

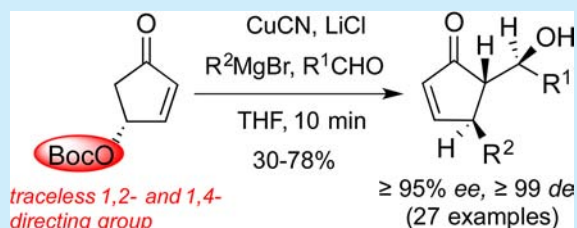
Traceless Stereinduction for the Enantiopure Synthesis of Substituted-2-Cyclopentenones

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S Supporting Information

ABSTRACT: The pseudoenantiomeric 4-O-Boc- and 4-OPMP-cyclopent-2-enones, readily available from hydroxymethylfuran on multigram scale, are demonstrated to be exceptional building blocks for the synthesis of enantiopure 4-alkyl-5-(1'-hydroxyalkyl) substituted 2-cyclopentenones and derivatives thereof. The 4-OR substituent acts as a traceless stereoredirecting element, conferring not only 1,2- but also 1,4-stereocontrol with excellent selectivity. The methodology developed here was applied for the rapid synthesis of natural products and biologically active 2-cyclopentenones such as TEI-9826, guaianes, and pseudoguaianolides.



Chiral 4-alkyl-5-(1-hydroxyalkyl) substituted-2-cyclopentenones **1** and derivatives thereof are important constituents of numerous natural products and biologically active drugs.¹ For example, prostaglandin analogues **2**² and **3**³ have strong antitumor activity, and especially the latter was shown to retain *in vivo* activity against *cis*-platin-resistant tumors (Figure 1). Sesquiterpenoid **4**^{4a} was found to be a submicromolar

Scheme 1. Synthesis of Chiral Cyclopentenones According to Noyori et al.

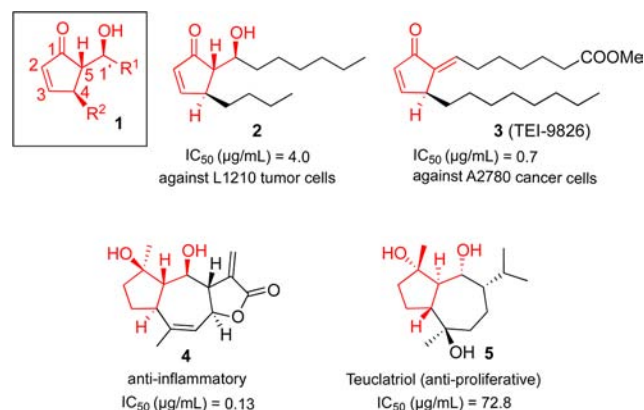
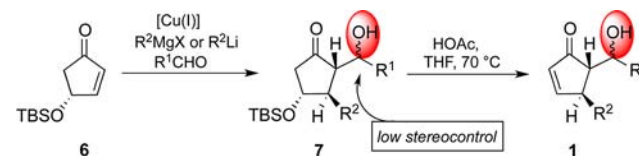


Figure 1. Bioactive substituted 2-cyclopentenones, pseudoguaianolide, and guaiane molecules.

inhibitor against LPS-induced nitric oxide production in RAW264.7 macrophages, while teuclatriol **5** shows strong antiproliferative effects on human activated peripheral blood lymphocytes.^{4b,c}

