a change from *nido* character, exemplified by $B_{10}H_{14}$, to *arachno* character, exemplified by $[B_{10}H_{14}]^{2-}$, is characterized by an inversion of the ordering of the ¹¹B resonance positions and an increase in the mean ${}^{11}B$ shielding of several p.p.m. ${}^{5-8}$ We have found these criteria are useful in assessing *nido versus arachno* character in a variety of ten-vertex non-heteroborane $^{8-10}$ and metallaborane^{11–15} systems, and also in assessing *nido versus* arachno character in ten-boron fragments in polyhedral systems with more than ten vertices,^{16,17} thereby providing insights into bonding patterns¹⁸ within these larger polyhedral types. With the current resurgence of general interest in polyhedral heteroborane types other than those containing metal or carbon atoms, it is of value to examine the n.m.r. properties of other tenvertex nido-arachno pairs to see whether the overall shielding differences apply also to them, and whether any other interesting features emerge. This has been carried here out for the open SB₉ fragment in the nido-6-SB₉H₁₁-arachno-[6- SB_9H_{12} pair [schematic structures (I) and (II) respectively; note that, in (I) and (II), BH(exo) H atoms, except for H(9) in (II), are not represented]. Only the ¹¹B n.m.r. spectra for these compounds have been previously reported, and only for the arachno species were these reasonably definitively assigned.19.20

Experimental

Samples of $6-SB_9H_{11}$ [compound (1)] and $Cs[6-SB_9H_{12}]$ [compound (2)] were prepared according to literature procedures ²¹ and examined under the conditions specified in the Tables of n.m.r. data. N.m.r. spectroscopy was performed at 9.4 T (400 MHz, ¹H; 128 MHz, ¹¹B) using commercially available instrumentation. The techniques of ¹H-{¹¹B},^{10,22,23} [¹¹B-¹¹B]-COSY,^{16,24-28} and [¹H-¹H]-COSY ^{27,29} spectroscopy used in this work were essentially as described elsewhere; use being made in the ¹H-{¹¹B} experiments of the procedure ^{10,11,30} in which a ¹H-{¹¹B} (off-resonance)} spectrum is subtracted from a ¹H-{¹¹B}(n-resonance)} spectrum in order to remove lines arising from ¹H resonances not coupled to the ¹¹B nucleus of interest. In the [¹¹B-¹¹B]-COSY and [¹H-¹H]-COSY ²⁹ experiments, {¹H(broad-band noise)} and {¹¹B(broad-band noise)</sup> decoupling, respectively, were applied, with the {¹¹B} decoupling gated off during the relaxation periods between the pulse-acquire sequences to minimise r.f.

Table 1. Details for the two-dimensional n.m.r. experiments

	$SB_{9}H_{11}(1)$		$Cs[SB_9H_{12}]$ (2)		
Compound	FIID 11D1				
COST experiment	[BB]	[.uu]	[BB]	[.uu]	
Data size					
$(t_2, t_1 / \text{words})$	512,128	1 024,128	256,64	512,128	
Transform size					
$(F_2, F_1 / \text{words})$	512,256	1 024,512	256,128	512,256	
t ₂ sweepwidth					
$(=2 \times t_1)$					
sweepwidth)/Hz	9 615.0	4 854.4	6 493.5	2 604.2	
Digital resolution					
for F_2 and F_1/Hz	37.6	9.5	50.7	10.2	
No. of transients					
per t_1 increment	64	256	64	144	
Recycling time/s	0.12	1.1	0.12	1.1	
Mixing pulse/°	45	45	45	45	
Window	sine-bell	sine-bell	sine-bell	sine-bell	
(both t_2 and t_1)	squared	squared	squared	squared	
Decoupling	{ ¹ H}-	$\{^{11}\mathbf{B}\}$ gated	{ ¹ H}-	{ ¹¹ B}gated	
Zeeeapung	continuous	(2)Batoa	continuous	, 2, gatea	



Figure 1. Schematic representation of the ten-vertex *nido-arachno* decaboranyl cluster geometry showing the numbering scheme. *nido*-6-SB₉H₁₁ (1) has bridging H atoms at B(8)B(9) and B(9)B(10) [see structure (1)] and *arachno*-[6-SB₉H₁₂]⁻ (2) has bridging atoms at B(5)B(10) and B(7)B(8), together with an *endo*-terminal H atom at B(9) [see structure (11)]



Figure 2. 128-MHz ¹¹B n.m.r. spectra for *nido*-6-SB₉H₁₁ (1) in CD₂Cl₂ solution. (a) Normal ¹¹B spectrum, (b) normal spectrum with {¹H(broad-band noise)} decoupling. In (a) a short pre-pulse delay time together with a long $T_1[^{11}B(4)]$ (see Table 2) has resulted in an artificially reduced intensity for the ¹¹B(4) resonance. Additional fine structure arising from ¹J(¹¹B-¹H_u), was apparent for B(9) and B(8,10) in the ¹¹B spectrum [(a)] by use of resolution-enhancement techniques (Table 2, footnote g). (c) [¹¹B-¹¹B]-COSY 45 spectrum recorded with {¹H(broad-band noise)} decoupling

(radio frequency) warming of the sample. The experimental parameters for these COSY experiments are summarized in Table 1. Other spectroscopy was straightforward, longitudinal relaxation times $T_1(^{11}B)$ being measured by the π -delay- $\frac{\pi}{2}$ -acquire inversion-recovery method, and chemical shifts, δ , being quoted positive to low field (high frequency) of Ξ 100 for ¹H and Ξ 32.083 971 MHz [nominally BF₃(OEt₂) in CDCl₃]⁴ for ¹¹B (Ξ being defined as in ref. 31). Chemical shifts were measured using solvent deuteron or residual proton resonances as internal secondary standards.

