Organic Syntheses via Transition-Metal Complexes. 119.¹ α-Methylenation and α-Alkenylation of α,β-Unsaturated Amides by Means of Carbene Tungsten Complexes: A Novel Baylis–Hillman Type Reaction

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Fischer carbene complexes were found to undergo Baylis–Hillman type additions to α,β unsaturated acid amides. Reactions of (non-CH-acidic) carbene tungsten complexes 1a-cwith (CH-acidic) but-2-enoic acid amides 2 in the presence of POCl₃/Et₃N resulted in an α -addition of the carbene ligand to the amide backbone and formation of cross-conjugated aminocarbene complexes 3. Reaction of CH-acidic carbene tungsten complexes 1d-h gave the cross-conjugated aminocarbene complexes 9 (which are double-bond isomers of compounds 3) together with (cyclobutenyl)carbene complexes 10, by α - and β -addition, respectively, to the but-2-enoic amides 2. Compounds 9 and 10 are valence isomers, but they do not interconvert thermally.

The α -alkylation of α , β -unsaturated carbonyl compounds by tertiary amine- or phosphine-catalyzed coupling with aldehydes, the so-called Baylis–Hillman reaction, is among the most useful C–C bond-forming reactions in organic synthesis (eq 1 in Scheme 1).² Much progress³ has been achieved since Baylis and Hillman first reported the reaction of acetaldehyde with ethyl acrylate and acrylonitrile in the presence of catalytic amounts of a strong Lewis base such as 1,4-diazabicyclo-[2.2.2]octane in 1972.⁴ The hydroxymethylene adducts thus generated contain a structural element that is widely present in products of biological and medicinal

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interest.⁵ The α , β -unsaturated carbonyl compounds which were shown to undergo a Baylis—Hillman reaction include acrylonitrile, acrolein, acrylates, and α , β -unsaturated ketones, but to date, only very few examples are known of acrylamides.⁶

We report on Baylis–Hillman type additions of Fischer carbene complexes to the α -position of α , β -unsatur-

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[†] X-ray structure analysis.

⁽²⁾ For reviews see: (a) Ciganek, E. Org. React. 1997, 51, 201–350.
(b) Basavaiah, D.; Rao, P. D.; Hyma, R. S. Tetrahedron 1996, 52, 8001–8062. (c) Drewes, S. E.; Roos, G. H. P. Tetrahedron 1988, 44, 4653–4670.

⁽³⁾ For applications see: (a) Langer, P. Angew. Chem., Int. Ed. 2000, 39, 3049–3052. (b) Yu, C.; Hu, L. J. Org. Chem. 2001, 66, 5413–5418.
(c) Yu, C.; Hu, L. J. Org. Chem. 2002, 67, 219–223. (d) Brzezinski, L. J.; Rafel, S.; Leahy, J. W. J. Am. Chem. Soc. 1997, 119, 4317–4318.
(e) Miyakoshi, T.; Saito, S. Nippon Kagaku Kaishi 1983, 1623–1628; Chem. Abstr. 1984, 100, 156191g. (f) Marko, I. E.; Giles, P. R.; Hindley, N. J. Tetrahedron 1997, 53, 1015–1024. (g) Richter, H.; Jung, G. Tetrahedron Lett. 1998, 39, 2729–2730. (h) Barrett, A. G. M.; Cook, A. S.; Kamimura, A. Chem. Commun. 1998, 2533–2534. (i) Kunig, E. P.; Xu, L. H.; Romanens, P.; Bernardinelli, G. Tetrahedron Lett 1993, 349, 7049–7052. (j) Aggarwal, V. K.; Mereu, A.; Tarver, G. J.; McMague, R. J. Org. Chem. 1998, 63, 7183–7189. (k) Kawamura, M.; Kobayashi, S. Tetrahedron Lett. 1999, 40, 1539–1542. (l) Kataoka, T.; Iwama, T.; Tsujiyama, S.; Iwamura, T.; Watanaba, S. Synlett 1999, 197–198. (n) Kataoka, T.; Iwama, T.; Iwama, T.; Tsujiyama, S.; Iwamura, S.; Kanematsu, K.; Iwamura, T.; Watanabe, S. Chem. Lett. 1999, 257–258. (o) Kataoka, T.; Iwama, T.; Tsujiyama, S. Chem. Commun. 1998, 197–198. (p) Ono, M.; Nishimura, K.; Nagaoka, Y.; Tomioka, K. Tetrahedron Lett. 1999, 40, 1509–1512. (q) Li, G.-G.; Wei, H.-X.; Gao, J.-J.; Caputo, T. D. Tetrahedron Lett. 2000, 41, 1–5. (r) Kataoka, T.; Kuroshita, H.; Iwama, T.; Tsujiyama, S.; Iwamura, T.; Watanabe, S. Muraoka, O.; Tanabe, G. Tetrahedron 2000, 56, 4725–4731. (s) Li, G.-G.; Gao, J. J.; Wei, H.-X.; Enright, M. Org. Lett. 2000, 2, 617–620. (t) Iwabuchi, Y.; Nakatani, M.; Yokoyama, N.; Hatakeyama, S. J. Am. Chem. Soc. 1999, 121, 10219–10220.

ated acid amides (eq 2 in Scheme 1). While enolates are typically assumed to be reactive intermediates in Baylis—Hillman reactions, our approach involves the formation of enol intermediates from acid amides under the influence both of a Lewis base and a Lewis acid. To our knowledge, an "enol approach" has not been successfully applied before to such reactions.

Our enol approach is based on the condensation of acid amides with carbene complexes under Vilsmeyer conditions. Reactions of this type have proven most versatile in our hands and have been utilized for the generation of a broad range of interesting compounds from readily available starting components in most straightforward procedures and usually in good yields. Even though seemingly minor structural changes of the amide as well as of the carbene complex are expected to exhibit a strong influence on the overall reaction, it is obvious that the different reaction paths are strictly controlled and are easily distinguished on the basis of mechanistic considerations referring to the specific iminium compounds derived from the amides under Vilsmeyer conditions. Typical reaction paths include the condensation of methylethoxycarbene complexes (M = Cr, Mo, W) with tertiary non- α -CH-acidic amides, such as tertiary arylamides or tertiary formamides, to produce (3E)-4amino-1-metalla-1,3-dienes.⁷ Furthermore, condensation of methylethoxycarbene complexes (M = Cr, Mo, W) with secondary arylamides or even secondary alkylamides gave (3Z)-4-amino-1-metalla-1,3-dienes, while (in line with expectation) the condensation with secondary formamides took a completely different course and produced ketenimine complexes instead.⁸ Condensation of (non- α -CH-acidic) arylethoxycarbene complexes (M = Cr, Mo, W) with (α -CH-acidic) tertiary alkylamides afforded 2-amino-1-metalla-1,3-dienes by insertion of a C,C unit into the M=C bond, which was induced by the elimination of an α -hydrogen atom from the iminium intermediate.⁹ A manifold of fascinating products has been derived from α,β -unsaturated acid amides. In line with expectation, these reactions were strictly controlled by structural features, such as the presence or absence of acidic hydrogen atoms as well as substituents influencing a 2,1- and 4,1-addition, respectively, in an early reaction stage. Thus, condensation of methylethoxycarbene complexes with tertiary cinnamides produced (3E)-4-amino-1-metalla-1,3,5-hexatrienes and aminocyclopentadienes resulting thereof by π -cyclization, together with minor amounts of cyclohexanonylcarbene complexes, both resulting from initial 2,1-addition. These

(9) Aumann, R.; Hinterding, P. Chem. Ber. 1989, 122, 365-370.

products were not obtained from (primary alkyl)ethoxycarbene complexes, since 4,1-addition to the α,β unsaturated amide prevailed with these compounds and resulted in formation of cyclobutenylaminocarbene complexes as the only products.¹⁰ Furthermore, condensation of methylethoxycarbene complexes with secondary cinnamides produced (3*Z*)-amino-1-metalla-1,3,5-hexatrienes, while condensation of (primary alkyl)ethoxycarbene complexes afforded an efficient access to novel (*N*-enamino)carbene complexes.¹ The reaction on which we wish to report in this paper is based on the specific reactivity of tertiary α,β -unsaturated acid amides containing a CH-acidic unit (Scheme 2).

Reactions of but-2-enoic acid amides **2a**,**b** with equivalent amounts of phosphorus oxychloride, tungsten carbene complexes **1a**-**c**, and triethylamine in dichloromethane at 25 °C afford the yellow cross-conjugated α -methylenation products **3a**-**e** (Scheme 2). They form 3E/3Z isomeric mixtures with respect to the 2-amino-3-ethoxy-1-tungstabuta-1,3-diene unit (Scheme 2), in which the 3E isomers prevail. Both isomers (3E)-**3** and (3Z)-**3** exhibit zwitterionic carbiminium carbonylmetalate structures.¹¹ These are chiral, since they are strongly twisted due to the allyl effect of the ligand backbone (Figure 1).

It is assumed that iminium chlorides 4 and enamines 5 are generated initially and add to the carbene complexes 1. Interestingly, addition is observed to the carbene carbon atom to give intermediates 6 only, thus providing a good entry to the formation of 1-metallahexa-1,3,5-trienes and 1-metallahexa-1,3-dien-5-ynes. No products arising from β -addition to α , β -unsaturated carbene complexes 1b,c have been found so far. A typical Baylis-Hillman reaction would involve transformation of an adduct 6 into compound 6', but apparently the reaction course is outrun by a 1,3-migration of the W(CO)₅ unit to give the zwitterion 7 as precursor to the amino carbene complex 3 (Scheme 2). Compounds 3 undergo hydrolysis of the enol ether unit on extended contact with silica gel to afford the acyl derivatives 8 (eq 1 in Scheme 3), which are considered Baylis-Hillman type products, in which the oxygen atom of the amide moiety is replaced by the isolobal $W(CO)_5$ unit.

While pursuing studies, reactions of the butyl carbene complex **1f** instead of non-CH-acidic compounds **1a**–**c** were found to afford conspicuously small yields of the α -methylenation product **3f**. This observation prompted us to further investigate the influence of CH-acidic groups in compounds **1** on the overall reaction. It was explicitly shown by NMR spectra that the compound (3*E*)-**3f** is smoothly transformed into compound (3*Z*)-**9f** already at 20 °C within 15 h by a (supposedly catalyzed) 1,5-hydrogen transfer (eq 2 in Scheme 3). Metallatrienes **3a**–**e** containing R¹ = aryl, alkynyl, alkenyl instead of R¹ = alkyl proved to be quite stable thermally.