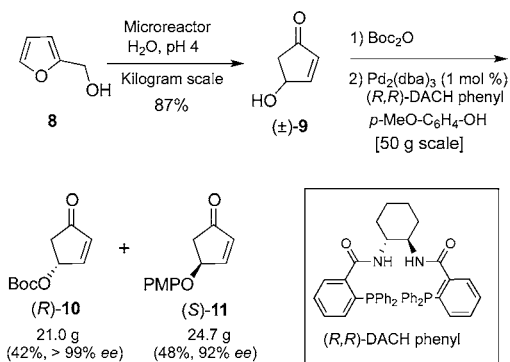
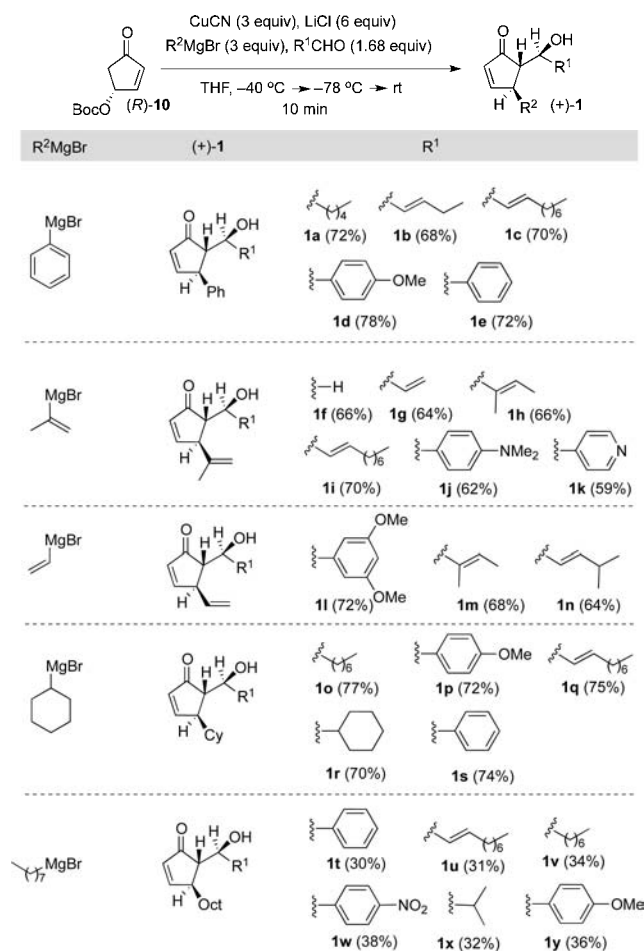
With the pioneering work of Noyori et al.⁵ utilizing 4-siloxy-2-cyclopentenone **6** as a key building block for the synthesis of prostaglandins (Scheme 1), a reliable strategy toward enantiopure cyclopentenones was established, being broadly applied by many.⁶ Conjugate *anti*-addition of nucleophiles to **6** in the 3-position controlled by the adjacent siloxy group

followed by an aldol reaction in the 2-position *anti* to the nucleophile just introduced was developed. Subsequently, the siloxy group can be eliminated to generate substituted 2-cyclopentenones. With the advent of asymmetric conjugate additions, it was demonstrated that 2-cyclopentenone can be used directly as a starting material for the synthesis of enantiopure cyclopentenones.^{7a} Other strategies toward the target structure have been reported as well;^{7b} however, mixtures of epimers at C-1', often in ratios close to 1:1, are generally obtained with only a few exceptions that require sterically demanding aldehydes.^{6d,7c}

We report here the pseudoenantiomeric building blocks (*R*)-**10** and (*S*)-**11**⁸ (Scheme 2) as exceptional starting materials for the one-flask synthesis of the target structure **1** with excellent enantio- and diastereocontrol including the C-1' position. The 4-oxo-substituents in (*R*)-**10** and (*S*)-**11** act as traceless stereoredirecting elements that relay not only 1,2- but also remote 1,4-stereocontrol in a cascade of nucleophile addition/aldol reaction/elimination (Table 1, Scheme 3). Moreover, it is demonstrated that racemic α -chiral aldehydes are resolved under the reaction conditions, allowing the one-step construction of cyclopentenones with 4-contiguous stereocenters (Scheme 5b).

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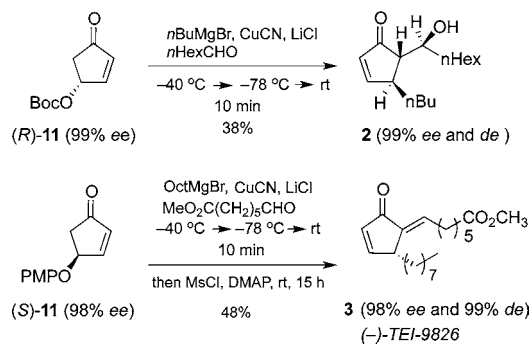
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Scheme 2. Multigram Synthesis of the Pseudoenantiomeric Building Blocks (*R*)-11 and (*S*)-11Table 1. Synthesis of Enantiopure Cyclopentenones (+)-1^a