Results

N.M.R. Behaviour of nido-6-SB₉H₁₁ (1).—The ¹¹B n.m.r. spectrum of *nido*-6-SB₉H₁₁ (1) showed a 2:1:2:2:1:1 relative intensity pattern consistent with ^{19.20} the molecular structure represented in (I) (Table 2 and Figure 2). The observation of coupling from the bridging protons to ¹¹B(9) and ¹¹B(8,10)



Figure 3. Sections of a 400-MHz $[^{1}H_{-}^{1}H]$ -COSY 45 spectrum for *nido*-6-SB₉H₁₁ (1) in CD₂Cl₂ solution, recorded with {¹¹B(broad-band noise)} decoupling, together with (*a*) the corresponding sections of the ¹H-{¹¹B(broad-band noise)} spectrum

(see footnote g in Table 2) assigned them to these positions and thence the observed $[^{11}B^{-11}B]$ -COSY²⁴ cross-peaks unambiguously assigned the whole ^{11}B spectrum (Figure 2). Definitive [¹¹B-¹¹B]-COSY correlations were observed for all nearest-neighbour connectivities, although that between the hydrogen-bridged ¹¹B(9) and ¹¹B(8,10) positions was somewhat weaker. This, however, is expected since internuclear spin-spin coupling constants between hydrogen-bridged boron atoms are generally small in magnitude.4,24,32,33 The correlation between the ${}^{11}B(2)$ and ${}^{11}B(5,7)$ positions that flank the electronegative sulphur atom S(6) was also weaker. Again, however, a weaker coupling has been generally noted before for interboron connectivities that flank more electronegative cluster atoms,⁴ for example carbon in carboranes,^{34,35} and sulphur, selenium, and tellurium in chalcogenaborane species,^{36,37} and may well result from a diversion of bonding electron density towards bonds with the electronegative element. Conversely, the observed strong correlation ${}^{11}B(9)-{}^{11}B(4)$ was manifested by a partially resolvable coupling constant ${}^{1}J[{}^{11}B(9)-{}^{11}B(4)]$ with a magnitude of ca. 30 Hz. As far as we are aware, this is the largest intracluster interboron coupling constant yet observed; much larger, for example, than has been observed for ${}^{1}J[{}^{11}B(9)-$ ¹¹B(4)] in ten-boron *nido* ten-vertex species, in which it takes values up to 19 Hz in the examples reported so far.^{4,38-40} This

				Annrovimate					
Assignment ^a	$\delta({}^{11}B)/p.p.m.{}^{b.c}CD_2Cl_2$ solution	$\delta(^{11}B)/p.p.m.^b$ C_6D_6 solution	Observed $[^{11}B^{-11}B]^{-1}$ COSY ^{<i>d.e</i>} correla- tions, CD ₂ Cl ₂ solution	$T_1({}^{11}B)/ms$ saturated CD_2Cl_2 solution ^f	^{−1} J(¹¹ B− ¹ H)/ Hz ^g	$\begin{array}{c} \delta(^1H)/\\ p.p.m.^{h}\\ CD_2Cl_2\\ solution \end{array}$	$\begin{array}{c} \delta({}^1H)/\\ p.p.m.^h\\ C_6D_6\\ solution \end{array}$	a.s.i.s.(¹ H)- (C ₆ D ₆)/ p.p.m. ⁱ	Observed [¹ H- ¹ H]- COSY correlations, CD ₂ Cl ₂ solution ^{<i>e</i>,<i>j</i>}
(5.7)	+25.3 (2 B)	+25.0	(1,3)s (8,10)s (2)w	17	170	+5.04	+ 4.97	+0.07	(1,3)m (8,10)s (2)w
(9)	$+17.7(1 \text{ B})^{k}$	+17.7	(8,10)mw (4)s	13	167, 38	+3.88	+3.57	+0.31	$(8,10)w (4)m (\mu)vs_2$
(1,3)	+6.6(2 B)	+ 6.8	(5,7)s (8,10)vs (4)m (2)s	42	148	+3.55	+ 3.85	-0.30	(5,7)m (8,10)s (4)s (2)m (µ)vs
(8,10)	-9.9 (2 B)	- 10.05	(5,7)s (9)mw (1,3)vs (4)s	28	158, 42	+ 2.64	+ 2.65	-0.01	$(5,7)s (9)w (1,3)s (4)s (2)vw?4 (\mu)vs2$
(4)	$-21.5 (1 B)^{t}$	-21.4	(9)s (1,3)m (8,10)s	55	157	+1.76	+ 2.08	-0.32	(9)m (1,3)s (8,10)s (μ)vs
(2) µ(8,9;9,10)	- 30.7 (1 B)	- 30.6	(5,7)w (1,3)s	17	182 38, 42	+0.64 -2.86	+0.78 -3.40	-0.14 + 0.54	(5,7)w $(1,3)$ m (8.10) vw (9)vs ₂ $(1,3)$ vs (8,10)vs (4) vs

