Synthesis, Structures, and Reactivities of Rhodium and Ruthenium Complexes with a Novel Chiral Cyclopentadienyl–Ferrocenyldiphenylphosphine Bidentate Ligand

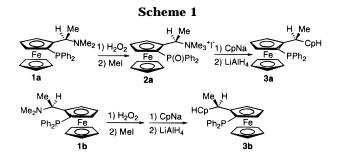
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Summary: Novel rhodium(I) and ruthenium(II) complexes with the chiral cyclopentadienyl–phosphine bidentate ligand PPFCp have been prepared and characterized crystallographically. The cationic ruthenium– vinylidene complex [$Ru(C=CHPh)(PPh_3)(PPFCp)$]PF₆, obtained from [$RuCl(PPh_3)(PPFCp)$] and phenylacetylene, reacts with 1-buten-3-ol to afford 3-methyl-1phenyl-4-penten-2-one in moderate yield with moderate enantioselectivity.

The molecular design of an appropriate chiral ligand is essential in transition-metal-catalyzed asymmetric reactions to obtain highly enantiomerically pure compounds. Consequently, many studies have currently focused on the preparation of new efficient ligands.¹ A variety of early-transition-metal-catalyzed asymmetric reactions using chiral cyclopentadienyl ligands have been reported during the last few years.² In particular, the chiral C2-symmetric ansa-metallocenes of titanium and zirconium have found notable application as catalysts for enantioselective catalytic reactions.² However, the ruthenium-catalyzed asymmetric reactions using chiral cyclopentadienyl ligands are limited, in spite of the development of unique catalytic activities of the η^{5} cyclopentadienyl (Cp) or η^5 -pentamethylcyclopentadienyl (Cp*) ruthenium complexes.^{3,4} In the Cp-containing ruthenium complexes, the cyclopentadienyl moiety is free to rotate about the cyclopentadienyl-ruthenium axis, so that a suitable chiral enviroment is not effectively made around the metal.⁵ Here we envisage the preparation of a new type of a chiral cyclopentadienyl-phosphine bidentate ligand to prevent the rotation



of the Cp ring. The preliminary results of the preparation of rhodium and ruthenium complexes containing the new chiral ligand and their application to asymmetric synthesis are described below.

Our investigations of the synthesis of optically active cyclopentadienyl-phosphine bidentate ligands started with the utilization of chiral ferrocenes with planar chirality because the latter ligands are of increasing importance in stoichiometric and catalytic asymmetric synthesis using transition-metal complexes.^{6,7} The optically active chiral ligand (S,S)-PPFCpH (3a) was prepared from the corresponding quaternary ammonium salt 2a by the introduction of a cyclopentadienyl moiety into the pseudo benzylic position of ferrocene and the following reduction of the phosphine oxide by LiAlH₄ (Scheme 1). The nucleophilic substitution reaction proceeded with the complete retention of configuration at the stereogenic center.⁸ Thus, the chiral ligand **3a** was synthesized by four steps in 46% overall yield. Similarly, (R,R)-PPFCpH (**3b**) was obtained from (S,R)-PPFA (1b) (Scheme 1).

The reaction of (R,R)-PPFCpH (**3b**), after being treated with an equimolar amount of *sec*-BuLi, with

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For reviews, see: (a) Catalytic Asymmetric Synthesis; Ojima, I., Ed.; VCH: New York, 1993. (b) Noyori, R. Asymmetric Catalysis in Organic Synthesis; Wiley: New York, 1994. (c) Brunner, H.; Zettimeier, W. Handbook of Enantioselective Catalysis with Transition Metal Compounds; VCH: Weinheim, Germany, 1993.
 (2) (a) Hoveyda, A. H.; Morken, J. P. Angew. Chem., Int. Ed. Engl.

