Practical Synthesis of Roscovitine and CR8

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Abstract:

Roscovitine and CR8 are potent inhibitors of cyclin-dependent kinases. A scalable synthesis of both inhibitors is described. In the case of CR8, the biarylmethylamine moiety was obtained as a stable and high-purity salt.

1. Introduction

Abnormalities in protein phosphorylation have been observed in numerous major human diseases, $\frac{1}{1}$ strongly encouraging the search for pharmacological inhibitors of protein kinases.² The 2,6,9-trisubstituted purines were among the first low-molecular weight inhibitors of cyclin-dependent kinases (CDKs).³ One of these, (*R*)-roscovitine, **10a** (CYC202, Seliciclib), has now reached clinical phase 2 trials against nonsmall cell lung cancer and nasopharyngeal cancer.⁴ In this context we have pursued an intensive search for improved analogues of roscovitine, taking into account their ability to interact with CDKs (guided by the CDK2/ and CDK5/roscovitine crystal structures), to maintain high kinase selectivity, and to induce cell death at lower concentrations. In this study,⁵ CR8, 10b, was identified as a highly efficacious agent, $25-50$ times more potent than (*R*)-roscovitine at inducing apoptotic cell death and its accompanying biochemical events and therefore suitable for preclinical development.6 We herein describe a protocol for a scalable synthesis of roscovitine and CR8 (Scheme 1).

2. Results and Discussion

We first focussed our work on the production of roscovitine. Roscovitine can be prepared in a simple three-step procedure (Scheme 1). In the last step, a chromatographic purification of roscovitine is required.7 The three-step procedure has been optimized with the goal of avoiding any chromatographic isolation of either the intermediates or the target compound. In the first step, 2,6-dichloropurine **7**, was converted into 6-benzylaminopurine **8a**, upon heating with benzylamine in *n*-butanol. Alkylation of **8a** was regioselectively accomplished in the usual way upon treatment with isopropylbromide in the presence of K_2CO_3 ⁵ Amination with (R) -2-aminobutanol was then performed. However, in this last step, the isolation of the material was plagued by the formation of side compounds (which had lower R_f than 10a in thinlayer chromatography (TLC)). The key feature affecting the reaction's outcome based on an in-depth study appeared to be the quantity of (*R*)-2-aminobutanol. The best compromise between reaction rate and isolated yield was obtained with 8 mol equiv. In the optimized conditions, **10a** was purified by crystallization without the need of chromatographic purification. Indeed, the reaction proceeded faster to completion with 10 mol equiv. However, the isolated yield was diminished (65% instead of 89%).

In the case of CR8, **10b**, a practical route to the key building block **6** had to be secured. Commercially available 4-pyridylbenzaldehyde, **1**, was converted into the oxime **2** using hydroxylamine hydrochloride and AcONa in EtOH at rt. Reduction of **2** was first performed using LiAlH4 in THF. TLC of the crude reaction mixture showed four to five spots less polar than that for the targeted amine **3**. When this crude amine

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^a Reagents and conditions: a) NH2OH·HCl, NaOAc; b) H2 Pd/C, or LiAlH4,THF; c) Boc2O, NEt3; d) CH2Cl2, TFA; e) benzylamines, BuOH, 110°C; f) 2-bromopropane, K₂CO₃, DMSO, 18-20 °C; g) (*R*)-2-aminobutan-1-ol, heat, 160°C.

was reacted in the next step, compound **7b** was isolated in moderate yield. Several attempts to purify **3** as salts were unsuccessful. Hydrogenation of **2** was then investigated. Although the TLC of the mixture looked clean, the NMR spectrum of the crude amine showed a 6:1 mixture of the desired amine **3** and an impurity. In order to identify this side product, the crude reaction mixture was submitted to protection with di-*tert*butyldicarbonate. Interestingly, the Boc protected derivative 4 , crystallized easily from Et₂O, and acylated impurity 5 was well rejected in the mother liquors. The bis-trifluoroacetate was not hygroscopic and could be kept at rt for at least six months. Compound **5** was identified as a piperidine derivative. The hydrogenation conditions were investigated in detail (reaction time, temperature, pressure, solvent). In particular, the influence of pressure on yield and selectivity was measured in the range of $1-10$ atm. The reaction rate was low below 3 atm. Noticeable amounts of the piperidine derivative $(12-15%)$ were formed at 10 atm. Under optimized conditions (5 atm, ¹⁸-24 h reaction time) some over-reduction could not be completely avoided, leading to $3-5\%$ 5 in the crude reaction mixture after acylation. The acylated derivative, **4**, was converted into the bis-trifluoroacetate salt **6** which was isolated in good overall yield (80-85% from the aldehyde) and purity (>98%). The bis-trifluoroacetate **6** was not hygroscopic and could be stored at rt for at least 6 months. Condensation of **6** with 2,6-dichloropurine was performed in the presence of 4 molar equiv of NEt₃. The last two synthetic steps were achieved using the same conditions for the synthesis of roscovitine to afford CR8 in $72-76\%$ overall yield.
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Interestingly, the piperidine **5** had been previously incorporated in other series of active compounds.⁸ Therefore, a selective route to this compound was investigated. The monoprotected piperidine **5** was easily obtained upon adding 1 molar equiv of AcOH in the hydrogenation medium.