Table 2. Measured ¹¹B and ¹H n.m.r. parameters for *nido*-6-SB₉H₁₁ (1), at 297 K unless otherwise indicated

^{*a*} By relative intensities, incidence of couplings to μ -H, and two-dimensional $[^{11}B^{-11}B]$ -COSY experiments. ^{*b*} ± 0.2 p.p.m. to low field (high frequency) of BF₃(OEt₂). ^{*c*} Relative intensities in parentheses. ^{*d*} {¹H(broad-band noise)} decoupling applied continuously (Table 1). ^{*c*} s = Strong, w = weak, m = intermediate, v = very. ^{*f*} 298 K. ^{*g*} ± 2 Hz agreement from both ¹H and ¹¹B spectra with resolution enhancement by apodization. ^{*h*} ± 0.05 Hz to low field of SiMe₄; ¹H resonances related to directly bound B atoms by ¹H-{¹¹B(selective)} experiments. ^{*i*} a.s.i.s. = aromatic solvent induced shielding (ref. 42) = [δ (¹H)(CD₂Cl₂) - δ (¹H)(C₆D₆)]. ^{*j*} Recorded with gated {¹¹B(broad-band noise)} decoupling (Table 1); subscripts refer to *n* in ^{*n*}J(¹H-¹H) when *n* \neq 3. ^{*k*} Multiplet structure, splitting *ca*. 31 Hz, presumably arising from ¹J[¹¹B(4)-¹¹B(9)] in view of strong [¹¹B-¹¹B]-COSY correlation. ^{*i*} Unresolved multiplet structure apparent; see footnote *k*.

Table 3. Measured ¹¹B and ¹H n.m.r. parameters for Cs[arachno-6-SB₉H₁₂] (2) in (CD₃)₂CO solution at 294 K

Assignment "	$\delta(^{11}B)/p.p.m.^{b}$	Observed [¹¹ B- ¹¹ B]-COSY correlations ^{c,d}	$^{1}J(^{11}B-^{1}H)/Hz^{e}$	$\delta(^{1}H)/p.p.m.^{f}$	Observed[¹ H- ¹ H]-COSY correlations ^{c,g}
(4)	+ 3.6 (1 B)	(9)m (8,10)s (1,3)m	138	+ 2.79	(9 <i>exo</i>)s (9 <i>endo</i>)ms, (8,10)m (1,3)m (μ)w
(5,7)	$-8.4 (2 B)^{h}$	(2)w (8,10)vw? (1,3)s	149 (and ca. 41)	+2.43	$[(8,10)m (1,3)s (\mu)s]^{i}$
(2)	-11.9(1 B)	(5,7)w (1,3)s	166	+2.42	$[(8,10)m (1,3)s (\mu)s]^{i}$
(9)	-14.9 (1 B)	(4)m (8,10)s	117 114	$+ 1.42(exo)^{j}$ + 0.56(endo) ^k	(4)s (9 <i>endo</i>)vs (8,10)m (μ)vw? (4)ms (9 <i>exo</i>)vs (μ)w?
(8,10)	$-33.8 (2 B)^{t}$	(4)s (5,7)w? (9)s (1,3)s	140 (and ca. 47)	+0.37	(4)m [(5,7)(2)]m ⁱ (9exo)m (1,3)s (μ)vs ₂
(1.3)	-37.1 (2 B)	(4)m (5,7)s (2)s (8,10)s	140	+0.76	(4)m $[(5,7)(2)]s^i$ (8,10)s (µ)m
μ(7,8;5,10)			ca. 36, ca. 47	-1.83	$\begin{array}{c} (4) & [(5,7)(2)] s^{i} (9 exo) v w? \\ (9 endo) & w? (8,10) v w_{2} (1,3) m \end{array}$

^{*a*} By relative intensities, incidence of couplings to μ -H, and two-dimensional [¹¹B⁻¹¹B]-COSY experiments. ^{*b*} ±0.5 p.m. to low field (high frequency) of BF₃(OEt₂); relative intensities in parentheses. ^{*c*} s = Strong, w = weak, m = intermediate, v = very, ? = uncertain. ^{*d*} {¹H(broad-band noise)} decoupling applied continuously (Table 1). ^{*e*} ±5 Hz; measured from resolution-enhanced ¹¹B and ¹H spectra. ^{*f*} ±0.05 Hz to low field of SiMe₄; unresolved multiplet structure on most peaks arising from various ^{*n*}J(¹H⁻¹H) of $\leq ca$. 5 Hz or so. See also footnotes *j* and *k*. ^{*g*} Recorded with gated {¹¹B(broad-band noise)} decoupling (Table 1); subscripts refer to *n* in ^{*n*}J(¹H⁻¹H) when $n \neq 3$. ^{*k*} Fine structure, splitting *ca*. 31 Hz, probably arising from 'J(¹H₂)] accidentally near-coincident, therefore any correlation between these resonances is not observed, and correlations from each resonance not specific. ^{*j*} Triplet structure in ¹H-{¹¹B} spectrum, splitting 7.8 Hz, probably arising from couplings ²J[¹H(9*exo*)-¹H(9*endo*)-¹H(4)]. ^{*k*} Triplet structure in ¹H-{¹¹B} spectrum, splitting 6.8 Hz, probably arising from couplings ²J[¹H(9*endo*)-¹H(9*endo*)-¹H(4)]. ^{*k*} Unresolved fine structure, splitting $\leq ca$. 25 Hz, probably arising from ¹J[¹¹B(8,10)-¹¹B(1,3)].