^{(2) (}a) Hoveyda, A. H.; Morken, J. P. Angew. Chem., Int. Ed. Engl.
1996, 35, 1262 and references cited therein. (b) Halterman, R. L. Chem. Rev. 1992, 92, 965 and references cited therein. (c) Hicks, F. A.; Buchwald, S. L. J. Am. Chem. Soc. 1996, 118, 11688. (d) Verdaguer,
X.; Lange, U. E. W.; Reding, M. T.; Buchwald, S. L. J. Am. Chem. Soc.
1996, 118, 6784. (e) Visser, M. S.; Heron, N. M.; Didiuk, M. T.; Sagal,
J. F.; Hoveyda, A. H. J. Am. Chem. Soc. 1996, 118, 4291. (f) Visser, M.
S.; Harrity, J. P. A.; Hoveyda, A. H. J. Am. Chem. Soc. 1996, 118, 3779.
(g) Kondakov, D. Y.; Negishi, E. J. Am. Chem. Soc. 1996, 118, 1577.
(3) (a) Trost, B. M. Chem. Ber. 1996, 129, 1313 and references cited

^{(3) (}a) Trost, B. M. Chem. Ber. 1996, 129, 1313 and references cited therein. (b) Murahashi, S.; Naota, T. Bull. Chem. Soc. Jpn. 1996, 69, 1805 and references cited therein. (c) Trost, B. M.; Portnoy, M.; Kurihara, H. J. Am. Chem. Soc. 1997, 119, 836. (d) Merlic, C. A.; Pauly, M. E. J. Am. Chem. Soc. 1996, 118, 11319.
(4) (a) Hidai, M.; Mizobe, Y. In Transition Metal Sulfur Chemistry; Stiefel E. L. Matematica K. Eds., ACS. Surgeoing. Science 650.

^{(4) (}a) Hidai, M.; Mizobe, Y. In *Transition Metal Sulfur Chemistry*; Stiefel, E. I., Matsumoto, K., Eds.; ACS Synposium Series 653; American Chemical Society: Washington, DC, 1996. (b) Matsuzaka, H.; Takagi, Y.; Ishii, Y.; Nishio, M.; Hidai, M. *Organometallics* **1995**, *14*, 2153. (c) Shimada, H.; Qü, J.-P.; Matsuzaka, H.; Ishii, Y.; Hidai, M. *Chem. Lett.* **1995**, 671. (d) Nishibayashi, Y.; Yamanashi, M.; Takagi, Y.; Hidai, M. *Chem. Commun.* **1997**, 859.

^{(5) (}a) Schwink, L.; Vettel, S.; Knochel, P. Organometallics **1995**, 14, 5000. (b) Komatsuzaki, N.; Uno, M.; Kikuchi, H.; Takahashi, S. Chem. Lett. **1996**, 677. (c) Uno, M.; Ando, K.; Komatsuzaki, N.; Takahashi, S. J. Chem. Soc., Chem. Commun. **1992**, 964. (d) Schofield, P. A.; Adams, H.; Bailey, N. A.; Cesarotti, E.; White, C. J. Organomet. Chem. **1991**, 412, 273. (e) Lindsay, C.; Cesarotti, E.; Adams, H.; Bailey, N. A.; White, C. Organometallics **1990**, 9, 2594. (f) Cesarotti, E.; Ciani, G.; Sironi, A. J. Organomet. Chem. **1981**, 216, 87.

⁽⁶⁾ For reviews, see: (a) *Ferrocenes*, Togni, A., Hayashi, T., Eds.; VCH: Weinheim, Germany, 1995. (b) Togni, A. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1475.

^{(7) (}a) Nishibayashi, Y.; Segawa, K.; Ohe, K.; Uemura, S. Organometallics 1995, 14, 5486. (b) Nishibayashi, Y.; Segawa, K.; Takada, H.; Ohe, K.; Uemura, S. Chem. Commun. 1996, 847. (c) Nishibayashi, Y.; Singh, J. D.; Fukuzawa, S.; Uemura, S. J. Org. Chem. 1995, 60, 4114. (d) Nishibayashi, Y.; Singh, J. D.; Segawa, K.; Fukuzawa, S.; Ohe, K.; Uemura, S. Organometallics 1996, 15, 370. (e) Nishibayashi, Y.; Arikawa, Y.; Ohe, K.; Uemura, S. J. Org. Chem. 1996, 61, 1172.

^{(8) (}a) Hayashi, T.; Mise, T.; Fukushima, M.; Kagotani, M.; Nagashima, N.; Hamada, Y.; Matsumoto, A.; Kawakami, S.; Konishi, M.; Yamamoto, K.; Kumada, M. *Bull. Chem. Soc. Jpn.* **1980**, *53*, 1138. (b) Gokel, G. W.; Marquarding, D.; Ugi, I. *J. Org. Chem.* **1972**, *37*, 3052.

