

*Volume 18, Number 26, December 20, 1999 © Copyright 1999*

*American Chemical Society*

## *Communications*

## **Third Generation of the Cp**′-**P Ligand: Highly Stereoselective Control of Central Chirality Arising at a Metal Center**

Yasutaka Kataoka,\* Yoko Iwato, Tsuneaki Yamagata, and Kazuhide Tani\*

*Department of Chemistry, Graduate School of Engineering Science, Osaka University, Toyonaka, Osaka 560-8531, Japan*

*Received October 7, 1999*

*Summary: A new type of Cp*′-*P ligand (the third generation of the Cp'* $-\tilde{P}$ *ligand),* [{ $3-(N\tilde{M})Ind-P_{1n=2}$ ]H ( $3a$ ), *in which an indenyl group, having a neomenthyl group, and a diphenylphosphino group are connected by an ethylene group, was designed and prepared. The presence of the two kinds of chiralities, indenyl-based planar chirality and stereogenic centers of the neomenthyl group, was an indispensable factor for inducing high stereoselectivity around the metal center.*

Hybrid ligands [Cp'-P]H, containing both a cyclopentadienyl derivative and a tertiary phosphine group connected by an appropriate spacer (Scheme 1), have attracted much attention due to the combined characters of their components, which could induce some unexpected reactivity and phenomena. $1-10$  So far, we have designed and prepared several types of  $[Cp'-P]$ H ligands and disclosed unique characters of their Rh, Ru, and Ir complexes.<sup>2</sup> The first one is the optically active [Cp′-P]H ligand **<sup>1</sup>**, in which stereogenic centers are in the spacer. The second one is the [(Ind-P)*n*]H ligand **<sup>2</sup>**, in which indenyl-based planar chirality will arise on coordinating to a metal center. Recently, we have found that the indenyl-based planar chirality (the latter ligand) can better control the chiral center arising at Rh by the oxidative addition of alkyl halides to its

<sup>(1)</sup> Charrier, C.; Mathey, F. *J. Organomet. Chem.* **1979**, *170*, C41. (2) (a) Saito, Y.; Yamagata, T.; Kataoka, Y.; Tani, K. The 41st Symposium on Organometallic Chemistry, Japan, 1994; Abst. PA.109. (b) Kataoka, Y.; Saito, Y.; Nagata, K.; Kitamura, K.; Shibahara, A.; Tani, K. *Chem. Lett.* **1995**, 833. (c) Kataoka, Y.; Saito, Y.; Shibahara,<br>A.; Tani, K. *Chem. Lett.* **1997**, 621. (d) Kataoka, Y.; Shibahara, A.;<br>Saito, Y.; Yamagata, T.; Tani, K. *Organometallics* **1998**, *17*, 4338.

<sup>(3) (</sup>a) Lebranc, J. C.; Moise, C.; Maisonnat, A.; Poilblanc, R.;<br>Charrier, C.; Mathey, F. *J. Organomet. Chem.* **1982**, 231, C43. (b) Lee,<br>I.; Dahan, F.; Maisonnat, A.; Poilblanc, R. *Organometallics* **1994**, 13,<br>2743. (c) *Chem.* **1997**, *532*, 159.

<sup>(4) (</sup>a) Kauffmann, T.; Olbrich, J. *Tetrahedron Lett.* **1984**, *25*, 1967. (b) Slawin, A. M. Z.; Williams, D. J.; Crosby, J.; Ramsden, J. A.; White, C. *J. Chem. Soc., Dalton Trans.* **1988**, 2491. (c) Nishibayashi, Y.; Takei, I.; Hidai, M. *Organometallics* **1997**, *16*, 3091. (d) Trost, B. M.; Vidal, B.; Thommen, M. *Chem. Eur. J.* **1999**, *5*, 1055.

