

the tin reagent  $(C_5Me_5)SnBu_3$ , which is required to obtain **1** in high yield,<sup>17</sup> does increase the expense and hazard of the overall procedure. However, it should be noted that  $ClSnBu_3$ , which is used to prepare  $(C_5Me_5)SnBu_3$ ,<sup>13</sup> is regenerated in the synthesis of **1** and, thus, can be recovered and reused in subsequent preparations. Dioxygen, the source of the oxo ligand in **2**, can be used as a pure reagent or can come from ambient air. In comparison, previously reported routes to **2** (Figure 1) are nowhere near as facile. These reactions trace back to two pentamethylcyclopentadienyl-substituted vanadium starting materials that are more difficult and costly to prepare and use than **1**,  $(\eta-C_5Me_5)V(CO)_4$ <sup>6,18</sup> and  $(\eta-C_5Me_5)_2V$ .<sup>19</sup> Several oxidation and/or chlorination reactions produce **2** from these starting materials or their derivatives (Figure 1), with yields ranging from 20–85%, but none rivals the essentially quantitative conversion of **1** to **2** reported here.

In addition to being a facile route to **2**, reaction 1 is a

(17) We have not succeeded in preparing **1** in any useful yield using other  $C_5Me_5$  reagents.

(18) Hoch, M.; Duch, A.; Rehder, D. *Inorg. Chem.* **1986**, *25*, 2907–2909. Herrmann, W. A.; Kalcher, W. *Chem. Ber.* **1982**, *115*, 3886–3889.

(19) Gambarotta, S.; Floriani, C.; Chiesi-Willa, A.; Guastini, C. *Inorg. Chem.* **1984**, *23*, 1739–1747. Robbins, J. L.; Edelstein, N.; Spencer, B.; Smart, J. C. *J. Am. Chem. Soc.* **1982**, *104*, 1882–1893.

interesting example of the clean use of  $O_2$  to produce a terminal oxo compound.<sup>20</sup> Overall, a net 2-electron oxidation of vanadium(III) in the starting material occurs upon formal addition of one oxygen atom, producing the vanadium(V) oxo product. Several different mechanistic scenarios that might occur in reaction 1 can be envisioned. Interest in understanding the metal-mediated chemistry of dioxygen is widespread,<sup>21</sup> suggesting it will be worth examining this reaction in more detail.

**Acknowledgment.** We gratefully acknowledge support of this work by the National Science Foundation (CHE 93-21196). T.R. thanks the Deutsche Forschungsgemeinschaft for a postdoctoral fellowship, and C.N. thanks the Fulbright Scholar Program for partial support.

OM960625Z

(20) Thiyagarajan, B.; Kerr, M. E.; Bruno, J. W. *Inorg. Chem.* **1995**, *34*, 3444–3452 and references therein.

(21) See, for example: Studies in Surface Science and Catalysis, 66. *Dioxygen Activation and Homogeneous Catalytic Oxidation*; Simandi, L. I., Ed.; Elsevier: Amsterdam, 1991. *Oxygen Complexes and Oxygen Activation by Transition Metals*; Martell, A. E., Sawyer, D. T., Eds.; Plenum: New York, 1988. Sheldon, R. A.; Kochi, J. K. *Metal-Catalyzed Oxidations of Organic Compounds*; Academic Press: New York, 1981.

## Additions and Corrections

1996, Volume 15

**Jennifer C. Green\* and Sergey Yu. Ketkov\*:** Electronic Structures of  $\eta^7$ -Cycloheptatrienyl  $\eta^5$ -Cyclopentadienyl Derivatives of Vanadium and Tantalum As Studied by Solution-Phase and Vapor-Phase Photoabsorption Spectroscopy in the Ultraviolet and Visible Regions.

Page 4747. The title of this paper should appear as given above.

OM9700573