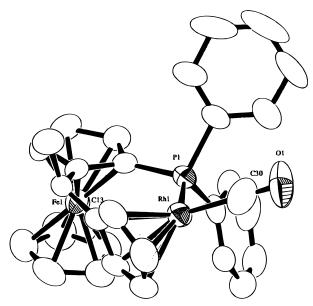
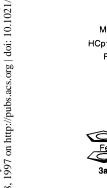
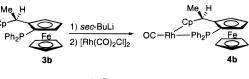


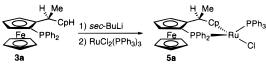
Figure 1. Molecular structure of 4b. Selected bond distances (Å) and angles (deg): Rh(1)-P(1) = 2.214(2), Rh-(1)-C(13) = 2.234(8), Rh(1)-C(30) = 1.819(9); P(1)-Rh-(1)-C(13) = 96.6(2), P(1)-Rh(1)-C(30) = 93.2(3), C(13)-C(13) = 96.6(2), P(1)-Rh(1)-C(30) = 93.2(3), C(13)-C(13) = 96.6(2), P(1)-Rh(1)-C(30) = 93.2(3), C(13)-C(13) = 96.6(2), C(13)-C(13)-C(13) = 96.6(2), C(13)-C(13)-C(13) = 96.6(2), C(13)-C(13)-C(13)-C(13) = 96.6(2), C(13)-Rh(1)-C(30) = 157.9(3).











 $[Rh(CO)_2Cl]_2$ in a 2:1 molar ratio in THF at room temperature for 12 h afforded the mononuclear rhodium(I) complex Rh(CO)[(S,R)-PPFCp] (4b) in 66% isolated yield (Scheme 2). The structure of 4b was determined by ¹H and ³¹P NMR.⁹ The cyclopentadienvl-phosphine chelation structure of 4b was unequivocally confirmed by X-ray analysis.¹⁰ An ORTEP drawing of one of the two independent molecules of 4b in each unit cell is shown in Figure 1 along with selected bond distances and angles. The rhodium atom has a distorted-trigonal-planar geometry.

The reaction of the lithium salt of **3a** with [RuCl₂-(PPh₃)₃] produced the ruthenium(II) complex [RuCl- $(PPh_3)[(R)-(S)-PPFCp]]$ (5a) as orange crystals after recrystallization from EtOH in 75% yield (Scheme 3). Both ¹H and ³¹P NMR revealed that the complex 5a was a pure diastereoisomer.¹¹ In order to determine the absolute configuration around the ruthenium atom, the single-crystal X-ray diffraction analysis of 5a was car-

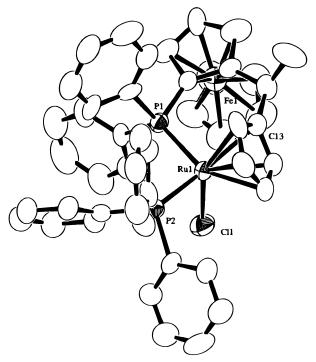


Figure 2. Molecular structure of 5a. Selected bond distances (Å) and angles (deg): Ru(1)-Cl(1) = 2.459(3), Ru-(1)-P(1) = 2.332(3), Ru(1)-P(2) = 2.309(4), Ru(1)-C(13)= 2.26(1); Cl(1)-Ru(1)-P(1) = 99.3(1), Cl(1)-Ru(1)-P(2)= 88.7(2), Cl(1)-R(1)-C(13) = 117.7(4), P(1)-Ru(1)-P(2)= 99.6(1), P(1)-Ru(1)-C(13) = 90.5(4), P(2)-Ru(1)-C(13)= 150.0(4).

ried out.¹² An ORTEP drawing of the molecular structure of 5a is shown in Figure 2 along with selected bond distances and angles. The stereogenic ruthenium center has an S configuration, provided that the priority sequence is in the order η^{5} -Cp > Cl > P(ferrocene)Ph₂ > PPh₃.¹³ Rhodium and ruthenium complexes with Cp-P bidentate chiral ligands derived from L-threitol or (S,S)-1,2-trans-bis(hydroxymethyl)cyclopentane have already been reported; however, the detailed structures were not determined.¹⁴

When the ruthenium complex 5a was treated with phenylacetylene at room temperature in the presence of NH₄PF₆, the ruthenium vinylidene complex [Ru- $(C=CHPh)(PPh_3)[(R)-(S)-PPFCp]]PF_6$ (**6a**) was obtained as orange crystals from acetone-diethyl ether in 96% yield (Scheme 4). ¹H and ³¹P NMR revealed that complex **6a** was a pure diastereomer.¹⁵ The X-ray