may be indicative of a stronger B(4)-B(9) bond in SB_9H_{11} than in $B_{10}H_{14}$, or at least indicative of a linkage containing more boron 2s character. It will be instructive to have quantummechanical calculations in this area. {See also large ${}^{1}J[{}^{11}B(1,3)-{}^{11}B(5,7)]$ in *arachno*-[6-SB₉H₁₂]⁻ mentioned below.}

It was evident from smaller observed ¹¹B(4) intensities in the initial ¹¹B accumulations that the relaxation time for this resonance was quite long [see (a) in Figure 2], and the subsequent measurement of $T_1(^{11}B)$ values for compound (1) (Table 2) showed that although they generally parallel those reported previously⁴¹ for *nido*-B₁₀H₁₄ they were, with the exception of $T_1[^{11}B(2)]$, in general much longer. However, the solution conditions differed (CD₂Cl₂ for 6-SB₉H₁₁ in the present work, CD₃C₆D₅ for B₁₀H₁₄ in ref. 41), and so the two sets of results are not strictly comparable. It is nevertheless apparent that $T_1({}^{11}B)$ for the 2-position is much shorter than would otherwise be expected, which can be accounted for by additional electric field gradient asymmetry at this nucleus presumably induced by the electronegative sulphur atom at the adjacent 6-position.

The proton resonances were traced to their directly bound boron positions by ¹H-{¹¹B} selective double-irradiation experiments. The ¹¹B and *exo-*¹H nuclear shieldings showed the expected ⁴ parallel (see Figure 5, later), with the slope $\delta(^{11}B):\delta(^{1}H)$ being *ca.* 14:1, and with the bridging protons at B(8)B(9) and B(9)B(10) resonating at $\Delta\sigma$ *ca.* + 6 p.p.m. above this general trend. In [¹H-¹H]-COSY experiments (Table 2 and Figure 3) the interproton correlations arising from ³J(¹H-¹H) and ²J(¹H-¹H_µ) confirmed the positional assignments estab-



Figure 4. 128 MHz ¹¹B n.m.r. spectra for Cs[*arachno*-6-SB₉H₁₂] (2) in $(CD_3)_2CO$ solution. (a) The straightforward ¹¹B spectrum, (b) the straightforward spectrum with {¹H(broad-band noise)} decoupling, and (c) a [¹¹B-¹¹B]-COSY 45 spectrum recorded with {¹H(broad-band noise)} decoupling

lished from the $[{}^{11}B{}^{-11}B]$ -COSY results. Interestingly, the relative intensities of the observed correlations often mirrored those of the corresponding ${}^{1}J({}^{11}B{}^{-11}B)$ correlations, with, for example, the ${}^{3}J[{}^{1}H(2){}^{-1}H(5,7)]$ and ${}^{3}J[{}^{1}H(9){}^{-1}H(8,10)]$ crosspeaks being significantly weaker than the others. A second point of interest was the observation of a possible (weak) long-range interproton correlation corresponding to ${}^{4}J[{}^{1}H(2){}^{-1}H(8,10)]$.

The ¹¹B and ¹H n.m.r. spectra for *nido*-6-SB₉H₁₁ (1) were also measured in C₆D₆ solution (Table 2). The aromatic-solvent induced shielding (a.s.i.s.)⁴² effects on the ¹¹B shifts were relatively small (*cf.* refs. 40 and 43), but those in the ¹H spectrum were relatively more significant. In general the latter ¹H a.s.i.s. effects were similar in sign and relative magnitude to those previously reported ^{23,41,43–45} for B₁₀H₁₄, but their absolute magnitudes were not so large as for B₁₀H₁₄ under similar conditions, presumably because the more asymmetric SB₉H₁₁ molecule does not fit so comfortably in the liquid lattice.

N.M.R. Behaviour of $Cs[arachno-6-SB_9H_{12}]$ (2).—Table 3 summarizes the measured ¹¹B and ¹H n.m.r. data for the *arachno* species $[6-SB_9H_{12}]^-$ in a $(CD_3)_2CO$ solution of $Cs[SB_9H_{12}]$ (2). Again the ¹¹B n.m.r. relative intensity pattern, 1:2:1:1:2:2, was consistent with the molecular symmetry



Figure 5. (a) A plot of $\delta(^{11}\text{B})$ versus $\delta(^{1}\text{H})$ for directly bound boron and hydrogen atoms in the *nido* compound (1) (\bigcirc) and the *arachno* compound (2) (\square); the line drawn has intercept $\delta(^{1}\text{H}) + 2.8$ p.p.m. and slope $\delta(^{11}\text{B})$: $\delta(^{1}\text{H}) = 16:1$, the slope of the correlation for (2) being somewhat greater than this at *ca*. 18:1 and that for (1) somewhat smaller at *ca*. 14:1. Points for the bridging protons are plotted versus the mean $\delta(^{11}\text{B})$ values for their directly bound boron atoms. (*b*) Stick representations of the chemical shifts and relative intensities of (1) and (2), together with $B_{10}H_{14}$ and $[B_{10}H_{14}]^{2-}$ for comparison; (——) adjacent (α) to the sulphur atom, (–––) antipodal and γ to the sulphur atom, and (····) *meta* (β) to the sulphur atom (*cf.* refs. 28 and S. Hěrmánek, J. Fusek, B. Štíbr, J. Plešek, and T. Jelínek, *Polyhedron*, 1986, 5, 1873)