^aAll products were obtained with $\geq 99\%$ *de* (¹H NMR of the crude products). 1a–1d, 1g–1i, 1o–1r, 1t, 1u: $\geq 99\%$ *ee*; 1j: 98% *ee*; 1l: 96% *ee*; 1m: $\geq 95\%$ *ee*; 1n: 95% *ee* (chiral HPLC against racemic reference samples; see Supporting Information). The relative stereochemistry of 1 was determined according to Kobayashi et al.^{7b} (see Supporting Information, Table S3).

(*R*)-10 and (*S*)-11 are readily available in enantiopure form from the bulk chemical furfurylalcohol (8) that can be converted to (±)-9 on kilogram scale by an acid catalyzed rearrangement carried out in a microreactor setup (Scheme 2).^{9a} Boc protection leads to (±)-9, which was resolved on a 50 g scale with *p*-methoxyphenol (0.5 equiv) utilizing Trost's

Scheme 3. One Flask Synthesis of Antitumor Agents 2 and 3 (TEI-9826)

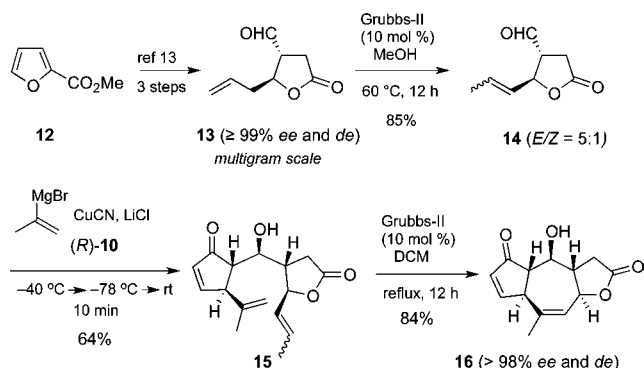


technology,^{9b–d} giving rise to (*R*)-10 (42%, $>99\%$ *ee*) and (*S*)-11 (48%, 92% *ee*) (Scheme 2). Higher enantiopurity for (*S*)-11 (96–98% *ee*) can be achieved lowering the amount of *p*-methoxyphenol to 0.2–0.4 equiv (Supporting Information, Table S1).

Copper(I)-catalyzed Grignard additions to (*R*)-10 followed by trapping of the resulting enolate with aldehydes were investigated next. Optimizing the reaction conditions¹⁰ with respect to copper sources, ligands, and solvents (Table S2, Supporting Information) revealed that CuCN·2LiCl in THF turned out to be best, directly giving rise to 1 with excellent diastereoselectivity ($\geq 99\%$ *de*, Table 1), irrespective of the Grignard reagent employed (aryl-, vinyl-, and alkyl) or the aldehydes used for trapping the resulting enolates (aromatic, α,β -unsaturated or aliphatic). The level of enantioselectivity is determined by the *anti*-selectivity of the Grignard reagent in the initial conjugate addition to (*R*)-10, and small differences in selectivity are observed correlating with its steric bulk. Introducing a vinyl group as the smallest nucleophile investigated (11–1n, Table 1), the enantioselectivity was $\geq 95\%$ *ee*, while octyl, isopropenyl, cyclohexyl, and phenyl generally gave $\geq 99\%$ *ee* with the exception of 1j (98% *ee*).