⁽⁹⁾ Spectroscopic and analytical data for **4b**: 1 H NMR δ 1.19 (d, 3H, J = 7.0 Hz), 3.65 (m, 1H), 3.93 (s, 5H), 3.95 (m, 1H), 4.18 (m, 1H), 4.42 (m, 1H), 5.27 (m, 1H), 5.46 (m, 1H), 5.61 (m, 1H), 5.76 (m, 1H), 7.2–7.9 (m, 10H); ³¹P NMR δ 45.7 (d, J = 197 Hz). Anal. Calcd for C₃₀H₂₆-

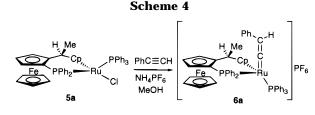
FeOPRh: C, 60.84; H, 4.43. Found: C, 60.59; H, 4.08. (10) Crystallographic data for **4b**: C₃0H₂₆FeOPRh, orthorhombic, space group $P_{2_12_12_1}$ (No.19), a = 15.614(6) Å, b = 33.912(9) Å, c = 9.544(7) Å, V = 5053(4) Å³, Z = 8, $D_{calcd} = 1.557$ g cm⁻³, $\mu(Mo Ka) = 0.040$ B = 0.022 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.022 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 cm⁻¹ B = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 for I = 0.040 B = 0.023 for 4100 unique reflections (I = 2.00 for I = 0.00 for I = 0.000 B = 0.000 for I = 0.000 for I13.09 cm⁻¹, R = 0.040, $R_w = 0.023$ for 4109 unique reflections (I > $3\sigma(I)$

⁽¹¹⁾ Spectroscopic and analytical data for 5a: ¹H NMR δ 1.33 (d, 3H, J = 7.0 Hz), 3.04 (m, 1H), 3.51 (m, 1H), 3.94 (m, 1H), 4.12 (s, 5H), 4.29 (m, 1H), 4.30 (m, 1H), 4.48 (m, 1H), 4.57 (m, 1H), 5.20 (m, 1H), 6.44 (m, 2H), 6.87 (m, 2H), 7.0–8.1 (m, 21H); ³¹P NMR δ 28.5 (d, J = 41 Hz), 45.5 (d, J = 41 Hz). Anal. Calcd for C₄₇H₄₁ClFeP₂Ru·2EtOH: C, 64.34; H, 5.61. Found: C, 64.10; H, 5.28.

⁽¹²⁾ Crystallographic data for 5a: C47H41ClFeP2Ru·2EtOH, monoclinic, space group P_{21} (No. 4), a = 10.992 (2) Å, b = 13.099 (2) Å, c = 15.595 (2) Å, $\beta = 93.43$ (2) °, V = 2241.4 (6) Å³, Z = 2, $D_{calcd} = 1.411$ g cm⁻³, μ (Mo K α) = 8.31 cm⁻¹, R = 0.074, $R_w = 0.084$ for 3960 unique reflections $(I > 3\sigma(I))$.

⁽¹³⁾ Stereoselective Synthesis; Helmchen, G., Hoffmann, R. W., Mulzer, J., Schaumann, E., Eds.; Houben-Weyl, Methods of Organic Chemistry E21a; Georg Thieme Verlag: Stuttgart, Germany, 1995. (14) Kataoka, Y.; Saito, Y.; Nagata, K.; Kitamura, K.; Shibahara, A.; Tani, K. *Chem. Lett.* **1995**, 833.

⁽¹⁵⁾ Spectroscopic and analytical data for **6a**: ¹H NMR δ 1.45 (d, 3H, J = 7.0 Hz), 3.07 (m, 1H), 3.97 (s, 5H), 4.19 (m, 1H), 4.30 (m, 1H), 4.67 (m, 1H), 4.86 (m, 1H), 5.13 (m, 1H), 5.27 (m, 1H), 5.35 (m, 1H), 5.74 (m, 1H), 6.89 (m, 2H), 6.8–7.4 (m, 28H); ³¹P NMR δ 31.8 (d, J = 27 Hz) and 44.9 (d, J = 27 Hz). Anal. Calcd for C₅₅H₄₈F₆FeP₃Ru: C, 61 162 H 4.51 61.58; H, 4.51. Found: C, 61.19; H, 4.89.