α-CH-Acidic Carbene Complexes 1d-i

Reactions of α -CH-acidic carbene complexes **1d**-**h** with the but-2-enoic acid amides **2a**,**b** in the presence of POCl₃/Et₃N gave three major products: 2-amino-1-tungstapenta-1,3-dienes (3*Z*)-**9**, which are double-bond

⁽⁴⁾ For the first example see: (a) Baylis, A. B.; Hillman, M. E. D. Ger. Offen. 2,155,113, 1972; *Chem. Abstr.* **1972**, *77*, 341740, Hillman, M. E. D.; Baylis, A. B. U.S. Patent 3,743,669, 1973. (b) Morita, K.; Suzuki, Z.; Hirose, H. *Bull. Chem. Soc. Jpn.* **1968**, *41*, 2815–2816. (5) See for example: (a) Hoveyda, A. H.; Evans, D. A.; Fu, G. C.

^{(6) (}a) Bode, M. L.; Kaye, P. T. J. Chem. Soc., Perkin Trans. 1 1993, 1809–1813.
(b) Yu, C.; Hu, L. J. Org. Chem. 2002, 67, 219–223.
(7) (a) Aumann, R.; Hinterding, P. Chem. Ber. 1990, 123, 611–620.

^{(7) (}a) Aumann, R.; Hinterding, P. Chem. Ber. 1990, 123, 611–620.
(b) For related studies see: Lattuado, L.; Licandro, E.; Maiorana, S.; Papagni, A.; Chiesi Villa, A.; Guastini, C. J. Chem. Soc., Chem. Commun. 1988, 1092–1093.

^{(8) (}a) Aumann, R.; Hinterding, P. *Chem. Ber.* **1990**, *123*, 2047–2051. (b) For a review on the formation of ketenimine complexes see: Aumann, R. *Angew. Chem.* **1988**, *100*, 1512–1524; *Angew. Chem., Int. Ed. Engl.* **1988**, *27*, 1456–1467.

⁽¹⁰⁾ Aumann, R.; Vogt, D.; Fu, X.; Fröhlich, R.; Schwab, P. Organometallics 2002, 21, 1637-1645.

⁽¹¹⁾ Aumann, R.; Roths, K.; Fröhlich, R. J. Organomet. Chem. 2001, 617–618, 322–328.





^{*a*} See Scheme 4 for this reaction.



Figure 1. Molecular structure of the 3-ethenyl-1-tungstahexa-1,3-dien-5-yne compound (3E)-**3d**. Selected bond lengths (Å), valence angles (deg), and dihedral angles (deg): W-C1 = 2.228(8), C1-C2 = 1.514(10), C1-N11 = 1.334(10), C2-C3 = 1.355(12), C2-C6 = 1.454(11), C3-C4 = 1.440(13), C3-O8 = 1.379(10), C4-C5 = 1.208(13), C5-C51 = 1.427(14), C6-C7 = 1.327(13); W-C1-C2 = 115.9(5), W-C1-N11 = 130.6(5), N11-C1-C2 = 113.4(7), C1-C2-C3 = 119.8(7), C1-C2-C6 = 118.9(7), C6-C2-C3 = 121.3(7), C2-C3-C4 = 122.6(8), C2-C3-O8 = 119.0-(7), O8-C3-C4 = 118.4(8), C3-C4-C5 = 177.6(11), C4-C5-C51 = 177.3(11), C2-C6-C7 = 124.4(8); W-C1-C2-C3 = -91.7(8), W-C1-C2-C6 = 89.0(7), C1-C2-C3-C4 = -0.5(13).

isomers of compounds **3**, and acyl derivatives **12**, apparently resulting thereof by hydrolysis (Scheme 4).

Scheme 3. Hydrolysis and Rearrangement of Compounds 3



Scheme 4. Aminocarbene Complexes 9 and 10: Stable Valence Isomers Generated from Tertiary Crotyl Amides and CH-Acidic Carbene Complexes 1d-h in Parallel Reactions



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^{*a*} Isolated chemical yields. ^{*b*}Molar ratio "*cis*"-**10**" *trans*"-**10** is shown in parentheses. ^{*c*}ETHF = 5-ethyltetrahydrofuran-2-ylidene–W(CO)₅. ^{*d*}Compound **3f** was isolated instead, if short reaction times were applied (see footnote *a* in Scheme 2), but compound **9f** has been obtained on extension of the reaction time (see Scheme 3), resulting from rearrangement of compound **3f**. ^{*e*}1-Ethoxy-2-phenylethene (**11g**) was obtained as major product by a competing base-induced elimination of the carbene ligand. ^{*f*}2-Hydroxybutan-1-yl.

Scheme 5. Reaction Paths Initiated by α - and β -Addition of CH-Acidic Carbene Complexes 1d-h to α , β -Unsaturated Amides 2



Furthermore, (2-ethoxycyclobutenyl)carbene complexes **10** were obtained, which are valence isomers of compounds **9**.

Side reactions included a base-induced elimination of enol ethers from carbene complexes 1d-h. Formation of the enol ether 11g was the major reaction of the benzyl carbene complex 1g, since the α -position of this compound is activated by two neighboring π systems. Reaction of the carbocyclic carbene complex 1i was shown to afford the open-chain system 12k, presumably resulting from hydrolysis of its precursor 9k. Formation of compounds **9** seems to involve precursors **13** and **14** (Scheme 5). Compounds **14** readily afford isomers **9** by double-bond isomerization (see Scheme 3), particularly since the W=CC=C(OEt) skeleton is not stabilized by π conjugation due to its gauche conformation (see Figure 1). The migration of the C=C bonds is highly stereoselective and yields compounds (3*Z*)-**9** as the only detected organometallic product. Compounds **9** are thermally quite stable but prove to be quite sensitive to hydrolysis on silica gel to give acyl derivatives (3*Z*)-**12** even on fast chromatography.

Scheme 6. α-Methylenation of Aliphatic Amides 15 with the (1-Alkynyl)carbene Complex 1b



^a Isolated chemical yields.

Even though (butadien-2-yl)carbene complexes (3*Z*)-9 and (cyclobutenyl)carbene complexes 10 are valence isomers, it was not possible to induce an interconversion of these compounds. According to NMR measurements, both compounds 9c and 10c were stable in C₆D₆ even at 70 °C for at least 2 h and did not undergo an electrocyclic process. It therefore is obvious that these compounds were not generated by isomerization but in parallel reactions (Scheme 5). While the compound (3*Z*)-9 seems to be obtained by an α -addition similar to that outlined in Scheme 2, compounds 10 are assumed to be obtained by the β -addition of a carbene complex 1d-h to the amide 2 in a process which has been recently described in more detail (Scheme 5).¹⁰

α-Methylenation of Aliphatic Acid Amides

Vilsmeyer conditions with POCl₃/Et₃N, which were applied in the enol approach to the Baylis–Hillman reaction, had already previously been found to be superior in condensation reactions of saturated acid amides with Fischer carbene complexes.^{7–9} To point out its outstanding performance in the regioselective generation also of highly unsaturated aminocarbene complexes, we wish to report on the formation of 1-tungs-tahexa-1,3-dien-5-ynes **16a**,**b** (Scheme 6).

Structure Elucidation

The aminocarbene complexes **3**, **8**–10, **12**, and **16** were characterized by ¹H and ¹³C NMR spectra, including ¹J(C,H) and ^{2,3}J(C,H) correlation experiments. The configuration of the cross-conjugated metallatrienes **3**, **8**, **9**, and **12** were routinely determined by NOE measurements.

Structural details of the compound (3E)-**3d** were collected by a crystal structure analysis (Figure 1). The 2-ethoxybut-1-en-3-yne moiety of the compound (3E)-**3d** is planar (C1-C2-C3-C4 = 0.5(13)°) and strongly twisted against the plane defined by the bonds to the carbene carbon atom (W-C1-C2-C3 = 91.7(8)°). All bond distances are within expectation.

The molecular structure of the acyl derivative (3Z)-**12a** shows an s-trans configuration of the C=CC=O



Figure 2. Molecular structure of the acyl derivative (3*Z*)-**12a**. Selected bond lengths (Å), valence angles (deg), and dihedral angles (deg): W-C1 = 2.249(3), C1-C2 = 1.504(4), C1-N7 = 1.304(4), C2-C5 = 1.496(5), C2-C3 = 1.331(5), C3-C4 = 1.490(5), C5-O5 = 1.214(4), C5-C6 = 1.498(5); W-C1-C2 = 114.9(2), W-C1-N7 = 130.8(2), N7-C1-C2 = 114.3(3), C1-C2-C5 = 114.8(3), C1-C2-C3 =122.9(3), C5-C2-C3 = 122.3(3), C2-C3-C4 = 125.6(3), C2-C5-O5 = 119.5(3), C2-C5-C6 = 119.7(3), O5-C5-C6 = 120.8(3); W-C1-C2-C3 = 91.6(3), W-C1-C2-C5 =-89.0(3), C1-C2-C3-C4 = 4.7(5), C1-C2-C5-O5 =-16.9(5).

unit, which is arranged perpendicular to the plane defined by the bonds to the carbone carbon atom (W- $C1-C2-C3 = 91.6(3)^{\circ}$) (Figure 2).

Conclusion

Since α,β -unsaturated aminocarbene complexes find wide application in organic synthesis, regio- and stereoselective generation of such compounds from readily available starting materiald has become a recent focus in research. Condensation reactions of Fischer carbene complexes with acid amides under Vilsmeyer conditions were shown to provide an efficient entry to the selective formation of very different types of aminocarbene complexes. We are currently unraveling the structural prerequisites of these condensation reactions. In the present paper it was shown that the condensation of non-CH-acidic Fischer carbene complexes 1a-c with (CH-acidic) α,β -unsaturated acid amides **2** affords the aminocarbene complexes **3** by α -methylenation of the amide skeleton. Corresponding reactions of CH-acidic compounds **1d**-**h** are different and give compounds **9**, which are double-bond isomers of compounds 3, together with (cyclobutenyl)carbene complexes 10.

Experimental Section

General Considerations. NMR: Bruker AM 360, Bruker AMX 400, and Varian U 600 instruments. All new compounds were routinely analyzed by ¹H, ¹³C, DEPT, (¹H,¹H)COSY, (¹H,¹³C)GHSQC, and (¹H,¹³C)GHMBC experiments on a Bruker AMX 400 instrument. IR: FT-IR Bio-Rad Digilab Division FTS-45 instrument. MS: Finnigan MAT8200 instrument. Elemental analyses: Heraeus CHN-O Rapid instrument. Column chromatography: Merck silica gel 60F. Flash chromatography was performed under an argon atmosphere. TLC: Merck silica gel $60F_{254}$. R_f values refer to TLC tests.