[structure (II)]. The ¹¹B(9) and ¹¹B(5,7)–¹¹B(8,10) resonances were identified as such by the bridge and *endo* ¹ $J(^{11}B-^{11}H)$ couplings, and the entire ¹¹B spectrum was thence assigned by [¹¹B–¹¹B]-COSY cross-correlations (Figure 4). As with the *nido* species (1) (Table 2 and Figure 2), [¹¹B–¹¹B]-COSY correlations were observed for all nearest-neighbour boron positions, although the relative intensities of cross-peaks for particular cluster-site pairs were generally different for (1) *versus* (2), as expected from the different *nido versus arachno* cluster electronic structures. Again, however, the ¹¹B(2)–¹¹B(5,7) correlation for the linkage that flanks electronegative sulphur was weak, and the B–H–B bridged linkage, now at ¹¹B(5,7)–¹¹B(8,10) in the *arachno* species (2), also exhibited only a weak correlation. Fine structure presumably arising from interboron coupling was observable in the ¹¹B-{¹H(broad-band noise)} spectrum for ¹¹B(5,7) and ¹¹B(8,10), the relative strengths of the [¹¹B-¹¹B]-COSY cross-peaks from each of these positions reasonably ascribing the splittings to the couplings ¹J[¹¹B(5,7)-¹¹B(1,3)] and, probably ¹J[¹¹B(8,10)-¹¹B(1,3)] respectively. The apparent magnitudes of these were *ca.* 30 and $\leq ca.$ 25 Hz respectively, again very large for observed interboron couplings within contiguous borane polyhedra.

The ¹H resonances of compound (2) were traced to their directly bound boron atoms by ¹H-{¹¹B} selective doubleirradiation experiments (Table 3) and as with compound (1) there was a general parallel between $\delta(^{11}B)$ and $\delta(^{1}H)(exo)$, the slope of the correlation, at $\delta(^{11}B)$: $\delta(^{1}H)$ ca. 18:1, being somewhat greater than observed for the *nido* species (1) (Figure 5). A higher $\delta(^{11}B)$: $\delta(^{1}H)$ slope for arachno versus nido species has been observed elsewhere,^{10,13} whereas *closo* species often have a lower value for $\delta(^{11}B)$: $\delta(^{1}H)$ (e.g. refs. 27, 36, and 37). However, an insufficient number of compounds has been examined to assess the generality of this behaviour. The endoterminal proton on B(9) resonates at $\Delta \sigma$ ca. +1 p.p.m. above this general trend, and $exo^{-1}H(9)$ at $\Delta\sigma$ ca. +3 p.p.m., both within expected ^{4,9} ranges (note that shielding, σ , takes the opposite sign to chemical shift, δ). Interproton correlations were observed for all the ${}^{3}J({}^{1}H-{}^{1}H)$ paths in the $[{}^{1}H-{}^{1}H]$ -COSY experiments (Table 3). These, together with the observed ${}^{2}J({}^{1}H-{}^{1}H)$ correlations involving the bridging protons, confirm the positional assignments. Fine structure arising from these interproton couplings was apparent on most of the ¹H resonance lines in the ¹H spectra that were recorded with complete $\{^{11}B(broad-band noise)\}$ decoupling, some of the fine structure for ${}^{1}H(9)(exo)$ and ${}^{1}H(9)(endo)$ being better resolved (footnotes *i* and *j* in Table 3).

Discussion

Comparison of N.M.R. Properties.—It is of interest to compare the n.m.r. shielding patterns of nido-6-SB₉H₁₁ (1) and arachno-[6-SB₉H₁₂]⁻ (2) with those of their binary borane analogues, nido-B₁₀H₁₄ and arachno-[B₁₀H₁₄]²⁻ respectively, in order to assess the effects of replacing the 6-boron positions with the sulphur heteroatom, and also to compare the shielding pattern of (1) with that of (2) to see whether the inversion of the ordering of the shielding pattern observed ⁴⁻⁸ for B₁₀H₁₄→ [B₁₀H₁₄]²⁻ is also observed for SB₉H₁₁→[SB₉H₁₂]⁻. These comparisons are summarized in Figure 5(b) and Figure 5(a) shows the $\delta(^{11}B)-\delta(^{1}H)$ correlation data as mentioned above.

