

(d, 3 H, CH₃, $J = 1.3$ Hz), 4.02 (q, 2 H, CH₂), 6.9–7.3 (m, 5 H, C₆H₄ and CH=); exact mass calcd for C₁₁H₁₃O₂B 204.0958, found 204.0930. The compound was directly compared with an authentic sample¹⁷ we prepared previously.

All these compounds are rather sensitive to air and moisture. For oils or low melting solids, correct elemental analyses could

not be obtained; however, spectroscopically pure compounds can be readily obtained by repeated Kugelrohr distillation.

Supplementary Material Available: Figures of ¹H NMR (400 MHz) spectra of β -(alkylthio)alkenyl-1,3,2-benzodioxaboroles in Table II (7 pages). Ordering information is given on any current masthead page.

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(17) Satoh, M.; Miyaura, N.; Suzuki, A. *Synthesis* 1987, 373–377.

Additions and Corrections

1992, Volume 11

Raymond N. Vrtis, Simon G. Bott, and Stephen J. Lippard*: Linear CO-Bridged Dinuclear Tantalum Siloxycarbyne Complexes.

Page 270. The unit cell parameters for [(dmpe)₂(CO)-TaCOZrCp*₂Cl] (4) should be changed to $a = 9.687$ (2) Å, $b = 12.288$ (2) Å, $c = 17.851$ (2) Å, $\alpha = 80.46$ (1)°, $\beta = 89.33$ (1)°, $\gamma = 78.46$ (1)°, and $V = 2053$ (1) Å³. This cell was used for solving and refining the structure, whereas the one reported is an equivalent, more conventional choice. Bond distances, angles, positional parameters, and discussion contained within the paper are unaffected by this substitution. In Table V, the phosphorus atoms numbered as P(1), P(2), P(3), and P(4) should be changed to P(3), P(4), P(2), and P(1), respectively.

Supplementary Material Available: Tables of anisotropic thermal parameters for the non-hydrogen atoms and positional and isotropic thermal parameters for the hydrogen atoms of 4 (6 pages); a table of observed and calculated structure factors (40 pages). Ordering information is given on any current masthead page.