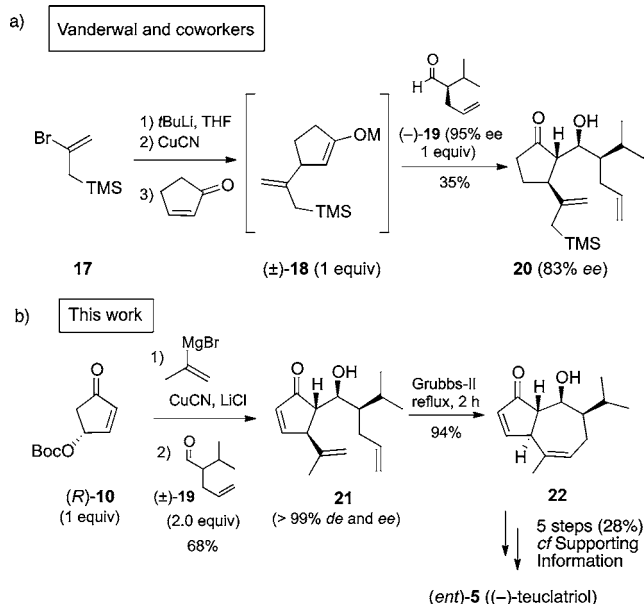
Using this strategy, the rapid assembly of biologically active prostaglandin derivatives could be accomplished. Starting from (*R*)-10 the antitumor agent 2² is accessible in one step (Scheme 3) via addition of *n*-butylmagnesium bromide followed by trapping with heptanal. The commercial anticancer drug TEI-9826 (3), which has been the target of many groups,^{3,11} can be assembled in unparalleled purity and step economy in a one-flask protocol from (*S*)-11 (Scheme 3). In general, (*S*)-11 undergoes the analogous cascade sequences (not shown) described in Table 1 for (*R*)-10 with equally high selectivities and yields.

Pseudoguaianolides and guainanes represent the largest group among naturally occurring sesquiterpene lactones.¹² We envisioned a new strategy to these structures applying the methodology described above (Scheme 4). To accomplish this goal, lactone 14 was required, which was synthesized from the known 13¹³ by a ruthenium-catalyzed double bond isomerization.¹⁴ The pseudoguaianolide core 16, which should be a suitable precursor for the anti-inflammatory compound 4 (Figure 1b), was then efficiently constructed by the three-component reaction between (*R*)-10, isopropenylmagnesium bromide, and 14, giving rise to 15 followed by ring closing metathesis to 16. Noteworthy, 16 is ultimately assembled from 8 and 12, being both derived from the renewable bulk chemical furfural. No protecting groups are required in the reaction

Scheme 4. Enantioselective, Protecting-Group-Free Synthesis of the Core Structure of Pseudoguaianolides


sequence, and all chiral information is introduced by catalytic asymmetric methodology (Scheme 4).

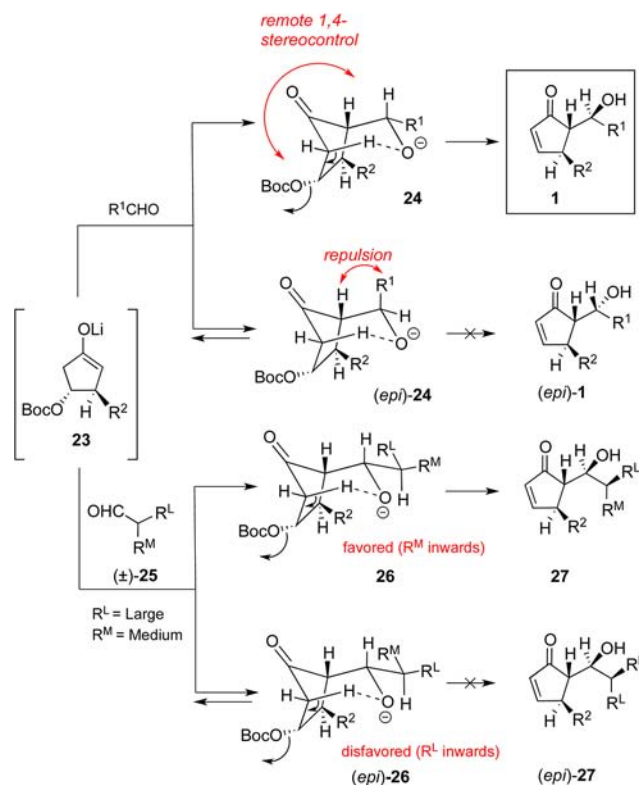
Last but not least, the enantioselective synthesis of (–)-teuclatriol (*ent*)-5 was achieved, taking the lead from the elegant studies of Vanderwal and co-workers reported for the synthesis of (–)-teucladiol (Scheme 5).¹⁵