(2-(Dimethylamino)-3-ethenyl-4-ethoxy-4-phenyl)-1-(pentacarbonyltungsta)-1,3-butadiene ((*3E*)-3a and (*3Z*)-3a) and (3-Benzoyl-2-(dimethylamino)-1-(pentacarbonyltungsta)-1,3-pentadiene ((*3Z*)-8a). To but-2-enoic acid dimethylamide (2a; 226 mg, 2.00 mmol) and 2 mL of dry dichloromethane in a 5 mL screw-top vessel was added phosphorus oxychloride (306 mg, 2.00 mmol) with stirring at 0 °C. Pentacarbonyl(1-ethoxybenzylidene)tungsten (**1a**; 458 mg, 1.00 mmol) and triethylamine (404 mg, 4.00 mmol) in 2 mL of dry dichloromethane was added after ca. 30 min at 0 °C, and the mixture was warmed to 25 °C. Workup after 24 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (3:1) afforded the compounds (3*Z*)-**3a** (80 mg, 14%, $R_f = 0.5$ in 2/1 *n*-pentane/dichloromethane), (3*E*)-**3a** (354 mg, 64%, $R_f = 0.3$ in 2/1 *n*-pentane/dichloromethane, yellow crystals from *n*-pentane at -20 °C, mp 97–98 °C), and **8a** (65 mg, 12%, $R_f = 0.5$ in 1/2 *n*-pentane/dichloromethane).

(3E)-3a.



¹H NMR (CDCl₃, 25 °C): δ 7.36–7.28 (5 H, m, Ph), 6.99 (1 H, dd, ${}^{3}J = 10.8$ Hz and 18.0, H₂C=CH), 5.31 (1 H, dd, ${}^{3}J = 10.8$ Hz, ${}^{2}J = 1.2$, cis-H H₂C=), 4.81 (1 H, dd, ${}^{3}J = 18.0$ Hz, ${}^{2}J =$ 1.2, trans-H H₂C=), 3.87 and 3.27 (3 H each, s each, 2 NCH₃), 3.82 and 3.27 (1 H each, m each, diastereotopic OCH₂), 1.31 (3 H, t, ${}^{3}J = 7$ Hz, OCH₂CH₃). 13 C NMR (CDCl₃): δ 256.2 (C_q, W=C), 202.6 and 198.2 (Cq each, 1:4, trans- and cis-CO W(CO)₅), 142.9 (C_q, C4), 137.9 (C_q, C3), 134.7 (C_q, *i*-C Ph); 128.8, 128.2, and 128.0 (2:1:2, CH each, Ph), 128.5 (CH, H₂C= H), 116.5 (CH₂, H₂C=CH), 67.6 (OCH₂), 53.1 and 44.1 (NCH₃ each), 15.2 (OCH₂CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2061.9 (18), 1970.3 (5), 1926.9 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m*/*e* (%)): 553 (35) $[M^+]$, 525 (20) $[M^+ - CO]$, 497 (40) $[M^+ - 2 CO]$, 469 (95) $[M^+ - 3 CO]$, 441 (20) $[M^+ - 4 CO]$, 413 (100) $[M^+ - 5]$ CO]. Anal. Calcd for C₂₀H₁₉NO₆W (553.2): C, 43.42; H, 3.46; N, 2.53. Found: C, 43.41; H, 3.29; N, 2.42. (3Z)-3a.



¹H NMR (CDCl₃, 25 °C): δ 7.40–7.35 (5 H, m, Ph), 6.01 (1 H, dd, ³*J* = 10.8 and 17.7 Hz, H₂C=*CH*), 4.92 (1 H, dd, ³*J* = 10.8 Hz, ²*J* = 1.2, *cis*-H H₂C=), 4.59 (1 H, dd, ³*J* = 17.7 Hz, ²*J* = 1.2, *trans*-H H₂C=), 3.90 and 3.32 (3 H each, s each, 2 NCH₃), 3.56 (2 H, q, ³*J* = 7 Hz, OCH₂), 1.17 (3 H, t, ³*J* = 7 Hz, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 254.6 (C_q, W=C), 203.5 and 198.7 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 146.7 (C_q, C4), 144.3 (C_q, C3), 135.7 (C_q, *i*-C Ph); 130.1, 128.4, and 127.1 (1: 2:2, CH each, Ph), 128.4 (CH, H₂C=*C*H), 113.6 (CH₂, H₂*C*=CH), 64.8 (OCH₂), 53.1 and 43.9 (1:1, NCH₃ each), 15.4 (OCH₂*C*H₃). IR (*n*-hexane; cm⁻¹ (%)): 2060.9 (20), 1966.0 (5), 1925.6 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 553 (20) [M⁺], 525 (15) [M⁺ - CO], 497 (20) [M⁺ - 2 CO], 469 (30) [M⁺ - 3 CO], 441 (45) [M⁺ - 4 CO], 413 (30) [M⁺ - 5 CO].

(3Z)-8a.



CH), 3.95 and 3.32 (3 H each, s each, 2 NCH₃), 1.76 (3 H, d, ${}^{3}J$ = 7 Hz, C–*C*H₃). ¹³C NMR (CDCl₃): δ 252.0 (C_q, W=C), 202.7 and 198.0 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 191.7 (C_q, C=O), 151.8 (C_q, *C*=CH), 137.0 (C_q, *i*-C Ph); 132.5, 129.7, and 128.4 (1:2:2, CH each, Ph), 128.7 (CH, =CH), 53.4 and 43.7 (NCH₃ each), 15.0 (C-*C*H₃). IR (*n*-hexane; cm⁻¹ (%)): 2064.1 (25), 1983.8 (3), 1971.8 (8), 1937.8 (100) (ν (C=O)), 1652.6 (5). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 497 (60) [M⁺ – CO], 469 (40) [M⁺ – 2 CO], 441 (55) [M⁺ – 3 CO], 413 (70) [M⁺ – 4 CO], 385 (90) [M⁺ – 5 CO]. Anal. Calcd for C₁₈H₁₅NO₆W (525.2): C, 41.17; H, 2.88; N, 2.67. Found: C, 41.22; H, 3.11; N, 2.54.

(3-Ethenyl-4-ethoxy-2-morpholino-4-phenyl)-1-(pentacarbonyltungsta)-1,3-butadiene ((*3E*)-3b and (3*Z*)-3b) and (3-Benzoyl-2-morpholino-1-(pentacarbonyltungsta)-1,3-pentadiene ((*3Z*)-8b). 1-(Morpholin-4-yl)but-2-en-1-one (2b; 310 mg, 2.00 mmol) was reacted with phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxybenzylidene)tungsten (1a; 458 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) as described above. Workup after 24 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (3/1) afforded a 20/1 mixture of compounds (3*E*)-3b and (3*Z*)-3b (384 mg, 65%, $R_f = 0.6$ in 2/1 *n*pentane/dichloromethane, yellow oil) and compound 8b (65 mg, 12%, $R_f = 0.6$ in 1/4 *n*-pentane/dichloromethane, yellow oil).

(3E)-3b. ¹H NMR (CDCl₃, 25 °C): δ 7.46–7.28 (5 H, m, Ph), 7.04 (1 H, dd, ${}^{3}J = 11.1$ and 18.3 Hz, H₂C=CH), 5.36 (1 H, d, ${}^{3}J = 11.1$ Hz, *cis*-H H₂C=), 4.88 (1 H, dd, ${}^{3}J = 18.3$ Hz, ${}^{2}J =$ 1.2, trans-H H₂C=); 4.61, 4.24, 3.77, and 3.50 (1 H each, m each, diastereotopic NCH₂); 3.93, 3.88, 3.49, and 3.37 (1 H each, m each, OCH2CH2N), 3.80 (2 H, m, diastereotopic OCH2-CH₃), 1.34 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 253.7 (C_q, W=C), 202.3 and 198.3 (C_q each, 1:4, trans- and cis-CO W-(CO)₅), 142.5 (C_q, C4), 136.2 (C_q, C3), 134.4 (C_q, *i*-C Ph); 129.4, 128.5, and 128.3 (1:2:2, CH each, Ph), 128.8 (CH, H₂C=CH), 117.3 (CH₂, H₂C=CH), 67.8 (OCH₂CH₃), 66.9 and 66.5 (OCH₂-CH₂ each), 62.2 and 53.6 (NCH₂ each), 15.3 (OCH₂CH₃). IR, MS, and CHN analytical data of a 20/1 mixture of compounds (3E)-**3b** and (3Z)-**3b** are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2062.9 (25), 1971.5 (5), 1935.5 (15), 1926.0 (100) (v(C≡O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 595 (30) [M⁺], 567 (20) [M⁺ - CO], 539 (55) $[M^+ - 2 \text{ CO}]$, 511 (100) $[M^+ - 3 \text{ CO}]$, 483 (20) $[M^+ - 4 \text{ CO}]$ CO], 455 (45) [M⁺ -5 CO]. Anal. Calcd for C₂₂H₂₁NO₇W (595.3): C, 44.39; H, 3.56; N, 2.35. Found: C, 43.96; H, 3.59; N, 2.34.

(3*Z*)-3**b.** ¹H NMR (CDCl₃, 25 °C): δ 7.46–7.28 (5 H, m, Ph), 6.04 (1 H, dd, ³*J* = 11.2 and 18.3 Hz, H₂C=C*H*), 4.93 (1 H, d, ³*J* = 11.2 Hz, *cis*-H H₂C=), 4.51 (1 H, dd, ³*J* = 18.3 Hz, *trans*-H H₂C=), 3.18 (2 H, m, diastereotopic OCH₂), morpholino,¹² 0.98 (3 H, t, OCH₂C*H*₃). ¹³C NMR (CDCl₃): δ (C_q, W=C),¹² 202.3 and 198.8 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 144.8 (C_q, C4), 135.2 (C_q, *i*-C Ph), 132.7 (C_q, C3); 129.9, 129.5, and 129.1 (1:2:2, CH each, Ph), 127.9 (CH, H₂C=*C*H), 114.0 (CH₂, H₂*C*= CH); 67.9, 67.3, and 64.9 (OCH₂ each), 64.8 and 53.1 (NCH₂ each), 15.1 (OCH₂*C*H₃).