<sup>(5) (</sup>a) Kettenbach, R. T.; Butenschön, H. *New J. Chem.* **1990**, 14, 599. (b) Butenschön, H.; Kettenbach, R. T.; Krüger, C. *Angew. Chem.*, *Int. Ed. Engl*. **1992**, *31*, 1066. (c) Kettenbach, R. T.; Bonrath, W.; Butenschön, H. *Chem. Ber.* 1993, 126, 1657. (d) Foerstner, J.; Olbrich, F.; Butenschön, H. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1234. (e)<br>Foerstner, J.; Kettenbach, R.; Goddard, R.; Butenschön, H. *Chem. Ber.* **1996**, *129*, 319. (f) Foerstner, J.; Kozhushkov, S.; Binger, P.; Wedemann, P.; Noltemeyer, M.; de Meijere, A.; Butenschön, H. *Chem.*<br>*Commun.* **1998**, 239. (g) Foerstner, J.; Kakoschke, A.; Stellfeldt, D.; Butenschön, H.; Wartchow, R. *Organometallics* **1998**, 17, 893.<br>
(6) (a) Fryzuk, M. D.; Mao, S. S. H.; Zaworotko, M. J.; MacGillivray,

L. R. *J. Am. Chem. Soc.* **1993**, *115*, 5336. (b) Bosch, B. E.; Erker, G.;

Fröhlich, R.; Meyer, O. *Organometallics* **1997**, *16*, 5449.<br>(7) (a) Wang, T.-F.; Juang, J.-P.; Wen, Y.-S. *J. Organomet. Chem.*<br>**1995**, *503*, 117. (b) Wang, T.-F.; Lai, C.-Y. *J. Organomet. Chem.* **1997**, *<sup>545</sup>*-*546*, 179. (8) Moblet, T. A.; Bergman, R. G. *J. Am. Chem. Soc.* **1998**, *120*, 3253.

<sup>(9)</sup> Lefort, L.; Crane, T. W.; Farwell, M. D.; Baruch, D. M.; Kaeuper,

J. A.; Lachicotte, R. J.; Jones, W. D. *Organometallics* **1998**, *17*, 3889. (10) (a) Barthel-Rosa, L. P.; Catalano, V. J.; Nelson, J. H. *J. Chem.*

*Soc., Chem. Commun.* **1995**, 1629. (b) Barthel-Rosa, L. P.; Catalano, V. J.; Maitra, K.; Nelson, J. H. *Organometallics* **1996**, *15*, 3924. (c) Antelmann, B.; Winterhalter, U.; Huttner, G.; Janssen, B. C.; Vogelgesang, J. *J. Organomet. Chem.* **<sup>1997</sup>**, *<sup>545</sup>*-*546*, 407.



carbonyl complex,  $(Cp'-P)RhCO,$ <sup>11</sup> than the chiral center in the spacer (the former ligand).<sup>2d</sup> Herein we report that the third generation of  $[Cp'-P]H$  ligand,  $[{3}$ - $(NM)Ind-P_{n=2}H$  (3a) or  $[{3-(NIM)Ind-P}_{n=2}]H$  (3c), in which an optically active substituent (a neomenthyl or a neoisomenthyl group) is attached to the indenyl ring of **2**, can control more efficiently the stereochemistry of the reaction of its metal complexes.

The general method for the preparation of **3a** and its rhodium complex  $[\eta^5:\eta^1-\{3-(NM)\}$ Ind-P $\}_n=2]$ RhCO (4a) is shown in Scheme 2. Experimental details are provided as Supporting Information. The rhodium complex **4a** was obtained as orange powders in 89% isolated yield from **3a**. 1H and 31P NMR showed that **4a** is a mixture of two diastereomers (major:minor  $= 62:38, 24%$  de). Recrystallization from hexanes at 4 °C gave only the major isomer of **4a** as yellow powders. The planar chirality of **4a**-major was found to be *S* configuration from the X-ray analysis (Figure 1a).<sup>12</sup> The cyclopentadienyl analogue  $[\eta^5:\eta^1-\{3-(NM)C_5H_3-P\}_{n=2}]\text{RhCO (4b)}$ was also prepared by a similar method using cyclopentadiene instead of indene, but its de was much lower (8% de). A combination of two kinds of substituents (a neomenthyl group and a benzene ring) is more effective



**Figure 1.** ORTEP drawing of the X-ray crystal structure: (a)  $(S_{\text{pl}})$ -[ $\eta^5$ : $\eta^1$ -{3-(NIM)Ind-P}<sub>*n*=2</sub>]RhCO (4a-major). (b)  $(S_{\text{pl}})$ -[ $\eta^5$ : $\eta^1$ -{3-(NIM)C<sub>5</sub>H<sub>3</sub>-P}<sub>*n*=2</sub>]RhCO (**4b**-major).

for controlling the planar chirality arising on complexation to the metal. The major isomer of **4b** could be isolated by recrystallization from hexanes at 5 °C. The X-ray analysis of **4b**-major (Figure 1b)13 showed that the relative location of the neomenthyl group and the pendant group (diphenylphosphinoethyl group) was different from that in **4a**-major, but the planar chirality was also *S* configuration according to a general system for designating an absolute configuration. When (+) isomenthol was used as the starting alcohol, [*η*5:*η*1-{3-  $(NIM)Ind-P\}_{n=2}$ ]RhCO (**4c**) was obtained with 74% de. The neoisomenthyl group caused much higher asymmetric induction with regard to the arising planar chirality than the neomenthyl group.

