

Photochemical Reactions of Polysilacycloalka-1,2-dienes: Synthesis of Polysila 1,3-Bridged Cyclopropenes

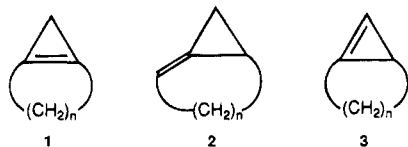
Fusao Hojo, Toshio Shimizu, and Wataru Ando*

Department of Chemistry, University of Tsukuba, Tsukuba, Ibaraki 305, Japan

Received May 26, 1994[®]

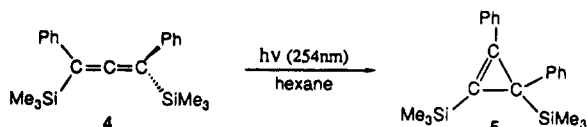
Summary: The photochemical reaction of the hexasilacyclonona-1,2-diene **6** provides the hexasilabicyclo[6.1.0]non-1(9)-ene **7** together with the corresponding ring-contracted derivatives pentasilacycloocta-1,2-diene **8** and pentasilabicyclo[5.1.0]oct-1(8)-ene **9**. Similarly, the photolysis of the tetrasilacyclohepta-1,2-diene **10** yields the tetrasilabicyclo[4.1.0]hept-1(7)-ene **11**; however, no ring-contracted compounds are observed in this case.

In recent years the chemistry of strained unsaturated cyclic compounds has been focused on molecules exhibiting unusual structures and reactivities;¹ e.g., the isolation of strained polysilacycloalkynes² and polysilacyclic allenes³ has been reported. Bicyclo[*n*.1.0]alkenes **1–3**, as representatives of highly strained olefins, are



of considerable interest.⁴ In the series of 1,2-bridged cyclopropenes (**1**), bicyclo[4.1.0]heptene derivatives ($n = 4$) are the smallest representatives which have been isolated so far.^{4,5} Previously we observed that such bicyclic systems can be effectively stabilized by introduction of suitable heteroatoms which enable us to isolate several heteroatom-substituted bicyclo[4.1.0]hept-1(6)-enes.^{2c,6} Bicyclo[4.1.0]hept-1(2)-ene derivatives (**2**; $n = 4$)^{4,7} and 1,3-bridged cyclopropene derivatives (**3**) such as bicyclo[6.1.0]nonene ($n = 6$) could be

Scheme 1



isolated.⁴ Nevertheless, up to now the isolation of smaller ring system has not been reported. Herewith, we wish to report the synthesis of polysila 1,3-bridged cyclopropenes by photochemical transformations of the corresponding polysilacyclic allenes.

As an example of an acyclic silyl-substituted allene, the conversion of 1,3-diphenyl-1,3-bis(trimethylsilyl)propadiene (**4**) was studied. Irradiation of a solution of **4** in hexane with a low-pressure mercury lamp afforded 1,3-diphenyl-2,3-bis(trimethylsilyl)cyclopropene (**5**)⁹ in 91% yield. Similarly, after photolysis of the hexasilacyclonona-1,2-diene **6**^{3a} under identical conditions the corresponding hexasilabicyclo[6.1.0]non-1(9)-ene **7**¹⁰ was obtained in 59% yield; additionally, the ring-contracted allene **8**¹¹ and pentasilabicyclo[5.1.0]oct-1(8)-ene **9**¹² were isolated in yields of 14% and 23%, respectively. On the other hand, irradiation of the tetrasilacyclohepta-1,2-diene **10**^{3a} gave the tetrasilabi-

(8) Compound **4**: mp 55–56 °C; colorless crystals. Anal. Calcd for $C_{21}H_{28}Si_2$: C, 74.91; H, 8.38. Found: C, 74.51; H, 8.16. HRMS (EI, 20 eV): *m/e* calcd for $C_{21}H_{28}Si_2$ 336.1730, found 336.1725. LRMS (EI, 70 eV): *m/e* 336 (M^+), 263, 248 (100%). ¹H NMR (C_6D_6 , 90 MHz): δ 0.28 (s, 18H), 6.9–7.8 (m, 10H). ¹³C NMR (C_6D_6 , 22.5 MHz): δ 0.02 (q), 95.1 (s), 125.9 (d), 127.6 (d), 128.6 (d), 137.2 (s), 209.5 (s). ²⁹Si NMR (C_6D_6 , 17.6 MHz): δ -4.1. IR (KBr): 1889 (C=C), 1595, 1493, 1251, 917 cm^{-1} . UV (hexane): λ_{max} (ϵ) 259 nm (13 000).