(3Z)-8b. ¹H NMR (CDCl₃): δ 7.87 (2 H, m, *o*-H Ph), 7.59 (1 H, *p*-H Ph), 7.51 (2 H, m, *m*-H Ph), 5.79 (1 H, q, ³J = 6.9 Hz, =CH), 4.70, 4.30, 3.88, and 3.63 (1 H each, m each, 2 NCH₂), 4.05, 3.99, 3.98, and 3.81 (1 H each, m each, 2 OCH₂CH₂N), 1.78 (3 H, d, ³J = 6.9 Hz, CH₃). ¹³C NMR (CDCl₃): δ 252.4 (C_q, W=C), 202.4 and 197.7 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 191.6 (C_q, C=O), 150.8 (C_q, C3), 136.6 (C_q, *i*-C Ph), 132.8, 129.7, and 128.5 (1:2:2, CH each, Ph), 128.4 (CH, *C*H=C), 67.7 and 67.6 (2 O*C*H₂CH₂N), 62.7 and 54.2 (2 NCH₂), 15.1 (CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2064.1 (20), 1983.8 (6), 1971.6 (5), 1938.6 (80), 1925.6 (16) (ν (C=O)); 1648.6 (6) (ν (C=O)), 1462.6 (100). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 539 (80) [M⁺ - CO], 511 (50) [M⁺ - 2 CO], 483 (55) [M⁺ - 3 CO], 455 (70) [M⁺ - 4 CO], 427 (100) [M⁺ - 5 CO].

(2-(Dimethylamino)-3-ethenyl-4-ethoxy-6-phenyl)-1-(pentacarbonyltungsta)-1,3-hexadien-5-yne ((*3E*)-3c and

⁽¹²⁾ Signals not located due to line broadening.

(3*Z*)-3c). But-2-enoic acid dimethylamide (**2a**; 226 mg, 2.00 mmol) was reacted with phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxy-3-phenyl-2-propynylidene)-tungsten (**1b**; 482 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) as described above. Workup after 24 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (3/1) afforded a 25/1 mixture of compounds (3*E*)-**3c** and (3*Z*)-**3c** (299 mg, 52%, $R_f = 0.5$ in 2/1 *n*-pentane/dichloromethane, crystals from 10/1 *n*-pentane/dichloromethane; mp 71 °C).

(3E)-3c.



¹H NMR (CDCl₃, 25 °C): δ 7.45-7.33 (5 H, m, Ph), 6.89 (1 H, dd, ${}^{3}J = 11.2$ and 18.0 Hz, H₂C=CH), 5.23 (1 H, dd, ${}^{3}J = 11.2$, $^{2}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 18.0$ Hz, $^{2}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 18.0$ Hz, $^{2}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 18.0$ Hz, $^{2}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-H H₂C=), 4.79 (1 H, dd, $^{3}J = 1.2$ Hz, *cis*-Hz, 1.2, *trans*-H H₂C=), 4.12 (2 H, m, diastereotopic OCH₂), 3.90 and 3.28 (3 H each, s each, 2 NCH₃), 1.35 (3 H, t, ${}^{3}J = 7$ Hz, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 253.1 (C_q, W=C), 203.2 and 198.3 (Cq each, 1:4, trans- and cis-CO W(CO)₅), 139.9 (Cq, C4); 131.3, 129.1, and 128.5 (2:1:2, CH each, Ph), 127.2 (Cq, C3), 126.4 (CH, H₂C=CH), 121.7 (C_q, *i*-C Ph), 116.6 (CH₂, H₂C= CH), 102.0 and 82.2 (C_q each, C≡C), 65.7 (OCH₂), 53.3 and 43.2 (NCH₃ each), 15.2 (OCH₂CH₃). IR, MS, and CHN analytical data of a 25/1 mixture of compounds (3E)-3c and (3Z)-3c are as follows. IR (n-hexane; cm⁻¹ (%)): 2062.7 (30), 1972.2 (5), 1935.5 (100) (v(C≡O)), 1496.7 (5). MS (70 eV; ¹⁸⁴W, m/e (%)): 577 (25) $[M^+]$, 521 (10) $[M^+ - 2 \text{ CO}]$, 493 (100) $[M^+ - 3 \text{ CO}]$ CO], 465 (15) [M⁺ - 4 CO], 437 (20) [M⁺ - 5 CO]. Anal. Calcd for C₂₂H₁₉NO₆W (577.2): C, 45.78; H, 3.32; N, 2.43. Found: C, 45.48; H, 3.25; N, 2.38.





¹H NMR (CDCl₃, 25 °C): δ 7.45–7.33 (5 H, m, Ph), 6.20 (1 H, dd, ³*J* = 10.8 and 17.7 Hz, H₂C=*CH*), 5.06 (1 H, dd, ³*J* = 10.8 Hz, *cis*-H *H*₂C=CH), 4.66 (1 H, dd, ³*J* = 17.7 Hz, *trans*-H *H*₂C=CH), 3.95 (2 H, q, OCH₂), 3.84 and 3.21 (3 H each, s each, 2 NCH₃), 1.31 (3 H, t, OCH₂C*H*₃). ¹³C NMR (CDCl₃): δ (C_q, W=C), ¹² 202.5 and 198.3 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 139.0 (C_q, C3); 131.4 and 129.4 (2:2, CH each, Ph), 128.1 (C_q, C2), 126.7 (CH, H₂C=*C*H), 124.8 (C_q, *i*-C Ph), 115.1 (CH₂, H₂*C*=CH), 98.3 and 80.5 (C_q each, C=C), 65.2 (OCH₂), 52.8 and 43.3 (NCH₃ each), 15.5 (OCH₂*C*H₃).

(3-Ethenyl-4-ethoxy-2-morpholino-6-phenyl)-1-(pentacarbonyltungsta)-1,3-hexadien-5-yne ((*3E*)-3d and (*3Z*)-3d). 1-(Morpholin-4-yl)but-2-en-1-one (**2b**; 310 mg, 2.00 mmol) was reacted with phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxy-3-phenyl-2-propynylidene)tungsten (**1b**; 482 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) as described above. Workup after 24 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (3/1) afforded a 20/1 mixture of compounds (3*E*)-**3d** and (3*Z*)-**3d** (303 mg, 49%, $R_f = 0.4$ in 2/1 *n*-pentane/dichloromethane, yellow crystals from 10/1 *n*-pentane/dichloromethane; mp 80 °C).

(3E)-3d. ¹H NMR (CDCl₃, 25 °C): δ 7.46–7.35 (5 H, m, Ph), 6.89 (1 H, dd, ${}^{3}J = 10.8$ and 18.0 Hz, H₂C=CH), 5.25 (1 H, dd, ${}^{3}J = 10.8$ Hz, ${}^{2}J = 1.2$, cis-H H₂C=), 4.79 (1 H, dd, ${}^{3}J = 18.0$ Hz, ${}^{2}J = 1.2$, trans-H H₂C=), 4.59, 4.40, 3.82, and 3.64 (1 H each, m each, 2 NCH₂), 4.17 and 4.09 (1 H each, m each, diastereotopic OCH₂CH₃), 4.02, 3.98, 3.94, and 3.91 (1 H each, m each, 2 OCH₂CH₂N), 1.36 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 252.0 (C_q, W=C), 202.9 and 198.3 (C_q each, 1:4, trans- and cis-CO W(CO)₅), 138.6 (Cq, C4), 131.2, 129.3, and 128.6 (2:1:2, CH each, Ph), 127.5 (C_q , C3), 126.9 (CH, $H_2C=$ *C*H), 121.5 (C_q, *i*-C Ph), 116.7 (CH₂, H₂*C*=CH), 102.8 and 82.7 (C_q each, C=C), 67.7 and 67.4 (2 OCH₂CH₂N), 65.9 (OCH₂: CH₃), 62.4 and 53.6 (2 NCH₂), 15.2 (OCH₂CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2062.7 (30), 1972.3 (5), 1935.5 (100), 1925.9 (11) (v-(C=O)). MS (70 eV; 184 W, m/e (%)): 619 (20) [M⁺], 563 (10) [M⁺] - 2 CO], 535 (100) [M⁺ - 3 CO], 507 (15) [M⁺ - 4 CO]. Anal. Calcd for C₂₄H₂₁NO₇W (619.3): C, 46.55; H, 3.42; N, 2.26. Found: C, 46.40; H, 3.24; N, 2.11. X-ray crystal structure analysis of compound **3d**: formula $C_{24}H_{21}NO_7W$, $M_r = 619.27$, yellow crystal, $0.35 \times 0.25 \times 0.10$ mm, a = 18.690(1) Å, b =9.434(1) Å, c = 15.646(1) Å, $\beta = 101.12(1)^{\circ}$, V = 2706.9(4) Å³, $\rho_{\text{calcd}} = 1.520 \text{ g cm}^{-3}, \mu = 43.05 \text{ cm}^{-1}, \text{ empirical absorption}$ correction via SORTAV (0.314 $\leq T \leq$ 0.673), Z = 4, monoclinic, space group $P2_1/c$ (No. 14), $\lambda = 0.710$ 73 Å, T = 198 K, ω and φ scans, 10 651 reflections collected ($\pm h$, $\pm k$, $\pm l$), (sin θ)/ λ = 0.68 Å^{-1} , 6771 independent ($R_{int} = 0.046$) and 4916 observed reflections ($I \ge 2\sigma(I)$), 299 refined parameters, R1 = 0.057, wR2 = 0.144, maximum (minimum) residual electron density 4.46 (-1.77) e Å³, very high remaining electron density around 0, 0, 0 and 0, 0.5, 0.5 that is perhaps disordered petroleum ether, hydrogens calculated and refined as riding atoms.¹³

(3*Z*)-3d. Some typical spectroscopic data which could be collected from the isomer mixture are as follows. ¹H NMR (CDCl₃, 25 °C): δ 6.22 (1 H, dd, ³*J* = 10.8 and 18.0 Hz, CH), 5.09 (1 H, dd, ³*J* = 10.8 Hz, *cis*-H H₂C=), 4.58 (1 H, d, ³*J* = 17.7 Hz, *trans*-H H₂C=), 1.31 (3 H, t, ³*J* = 7 Hz, OCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): 202.1 and 198.3 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 131.7, 131.5, 129.2, 128.5, 128.1 and 123.9 (alkenyl and phenyl), 67.7, 67.3, 65.2, 62.0 and 53.8 (OCH₂ and NCH₂).