Scheme 5. (a) Kinetic Resolution of Achiral Enolate (±)-18 Using Enantiopure Aldehyde (–)-19; (b) Kinetic Resolution of α-Chiral Aldehyde (±)-19 by the Enantiopure Enolate Derived from (R)-10 after Addition of *i*-PrMgBr


In their key step (Scheme 5a), the kinetic resolution of racemic enolate (±)-18 with enantioenriched, highly unstable aldehyde (–)-19 was demonstrated, leading to 20. Besides other diastereomers being formed in the crude, some erosion of enantiopurity in 20 most likely due to epimerization of aldehyde (–)-19 was observed under the reaction conditions. The authors therefore noted that *an ideal enantioselective synthesis would incorporate enantioenriched enolate 18, [but] that the asymmetric conjugate addition of sp^2 -hybridized organometallics to cyclopentenones is not a well-developed process.*¹⁵ We were pleased to find that (R)-10 offers a solution to this problem, allowing the efficient resolution of racemic aldehyde (±)-19—much more readily available than in its enantiopure

form—again with excellent selectivity (Scheme 5b). Thus, reacting (R)-10 with isopropenylmagnesium bromide and (±)-19 gave rise to 21 in diastereo- and enantiopure form. Ring closing metathesis allowed the construction of the guaiane core 22, from which (–)-teuclatriol 5 was obtained in five further steps (see Supporting Information) in 28% overall yield.

We attribute the excellent stereocontrol achieved for 1 at the C-1'-position to the terminating Boc-OH *anti*-elimination from intermediate 24, which we reckon is initially formed along with (*epi*)-24 by aldol reaction from 23 (Scheme 6). Compound 24

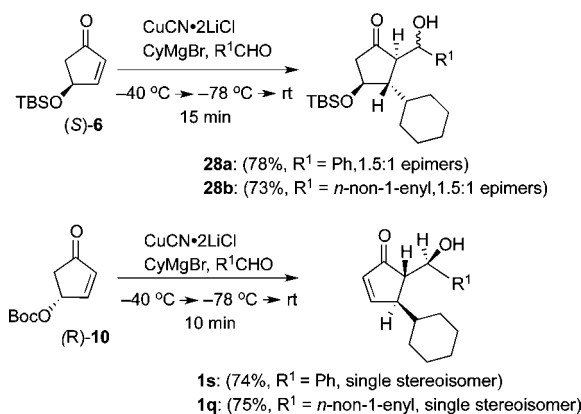
Scheme 6. Stereochemical Model for the Remote 1,4-Stereocontrol of the OBoc Group in Enolate 23


can adopt a favorable conformation that triggers elimination to 1, while, in (*epi*)-24, the conformation that is required to invoke Boc-OH elimination suffers from steric repulsion caused by the axial position of R¹. Thus, assuming reversibility of the aldol process leading to 24/(*epi*)-24, the OBoc-group ultimately acts as a traceless, 1,4-stereoinducing element.

This mechanistic proposal is corroborated by comparing the stereochemical outcome in the reaction of (R)-10 versus (S)-6 with cyclohexylmagnesium bromide followed by an aldol reaction with either benzaldehyde or *trans*-2-decanol under identical reaction conditions (Scheme 7). Extending the stereochemical model to α-chiral aldehydes (±)-25 suggests that 26 is favored over (*epi*)-26: Placing the smallest substituent (hydrogen) on the α-center axially to minimize 1,3-interactions with the cyclopentanone moiety allows R¹ in 26 to orient away from the chair conformation, being most favorable to trigger the elimination of Boc-OH (Scheme 6).

In conclusion, the readily available (R)-10 and (S)-11 allow the stereoselective synthesis of 2-cyclopentenones with excellent selectivity and operational simplicity, allowing the rapid assembly of natural product scaffolds with complex

Scheme 7. Comparison of the Remote Stereoinduction of OTBS in (S)-6 versus OBoc in (R)-10



architecture. The methodology described here offers a versatile approach for the asymmetric synthesis of chiral five-membered carbocycles, being ubiquitous constituents in natural products and drugs.