(9) Compound **5**: colorless oil. Anal. Calcd for $C_{21}H_{28}Si_2$: C, 74.91; H, 8.38. Found: C, 76.62; H, 8.04. HRMS (EI, 20 eV): *m/e* calcd for $C_{21}H_{28}Si_2$ 336.1730, found 336.1743. LRMS (EI, 70 eV): *m/e* 336 (M^+), 263, 248 (100%). ¹H NMR (C_6D_6 , 90 MHz): δ 0.21 (s, 9H), 0.23 (s, 9H), 7.0–7.8 (m, 10H). ¹³C NMR (C_6D_6 , 100 MHz): δ -0.6 (q), -0.5 (q), 24.6 (s), 118.7 (s), 125.5 (d), 126.8 (d), 128.4 (d), 128.8 (d), 129.5 (d), 130.1 (d), 130.9 (s), 135.2 (s), 149.8 (s). ²⁹Si NMR (C_6D_6 , 17.6 MHz): δ -10.5, 0.9. IR (neat): 1596, 1493, 1250 cm^{-1} .

(10) Compound **7**: colorless oil. Anal. Calcd for $C_{27}H_{40}Si_6$: C, 60.15; H, 8.60. Found: C, 59.99; H, 8.42. HRMS (EI, 20 eV): *m/e* calcd for $C_{27}H_{40}Si_6$ 538.2215, found 538.2201. LRMS (EI, 70 eV): *m/e* 538 (M^+), 465, 392 (100%). ¹H NMR (C_6D_6 , 400 MHz): δ 0.14 (s, 3H), 0.261 (s, 3H), 0.264 (s, 6H), 0.29 (s, 6H), 0.31 (s, 3H), 0.32 (s, 3H), 0.35 (s, 3H), 0.43 (s, 3H), 0.58 (s, 3H), 0.70 (s, 3H), 7.111 (t, $J = 7.2$ Hz, 1H), 7.116 (t, $J = 7.2$ Hz, 1H), 7.220 (t, $J = 7.2$ Hz, 2H), 7.223 (t, $J = 7.2$ Hz, 2H), 7.47 (d, $J = 7.2$ Hz, 2H), 7.74 (d, $J = 7.2$ Hz, 2H). ¹³C NMR (C_6D_6 , 100 MHz): δ -5.01 (q), -4.95 (q), -4.8 (q), -4.7 (2 × q), -4.6 (q), -4.3 (q), -3.8 (q), -2.1 (q), -0.7 (q), -0.3 (q), 0.8 (q), 24.6 (s), 119.3 (s), 125.7 (d), 128.3 (d), 128.5 (d), 128.6 (d), 129.5 (d), 130.2 (d), 130.7 (s), 136.6 (s), 149.8 (s). ²⁹Si NMR (C_6D_6 , 79 MHz): δ -45.2, -41.8, -41.2, -38.2, -25.8, -8.5. IR (neat): 1597, 1489, 1446, 1402, 1249 cm^{-1} . UV (cyclohexane): λ_{max} (ϵ) 301 (12 300), 323 (12 800).

(11) Compound **8**: colorless crystals; mp 156–157 °C. Anal. Calcd for $C_{25}H_{40}Si_5$: C, 62.42; H, 8.38. Found: C, 62.09; H, 8.16. HRMS (EI, 20 eV): *m/e* calcd for $C_{25}H_{40}Si_5$ 480.1976, found 480.1970. LRMS (EI, 70 eV): *m/e* 480 (M^+), 407. ¹H NMR (C_6D_6 , 400 MHz): δ 0.24 (s, 6H), 0.30 (s, 6H), 0.33 (s, 6H), 0.47 (s, 6H), 0.54 (s, 6H), 7.10 (t, $J = 7.8$ Hz, 2H), 7.25 (t, $J = 7.8$ Hz, 4H), 7.59 (t, $J = 7.8$ Hz, 4H). ¹³C NMR (C_6D_6 , 100 MHz): δ -5.3 (q), -4.93 (q), -4.88 (q), -1.5 (q), -0.6 (q), 95.2 (s), 126.9 (d), 128.5 (d), 129.3 (d), 137.8 (s), 210.4 (s). ²⁹Si NMR (C_6D_6 , 79 MHz): δ -42.5, -42.3, -16.0. IR (KBr): 1888 (C=C), 1596, 1490, 1247 cm^{-1} . UV (hexane): λ_{max} (ϵ) 265 nm (22 900).

