## High-nuclearity cobaltadendrimers

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New dendritic polyalkynes have been prepared and reacted with  $[\text{Co}_2(\text{CO})_8]$  to give cobaltadendrimers containing up to 40 cobalt atoms.

Metalladendrimers are of intrinsic structural and synthetic interest <sup>1-3</sup> and offer potential applications as light-collecting devices, information storage devices and polyfunctional catalysts. We have prepared compounds in which the metals are part of the backbone dendritic connectivity, <sup>1</sup> or in which they decorate the dendrimer. <sup>4</sup> The generation of C<sub>2</sub>Co<sub>2</sub>(CO)<sub>6</sub> clusters from alkynes is a facile method for the introduction of multiple metal centres which has not been widely used in metalladendrimer chemistry. <sup>5</sup> We now report an extension of our previous work (which led to a starburst decorated with C<sub>2</sub>Co<sub>2</sub>(CO)<sub>6</sub> through to the third generation) <sup>6</sup> to genuinely dendritic systems.

Our synthetic strategy involves the preparation of a starburst or dendritic polyalkyne and post-functionalization with [Co<sub>2</sub>-(CO)<sub>8</sub>].<sup>6</sup> The doubly-protected compound 1 was prepared in 53% yield† by the sequential reaction of 1,3,5-tribromobenzene with two equivalents of (TIPS)C=CH (TIPS = triisopropylsilyl) and a three-fold excess of (TMS)C=CH (TMS = trimethylsilyl) in each case in the presence of [(Ph<sub>3</sub>P)<sub>2</sub>PdCl<sub>2</sub>], CuI and NEt<sub>3</sub>,

followed by alkaline cleavage of the TMS group. The use of the base-stable TIPS protecting group is critical to the success of this strategy and allows the preparation of asymmetrically substituted derivatives. The reaction of  $C(p-C_6H_4I)_4^7$  with an excess of 1 under similar palladium(II)-catalysed coupling conditions followed by deprotection of 2 with [ $^nBu_4N$ ]F in THF yielded the intensely luminescent ( $\lambda_{max}$  352 nm) dendritic dodecaalkyne 3 as white crystals.‡ Upon stirring 3 with [ $Co_2(CO)_8$ ] in  $CH_2Cl_2$ , a dark coloured solution was obtained from which the dendritic tetracosacobalt complex 4 was isolated as deep red crystals in 28% yield.§

Linear extension of these systems proved to be facile. The reaction of the polyalkyne 3 with  $p\text{-}\text{IC}_6\text{H}_4\text{C}\equiv\text{C}(\text{TMS})$  under standard Pd-coupling conditions yielded the protected dendritic icosaalkyne 5 as yellow crystals. Subsequent basic deprotection gave icosaalkyne 6, the reaction of which with an excess of  $[\text{Co}_2(\text{CO})_8]$  produced the deep red crystalline tetracontacobalt compound  $7\P$  (Fig. 1). This latter compound and all others described were characterized by the normal spectroscopic and analytical methods adopted for 'small molecules' even though modelling indicates that 7 has a diameter of  $\approx 35$  nm.

We are currently investigating the chemical and structural aspects of these novel high-nuclearity species.

Fig. 1 Proposed structure of compound 7.

TIPS

TIPS

TIPS

$$\begin{array}{c}
2 & R = TIPS \\
3 & R = H \\
5 & R = 4 \cdot C_0 H_4 C \equiv C(TMS) \\
6 & R = 4 \cdot C_0 H_4 C \equiv CH
\end{array}$$

$$\begin{array}{c}
CO_{3} & CO(CO)_{3} \\
CO_{3} & CO(CO)_{3}
\end{array}$$

$$\begin{array}{c}
CO_{3} & CO(CO)_{3} \\
CO(CO)_{3} & CO(CO)_{3}
\end{array}$$

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## **Notes and references**

† 1: 1,3,5-Br<sub>3</sub>C<sub>6</sub>H<sub>3</sub> (1.35 g, 4.28 mmol), (TIPS)C≡CH (1.64 g, 8.99 mmol), CuI (81.5 mg, 0.43 mmol) and [(PPh<sub>3</sub>)<sub>2</sub>PdCl<sub>2</sub>] (300 mg, 0.43 mmol) were stirred in NEt<sub>3</sub> (50 ml) for 4 h at 35 °C. Treatment with (TMS)C≡CH (1.26 g, 12.8 mmol) under analogous conditions to those described above (reaction time 12 h), followed by chromatographic work-up gave the protected intermediate; this was dissolved in THF (120 ml) and 1 M NaOH (150 ml) added; the solution stirred for 3.5 h. After extraction, the residue was purified by column chromatography to give a colourless oil (1.06 g; 53%). ¹H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  7.52–7.50 (m, 3H, Ar), 3.08 (s, 1H, CCH), 1.12 (s, 42H, TIPS); ¹³C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  135.2, 124.2, 122.6, 105.1, 92.4, 82.0, 78.3, 18.7 (TIPS), 11.3 (TIPS); MS (MALDI-TOF) mlz 502 [M + K]<sup>+</sup>, 486 [M + Na]<sup>+</sup>.