In summary, the synthesis of both CDK inhibitors could be performed at a 30-50 g scale without relying on timeconsuming and costly chromatography purification steps. In the particular case of CR8, the presented procedure circumvents the low solubility of biaryls.9 Biarylamines, being privileged pharmacophore structures, are part of numerous biologically active products;10 therefore, the herein reported simple procedure could find other applications.

Experimental Section

Melting points were determined on a Kofler hot-stage (Reichert) and are uncorrected. NMR spectra were recorded on Bruker Avance 400 MHz (100 MHz for 13C NMR). Chemical shifts are given in ppm downfield of tetramethylsilane (TMS) used as an internal standard. Reactions were monitored

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by TLC using Merk silica gel 60F-254 thin layer plates. Column chromatography was carried out on SDS Chromagel 60 ACC, ⁴⁰-⁶³ *^µ*m. The HPLC analyses were carried out on a system consisting of a Waters 600 system controller, a Jones chromatography heater-chiller oven and a Waters 994 photodiode array detector.

4-(2-Pyridyl)benzaldehyde Oxime (2). To a solution of **1** $(36.6 \text{ g}, 0.2 \text{ mol})$ in 50 mL of EtOH and 20 mL of H₂O were added hydroxylamine hydrochloride (15.27, 0.21 mol) and AcONa, $3H₂O$ (29.90 g, 0.21 mol). The resulting slurry was stirred for 12 h at rt. The mixture was concentrated to dryness, and the resulting solid was washed with H₂O (2×5 mL) and EtOH, $(1 \times 5 \text{ mL})$. Yield 97%, mp 138–139 °C: ¹H NMR
(CDCl): δ 7.16 (m 1H H …), 7.60 (d $I = 8.08$ Hz 2H (CDCl₃): δ 7.16 (m, 1H, H_{pyridyl}), 7.60 (d, $J = 8.08$ Hz 2H, H_{phenyl}), 7.67 (m, 2H, H_{pyridyl}), 7.91 (d, $J = 8.08$ Hz, 2H, H_{phenyl}), 8.08 (s, 1H, \underline{H} –C=N), 8.61 (d, *J* = 3.28 Hz, 1H, H_{pyridyl}), 10.70 (bs, 1H, OH). ¹³C NMR (CDCl₃): δ 119.45, 121.49, 125.60, 125.95, 128.98, 132.60, 136.02, 138.74, 147.08, 148.54.

*tert***-Butyloxycarbonyl-4-(2-pyridyl)phenylmethylamine (4).** To a solution of **2** (19.8 g, 0.1 mol) in 200 mL of EtOH was added 1 g of 10% Pd/C, and the suspension was submitted to hydrogenation at 5 atm for 18 h at rt. The catalyst was filtered on celite and washed with EtOH (2×20 mL). The solution was concentrated in vacuo, and the resulting viscous oil, **3**, was used in the next step without any purification. The crude amine, **3**, was dissolved in 150 mL of THF. The solution was cooled to 0 °C; NEt3 (17 mL, 0.11 mol) and di*tert*-butyldicarbonate, (Boc)2O (24 g, 0.11 mol) were added. The resulting solution was stirred at rt for 4 h. The mixture was concentrated in vacuo and triturated with 30 mL of $Et₂O$. The resulting solid was washed with Et₂O (2 \times 10 mL). Yield: 89%, mp 87-89 °C. ¹H NMR (CDCl₃): δ 1.47(s, 9H, C(CH₃)₃), 4.37(d, 2H *J* = 5.3 Hz, CH2), 4.93 (br, 1H, NH), 7.22 (m, 1H, Hpyridyl), 7.38 (d, *J* $= 8.08$ Hz, 2H, H_{phenyl}), 7.73 (m, 2H, H_{pyridyl}), 7.95 (d, 2H, H_{phenyl}), 8.68 (d, $J = 4.08$ Hz, 1H, H_{pyridyl}). ¹³C NMR (CDCl₃): *δ* 28.25, 44.15, 78.86, 120.05, 121.95, 126.87, 127.38, 136.59, 138.29, 139.65, 149.53, 155.49, 156.69. Anal. (C₁₇H₂₀N₂O₂) Calc: C, 71.81; H, 7.09; N, 9.85; found: C, 71.45; H, 6.98; N, 9.66.