■ ASSOCIATED CONTENT

Supporting Information

Experimental details can be found in the Supporting Information (procedures, analytical data, copies of NMR and HPLC spectra). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) *Leading Review*: Aitken, D. J.; Eijlsberg, H.; Frongia, A.; Ollivier, J.; Piras, P. P. *Synthesis* **2014**, *46*, 1–24.
- (2) Kato, T.; Fukushima, M.; Kurozumi, S.; Noyori, R. *Cancer Res.* **1986**, *46*, 3538–3542.
- (3) (a) Sasaki, H.; Niimi, S.; Akiyama, M.; Tanaka, T.; Hazato, A.; Kurozumi, S.; Fukushima, S.; Fukushima, M. *Cancer Res.* **1999**, *59*, 3919–3922. (b) Fukushima, S.; Takeuchi, Y.; Kishimoto, S.; Yamashita, S.; Uetsuki, K.; Shirakawa, S.; Suzuki, M.; Furuta, K.; Noyori, R.; Sasaki, H.; Kikuchi, Y.; Kita, T.; Yamori, T.; Sawada, J.; Kojima, M.; Hazato, A.; Kurozumi, S.; Fukushima, M. *Anti-Cancer Drugs* **2001**, *12*, 221–234. (c) Fukushima, S.; Kishimoto, S.; Takeuchi, Y.; Fukushima, M. *Drug Delivery Rev.* **2000**, *45*, 65–75. (d) Furuta, K.; Tomokiyo, K.; Satoh, T.; Watanabe, Y.; Suzuki, M. *ChemBioChem* **2000**, *1*, 283–286. (e) Furuta, K.; Maeda, M.; Hirata, Y.; Shibata, S.; Kiuchi, K.; Suzuki, M. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 5487–5491.
- (4) (a) Cheng, X.; Zeng, Q.; Ren, J.; Qin, J.; Zhang, S.; Shen, Y.; Zhu, J.; Zhang, F.; Chang, R.; Zhu, Y.; Zhang, W.; Jin, H. *Eur. J. Med. Chem.* **2011**, *46*, 5408–5415. (b) Bruno, M.; Tore, M. C.; Rodriguez, B.; Omar, A. A. *Phytochemistry* **1993**, *34*, 245–247. (c) Ziaei, A.; Amirghofran, Z.; Zapp, J.; Ramezani, M. *Iran. J. Immunol.* **2011**, *8*, 226–235.
- (5) (a) Suzuki, M.; Kawagishi, T.; Noyori, R. *Tetrahedron Lett.* **1982**, *23*, 5563–5566. (b) Suzuki, M.; Yanagisawa, A.; Noyori, R.

Tetrahedron Lett. **1984**, *25*, 1383–1386. (c) Noyori, R.; Suzuki, M. *Angew. Chem., Int. Ed. Engl.* **1984**, *23*, 847–876. (d) Suzuki, M.; Yanagisawa, A.; Noyori, R. *Tetrahedron Lett.* **1983**, *24*, 1187–1188.

(6) (a) Heng, K. K.; Smith, R. A. J. *Tetrahedron* **1979**, *35*, 425–435. (b) Suzuki, M.; Kawagishi, T.; Suzuki, T.; Noyori, R. *Tetrahedron Lett.* **1982**, *23*, 4057–4060. (c) Yamada, K. I.; Arai, T.; Sasai, H.; Shibasaki, M. *J. Org. Chem.* **1998**, *63*, 3666–3672. (d) Lipshutz, B. H.; Wood, M. R. *J. Am. Chem. Soc.* **1994**, *116*, 11689–11702.

(7) (a) Nicolaou, K. C.; Tang, W.; Dagneau, P.; Faraoni, R. *Angew. Chem., Int. Ed.* **2005**, *44*, 3874–3879. (b) Kobayashi, Y.; Muruges, M. G.; Nakano, M.; Takahisa, E.; Usmani, S. B.; Ainai, T. *J. Org. Chem.* **2002**, *67*, 7110–7123. (c) Arnold, L. A.; Naasz, R.; Minnaard, A. J.; Feringa, B. L. *J. Org. Chem.* **2002**, *67*, 7244–7254.

