

# Additions and Corrections

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1995, Volume 14

**Daniel E. Barber, Kurethara S. Bose, Michal Sabat, Ekkehard Sinn, and Bruce A. Averill\*:** Synthesis, Structure, and Characterization of the  $[\text{Fe}_5\text{S}_4(\text{CO})_{12}]^{2-}$  Ions and Studies on the Oxidative Conversion of the Dianion to  $[\text{Fe}_6\text{S}_6(\text{CO})_{12}]^{2-}$ .

Page 3229. During the editing and revision process, the name of one of the authors was inadvertently omitted. The correct authorship is as given above.

1996, Volume 15

**Donald W. Carpenetti, Lioba Kloppenburg, Justin T. Kupec, and Jeffrey L. Petersen\*:** Application of Amine Elimination for the Efficient Preparation of Electrophilic *ansa*-Monocyclopentadienyl Group 4 Complexes Containing an Appended Amido Functionality. Structural Characterization of  $[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{ZrCl}_2(\text{NMe}_2\text{H})]$ .

Page 1573. The NMR data reported in ref 13 for  $\{[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{ZrCl}(\mu\text{-Cl})]_2\}$  are for a different compound. The four distinct cyclopentadienyl ring proton resonances that we initially assigned to  $\{[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{ZrCl}(\mu\text{-Cl})]_2\}$  in the  $^1\text{H}$  NMR spectrum of the product mixture led us to believe that its dimeric structure remains intact in solution. This is apparently not the case. We have since prepared  $\{[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{ZrCl}(\mu\text{-Cl})]_2\}$  by the reaction of 2 equiv of  $\text{SiMe}_3\text{Cl}$  with  $[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{Zr}(\text{NMe}_2)_2]$  and have verified its dimeric structure in the solid state by X-ray crystallography. The correct solution NMR data in  $\text{CDCl}_3$  for  $\{[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{ZrCl}(\mu\text{-Cl})]_2\}$  are as follows.  $^1\text{H}$  NMR spectrum:  $\delta$  6.91, 6.44 ( $\text{C}_5\text{H}_4$ , t, 2.2), 1.38 ( $\text{NCMe}_3$ , s), 0.54 ( $\text{SiMe}_2$ , s). Gated nondecoupled  $^{13}\text{C}$  NMR spectrum:  $\delta$  122.1, 121.2 (proximal and distal carbons of  $\text{C}_5\text{H}_4$ , d, 175), 109.8 (bridgehead C, s), 57.7 ( $\text{NCMe}_3$ , s), 32.6 ( $\text{NCMe}_3$ , q, 125), 0.90 ( $\text{SiMe}_2$ , q, 120). The chemical shifts of these NMR resonances for  $\{[(\text{C}_5\text{H}_4)\text{SiMe}_2(\text{N}-t\text{-Bu})\text{ZrCl}(\mu\text{-Cl})]_2\}$  are identical with the corresponding resonances given in ref 14 for the zirconate compound, thereby indicating that both compounds afford the same zirconium species upon dissolution in  $\text{CDCl}_3$ .