(2-(Dimethylamino)-3-ethenyl-4-ethoxy-6-phenyl)-1-(pentacarbonyltungsta)-1,3,5-hexatrienes ((*3E*,5*E*)-3e and (*3Z*,5*E*)-3e). But-2-enoic acid dimethylamide (2a; 226 mg, 2.00 mmol) was reacted with phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxy-3-phenyl-2-propenylidene)tungsten (1c; 484 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) as described above for 2 h. Immediate workup by flash column chromatography on silica gel with *n*-pentane/ dichloromethane (3/1) afforded a 12/1 mixture of compounds (3*E*,5*E*)-3e and (3*Z*,5*E*)-3e (255 mg, 44%, $R_f = 0.6$ in 2/1 *n*-pentane/dichloromethane, yellow oil).

(3*E*,5*E*)-3e.



¹H NMR (CDCl₃, 25 °C): δ 7.42–7.25 (5 H, m, Ph), 6.83 (1 H,

⁽¹³⁾ Data sets were collected with a Nonius KappaCCD diffractometer, equipped with a Nonius FR591 rotating anode generator. Programs used: data collection COLLECT (Nonius BV, 1998), data reduction Denzo-SMN (Otwinowski, Z.; Minor, W. Methods Enzymol. **1997**, 276, 307–326), absorption correction SORTAV (Blessing, R. H. Acta Crystallogr. **1995**, A51, 33–37. Blessing, R. H. J. Appl. Crystallogr. **1997**, 30, 421–426), structure solution SHELXS-97 (Sheldrick, G. M. Acta Crystallogr. **1990**, A46, 467–473), structure refinement SHELXL-97 (Sheldrick, G. M. Universität Göttingen, Göttingen, Germany, 1997), graphics DIAMOND (Brandenburg, K. Universität Bonn, Bonn, Germany, 1997).

d, ${}^{3}J = 15.6$ Hz, 5-H), 6.79 (1 H, dd, ${}^{3}J = 10.8$ and 18.0 Hz, $H_2C=CH$, 6.35 (1 H, d, ${}^{3}J=15.6$ Hz, 6-H), 5.30 and 4.82 (1 H each, dd each, ³*J* = 10.8 and 18.0 Hz, ²*J* = 0.9, *H*₂C=CH), 4.01 and 3.88 (1 H each, m each, diastereotopic 4-OCH₂), 3.94 and 3.21 (3 H each, s each, 2 NCH₃), 1.43 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): 255.6 (C_q, W=C), 202.8 and 198.5 (C_q, 1:4, trans- and cis-CO W(CO)_5), 142.6 (Cq, C-4), 139.9 (Cq, C-3), 136.7 (Cq, *i*-C Ph), 130.1 and 121.6 (CH each, HC=CHPh), 128.9, 128.3, and 127.0 (2:1:2, CH each, Ph), 127.7 (H₂C=CH), 117.2 (H₂C=CH), 69.4 (OCH₂), 53.4 and 43.6 (NCH₃ each), 15.6 (OCH₂CH₃). IR and MS data of a 12/1 mixture of (3E,5E)-3e and (3Z,5E)-3e are as follows. IR (n-hexane; cm⁻¹ (%)): 2062.5 (5), 1983.8 (5), 1935.3 (15), 1927.1 (100) (v(C≡O)). MS (70 eV; 184 W, m/e (%)): 579 (5) [M⁺], 551 (5) [M⁺ - 1 CO], 523 (10) [M⁺ - 2 CO], 495 (10) [M⁺ - 3 CO], 467 (20) [M⁺ - 4 CO], 439 (20) $[M^+ - 5 CO], 255 (100).$

(3E,5E)-3e.



¹H NMR (CDCl₃, 25 °C): δ 7.42–7.20 (5 H, m, Ph), 6.89 (1 H, d, ³*J* = 15.6 Hz, 5-H), 6.60 (1 H, d, ³*J* = 15.6 Hz, 6-H), 6.58 (1 H, dd, ³*J* = 11.1 and 17.4 Hz, H₂C=C*H*), 5.15 and 4.71 (1 H each, dd each, ³*J* = 11.1 and 17.4 Hz, ²*J* = 0.9 Hz, *H*₂C=CH), 3.87 (2 H, m, diastereotopic OCH₂), 3.86 and 3.34 (3 H each, s each, 2 NCH₃), 1.44 (3 H, t, OCH₂C*H*₃). ¹³C NMR (CDCl₃, 25 °C; partial assignment only, due to overlapping signals): 251.4 (C_q, W=C), 202.8 and 198.6 (C_q, 1:4, *trans*- and *cis*-CO W(CO)₅), 132.2, 131.4, 128.4, 126.8, 126.2, 119.8, 117.5 (*C*=*C* and Ph), 69.2 (OCH₂), 53.4 and 44.3 (NCH₃ each), 14.6 (OCH₂*C*H₃).

3-Acetyl-2-(dimethylamino)-1-(pentacarbonyltungsta)-1,3-pentadiene ((3*Z***)-12a).** But-2-enoic acid dimethylamide (**2a**; 113 mg, 1.00 mmol) was reacted with phosphorus oxychloride (153 mg, 1.00 mmol), pentacarbonyl(1-ethoxy-1-ethylidene)tungsten (**1d**; 198 mg, 0.50 mmol), and triethylamine (202 mg, 2.00 mmol) as described above. Workup after 12 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (1/2) afforded compound (3*Z*)-**12a** (153 mg, 66%, R_t = 0.4 *n*-pentane/dichloromethane (1/2), yellow crystals from 10/1 *n*-pentane/dichloromethane at -20 °C, mp 121-122 °C).

(3*Z*)-12a.



¹H NMR (CDCl₃, 25 °C): δ 6.10 (1 H, q, ³J = 6.9 Hz, =CH), 3.87 and 3.18 (3 H each, s each, 2 NCH₃), 2.38 (3 H, s, CH₃-CO), 1.73 (3 H, d, ${}^{3}J$ = 6.9 Hz, CH₃). 13 C NMR (CDCl₃): δ 253.3 (C_q, W=C), 203.0 and 198.1 (C_q each, 1:4, trans- and cis-CO W(CO)₅), 195.3 (C_q, C=O), 153.8 (C_q, C3), 126.1 (CH, =CH), 53.5 and 43.5 (NCH₃ each), 26.1 (CH₃CO), 15.2 (CH₃). IR (nhexane; cm⁻¹ (%)): 2064.1 (20), 1983.8 (10), 1931.5 (100) (v-(C≡O)). MS (70 eV; ¹⁸⁴W, *m*/*e* (%)): 463 (5) [M⁺], 435 (45) [M⁺ - CO], 407 (20) [M⁺ - 2 CO], 379 (50) [M⁺ - 3 CO], 351 (60) $[M^+-4$ CO], 323 (60) $[M^+-5$ CO]. Anal. Calcd for $C_{13}H_{13}\text{-}$ NO₆W (463.1): C, 33.72; H, 2.83; N, 3.02. Found: C, 33.67; H, 2.55; N, 2.94. X-ray crystal structure analysis of compound **12a**: formula $C_{13}H_{13}NO_6W$, $M_r = 463.09$, yellow crystal 0.40 $\times 0.25 \times 0.20$ mm, a = 10.603(1) Å, b = 11.790(1) Å, c =12.614(1) Å, $\beta = 94.36(1)^{\circ}$, V = 1572.3(2) Å³, $\rho_{calcd} = 1.956$ g cm⁻³, $\mu = 73.71$ cm⁻¹, empirical absorption correction via SORTAV (0.157 $\leq T \leq 0.320$), Z = 4, monoclinic, space group $P2_1/c$ (No. 14), $\lambda = 0.710$ 73 Å, T = 298 K, ω and φ scans, 6607 reflections collected ($\pm h$, $\pm k$, $\pm h$, (sin θ)/ $\lambda = 0.66$ Å⁻¹, 3723 independent ($R_{int} = 0.018$) and 3331 observed reflections ($I \geq 2\sigma(I)$), 194 refined parameters, R1 = 0.022, wR2 = 0.048, maximum (minimum) residual electron density 0.60 (-1.06) e Å⁻³, hydrogens calculated and refined as riding atoms.¹³

3-Acetyl-2-morpholino-1-(pentacarbonyltungsta)-1,3pentadiene (12b). 1-(Morpholin-4-yl)-but-2-en-1-one (**2b**; 155 mg, 1.00 mmol) was reacted with phosphorus oxychloride (153 mg, 1.00 mmol), pentacarbonyl(1-ethoxy-1-ethylidene)tungsten (**1d**; 198 mg, 0.50 mmol), and triethylamine (202 mg, 2.00 mmol) as described above. Workup after 12 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (2/1) afforded the compound (3*Z*)-**12b** (148 mg, 59%, $R_f = 0.4$ in dichloromethane, yellow crystals from 10/1 *n*-pentane/dichloromethane at -20 °C, mp 119–120 °C).

12b. ¹H NMR (CDCl₃, 25 °C): δ 6.11 (1 H, q, ³J = 7 Hz, =CH), 4.60, 4.27, 3.80, and 3.54 (1 H each, m each, 2 NCH₂), 4.01 and 3.73 (2 H each, m each, 2 OCH₂CH₂N), 2.40 (3 H, s, CH₃CO), 1.75 (3 H, d, ³J = 7 Hz, 4-CH₃). ¹³C NMR (CDCl₃): δ 250.7 (C_q, W=C), 202.7 and 197.8 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 195.4 (C_q, C=O), 152.9 (C_q, C3), 126.2 (CH, H₂C=CH), 67.7 and 67.3 (2 OCH₂CH₂N), 62.4 and 53.8 (2 NCH₂), 26.1 (CH₃CO), 15.4 (4-CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2062.0 (75), 1975.3 (30), 1908.9 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m*/*e* (%)): 505 (5) [M⁺], 477 (70) [M⁺ – CO], 449 (35) [M⁺ – 2 CO], 421 (85) [M⁺ – 3 CO], 393 (80) [M⁺ – 4 CO], 365 (100) [M⁺ – 5 CO]. Anal. Calcd for C₁₅H₁₅NO₇W (505.1): C, 35.67; H, 2.99; N, 2.77. Found: C, 35.53; H, 2.60; N, 2.66.