(8) Noyori, R.; Suzuki, M.; Kurozumi, S. (Jpn. Kokai Tokkyo Koho), JPA 19860613, 1986.

(9) (a) Ulbrich, K.; Kreitmeier, P.; Reiser, O. *Synlett* **2010**, *13*, 2037–2040. (b) Ulbrich, K.; Kreitmeier, P.; Vilaivan, T.; Reiser, O. *J. Org. Chem.* **2013**, *78*, 4202–4206. (c) Trost, B. M.; Toste, F. D. *J. Am. Chem. Soc.* **1999**, *121*, 3543–3544. (d) Trost, B. M.; Masters, J. T.; Lumb, J. P.; Fateen, D. *Chem. Sci.* **2014**, *5*, 1354–1360.

(10) (a) Suzuki, M.; Suzuki, T.; Kawagishi, M.; Noyori, R. *Tetrahedron Lett.* **1980**, *21*, 1247–1250. (b) Suzuki, M.; Kawagishi, T.; Yanagisawa, A.; Suzuki, T.; Okamura, N.; Noyori, R. *Bull. Chem. Soc. Jpn.* **1988**, *61*, 1299–1312. (c) Lipshutz, B. H.; Wilhelm, R. S.; Kozlowski, J. A. *J. Org. Chem.* **1984**, *49*, 3938–3942. (d) Knochel, P.; Dohle, W.; Gommermann, N.; Kneisel, F. F.; Kopp, F.; Korn, T.; Sapountzis, I.; Vu, V. A. *Angew. Chem., Int. Ed.* **2003**, *42*, 4302–4320. (e) Kobayashi, Y.; Nakada, M. *Angew. Chem., Int. Ed.* **2013**, *52*, 7569–7573. (f) Alexakis, A.; Bäckvall, J. E.; Krause, N.; Pámies, O.; Diéguez, M. *Chem. Rev.* **2008**, *108*, 2796–2823.

(11) (a) Schelwies, M.; Dübon, P.; Helmchen, G. *Angew. Chem., Int. Ed.* **2006**, *45*, 2466–2469. (b) Chen, M.; Hartwig, J. F. *Angew. Chem., Int. Ed.* **2014**, *53*, 8691–8695. (c) Żurawiński, R.; Mikina, M.; Mikołajczyk, M. *Tetrahedron: Asymmetry* **2010**, *21*, 2794–2799. (d) Iqbal, M.; Evans, P. *Tetrahedron Lett.* **2003**, *44*, 5741–5745. (e) Weaving, R.; Roulland, E.; Monneret, C.; Florent, J. C. *Tetrahedron Lett.* **2003**, *44*, 2579–2581. (f) Sugiura, M.; Kinoshita, R.; Nakajima, M. *Org. Lett.* **2014**, *16*, 5172–5175.

(12) (a) Schall, A.; Reiser, O. *Eur. J. Org. Chem.* **2008**, 2353–2364. (b) Fraga, B. M. *Nat. Prod. Rep.* **2007**, *24*, 1350–1381. (c) Gao, F.; Wang, H.; Mabry, T. J.; Bierner, M. W. *Phytochemistry* **1990**, *29*, 895–899.

(13) (a) Chhor, R. B.; Nosse, B.; Sörgel, S.; Böhm, C.; Seitz, M.; Reiser, O. *Chem.—Eur. J.* **2003**, *9*, 260–270. (b) Nosse, B.; Chhor, R. B.; Jeong, W. B.; Böhm, C.; Reiser, O. *Org. Lett.* **2003**, *5*, 941–944. (c) Böhm, C.; Schinnerl, M.; Bubert, C.; Zabel, M.; Labahn, T.; Parisini, E.; Reiser, O. *Eur. J. Org. Chem.* **2000**, 2955–2965.

(14) Hanessian, S.; Giroux, S.; Larsson, A. *Org. Lett.* **2006**, *8*, 5481–5484.

(15) (a) Dowling, M. S.; Vanderwal, C. D. *J. Am. Chem. Soc.* **2009**, *131*, 15090–15091. (b) Dowling, M. S.; Vanderwal, C. D. *J. Org. Chem.* **2010**, *75*, 6908–6922.