* Abstract published in *Advance ACS Abstracts*, August 1, 1994.

(1) (a) Special issue on strained organic compounds: *Chem. Rev.* **1989**, *89*. (b) Greenberg, A.; Liebman, J. F. *Strained Organic Compounds*; Academic Press: New York, 1978.

(2) (a) Ando, W.; Nakayama, N.; Kabe, Y.; Shimizu, T. *Tetrahedron Lett.* **1990**, *31*, 3597. (b) Ando, W.; Hojo, F.; Sekigawa, S.; Nakayama, N.; Shimizu, T. *Organometallics* **1992**, *11*, 1009. (c) Hojo, F.; Sekigawa, S.; Nakayama, N.; Shimizu, T.; Ando, W. *Organometallics* **1993**, *12*, 803. (d) Pang, Y.; Schneider, A.; Barton, T. J.; Gordon, M. S.; Carroll, M. T. *J. Am. Chem. Soc.* **1992**, *114*, 4920.

(3) (a) Shimizu, T.; Hojo, F.; Ando, W. *J. Am. Chem. Soc.* **1993**, *115*, 3111. (b) Hojo, F.; Shimizu, T.; Ando, W. *Chem. Lett.* **1993**, 1171. (c) Petrich, S. A.; Pang, Y.; Young, V. G., Jr.; Barton, T. J. *J. Am. Chem. Soc.* **1993**, *115*, 1591. (d) Pang, Y.; Petrich, S. A.; Young, V. G., Jr.; Gordon, M. S.; Barton, T. J. *J. Am. Chem. Soc.* **1993**, *115*, 2534. (e) Lin, J.; Pang, Y.; Young, V. G., Jr.; Barton, T. J. *J. Am. Chem. Soc.* **1993**, *115*, 3794.

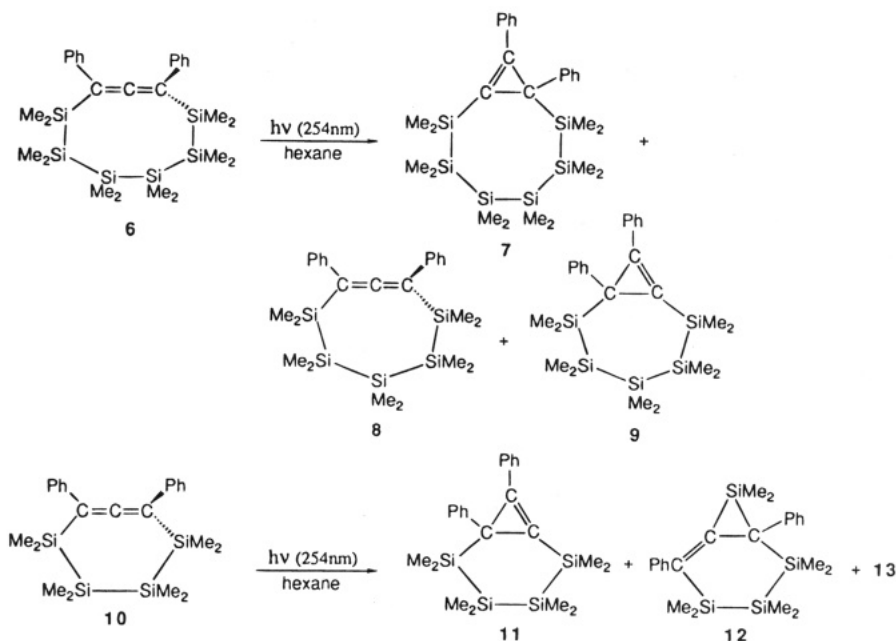
(4) (a) Billups, W. E.; Haley, M. M.; Lee, G.-A. *Chem. Rev.* **1989**, *89*, 1147. (b) Baird, M. S. Functionalized Cyclopropenes as Synthetic Intermediates. In *Topics in Current Chemistry*; de Meijere, A., Ed.; Springer-Verlag: Berlin, 1988. (c) Halton, B.; Banwell, M. G. Cyclopropenes. In *The Chemistry of the Cyclopropyl Group*; Rappoport, Z., Ed.; Wiley: New York, 1987. (d) Köbrich, G. *Angew. Chem.* **1973**, *85*, 494; *Angew. Chem., Int. Ed. Engl.* **1973**, *12*, 464.