(4E)-1-(Dimethylamino)-4-ethoxy-3-(syn-ethylidene)-1-(pentacarbonyltungsta)-1,4-hexadiene ((4Z,syn)-9c), 2-(Dimethylamino)(2-ethoxy-3,4-dimethyl-1-cyclobutenyl)-1-(pentacarbonyltungsta)ethenes $(3S^*, 4R^*)$ -10c and $(3R^*, 4R^*)$ -10c, and 2-(Dimethylamino)-3-ethylidene-1-(pentacarbonyltungsta)-1-hexen-4-one (12c). But-2-enoic acid dimethylamide (2a; 113 mg, 1.00 mmol) was reacted with phosphorus oxychloride (153 mg, 1.00 mmol), pentacarbonyl-(1-ethoxy-1-propylidene)tungsten (1e; 205 mg, 0.50 mmol), and triethylamine (202 mg, 2.00 mmol) as described above. Workup after 12 h at 25 °C by flash column chromatography on silica gel with n-pentane/dichloromethane (2/1) afforded a 5/3 mixture of compounds $(3S^*, 4R^*)$ -10c and $(3R^*, 4R^*)$ -10c (72 mg, 29%, $R_f = 0.8$ *n*-pentane/dichloromethane 2:1, yellow oil), compound (4*Z*,*syn*)-**9c** (56 mg, 22%, $R_f = 0.5$ in 2/1 *n*-pentane/ dichloromethane, yellowish powder from n-pentane, mp 57-58 °C), and compound **12c** (19 mg, 8%, $R_f = 0.4 \ 1/2 \ n$ -pentane/ dichloromethane, yellow crystals from 10/1 n-pentane/dichloromethane at -20 °C, mp 121-122 °C).

(4*Z*,*syn*)-9c.



¹H NMR (CDCl₃, 25 °C): δ 5.32 (1 H, q, ³*J* = 7 Hz, 1'-H), 4.46 (1 H, q, ³*J* = 7 Hz, 5-H), 3.88 and 3.24 (3 H each, s each, 2 NCH₃), 3.86 (2 H, m, diastereotopic OCH₂), 1.71 (3 H, d, ³*J* = 7 Hz, 6-H₃), 1.55 (3 H, d, ³*J* = 7 Hz, 2'-H₃), 1.36 (3 H, t, ³*J* = 7 Hz, OCH₂C*H*₃). ¹³C NMR (CDCl₃): δ 256.3 (C_q, W=C), 202.5 and 198.5 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 148.8 (C_q, C4), 147.5 (C_q, C3), 114.2 (CH, C5), 110.4 (CH, C1'), 67.0 (OCH₂), 53.3 and 42.8 (NCH₃ each), 15.6 (OCH₂*C*H₃), 14.1 (CH₃, C2'), 11.1 (CH₃, C6). IR (*n*-hexane; cm⁻¹ (%)): 2062.6 (20), 1969.1 (5), 1934.1 (98), 1926.4 (15) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m*/*e* (%)): 505 (10) [M⁺], 477 (10) [M⁺ - CO], 449 (25) [M⁺ - 2 CO], 365 (35) [M⁺ - 5 CO]. Anal. Calcd for C₁₆H₁₉-NO₆W (505.2): C, 38.04; H, 3.79; N, 2.77. Found: C, 38.01; H, 3.58; N, 2.60.

(3S*,4R*)-10c.



¹H NMR (CDCl₃, 25 °C): δ 3.94 (2 H, m, diastereotopic OCH₂), 3.77 and 3.41 (3 H each, s each, 2 NCH₃), 3.33 (1 H, qd, ${}^{3}J =$ 6.6 and 4.8 Hz, 3'-H), 3.25 (1 H, qd, ${}^{3}J = 7$ and 4.8 Hz, 4'-H), 1.30 (3 H, t, OCH₂CH₃), 1.15 (3 H, d, ${}^{3}J$ = 7 Hz, 4'-CH₃), 1.08 (3 H, d, ${}^{3}J$ = 6.6 Hz, 3'-CH₃). ${}^{13}C$ NMR (CDCl₃): δ 241.9 (C_a, W=C), 203.5 and 199.0 (C_q each, 1:4, *trans-* and *cis-*CO W(CO)₅), 143.7 (Cq, C2'), 128.7 (Cq, C1'), 64.8 (OCH₂), 53.5 and 44.8 (NCH3 each), 38.0 (CH, C4'), 37.3 (CH, C3'), 15.6 (OCH2CH3), 15.4 (4'-CH3), 13.6 (3'-CH3). IR, MS, and CHN analytical data of a 5/3 mixture of compounds $(3S^*, 4R^*)$ -10c and $(3R^*, 4R^*)$ -10c are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2059.9 (15), 1965.8 (1), 1927.6 (98) (ν (C=O)). MS (70 eV; ¹⁸⁴W, m/e (%)): 505 (35) [M⁺], 477 (40) [M⁺ - CO], 449 (30) [M⁺ -2 CO], 393 (40) [M⁺ - 4 CO]. Anal. Calcd for C₁₆H₁₉NO₆W (505.2): C, 38.04; H, 3.79; N, 2.77. Found: C, 37.97; H, 3.90; N, 2.81.

(3R*,4R*)-10c.



¹H NMR (CDCl₃, 25 °C): δ 3.93 (2 H, m, diastereotopic OCH₂), 3.80 and 3.40 (3 H each, s each, 2 NCH₃), 2.74 (1 H, qd, ³*J* = 6.6 and 1.2 Hz, 4'-H), 2.56 (1 H, qd, ³*J* = 6.9 and 1.2 Hz, 3'-H), 1.37 (3 H, d, ³*J* = 6.9 Hz, 3'-CH₃), 1.33 (3 H, t, OCH₂C*H*₃), 1.18 (3 H, d, ³*J* = 6.6 Hz, 4'-CH₃). ¹³C NMR (CDCl₃): δ 243.7 (C_q, W=C), 203.3 and 198.9 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 143.6 (C_q, C2'), 126.3 (C_q, C1'), 64.6 (OCH₂), 53.4 and 45.1 (NCH₃ each), 44.3 (CH, C3'), 42.5 (CH, C4'), 19.7 (4'-CH₃), 16.3 (3'-CH₃), 15.7 (OCH₂CH₃).

12c.



¹H NMR (CDCl₃ 25 °C): δ 6.10 (1 H, q, ³*J* = 7 Hz, 1'-H), 3.91 and 3.22 (3 H each, s each, 2 NCH₃), 2.79 (2 H, m, diastereotopic CH₂CO), 1.75 (3 H, d, ³*J* = 7 Hz, 2'-H₃), 1.17 (3 H, t, ³*J* = 7 Hz, 6-H₃). ¹³C NMR (CDCl₃): δ 253.6 (C_q, W=C), 202.9 and 198.1 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 198.0 (C_q, C=O), 153.3 (C_q, C3), 124.5 (CH, C1'), 53.3 and 43.5 (NCH₃ each), 31.5 (CH₂, C5), 8.1 (CH₃, C6), 15.1 (CH₃, C2'). IR (*n*-hexane; cm⁻¹ (%)): 2063.9 (20), 1970.5 (2), 1931.6 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 477 (5) [M⁺], 449 (20) [M⁺ - CO], 421 (10) [M⁺ - 2 CO], 393 (25) [M⁺ - 3 CO], 365 (30) [M⁺ - 4 CO], 337 (25) [M⁺ - 5 CO]. Anal. Calcd for C₁₄H₁₅-NO₆W (477.1): C, 35.24; H, 3.17; N, 2.94. Found: C, 35.16; H, 3.00; N, 2.90.

(4*E*)-4-Ethoxy-3-(*syn*-ethylidene)-2-morpholino-1-(pentacarbonyltungsta)-1,4-hexadiene ((4*Z*,*syn*)-9d), 2-(2-Ethoxy-3,4-dimethyl-1-cyclobutenyl)-2-morpholino-1-(pentacarbonyltungsta)ethenes (3*S**,4*R**)-10d and (3*R**,4*R**)-10d, and 3-Ethylidene-2-morpholino-1-(pentacarbonyltungsta)-1-hexen-4-one (12d). 1-(Morpholin-4-yl)but-2-en-1-one (2b; 155 mg, 1.00 mmol) was reacted with phosphorus oxychloride (153 mg, 1.00 mmol), pentacarbonyl(1-ethoxy-1propenylidene)tungsten (1e; 205 mg, 0.50 mmol), and triethylamine (202 mg, 2.00 mmol) as described above. Workup after 12 h at 25 °C by flash column chromatography on silica gel with *n*-pentane/dichloromethane (2/1) afforded a 5/1 mixture of compounds $(3S^*, 4R^*)$ -**10d** and $(3R^*, 4R^*)$ -**10d** (42 mg, 15%, $R_f = 0.6$ in 1/1 *n*-pentane/dichloromethane, yellow oil), compound (4*Z*, *syn*)-**9d** (47 mg, 17%, $R_f = 0.6$ in 1/1 *n*-pentane/dichloromethane), and compound **12d** (52 mg, 20%, $R_f = 0.3$ in 1/2 *n*-pentane/dichloromethane).

(4Z,syn)-9d. ¹H NMR (CDCl₃, 25 °C): δ 5.40 (1 H, q, ³J = 7 Hz, 1'-H), 4.52 (1 H, q, ³J = 7 Hz, 5-H), 4.54, 4.25, 3.82, and 3.70 (1 H each, m each, 2 NCH₂), 3.96, 3.80, and 3.78 (2 H each, m each, diastereotopic OCH₂CH₃ and 2 OCH₂CH₂N), 1.75 (3 H, d, ³J = 7 Hz, 6-H₃), 1.63 (3 H, d, ³J = 7 Hz, 2'-H₃), 1.40 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 255.0 (C_q, W=C), 202.6 and 198.4 (C_q each, 1:4, *trans-* and *cis-*CO W(CO)₅), 149.4 (C_q, C4), 146.4 (C_q, C3), 114.9 (CH, C5), 111.1 (CH, C1'), 67.8 and 67.2 (OCH₂CH₂N each), 67.1 (OCH₂CH₃), 62.5 and 53.0 (NCH₂ each), 15.7 (OCH₂CH₃), 14.7 and 11.3 (CH₃ each, C2' and C6). IR (*n*-hexane; cm⁻¹ (%)): 2062.6 (15), 1968.4 (2), 1934.7 (90) (ν(C≡O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 547 (25) [M⁺], 519 (10) [M⁺ − CO], 491 (35) [M⁺ − 2 CO], 463 (10) [M⁺ − 3 CO], 407 (30) [M⁺ − 5 CO]. Anal. Calcd for C₁₈H₂₁NO₇W (547.2): C, 39.51; H, 3.87; N, 2.56. Found: C, 39.86; H, 3.72; N, 2.47.