There are marked effects upon replacement of BH(6) (plus its associated bridging hydrogen atoms) in $B_{10}H_{14}$ by sulphur to give the $nido-6-SB_9H_{11}$ species (1). The principal effect is a ca. 25 p.p.m. deshielding of the ¹¹B(5,7) resonance adjacent to the sulphur, with the other adjacent site, at ${}^{11}B(2)$, being relatively little affected; this could be a consequence of the removal of bridging (5,6; 6,7) hydrogen atoms as well as the introduction of the heteroatom, changes in protonation and in effective protonation of polyhedral systems being known to produce significant consequential changes in electronic structure in particular cases.^{2-4,7,46,47} Similar changes in δ [¹¹B(5,7)], but not generally so marked, occur in the replacement of nido-decaboranyl BH(6) (but now not also of its associated bridging hydrogen atoms) by a variety of transition-metal centres.^{25,48,49} Associated with this α -deshielding effect at **B**(5,7) is a significant β -shielding increase at the ¹¹**B**(8,10) nuclei, and there are also significant γ decreases at the more distant antipodal ¹¹B(4) and open-face ¹¹B(9) positions. A large antipodal effect is not unexpected, $2^{-4,50-52}$ but the decrease at the ¹¹B(9) position is of interest, and the occurrence of significant electronic changes at these sites distant from the

sulphur atom is also manifested in the anomalously large value for ${}^{1}J[{}^{11}B(4)-{}^{11}B(9)]$ as discussed above.

Interestingly, the shielding changes in the arachno- $[B_{10}H_{14}]^{2-}$ skeleton upon replacement of a boron centre by sulphur to give the arachno- $[6-SB_9H_{12}]^-$ species (2) are largely similar to those in the *nido* skeleton in spite of their markedly different electronic structures. Thus there are marked adjacent and antipodal deshielding effects at B(5,7) and B(4) respectively, with a β -shielding effect at B(8,10), whereas the adjacent B(2) and β -B(1,3) positions are relatively little affected. The change at B(5,7) is not so large as in the *nido* species, however, perhaps because no change in bridging hydrogen atoms is involved in the replacement. By contrast the most distant open-face γ position at B(9) undergoes a profound deshielding, of >20p.p.m., which is unexpected and, taken with the marked changes for ${}^{11}B(5,7)$ and ${}^{11}B(8,10)$, may well reflect a substantial concerted electronic change in an extended open-face molecular orbital in the arachno ten-vertex system.

It can also be seen from Figure 5 that the basic *nido versus* arachno ten-vertex cluster-shielding patterns are retained in the two thiaboranes examined. Although the substituent effects discussed in the last two paragraphs tend to perturb the pattern somewhat, the basic inversion of the cluster nuclear magnetic shielding pattern on effecting the *nido-arachno* interconversion is retained, and the weighted mean ¹¹B chemical shift in the *arachno* species (2), at -20.2 p.p.m., is at significantly higher field than that of the *nido* species (1), at +1.1 p.p.m. The corresponding values for *arachno*-[B₁₀H₁₄]²⁻ and *nido*-B₁₀H₁₄ are -26.0 and -2.6 p.p.m. respectively.

Identification of arachno- $[SB_9H_{11}(OH)]^-$ (3).—After standing in acetone solution for some months, partial decomposition of compound (2) was noted. The n.m.r. spectra of the products were obtained by difference spectroscopy and were interpretable in terms of only one polyhedral boron-containing product [species (3), Table 4, and Figure 6], together with, probably, small amounts of borate products giving rise to resonances⁵³ with $\delta(^{11}B)$ ca. + 19 p.p.m. Apart from these borate resonances, the overall ¹¹B n.m.r. behaviour of species (3) was very similar to that of (2), but with the COSY correlations indicating a different shielding ordering of the three resonances of relative intensity 1 B, the incidence and intensity of the COSY correlations between the numbered cluster positions then being essentially the same as those observed for (1).

Table 4. Measured n.m.r. properties for the species identified as $[endo-9-(OH)-arachno-6-SB_9H_{11}]^-$ (3) in $(CD_3)_2CO$ solution at 297 K

Assignment ^a	$\delta(^{11}B)/p.p.m.^{b}$	Δσ(¹¹ B)/ p.p.m. ^c	$\delta(^{1}H)/p.p.m.^{d}$
(9)	+7.9(1 B)	-22.8	$+3.89(OH),^{e}+2.02$
(5,7)	-5.6(2 B)	-2.8	+2.62
(4)	-7.9(1 B)	+11.5	+1.98
(2)	-23.9(1 B)	+12.0	+1.83
(8,10)	-34.4 (2 B)	+0.6	+0.52
(1,3)	- 39.9 (2 B)	+2.3	+0.60
μ(5,10;7,8)	. ,		1.79

^{*a*} By relative intensities and observed $[{}^{11}B{}^{-11}B]$ -COSY correlations, the latter being similar in incidence and relative intensity to those observed for the unsubstituted species (2) (Table 3). ^{*b*} ±0.5 p.p.m.; relative intensities in parentheses. ^{*c*} $\Delta \sigma = [\delta({}^{11}B)(2) - \delta({}^{11}B)(3)]$, and is the increase in ${}^{11}B$ shielding observed on substituting compound (2) (Table 3) with *endo*-9-hydroxy. ^{*d*} ±0.05 p.p.m., $\delta({}^{1}H)$ related to directly bound boron positions by ${}^{1}H{}_{-}{}^{(11}B(\text{selective})\}$ experiments. ^{*e*} Only weakly sharpened in ${}^{1}H{}_{-}{}^{(11}B{}_{-}O{}^{-1}H)$.