*tert***-Butyloxycarbonyl-4-(2-piperidyl)benzylamine (5).** In the reduction step 1 molar equiv of $CH₃COOH$ was added. Otherwise, the procedure was accomplished according to the method described in the preceding section. Compound **4** did not crystallize and was purified by column chromatography using $CH_2Cl_2/MeOH/NEt_3$ (98:2:0.1) as eluent. Yield 65%, oil. ¹H NMR (CDCl₃): δ 1.44 (s, 9H, *tert*-Bu), 1.43–1.93 (m, 6H,
H₁ (b) 2.78 (td, 1H, *I* = 11.6, 3.1 Hz, H₂(b) 3.19 (d, 1H H_{piperidyl}), 2.78 (td, 1H, *J* = 11.6, 3.1 Hz, H_{piperidyl}), 3.19 (d, 1H, $J = 11.6$ Hz, H_{piperidyl}), 3.58 (dd, 1H, $J = 10.4$, 2.4 Hz, H_{piperidyl}), 4.24 (d, 2H, $J = 5.8$ Hz, CH₂), 4.81 (br, 1H, NH), 7.19 (d, $J =$ 7.5 Hz, 2H, H_{phenyl}), 7.38 (d, 2H, H_{phenyl}); ¹³C NMR (CDCl₃): *δ* 5.2, 25.7, 28.33, 30.22, 34.8, 44.3, 47.65, 61.92, 126.85, 127.41, 137.58, 144.49, 155.83. Anal. $(C_{17}H_{26}N_2O_2)$ Calc: C, 70.31; H, 9.02; N, 9.65; found: C, 70.39; H, 9.24; N, 9.48.

4-(2-Pyridyl)benzylamine Ditrifluoroacetate (6). To a cooled (0 °C) solution of **3** (18 g, 0.063 mol) in 85 mL of CH_2Cl_2 was added 85 mL of anhydrous CF_3COOH . The cooling bath was removed, and the mixture was stirred at rt for 2 h. The mixture was evaporated to dryness, and the residue was first triturated with 20 mL of cyclohexane. The cyclohexane was decanted, and 20 mL of $Et₂O$ was added. The salt crystallized and was washed with 10 mL of Et₂O. Yield 87%, mp 165–168 °C. ¹H NMR (CDCl₃): *δ* 4.11(s, 2H, CH₂), 4.99
(br 3H NH) 7.23 (t 1H H _{ini}), 7.38 (m 2H H ini), 7.73 (br, 3H, NH), 7.23 (t, 1H, H_{pyridyl}), 7.38 (m, 2H, H_{phenyl}), 7.73 (dt, 2H, H_{pyridyl}), 7.94 (d, $J = 7.83$ Hz, 2H H_{phenyl}), 8.61 (d, $J =$ 4.08 Hz, 1H, $H_{pyridyl}$), 8.7 (br s, 3H H₃N). ¹³C NMR (DMSO*d*₆): *δ* 41.28, 115.42 (q, *J_{C-F}* = 293 Hz), 120.16, 122.41, 126.49, 128.59, 134.41, 137.40, 137.43, 148.29, 154.29, 157.77 (q, J_{C-C-F} = 35 Hz). Anal. (C₁₆H₁₄N₂O₄F₆) Calc: C, 46.61; H, 3.42; N, 6.79; found: 46.84; H, 3.53; N, 6.43.