(5) Closs, G. L.; Böll, W. A.; Heyn, H.; Dev, V. J. *Am. Chem. Soc.* **1968**, *90*, 173 and references cited therein.

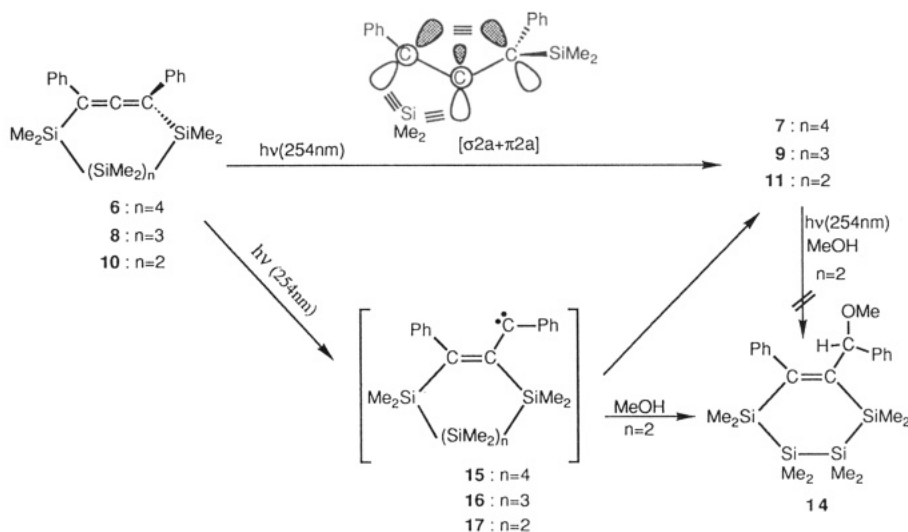
(6) (a) Ando, W.; Hanyu, Y.; Takata, T.; Ueno, K. *J. Am. Chem. Soc.* **1982**, *104*, 4981. (b) Ando, W.; Hanyu, Y.; Takata, T.; Ueno, K. *J. Am. Chem. Soc.* **1984**, *106*, 2216.

(7) Baumann, M.; Köbrich, G. *Tetrahedron Lett.* **1974**, 1217.

Scheme 2



Scheme 3



cyclo[4.1.0]hept-1(7)-ene **11**¹³ in 45% yield and the tetrasilabicyclo[4.1.0]hept-1(2)-ene **12**¹⁴ in 20% yield.¹⁵

(12) Compound **9**: colorless oil. Anal. Calcd for $C_{25}H_{40}Si_5$: C, 62.42; H, 8.38. Found: C, 62.20; H, 8.14. HRMS (EI, 20 eV): m/e calcd for $C_{25}H_{40}Si_5$ 480.1976, found 480.2007. LRMS (EI, 70 eV): m/e 480 (M^+), 407, 392 (100%). 1H NMR (C_6D_6 , 400 MHz): δ 0.12 (s, 3H), 0.277 (s, 3H), 0.282 (s, 3H), 0.285 (s, 3H), 0.31 (s, 3H), 0.33 (s, 3H), 0.34 (s, 3H), 0.42 (s, 3H), 0.60 (s, 3H), 0.63 (s, 3H), 7.10 (t, $J = 8.2$ Hz, 1H), 7.11 (t, $J = 8.2$ Hz, 1H), 7.22 (t, $J = 8.2$ Hz, 2H), 7.26 (t, $J = 8.2$ Hz, 2H), 7.51 (d, $J = 8.2$ Hz, 2H), 7.75 (d, $J = 8.2$ Hz, 2H). ^{13}C NMR (C_6D_6 , 100 MHz): δ -6.0 (q), -5.9 (q), -5.3 (q), -4.9 (q), -4.8 (q), -4.7 (q), -1.9 (q), -1.7 (q), -0.7 (q), -0.4 (q), 24.6 (s), 118.4 (s), 125.6 (d), 128.3 (d), 128.9 (d), 129.4 (d), 129.5 (d), 130.0 (d), 130.6 (s), 136.3 (s), 149.8 (s). ^{29}Si NMR (C_6D_6 , 79 MHz): δ -44.5, -42.9, -38.0, -25.5, -7.9. IR (neat): 1490, 1445, 1447 cm^{-1} . UV (hexane): λ_{max} (ϵ) 303 (13 000), 318 nm (12 700).