(3S*,4R*)-10d. ¹H NMR (CDCl₃, 25 °C): δ 4.43, 4.23, 3.86, and 3.76 (1 H each, m each, 2 NCH2), 3.97, 3.95, and 3.84 (2 H each, m each, diastereotopic OCH₂CH₃ and 2 OCH₂CH₂N), 3.38 (1 H, qd, ${}^{3}J = 5$ Hz and 6.8, 3'-H), 3.27 (1 H, qd, ${}^{3}J = 5$ and 7 Hz, 4'-H), 1.28 (3 H, t, OCH₂CH₃), 1.12 (3 H, d, ${}^{3}J = 7$ Hz, 4'-CH₃), 1.04 (3 H, d, ${}^{3}J = 6.8$ Hz, 3'-CH₃). ${}^{13}C$ NMR (CDCl₃): δ 240.4 (C_q, W=C), 203.1 and 198.6 (C_q each, 1:4, trans- and cis-CO W(CO)₅), 142.6 (Cq, C2'), 127.8 (Cq, C1'), 67.9 and 67.8 (2 OCH2CH2N), 64.8 (OCH2CH3), 62.7 and 54.4 (2 NCH2), 42.6 (CH, C4'), 37.6 (CH, C3'), 15.5 (OCH2CH3), 15.4 and 13.7 (3'-CH₃ and 4'-CH₃). IR, MS, and CHN analytical data of a 5/1 mixture of compounds $(3S^*, 4R^*)$ -10d and $(3R^*, 4R^*)$ -**10d** are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2060.2 (15), 1934.5 (2), 1929.0 (100) (v(C≡O)). MS (70 eV; ¹⁸⁴W, m/e (%)): 547 (50) $[M^+]$, 519 (40) $[M^+ - CO]$, 491 (35) $[M^+ - 2 CO]$, 463 (30) $[M^+$ - 3 CO], 407 (100) [M⁺ - 5 CO]. Anal. Calcd for C₁₈H₂₁NO₇W (547.2): C, 39.51; H, 3.87; N, 2.56. Found: C, 39.94; H, 3.88; N, 2.98.

(3*R**,4*R**)-10d. ¹H NMR (CDCl₃, 25 °C): δ 4.32, 4.09, 3.90, and 3.74 (1 H each, m each, 2 NCH₂), 4.00–3.70 (6 H, m, diastereotopic OC*H*₂CH₃ and 2 OC*H*₂CH₂N), 2.90 (1 H, qd, ³*J* = 7 and 1.2 Hz, 3'-H), 2.77 (1 H, qd, ³*J* = 6.8 and 1.2 Hz, 4'-H), 1.37 (3 H, d, ³*J* = 6.8 Hz, 3'-CH₃), 1.29 (3 H, t, ³*J* = 7 Hz, OCH₂CH₃), 1.15 (3 H, d, ³*J* = 6.8 Hz, 4'-CH₃). ¹³C NMR (CDCl₃, 25 °C): δ 242.5 (C_q, W=C), 203.0 and 198.7 (C_q each, 1:4, *trans*-and *cis*-CO W(CO)₅), 142.9 (C_q, C2'), 125.2 (C_q, C1'), 67.9 and 67.8 (2 O*C*H₂CH₂N), 64.7 (O*C*H₂CH₃), 62.8 and 54.8 (2 NCH₂), 44.5 (CH, C3'), 38.1 (CH, C4'), 19.7 and 16.8 (3'-CH₃ and 4'-CH₃), 15.6 (OCH₂*C*H₃).

12d. ¹H NMR (CDCl₃, 25 °C): δ 6.08 (1 H, q, ³*J* = 7 Hz, 1'-H), 4.60, 4.29, 3.79, and 3.57 (1 H each, m each, 2 NCH₂), 4.01 and 3.72 (2 H each, m each, 2 OC*H*₂CH₂N), 2.79 (2 H, q, ³*J* = 7 Hz, CH₂CO), 1.74 (3 H, d, ³*J* = 7 Hz, 2'-H₃), 1.13 (3 H, t, ³*J* = 7 Hz, 6-H₃). ¹³C NMR (CDCl₃): δ 251.0 (C_q, W=C), 202.7 and 197.9 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 198.3 (C_q, C=O), 152.4 (C_q, C3), 124.7 (CH, C1'), 68.1 and 67.8 (O*C*H₂-CH₂N each), 62.5 and 53.8 (NCH₂ each), 31.6 (*C*H₂CO), 15.3 (CH₃, C2'), 8.0 (*C*H₃CH₂CO). IR (*n*-hexane; cm⁻¹ (%)): 2063.9 (22), 1933.9 (100) (ν (C=O)), 1653.0 (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 519 (5) [M⁺], 491 (30) [M⁺ - CO], 463 (15) [M⁺ - 2 CO], 435 (40) [M⁺ - 3 CO], 407 (30) [M⁺ - 4 CO], 379 (40) [M⁺ - 5 CO]. Anal. Calcd for C₁₅H₁₅NO₇W (519.2): C, 37.02; H, 3.30; N, 2.70. Found: C, 37.61; H, 3.73; N, 2.45.

(4*E*)-2-(Dimethylamino)-4-ethoxy-3-(*syn*-ethylidene)-1-(pentacarbonyltungsta)-1,4-octadiene ((4*Z*,*syn*)-9e), 2-(Dimethylamino)-2-(2-ethoxy-4-methyl-3-(*n*-propyl)-1-cyclobutenyl)-1-(pentacarbonyltungsta)ethenes (3*S**,4*R**)-10e and (3*R**,4*R**)-10e, and 2-(Dimethylamino)-3-ethylidene-1-(pentacarbonyltungsta)-1-octen-4-one (12e). But-2-enoic acid dimethylamide (**2a**; 226 mg, 2.00 mmol), phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxy-1-pentenylidene)tungsten (**1f**; 438 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) were reacted as described above to give a 4/1 mixture of compounds ($3S^*, 4R^*$)-**10e** and ($3R^*, 4R^*$)-**10e** (168 mg, 32%, $R_f = 0.6$ in 2/1 *n*-pentane/dichloromethane, yellow oil), compound (4Z, syn)-**9e** (106 mg, 20%, $R_f = 0.4$ in 2/1 *n*-pentane/dichloromethane), and compound **12e** (30 mg, 6%, $R_f = 0.5$ in 1/2 *n*-pentane/dichloromethane).

(4Z,syn)-9e.



¹H NMR (CDCl₃ 25 °C): δ 5.32 (1 H, q, ³*J* = 7 Hz, 1'-H), 4.37 (1 H, t, ³*J* = 7 Hz, 5-H), 3.87 and 3.24 (3 H each, s each, 2 NCH₃), 3.83 (2 H, m, diastereotopic OC*H*₂CH₃), 2.16 and 1.25 (2 H each, m each, 6-H₂ and 7-H₂), 1.72 (3 H, d, ³*J* = 7 Hz, 2'-H₃), 1.35 (3 H, t, OCH₂*C*H₃), 0.92 (3 H, t, ³*J* = 7 Hz, 8-H₃). ¹³C NMR (CDCl₃): δ 256.8 (C_q, W=C), 202.7 and 198.1 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 148.0 (C_q, C4), 147.6 (C_q, C3), 119.7 (CH, C5), 110.7 (CH, C1'), 67.5 (OCH₂), 53.3 and 42.7 (NCH₃ each), 27.8, 22.5, and 14.0 (CH₂CH₂CH₃), 15.6 (OCH₂*C*H₃), 14.3 (CH₃, C2'). IR (*n*-hexane; cm⁻¹ (%)): 2062.2 (20), 1967.7 (2), 1933.0 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 533 (30) [M⁺], 477 (10) [M⁺ - 2 CO], 421 (20) [M⁺ - 4 CO], 393 (20) [M⁺ - 5 CO], 164 (100). Anal. Calcd for C₁₈H₂₃-NO₆W (533.2): C, 40.54; H, 4.35; N, 2.63. Found: C, 40.76; H, 4.21; N, 2.63.

(3S*,4R*)-10e. ¹H NMR (CDCl₃, 25 °C): δ 3.86 (2 H, m, diastereotopic OCH₂), 3.76 and 3.41 (3 H each, s each, 2 NCH₃), 3.34 (1 H, qd, ${}^{3}J = 6.9$ and 4.8 Hz, 4'-H), 3.08 (1 H, qd, ${}^{3}J =$ 5.1 and 4.8 Hz, 3'-H), 1.60-1.20 (4 H, m, 3'-CH₂CH₂CH₃), 1.26 (3 H, t, OCH₂CH₃), 1.08 (3 H, t, ${}^{3}J = 7$ Hz, 3'-CH₂CH₂CH₃), 1.06 (3 H, d, ${}^{3}J$ = 6.9 Hz, 4'-CH₃). 13 C NMR (CDCl₃): δ 243.5 (Cq, W=C), 203.4 and 199.0 (Cq each, 1:4, trans- and cis-CO W(CO)₅), 142.0 (C_q, C2'), 129.2 (C_q, C1'), 64.8 (OCH₂), 53.2 and 44.8 (NCH3 each), 43.7 (CH, C3'), 37.5 (CH, C4'); 30.8, 21.4, and 14.3 (3'-CH2CH2CH3), 15.6 (OCH2CH3), 15.3 (4'-CH3). IR, MS, and CHN analytical data of a 4/1 mixture of compounds (3*S**,4*R**)-10e and (3*R**,4*R**)-10e are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2060.1 (15), 1966.5 (2), 1928.0 (100) (ν (C=O)). MS (70 eV; 184 W, m/e (%)): 533 (35) [M⁺], 505 (30) [M⁺ - CO], 477 (30) [M⁺ - 2 CO], 421 (50) [M⁺ - 4 CO], 393 (100) [M⁺ - 5 CO]. Anal. Calcd for C₁₈H₂₃NO₆W (533.2): C, 40.54; H, 4.35; N, 2.63. Found: C, 40.52; H, 4.52; N, 2.75.

(3*R**,4*R**)-10e. ¹H NMR (CDCl₃, 25 °C): δ 3.82 (2 H, m, diastereotopic OCH₂), 3.77 and 3.39 (3 H each, s each, 2 NCH₃), 2.78 (1 H, qd, ³*J* = 6.6 and 1.2 Hz, 4'-H), 2.45 (1 H, qd, ³*J* = 5.1 and 1.2 Hz, 3'-H), 1.60–1.20 (4 H, m, 3'-CH₂CH₂CH₃), 1.29 (3 H, t, OCH₂CH₃), 1.17 (3 H, d, ³*J* = 6.6 Hz, 4'-CH₃), 0.96 (3 H, t, ³*J* = 7 Hz, 3'-CH₂CH₂CH₃). ¹³C NMR (CDCl₃): δ 244.5 (C_q, W=C), 202.5 and 199.0 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 143.1 (C_q, C2'), 126.7 (C_q, C1'), 64.9 (OCH₂), 53.5 and 45.1 (NCH₃ each), 49.7 (CH, C3'), 41.1 (CH, C4'), 34.7, 21.5, and 14.4 (3'-CH₂CH₂CH₃), 20.3 (4'-CH₃), 16.0 (OCH₂CH₃).