Some representative results on the reaction of the Rh carbonyl complexes **4** with alkyl halides are shown in Table 1. The complex **4a**-major reacted with MeI in  $CH_2Cl_2$  at room temperature for 2 h to afford the rhodium(III) acyl complex [ $η$ <sup>5</sup>: $η$ <sup>1</sup>-{(3-NM)Ind-P}<sub>*n*=2</sub>]RhI-(COMe) (**5a**) in 93% yield.14 1H and 31P NMR spectra of the reaction mixture showed that the acyl complex **5a** was a mixture of two diastereomers  $(R_{\rm pl}, R_{\rm Rh})$  and *R*pl,*S*Rh)15 in 96% de (run 1). The complex **4a**-major also reacted with EtI or PhCH<sub>2</sub>Br to afford the corresponding acyl complex with high diastereoselectivity in good yield (runs 2 and 3). A combination of the planar chirality (indenyl-based chirality) and the stereogenic centers of the neoisomenthyl group was highly effective for controlling the evolving central chirality (metal-centered chirality) in the oxidative addition. For example, when the cyclopentadienyl derivative **4b**-major, which has a cyclopentadienyl-based planar chirality and an optically active substituent on the cyclopentadienyl ring, reacted with MeI under the same conditions, **4b**-major was consumed within 1 h and the corresponding acyl complex **5b** was obtained in 79% yield, but its de was only 56%. The oxidative addition of MeI toward the Rh

<sup>(11) (</sup>Cp'-P) shows the anion of the [Cp'-P]H ligand.<br>
(12) Crystallographic data for **4a**-major: C<sub>34</sub>H<sub>38</sub>OPRh, (*M*<sub>τ</sub> = 596.52),<br>
orthorhombic,  $P2_12_12_1$  (No. 19),  $a = 16.308(6)$  Å,  $b = 26.862(5)$  Å,  $c =$ <br>
6.661(5)

<sup>(13)</sup> Crystallographic data for **4b**-major: C<sub>30</sub>H<sub>36</sub>OPRh, (*M*<sub>r</sub> = 546.47),<br>hexagonal, *P*6<sub>1</sub> (No. 169), *a* = 18.8504(13) Å, *b* = 18.8504(13) Å, *c* =<br>13.7819(14) Å,  $\alpha$  = 90.0°,  $\beta$  = 90.0°,  $\nu$  = 120.0°,  $V$  = 4 13.7819(14) Å,  $\alpha = 90.0^{\circ}$ ,  $\beta = 90.0^{\circ}$ ,  $\gamma = 120.0^{\circ}$ ,  $V = 42141.1(6)$  Å<sup>3</sup>,  $Z = 6$ ,  $D_{\text{cald}} = 1.284$  Mg/m<sup>3</sup>,  $\mu = 0.679$  mm<sup>-1</sup>,  $R_{\text{int}} = 0.0175$ , R1(all) = 0.0360 (>2*o*(*l*)), wR2(all) = 0.1291, wR2(obsd) = 0.1119  $(>2\sigma(I))$ .

<sup>(14)</sup> Excess amount of methyl iodide (0.27 mL, 4.3 mmol) was added to a solution of  $4a$ -major (26 mg, 0.044 mmol) in  $CH_2Cl_2$  (5 mL) at 25 °C. The reaction mixture was stirred for 2 h, and then the solvent was removed in vacuo to afford  $5a$  (major:minor  $= 98:2$ ,  $96\%$  de; the ratio was determined by 31P NMR). The resulting solid was purified by column chromatography on silica gel (eluent:hexane/AcOEt = 5:1) to<br>give **5a** (30 mg, 0.041 mmol, 93%, 96% de) as deep red powders. Mp: 108–112 °C (dec). <sup>31</sup>P{<sup>1</sup>H} NMR (CDCl<sub>3</sub>): δ 55.5 (d, *J*<sub>P-Rh</sub> = 171 Hz, major), 57.6 (d, *J*<sub>P-Rh</sub> = 173 Hz, minor). Anal. Found: C, 56.56; H, 5.82. Calcd for C<sub>35</sub>H<sub>41</sub>IOPRh: C, 56.92; H, 5.60%.