(13) Compound **11**: colorless crystals; mp 59–61 °C. Anal. Calcd for $C_{23}H_{34}Si_4$: C, 65.32; H, 8.11. Found: C, 65.53; H, 8.14. HRMS (EI, 20 eV): m/e calcd for $C_{23}H_{34}Si_4$ 422.1738, found 422.1740. LRMS (EI, 70 eV): m/e 422 (M^+). 1H NMR (C_6D_6 , 400 MHz): δ 0.10 (s, 3H), 0.27 (s, 3H), 0.30 (s, 3H), 0.31 (s, 3H), 0.36 (s, 3H), 0.38 (s, 3H), 0.48 (s, 3H), 0.58 (s, 3H), 7.0–7.8 (m, 10H). ^{13}C NMR (C_6D_6 , 100 MHz): δ -6.0 (q), -5.9 (q), -5.6 (q), -5.5 (q), -2.2 (q), -1.9 (q), -0.9 (q), -0.8 (q), 24.9 (s), 119.1 (s), 125.5 (d), 126.8 (d), 128.5 (d), 128.7 (d), 129.4 (d), 129.8 (d), 131.2 (s), 136.3 (s), 149.8 (s). ^{29}Si NMR (C_6D_6 , 79 MHz): δ -46.5, -44.9, -26.6, -11.1. IR (KBr): 1491, 1447 cm^{-1} . UV (hexane): λ_{max} (ϵ) 310 nm (12 000).

Stierman and Johnson et al. studied several photorearrangements of carbocyclic allenes giving the bicyclic [6.1.0] systems.¹⁶ In the photoreactions of **4**, **6**, and **10**, the pathway is considered to be via a vinylcarbene intermediate or to be a concerted process. To clarify the reaction mechanism, a methanol solution of the

(14) Compound **12**: colorless oil. Anal. Calcd for $C_{23}H_{34}Si_4$: C, 65.32; H, 8.11. Found: C, 65.22; H, 8.18. HRMS (EI, 20 eV): m/e calcd for $C_{23}H_{34}Si_4$ 422.1738, found 422.1731. LRMS (EI, 70 eV): m/e 422 (M^+). 1H NMR (C_6D_6 , 400 MHz): δ -0.14 (s, 3H), -0.07 (s, 3H), 0.09 (s, 3H), 0.13 (s, 3H), 0.30 (s, 3H), 0.45 (s, 3H), 0.56 (s, 3H), 0.72 (s, 3H), 6.9–7.4 (m, 10H). ^{13}C NMR (C_6D_6 , 100 MHz): δ -7.6 (q), -5.9 (q), -2.2 (q), -1.2 (2 \times q), 0.2 (q), 0.7 (q), 1.4 (q), 45.3 (s), 124.2 (d), 127.3 (d), 127.9 (d), 128.56 (d), 128.63 (d), 128.8 (d), 141.5 (s), 147.8 (s), 168.6 (s), 170.1 (s). ^{29}Si NMR (C_6D_6 , 17.6 MHz): δ -46.2, -24.1, -8.6, 12.4. IR (neat): 1597, 1493, 1479, 1247, 1180, 1075 cm^{-1} .