Figure 6. 128 MHz ¹¹B-{¹H(broad-band noise)} difference spectrum showing the [endo-9-(OH)-arachno-6-SB₉H₁₁]⁻ anion [species (3), numbered as in Figure 1] and borate resonance peaks arising from the hydrolysis of the [arachno-6-SB₉H₁₂]⁻ anion (2) in acetone solution. The spectrum was obtained by the subtraction of a spectrum of compound (2) (e.g. Figure 4) from a partially hydrolysed mixture. The features at the chemical shifts of species (2) (marked 'F') arise from an imperfect subtraction that arose because of unsuspected field homogeneity changes that occurred in the months between recording the two spectra



This consideration, together with the absence of an endoterminal ¹H resonance associated with ¹¹B(9) (see ref. 10) then indicated that species (3) is a 9-endo-substituted arachno-[6- $SB_{9}H_{11}$ - cluster anion closely related to (2). The substituent is reasonably indentifiable as endo-9-hydroxy on the basis of the ca. 20 p.p.m. negative substituent chemical shift effect $\Delta\sigma$ (reasonably ascribable to alkoxy, hydroxy, or oxy),^{4,11,23,54,55} together with a marginal sharpening of the resonance at $\delta(^{1}H)$ +3.89 [1 H, reasonably ascribable to OH] in ${}^{1}H{}^{{11}}B(9)$ selective irradiation experiments. An anion of this [9-(OH) $arachno-6-SB_{9}H_{11}]^{-1}$ formulation has previously been reported.²⁰ The essential identity of the ¹¹B shielding behaviour of (3) with this previously reported species confirms the formulation. This proposed identification of (3) as [endo-9-(OH)-arachno-6-SB₉H₁₁]⁻ together with the other observed ¹¹B data would imply positive hydroxy substituent shielding effects $\Delta \sigma$ of ca. + 10 p.p.m. for ¹¹B(8,10) (adjacent) and ¹¹B(2) (antipodal) [both within reasonably expected ranges],^{23,54} the other positions being largely unperturbed from those of the unsubstituted species (2).

The hydroxy anion (3) is probably formed *via* hydrolysis by fortuitous moisture in the samples of acetone solvent used. Its incidence and the formation of the *endo* configuration, as opposed to the *exo* configuration and/or a general decomposition to borate, being presumably stabilized by $O-H \cdots S$ interaction with the 6-sulphur *endo* lone pair as indicated in (III). The *endo,endo* chelating effect resulting in cluster stabilization has been noted elsewhere in ten-vertex *arachno* chemistry.^{8,10,56}

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References

- 1 G. R. Eaton and W. N. Lipscomb, 'N.M.R. Studies of Boron Hydrides and Related Compounds,' Benjamin, New York and Amsterdam, 1969.
- 2 L. J. Todd and A. R. Siedle, Prog. Nucl. Magn. Reson. Spectrosc., 1979, 13, 87; A. R. Siedle, Annu. Rep. N.M.R. Spectrosc., 1982, 12, 177.
- 3 S. Heřmánek, T. Jelínek, J. Plešek, B. Štíbr, J. Fusek, and F. Mareš, in 'Boron (IMEBORON VI),' ed. S. Heřmánek, World Scientific Press, Singapore, 1987, pp. 26–73.
- 4 J. D. Kennedy, 'Boron,' ch. 8 in 'Multinuclear N.M.R.,' ed. J. Mason, Plenum, London and New York, 1987, pp. 221–258, and refs. therein.
- 5 D. E. Hyatt, F. R. Scholer, and L. J. Todd, *Inorg. Chem.*, 1967, 6, 630.
- 6 W. N. Lipscomb, R. J. Wiersema, and M. F. Hawthorne, *Inorg. Chem.*, 1972, 11, 651.
- 7 R. R. Rietz, A. R. Siedle, R. D. Schaeffer, and L. J. Todd, *Inorg. Chem.*, 1973, 12, 2100.
- 8 M. A. Beckett and J. D. Kennedy, J. Chem. Soc., Chem. Commun., 1983, 575.
- 9 R. Ahmad, J. E. Crook, N. N. Greenwood, and J. D. Kennedy, J. Chem. Soc., Dalton Trans., 1986, 2433.
- 10 X. L. R. Fontaine and J. D. Kennedy, J. Chem. Soc., Dalton Trans., 1987, 1573.
- 11 S. K. Boocock, N. N. Greenwood, M. J. Hails, J. D. Kennedy, and W. S. McDonald, J. Chem. Soc., Dalton Trans., 1981, 1415.
- 12 M. A. Beckett, J. E. Crook, N. N. Greenwood, and J. D. Kennedy, J. Chem. Soc., Dalton Trans., 1984, 1427.
- 13 X. L. R. Fontaine, N. N. Greenwood, J. D. Kennedy, P. MacKinnon, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., in the press.
- 14 J. Bould, N. N. Greenwood, and J. D. Kennedy, J. Chem. Soc., Dalton Trans., 1984, 2477.
- 15 M. Bown, X. L. R. Fontaine, N. N. Greenwood, J. D. Kennedy, and P. MacKinnon, J. Chem. Soc., Chem. Commun., 1987, 817.
- 16 X. L. R. Fontaine, H. Fowkes, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., 1987, 2417.
- 17 M. Bown, X. L. R. Fontaine, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., 1987, 1169.
- J. D. Kennedy, Prog. Inorg. Chem., 1986, 34, 211.
 I. J. Todd, A. R. Garber, G. M. Bodner, A. R. Siedle, and R. Wright,
- unpublished work, as cited in ref. 2.
- 20 A. R. Siedle, G. M. Bodner, A. R. Garber, and L. J. Todd, *Inorg. Chem.*, 1977, **13**, 1756; A. R. Siedle, D. McDowell and L. J. Todd, *ibid.*, 1974, **13**, 2735.
- 21 R. W. Rudolph and W. R. Pretzer, Inorg. Synth., 1983, 22, 226.
- 22 J. D. Kennedy and B. Wrackmeyer, J. Magn. Reson., 1980, 38, 529.
- 23 J. D. Kennedy and N. N. Greenwood, Inorg. Chim. Acta, 1980, 38, 93.
- 24 T. L. Venable, W. C. Hutton, and R. N. Grimes, J. Am. Chem. Soc., 1984, 106, 29.
- 25 M. Bown, X. L. R. Fontaine, N. N. Greenwood, and J. D. Kennedy, J. Organomet. Chem., 1987, 325, 233.
- 26 X. L. R. Fontaine, H. Fowkes, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., 1986, 547.
- 27 X. L. R. Fontaine, H. Fowkes, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., 1987, 1431.
- 28 M. A. Beckett, M. Bown, X. L. R. Fontaine, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., in the press.
- 29 X. L. R. Fontaine and J. D. Kennedy, J. Chem. Soc., Chem. Commun., 1986, 779.
- 30 J. D. Kennedy and J. Staves, Z. Naturforsch., Teil B, 1979, 34, 808.
- 31 W. McFarlane, Proc. R. Soc. London, Ser. A, 1968, 306, 185.
- 32 T. C. Farrar and G. R. Quinting, Inorg. Chem., 1985, 24, 1941.
- 33 R. F. Sprecher and J. C. Carter, J. Am. Chem. Soc., 1973, 95, 2369.
- 34 E. W. Corcoran and L. G. Sneddon, Inorg. Chem., 1985, 107, 7446.
- 35 R. J. Astheimer and L. G. Sneddon, *Inorg. Chem.*, 1983, 22, 1928 and refs. therein.