Synthesis of 6-Arylmethylamino-2-chloropurines. To a solution of **1** (23 g, 0.1 mol) in 200 mL of *n*-BuOH were added the primary amine (0.12 mol) and NEt₃ (synthesis of **8a**: 22.5 mL, 0.16 mol; synthesis of **8b**: 70 mL, 0.50 mol). After 3 h heating at 110 °C, the mixture was cooled to 20 °C, and the solid was filtrated, washed with 5 mL of cold *n*-BuOH, and dried in vacuo.

2-Chloro-6-phenylmethylaminopurine (8a): yield 95%; mp > 250 °C. ¹H NMR (DMSO-*d*₆): δ 4.80 (s, 2H, CH₂), 7.45-7.70(m, 5H, H_{phenyl}), 7.91(s, 1H, H-8). (Solubility in $DMSO_{d6}$ or in CD₃OD was to low for recording a ¹³C NMR spectrum). Anal. (C₁₂H₁₀N₅Cl) Calc: C, 55.50; H, 3.88; N, 26.97; found: C, 55.81; H, 3.73; N, 26.41.

2-Chloro-6-[4-(2-pyridyl)phenylmethylamino]purine (8b): yield 93%; mp > 250 °C. ¹ H NMR (DMSO-*d*6): *δ* 4.80 (s, 2H, CH₂), 7.20 (m, 1H, H_{pyridyl}), 7.45 (d, 2H, $J = 8$ Hz, H_{phenyl}), 7.72 (m, 2H, Hpyridyl), 7.95 (m, 3H, Hphenyl and H-8), 8.54 (d, 1H, $J = 4.8$ Hz, H_{pyridyl}). Anal. (C₁₇H₁₃N₆Cl) Calc: C, 60.63; H, 3.89; N, 24.95; found: C, 60.80; H, 3.98; N, 24.61.

Alkylation of Chloropurines 9. To a solution of **8** (0.08 mol) in DMSO (100 mL) at $18-20$ °C were added K_2CO_3 (35.52 g, 0.24 mol) and 2-bromopropane (19 mL, 0.2 mol). After 5 h stirring at $18-20$ °C, 2-bromopropane (4.7 mL, 0.05) mL) was added, and the stirring continued at the same temperature for 5 h. After addition of 200 mL of cold $(5 \degree C)$ H₂O, the mixture was extracted with EtOAc $(3 \times 100 \text{ mL})$, and the combined organic layers were washed with brine (3×50) mL), dried, and dried over Na2SO4. Derivatives **9**, which crystallized upon concentration, were triturated once with 5 mL of 2-propanol and were collected by filtration.

2-Chloro-6-phenylmethylamino-9-iso-propylpurine (9a): yield 85%, mp 180–182 °C. ¹H NMR (DMSO): δ 1.58 (d, 6H $I = 6.8$ Hz CH(CH))) 4.90 (bapt 1H CH(CH))) 6H, $J = 6.8$ Hz, CH(CH₃)₂), 4.90 (hept, 1H, CH(CH₃)₂), 6-7.25-7.50 (m, 5H, H_{phenyl}), 8.16 (s, 1H, H-8). ¹³C NMR (DMSO): *δ* 21.81, 43.19, 46.04, 117.98, 126.44, 126.85, 127.67, 137.03, 137.97, 148.93, 153.03, 154.45. Anal. (C₁₅H₁₆N₅Cl) Calc: C, 59.70; H, 5.34; N, 23.21; found: C, 59.60; H, 5.22; N, 23.36.

2-Chloro-6-[4-(2-pyridyl)phenylmethylamino]-9-iso-propylpurine (9b): yield 86%, mp 187–189 °C. ¹H NMR
(CDCL): δ 1.58 (d 6H $I = 6.8$ Hz CH(CH))) 4.79 (hent (CDCl₃): δ 1.58 (d, 6H, $J = 6.8$ Hz, CH(CH_3)₂), 4.79 (hept, 1H, C*H*(CH3)2), 4.85 (bs, 2H, NHC*H*2), 6.59 (bs, 1H, N*H*CH2), 7.20-7.23 (m, 1H, H_{pyridyl}), 7.49 (d, 2H, $J = 8$ Hz, H_{phenyl}), 7.73-7.71 (m, 2H, Hpyridyl), 7.79 (s, 1H, H-8), 7.98 (d, 2H, H_{phenyl}), 8.71 (d, 1H, $J = 4.8$ Hz, H_{pyridyl}). 22.52, 44.07, 46.78, 118.72, 120.25, 122.08, 127.01, 128.04, 136.66, 137.67, 138.54, 139.02, 149.57, 154.15, 155.15, 156.67. Anal. (C₂₀H₁₉N₆Cl) Calc: C, 63.41; H, 5.05; N, 22.18; found: C, 63.27; H, 5.20; N, 22.35.