(15) In this reaction the isomer **13** was observed by GCMS (8% GC yield). Although attempts to isolate compound **13** failed, it is most likely another isomeric product, as indicated by the mass spectrum. (16) (a) Stierman, T. J.; Johnson, R. P. *J. Am. Chem. Soc.* **1983**, *105*, 2492. (b) Stierman, T. J.; Johnson, R. P. *J. Am. Chem. Soc.* **1985**, *107*, 3971. (c) Price, J. D.; Johnson, R. P. *J. Am. Chem. Soc.* **1985**, *107*, 2187. (d) Stierman, T. J.; Shakespeare, W. C.; Johnson, R. P. *J. Org. Chem.* **1990**, *55*, 1043. (e) Price, D. P.; Johnson, R. P. *J. Org. Chem.* **1991**, *56*, 6372.

tetrasilacyclohexa-1,2-diene **10** was irradiated with a low-pressure mercury lamp; the methanol-incorporated product **14**¹⁷ was obtained in 54% yield, together with the bicyclo[4.1.0]heptene derivative **11** in 8% yield. The formation of methanol adduct **14** can be explained by the methanol trapping of the vinylcarbene **17**, which was generated by a 1,2-silyl shift. When bicyclo[4.1.0]heptene derivative **11** was irradiated in methanol, the methanol-incorporated product **14** was not observed. On the other hand, when the hexasilacyclonona-1,2-diene **6** was irradiated in methanol, **7–9** were obtained in 81, 6, and 9% yields, respectively, and solvent dependency was not observed in hexane and methanol. Nevertheless, no methanol-incorporated product was found by ¹H NMR and GCMS. It is anticipated that there could be

(17) Compound **14**: colorless oil. HRMS (EI, 20 eV): *m/e* calcd for C₂₄H₃₈Si₄O 454.2000, found 454.2019. LRMS (EI, 70 eV): *m/e* 454 (M⁺), 439, 335, 319, 135. ¹H NMR (C₆D₆, 90 MHz): δ -0.03 (s, 3H), 0.04 (s, 3H), 0.15 (s, 3H), 0.29 (s, 3H), 0.336 (s, 3H), 0.342 (s, 3H), 0.39 (s, 3H), 0.69 (s, 3H), 3.28 (s, 3H), 4.03 (s, 1H), 6.9–7.4 (m, 6H), 7.6–7.8 (m, 4H). ¹³C NMR (CDCl₃, 22.5 MHz): δ -8.9 (q), -8.6 (q), -3.0 (q), -2.1 (q), -1.9 (q), -1.6 (q), -1.2 (q), -0.8 (q), 44.8 (d), 50.1 (q), 125.0 (d), 125.3 (d), 127.4 (d), 128.0 (2 × d), 130.6 (d), 141.8 (s), 144.6 (s), 159.5 (s), 160.2 (s). ²⁹Si NMR (C₆D₆, 17.6 MHz): δ -54.1, -15.9, -12.0, 13.6. IR (neat): 2370, 1597, 1251, 1085 cm⁻¹. UV (hexane): λ_{max} (ε) 225 (24 300), 264 (3700), 272 nm (3000).

two reasons for being unable to trap the vinylcarbene intermediate. One is that the mechanism might be concerted; another may be due to the very short lifetime of the corresponding vinylcarbene intermediates **15** and **16**.¹⁸

Acknowledgment. This work was supported by Research Fellowships of the Japan Society for the Promotion of Science for Young Scientists and Grant-in-Aid for Scientific Research from the Ministry of Education, Science, and Culture, Japan. We thank Shin-Etsu Chemical Co. Ltd. for a gift of organosilicon reagents.

Supplementary Material Available: Text giving detailed information on the reaction conditions and product isolation (3 pages). Ordering information is given on any current masthead page.

OM940405Q

(18) The different lifetimes of vinylcarbene intermediates **15–17** are due to easy rearrangements to polysilabicyclo[*n*.1.0]alkenes **7**, **9**, and **11**. Since the tetrasilabicyclo[4.1.0]heptene derivative **11** is a very strained compound, the corresponding vinylcarbene, generated by the photolysis of **10**, has a sufficiently long lifetime to react with methanol. However, since **7** and **9** are less strained than **11**, the corresponding vinylcarbenes of **15** and **16** do not furnish methanol addition products, as the lifetimes of **15** and **16** are too short for reaction with methanol.