**Table 1. Reaction of [***η***5:***η***1-(Cp**′-**P)]RhCO with RX***<sup>a</sup>*

|       |                                    |                      |             | $[\eta^5:\eta^1$ -(Cp'P)]Rh(X)COR |                              |
|-------|------------------------------------|----------------------|-------------|-----------------------------------|------------------------------|
| entry | $[\eta^5:\eta^1$ -(Cp'P)]-<br>RhCO | RX                   | time<br>(h) | yield<br>$(%)^b$                  | major:<br>minor <sup>c</sup> |
|       | 4a-major                           | MeI                  |             | 93                                | 98:2                         |
| 2     | 4a-major                           | EtI                  | 72          | 91                                | 96:4                         |
| 3     | 4a-major                           | PhCH <sub>2</sub> Br | 12          | 92                                | 95:5                         |
| 4     | 4 <b>b</b> -major                  | MeI                  |             | 79                                | 78:22                        |
| 5     | 4c <sup>d</sup>                    | MeI                  |             | 95                                | 96:4e                        |

*<sup>a</sup>* [*η*5:*η*1-(Cp′-P)]RhCO was treated with excess RX in CH2Cl2 at room temperature. *<sup>b</sup>* Isolated yield. *<sup>c</sup>* The ratio was determined by 1H NMR and/or 31P NMR of the reaction mixture. *<sup>d</sup>* 86% de. *<sup>e</sup>* The ratio of the product obtained from **4c**-major.

carbonyl complex having the Cp'-P ligand of the second generation,  $[\eta^5:\eta^1-(\text{Ind}-P)_{n=2}]\text{RhCO}$ , gave the corresponding acyl complex with lower diastereoselectivity (34% de).2d Both indenyl-based chiralirty and stereogenic centers of a substituent of the indenyl group are essential factors for inducing the high stereoselectivity around the metal in the oxidative addition.

Reaction of  $4c$  with  $86\%$  de  $(4c$ -major: $4c$ -minor  $= 93$ : 7) with MeI under the same conditions gave the acyl complex  $[η<sup>5</sup>:η<sup>1</sup> - {3-(NM)Ind-P}<sub>ln=2</sub>]RhI(COMe)$  (5c) in 95% yield (run 5). From the 31P NMR of the reaction mixture, **5c** was found to contain four diastereomers (two sets of the diastereomers were produced from **4c**major and **4c**-minor, respectively). The ratio of the diastereomers derived from **4c**-major (**5c**-major) was 96: 4, and that from **4c**-minor (**5c**-minor) was ca. 82:18.16 After purification by silica gel column chromatography, **5c**-major with 92% de was isolated in 75% yield. Reaction of the complex **5c**-major (92% de) with AgPF6 in  $CH_2Cl_2$  at room temperature for 2 h gave the cationic complex **6** in 87% yield with 92% de (eq 1). The ratio of



the diastereomers was much the same before and after the reaction, indicating that the stereospecific retromigratory insertion of CO proceeded.<sup>17</sup> Now we are investigating the application of the optically active complex shown in this paper to new stereospecific or asymmetric catalytic reactions.

**Acknowledgment.** This work was supported by a Grant-in-Aid for Scientific Research from the Ministry of Education, Science, Sports, and Culture of Japan.

**Supporting Information Available:** Text giving synthetic procedures and spectroscopic and analytical data for all new compounds and text and tables giving X-ray structural information on **4a**-major and **4b**-minor. The material is available free of charge via the Internet at http://pubs.acs.org.

## OM990800M

(18) Unpublished results.

<sup>(15)</sup> The chirality symbol with subscript "pl" indicates planar chirality, and that with subscript "Rh" means the chirality around the metal center.

<sup>(16)</sup> The integral value of the minor product derived from **5c**-minor contained errors to some extent.

<sup>(17)</sup> From the separate experiments using  $(R^*_{\text{pl}}, R^*_{\text{Rh}})$ -[ $\eta^5:\eta^1$ -(Ind- $P$ <sub>*n*=2</sub>]RhI(COMe), we have elucidated that the methyl group migrated from the acyl group to the Rh.18 Thus, the structure of the major isomer of **6** will be shown as in eq 1.