 $\ddagger$  3: compound 1 (0.37 g, 0.80 mmol),  $C(p-C_6H_4I)_4$  (0.16 g, 0.19 mmol), CuI (10.8 mg, 0.06 mmol) and [(Ph<sub>3</sub>P)<sub>2</sub>PdCl<sub>2</sub>] (40.0 mg, 0.06 mmol) were stirred in NEt<sub>3</sub> (10 ml) and DMF (20 ml) for 70 h at 35 °C. Chromatographic work-up gave **2** as yellow crystals (0.35 g, 84.5%). Deprotection using [<sup>n</sup>Bu<sub>4</sub>N]F (1.60 mmol) in THF (50 ml, room temperature, 4 h) yielded 3 as white crystals (62 mg, 42%). <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.61 (d, J 1.4 Hz, 8H), 7.56 (t, J 1.5 Hz, 4H), 7.45 (d, J 8.6 Hz, 8H), 7.21 (d, J 8.5 Hz, 8H), 3.11 (s, 8H, CCH); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) δ 146.2, 135.1, 131.3, 130.9, 124.0, 122.9, 120.8, 90.3, 87.9, 81.8, 78.5, 48.4; MS (MALDI-TOF) m/z 1828 [2M]<sup>+</sup>, 914 [M]<sup>+</sup>. § 4: alkyne 3 (54.3 mg, 0.06 mmol) and Co<sub>2</sub>(CO)<sub>8</sub> (0.49 g, 1.43 mmol) were stirred in CH<sub>2</sub>Cl<sub>2</sub> (10 ml) for 1.5 h at room temperature, the solvent removed, and the residue purified by column chromatography to give a deep red crystalline solid (71.9 mg, 28%). IR (KBr disc, cm $^{-1}$ )  $v_{CO}$  2093 s, 2055 vs, 2020 vs;  $^{1}$ H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  7.67 (d, J 2.0 Hz, 8H), 7.57 (d, J 8.3 Hz, 8H), 7.50 (t, J 1.7 Hz, 4H), 7.18 (d, J 8.8 Hz, 8H), 6.37 (s, 8H,  $H_{\text{cluster}}$ ); <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>)  $\delta$  199.0 (CO), 145.7, 140.4, 139.3, 135.9, 131.7, 130.4, 129.2, 128.3, 91.4, 90.3, 88.5, 72.8, 64.8; MS (MALDI-TOF) m/z 4317 [M - CO] $^+$ .

¶ 6 and 7: alkyne 3 (62 mg, 68  $\mu$ mol), *p*-IC<sub>6</sub>H<sub>4</sub>C≡C(TMS) (0.24 g, 0.82 mmol), CuI (10.4 mg, 0.05 mmol) and [(PPh<sub>3</sub>)<sub>2</sub>PdCl<sub>2</sub>] (38.1 mg, 0.05 mmol) were stirred in dry NEt<sub>3</sub> (5 ml) for 100 h at 39 °C. Chromatographic work-up gave compound 5 as yellow crystals (67 mg, 43%); it was dissolved in THF (30 ml), and 1 M NaOH (30 ml) added; the solution was stirred for 4 h. Water was added and after extraction with CH<sub>2</sub>Cl<sub>2</sub>, the residue was purified by column chromatography to give 6 as white crystals (21.3 mg, 43%). Reaction of 6 (15.4 mg, 8.98 µmol) with [Co<sub>2</sub>(CO)<sub>8</sub>] under the same conditions as for 4 gave 7 as deep red crystals (27 mg, 41%). **6**:  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.65 (br s, 12H), 7.48–7.46 (m, 40H), 7.24 (d, J 8.6 Hz, 8H), 3.19 (s, 8H, CC*H*); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>)  $\delta$  146.2, 134.3, 132.1, 131.5, 131.3, 130.9, 124.0, 123.8, 123.1, 122.3, 120.9, 90.3, 90.1, 89.6, 88.1, 83.1, 79.2; MS (MALDI-TOF) m/z 1713 [M]<sup>+</sup>. 7: IR (KBr disc, cm<sup>-1</sup>)  $\nu_{CO}$  2091 s, 2055 vs, 2020 vs; <sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>)  $\delta$  7.79 (d, J 1.6 Hz, 8H), 7.72 (t, J 1.7 Hz, 4H), 7.62 (d, J 8.6 Hz, 8H), 7.56 (d, J 8.4 Hz, 16H), 7.48 (d, J 8.5 Hz, 16H), 7.24 (d, J 8.6 Hz, 8H), 6.46 (s, 8H, CCH); <sup>13</sup>C NMR (101 MHz, CD<sub>2</sub>Cl<sub>2</sub>) δ 199.8 (CO), 199.4 (CO), 199.2 (CO), 146.2, 140.9, 140.6, 138.4, 137.9, 136.3, 131.9, 131.2, 129.9, 129.4, 128.8, 91.8, 91.7, 90.9, 90.8, 89.6, 73.4, 69.5; MS (MALDI-TOF) m/z 7298 [M - 5CO]<sup>+</sup>,  $7144 [M - 10 CO]^{+}$ 

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