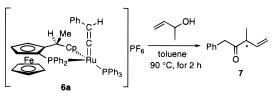
diffraction analysis of **6a** was performed to determine the absolute configuration around the ruthenium atom.¹⁶ It confirms that this reaction proceeds with inversion of configuration at the ruthenium center. This is in sharp contrast to the reaction of cyclopentadienyl ruthenium complexes containing chiral diphosphine ligands with terminal alkynes such as CHIRAPHOS and PROPHOS, where the reaction is completely stereospecific with retention of configuration at the ruthenium center.¹⁷

Recently, Trost et al. reported the reconstitutive condensation of both allylic alcohols and terminal alkynes by using [CpRuCl(PPh₃)₂], where the reaction was considered to proceed via a ruthenium vinylidene.¹⁸ This reconstitutive condensation did not proceed when cyclopentadienylruthenium complexes with chelate diphosphines were used.^{18,19} In this context, we employed complex **6a** as a catalyst for this asymmetric condensation. Treatment of the ruthenium vinylidene complex **6a** with excess amounts of 1-buten-3-ol at 90 °C for 2 h in toluene resulted in the formation of 3-methyl-1-phenyl-4-penten-2-one (**7**) in 40% yield with 65% enantiomeric excess (ee) (Scheme 5).²⁰ This provides the first example of an asymmetric reaction via a

(17) (a) Davies, S. G.; McNally, J. P.; Smallridge, A. J. Adv. Organomet. Chem. **1990**, 30, 1. (b) Consiglio, G.; Morandini, F. Chem. Rev. **1987**, 87, 761.

(18) Trost, B. M.; Kulawiec, R. J. J. Am. Chem. Soc. **1992**, 114, 5579. (19) Sato, M.; Asai, M. J. Organomet. Chem. **1996**, 508, 121.

Scheme 5



chiral vinylidene complex as the chiral source. Furthermore, the asymmetric reconstitutive condensation between phenylacetylene and 1-buten-3-ol was performed in the presence of 10 mol % of 5a and 20 mol % of NH₄PF₆ at 90 °C for 2 h in toluene. The chemical yield of 7 was 10% with an ee value of 25%.²⁰ The difference in enantioselectivity between the stoichiometric and catalytic reactions suggests that the reaction of 5a with phenylacetylene does not proceed stereospecifically at 90 °C. Actually, it was observed that treatment of 5a with an excess amount of phenylacetylene in the presence of NH₄PF₆ at 90 °C immediately afforded a mixture of ruthenium vinylidene diastereomers (the ratio of diastereomers observed by ¹H NMR spectroscopy was ca. 70:20:10). Furthermore, treatment of these ruthenium vinylidene diastereomers with excess amounts of 1-buten-3-ol at 90 °C for 2 h in toluene resulted in the formation of 7 in 55% yield with 21% ee. In order to achieve a higher selectivity in this catalytic reaction, this problem has to be solved. Further studies of the catalytic reaction²¹ using **5** or **6** and the determination of the absolute configuration of 7 are now in progress.

Supporting Information Available: Text, tables, and figures giving details of the X-ray structure study of complexes **4b**, **5a**, and **6a** (74 pages). Ordering information is given on any current masthead page.

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⁽¹⁶⁾ Preliminary crystallographic data for **6a**: $C_{55}H_{48}F_6FeP_3Ru$: Et₂O, orthorhombic, space group $P2_12_12_1$ (No. 19), a = 18.794(8) Å, b = 21.164(5) Å, c = 13.759(6) Å, V = 5472(3) Å³, Z = 4, $D_{calcd} = 1.392$ g cm⁻³, μ (Mo K α) = 6.88 cm⁻¹, R = 0.078, Rw = 0.085 for 1944 unique reflections ($I > 3\sigma(I)$). The carbon atoms of cyclopentadienyl rings and ruthenium, iron, and phosphine atoms were refined anisotropically. All other non-hydrogen atoms were refined isotropically.

⁽²⁰⁾ The low yield of **7** for both the stoichiometric and catalytic reactions was due to the formation of unidentified side products.

⁽²¹⁾ The preliminary results of a few catalytic reactions using **5** are as follows: 1-octyne and 1-buten-3-ol (20%, 34% ee), 1-octyne and 4-methyl-1-penten-3-ol (10%, 27% ee). The results will be reported in due course.