- 36 G. Ferguson, Faridoon, O. Ni Dhubhghaill, T. R. Spalding, J. D. Kennedy, and X. L. R. Fontaine, unpublished work.
- 37 Faridoon, O. Ni Dhubhghaill, T. R. Spalding, G. Ferguson, X. L. R. Fontaine, and J. D. Kennedy, unpublished work.
- 38 A. O. Clouse, D. C. Moody, R. R. Rietz, T. Roseberry, and R. Schaeffer, J. Am. Chem. Soc., 1973, 95, 2496.
- 39 E. J. Stampf, A. R. Garber, J. D. Odom, and P. D. Ellis, J. Am. Chem. Soc., 1976, **98**, 6550.
- 40 D. F. Gaines, C. K. Nelson, J. C. Kunz, J. H. Morris, and D. Reed, Inorg. Chem., 1984, 23, 3252.
- 41 T. C. Gibb and J. D. Kennedy, J. Chem. Soc., Faraday Trans. 2, 1982, 525.
- 42 T. Onak, W. Inman, H. Rosendo, E. W. Distefano, and J. Nurse, J. Am. Chem. Soc., 1977, 99, 6488.
- 43 T. L. Venable, C. T. Brewer, and R. N. Grimes, *Inorg. Chem.*, 1985, 24, 4751.
- 44 S. K. Boocock, N. N. Greenwood, J. D. Kennedy, and D. Taylorson, J. Chem. Soc., Chem. Commun., 1979, 16.
- 45 S. K. Boocock, N. N. Greenwood, J. D. Kennedy, W. S. McDonald, and J. Staves, J. Chem. Soc., Dalton Trans., 1980, 790.
- 46 P. M. Garret, G. S. Ditta, and M. F. Hawthorne, J. Am. Chem. Soc., 1971, 93, 1265.
- 47 B. Stíbr, S. Hěrmánek, Z. Janousek, Z. Plzák, J. Dolánsky, and J. Plesek, *Polyhedron*, 1982, 1, 822.

- 48 M. A. Beckett, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., 1986, 795.
- 49 N. N. Greenwood, J. D. Kennedy, I. Macpherson, and M. Thornton-Pett, Z. Anorg. Allg. Chem., 1986, 54, 45.
- 50 W. L. Smith, B. J. Meneghelli, N. McClure, and R. W. Rudolph, J. Am. Chem. Soc., 1976, 98, 624.
- 51 W. F. Wright, A. R. Garber, and L. J. Todd, J. Magn. Reson., 1978, 30, 595.
- 52 F. Teixidor, C. Viñas, and R. Rudolph, Inorg. Chem., 1986, 25, 3339.
- 53 H. Nöth and B. Wrackmeyer, 'Nuclear Magnetic Resonance Spectroscopy of Boron Compounds,' Springer, Berlin, Heidelberg and New York, 1978, and refs. therein.
- 54 M. Bown, X. L. R. Fontaine, N. N. Greenwood, J. D. Kennedy, and M. Thornton-Pett, J. Chem. Soc., Dalton Trans., 1988, 925.
- 55 R. Ahmad, N. N. Greenwood, and J. D. Kennedy, unpublished work; R. Ahmad, Ph.D. Thesis, University of Leeds, 1982.
- 56 M. Thornton-Pett, M. A. Beckett, and J. D. Kennedy, J. Chem. Soc., Dalton Trans., 1986, 303.

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