Amination by Nucleophilic Substitution of the 2-Chlorine (10). A mixture of chloropurines, **9** (0.138 mol), and (*R*)-2 aminobutan-1-ol (104.10 mL, 1.11 mol) was heated under N_2 at 160 °C for 8 h. After cooling, $H₂O$ (100 mL) was added, and the mixture was extracted with EtOAc $(4 \times 100 \text{ mL})$. The organic layer was washed with warm (50 °C) H₂O (2 \times 50 mL), dried, and evaporated until dryness. Crystallization occurred after addition of 10 mL of EtOAc. The crystals of trisubstituted purines **10** were collected by filtration.

(*R***)-2-(1-Hydroxybut-2-ylamino)-6-(benzylaminopurine)- 9-isopropylpurine. Roscovitine (10a):** yield 89%, mp 108-¹¹⁰ ^oC. ¹H NMR (CDCl₃): δ 0.96 (m, 3H), 1.44 (d, 6H, CH(CH₃)₂). 1.53 (m, 2H, C*H*2CH3), 3.57(m, C*H*NH), 3.75(m, 1H, HOC*H*), 3.84 (m,1H, HOC*H*), 4.52 (hept, $J = 6.5$ Hz, 1H, C*H*(CH₃)₂), 4.71 (bs, 2H, CH₂), 4.91 (bs, 1H), 7.28–7.30 (m, 6H).¹³C NMR (CDCl3): *δ* 10.84, 22.44, 22.50, 24.90, 44.24, 56.13, 68.06, 114.55, 127.16, 127.58, 128.45, 134.43, 138.90, 154.84, 159.97. Anal. ($C_{20}H_{20}N_6O$) Calc: C, 61.52; H, 6.45; N, 26.90; found: C, 61.64; H, 6.25; N, 26.64.

(*R***)-2-(1-Hydroxybut-2-ylamino)-6-[4-(2-pyridyl)phenylmethylamino]-9-iso-propylpurine. CR8 (10b):** yield 92%, mp 89–93 °C. ¹H NMR (CDCl₃): δ 0.95 (t, 3H, *J* = 7.2 Hz, *CHCH*₀) 1.47 (d 6H *J* = 6.8 Hz *CH(CH*₀)) 1.50–1.60 (m CH_3CH_2), 1.47 (d, 6H, $J = 6.8$ Hz, CH(CH₃)₂), 1.50-1.60 (m, 2H, CH₃CH₂), 3.56 (dd, 1H, $J = 10.8$ and $J' = 2.8$ Hz, C*H*₂OH), 3.75 (dd, 1H, $J = 10.8$ Hz, and $J' = 2.4$, C*H*₂OH), 3.79-3.85 (m, 1H, C*H*NH), 4.54 (hept, 1H, C*H*(CH3)2), 4.76 (m, 2H, NHC*H*2), 4.90 (bs, 1H, CHN*H*), 6.10 (bs, 1H, N*H*CH2), 7.16 (td, 1H, $J = 5.4$ and $J' = 1.6$ Hz, H_{pyridyl}), 7.41 (d, 2H, *J* $= 8.4$ Hz, H_{phenyl}), 7.45 (s, 1H, 8-H), 7.63-7.70 (m, 2H, H_{pyridyl}), 7.89 (d, 2H, $J = 8.4$ Hz, H_{phenyl}), 8.62 (d, 1H, $J = 4.8$ Hz, Hpyridyl). 13C NMR (CDCl3): *δ* 10.5, 22.3, 22.4, 24.1, 38.0, 45.9, 55.0, 66.8, 113.6, 120.0, 120.9, 125.5, 126.8 (×2), 127.7, 130.0, 134.1, 135.9, 137.7, 140.0, 149.1, 154.1, 156.8, 159.1. Anal. $(C_{24}H_{29}N_7O)$ Calc: C, 66.80; H, 6.77; N, 22.72; found: C, 66.43; H, 6.65; N, 22.51.

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Supporting Information Available

NMR spectra of 4-(2-pyridyl)benzylamine ditrifluoroacetate (**6**), roscovitine (**10a**), and CR8 (**10b**). This material is available free of charge via the Internet at http://pubs.acs.org.

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