12e.



¹H NMR (CDCl₃, 25 °C): δ 6.9 (1 H, q, ³*J* = 7 Hz, CH), 3.88 and 3.17 (3 H each, s each, 2 NCH₃), 2.73, 1.63, and 1.36 (2 H each, m each, CH₂CH₂CH₂CH₃), 1.72 (3 H, d, ³*J* = 7 Hz, CH₃), 0.93 (3 H, t, ³*J* = 7 Hz, (CH₂)₃CH₃). ¹³C NMR (CDCl₃, 25 °C):

δ 253.9 (C_q, W=C), 202.9 and 198.2 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 197.7 (C_q, C=O), 153.5 (C_q, C=CH), 124.9 (CH, C=CH), 53.3 and 43.4 (NCH₃ each), 37.9, 26.3, 22.4, and 13.8 (CH₂CH₂CH₂CH₃), 15.1 (CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2063.7 (20), 1971.1 (3), 1931.0 (100) (ν(C=O)), 1675.4 (6), 1539.3 (5). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 505 (5) [M⁺], 477 (20) [M⁺ - CO], 449 (10) [M⁺ - 2 CO], 421 (30) [M⁺ - 3 CO], 393 (20) [M⁺ - 4 CO], 365 (10) [M⁺ - 5 CO].

(3E)-4-Ethoxy-3-ethenyl-2-morpholino-1-(pentacarbonyltungsta)-1,3-octadiene ((3E)-3f), (4E)-4-Ethoxy-3-(synethylidene)-2-morpholino-1-(pentacarbonyltungsta)-1,4octadiene ((4Z,syn)-9f), 2-(2-Ethoxy-4-methyl-3-(n-propyl)-1-cyclobutenyl)-2-morpholino-1-(pentacarbonyltungsta)ethenes (3.5*, 4.R*)-10f and (3.R*, 4.R*)-10f, and 3-Ethylidene-2-morpholino-1-(pentacarbonyltungsta)-1-octen-4-one (12f). 1-(Morpholin-4-yl)but-2-en-1-one (2b; 310 mg, 2.00 mmol) was reacted with phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxy-1-pentenylidene)tungsten (1f; 438 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) as described above to give a 20/1 mixture of compounds $(3S^*, 4R^*)$ -10f and $(3R^*, 4R^*)$ -10f (64 mg, 11%, $R_f = 0.5$ in 2/1 *n*-pentane/dichloromethane, yellow oil), compound (3*E*)-**3f** (112 mg, 20%, $R_f = 0.3$ in 3/2 *n*-pentane/dichloromethane), and compound **12f** (100 mg, 18%, $R_f = 0.5$ in 1/2 *n*-pentane/ dichloromethane). According to NMR spectra, compound (3E)-**3f** in C_6D_6 is smoothly transformed into compound (4*Z*)-**9f**. (3*E*)-3f.



¹H NMR (CDCl₃, 25 °C): δ 6.81 (1 H, dd, ³J = 18.0 and 11.4 Hz, 1'-H), 5.15 (1 H, d, ³J = 11.4 Hz, *cis*-H H₂C=), 4.65 (1 H, d, ${}^{3}J = 18.0$ Hz, trans-H $H_{2}C=$), 4.25 and 4.05 (1 H each, m each, NCH₂), 3.98-3.68 (6 H, m, NCH₂ and 2 OCH₂CH₂N), 3.61 (2 H, m, diastereotopic OCH₂CH₃), 2.37 and 1.95 (1 H each, m each, diastereotopic 5-H₂), 1.62-1.17 (4 H, m, 6-H₂) and 7-H₂), 1.33 (3 H, t, ${}^{3}J =$ 7 Hz, OCH₂CH₃), 0.93 (3 H, t, ${}^{3}J$ = 7 Hz, 8-H₃). ¹³C NMR (CDCl₃): δ 255.4 (C_q, W=C), 202.7 and 198.4 (Cq each, 1:4, trans- and cis-CO W(CO)₅), 146.0 and 134.0 (C_q each, C3 and C4), 128.0 (CH, C1'), 114.8 (CH₂, C2'), 67.8 and 67.7 (2 OCH₂CH₂N), 65.1 (OCH₂CH₃), 62.5 and 53.4 (NCH₂ each), 30.7, 28.7 and 23.0 (CH₂ each, C-5-C-7), 15.4 (OCH₂CH₃), 13.8 (CH₃, C8). IR (*n*-hexane; cm⁻¹ (%)): 2062.6 (25), 1969.7 (5), 1935.5 (100), 1928.2 (20) (v(C≡O)). MS (70 eV; 184 W, m/e (%)): 575 (10) [M^+], 547 (5) [M^+ - CO], 519 (15) $[\mathrm{M^{+}}-2$ CO], 491 (5) $[\mathrm{M^{+}}-3$ CO], 463 (5) $[\mathrm{M^{+}}-4$ CO], 435 (15) $[M^+ - 5 CO]$, 206 (100).

(4*Z*,*syn*)-9f. ¹H NMR (CDCl₃, 25 °C): δ 6.02 (1 H, q, ${}^{3}J = 7$ Hz, 1'-H), 4.62, 4.21, 4.00, and 3.9–3.4 (1:1:2:6 H, m each, morpholinyl and OCH₂), 4.07 (1 H, t, ${}^{3}J = 7$ Hz, 5-H), 2.80 (2 H, m, 6-H₂), 1.92 (3 H, d, ${}^{3}J = 7$ Hz, 2'-H₃), 1.45 (2 H, m, 7-H₂), 1.20 (3 H, t, OCH₂CH₃), 0.95 (3 H, t, ${}^{3}J = 7$ Hz, 8-H₃). ¹³C NMR (CDCl₃): δ 251.2 (C_q, W=C), 202.7 and 198.0 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 152.7 (C_q, C4), 147.6 (C_q, C3), 125.7 (CH, C5), 110.7 (CH, C1'), 67.7 (2 OCH₂CH₂N), 62.5 (O*C*H₂CH₃), 62.5 and 53.8 (NCH₂ each), 30.7, 26.1, and 22.9 (CH₂ each, C-5–C-7), 15.3 (OCH₂CH₃), 13.8 (CH₃, C2'). IR (*n*-hexane; cm⁻¹ (%)): 2062.0 (20), 1968.0 (5), 1933.0 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m*/*e* (%)): 575 (10) [M⁺], 547 (10) [M⁺ – CO], 519 (25) [M⁺ – 2 CO], 491 (5) [M⁺ – 3 CO], 463 (15) [M⁺ – 4 CO], 435 (15) [M⁺ – 5 CO], 206 (100).

(3*S**,4*R**)-10f. ¹H NMR (CDCl₃, 25 °C): δ 4.37, 4.29, 4.03, and 3.73 (1 H each, m each, 2 NCH₂), 4.00–3.80 (6 H, m, diastereotopic OC*H*₂CH₃ and 2 OC*H*₂CH₂N), 3.41 (1 H, qd, ³*J* = 6.9 and 4.8 Hz, 4'-H), 3.12 (1 H, qd, ³*J* = 6.4 and 4.8 Hz,

3'-H), 1.55–1.25 (4 H, m, $CH_2CH_2CH_3$), 1.26 (3 H, t, OCH_2CH_3), 1.06 (3 H, d, ${}^{3}J = 6.9$ Hz, 4'-CH₃), 0.94 (3 H, t, ${}^{3}J = 7$ Hz, $CH_2CH_2CH_3$). ${}^{13}C$ NMR (CDCl₃): δ 241.2 (C_q, W=C), 202.8 and 198.3 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 141.5 (C_q, C2'), 128.5 (C_q, C1'), 68.0 and 67.8 (2 OCH_2CH_2N), 65.0 (OCH_2CH_3), 62.6 and 54.5 (NCH₂ each), 43.8 (CH, C3'), 37.8 (CH, C4'); 30.8, 21.3, and 14.3 ($CH_2CH_2CH_3$), 15.6 (OCH_2CH_3), 15.3 (4'-CH₃). IR, MS, and CHN analytical data of a ca. 20/1 mixture of compounds ($3S^*, 4R^*$)-**10f** and ($3R^*, 4R^*$)-**10f** are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2060.2 (20), 1967.0 (2), 1929.2 (100) (ν -(C=O)). MS (70 eV; ${}^{184}W$, m/e (%)): 575 (40) [M⁺], 547 (40) [M⁺ - CO], 519 (45) [M⁺ - 2 CO], 491 (25) [M⁺ - 3 CO], 463 (45) [M⁺ - 4 CO], 435 (100) [M⁺ - 5 CO]. Anal. Calcd for C₂₀H₂₅NO₇W (575.3): C, 41.76; H, 4.38; N, 2.43. Found: C, 42.24; H, 4.40; N, 2.26.

(3*R**,4*R**)-10f. Some typical signals which could be collected from the reaction mixture are given. ¹H NMR (CDCl₃, 25 °C): δ 2.92 (1 H, qd, ³*J* = 6.9 Hz, ³*J* = 1.2 Hz, 4'-H), 2.42 (1H, qd, ³*J* = 6.4 Hz, ³*J* = 1.2 Hz, 3'-H). ¹³C NMR (CDCl₃, 25 °C): δ 68.3, 67.8, 55.5 (morpholino), 48.3 and 39.1 (CH, C3' and C4').

12f. ¹H NMR (CDCl₃, 25 °C): δ 6.11 (1 H, q, ³J = 7 Hz, 1'-H), 4.61, 4.28, 3.79, and 3.54 (1 H each, m each, 2 NCH₂), 4.00 and 3.73 (2 H each, m each, 2 OCH₂CH₂N), 2.75, 1.61, and 1.37 (2 H each, m each, CH₂CH₂CH₂CH₃), 1.74 (3 H, d, ³J = 7 Hz, 2'-H₃), 0.93 (3 H, t, ³J = 7 Hz, 8-H₃). ¹³C NMR (CDCl₃): δ 251.1 (C_q, W=C), 202.7 and 197.9 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 197.8 (C_q, C=O), 152.6 (C_q, C3), 125.1 (CH, C1'), 67.8 and 67.4 (O*C*H₂CH₂N each), 62.5 and 53.8 (NCH₂ each), 37.9, 26.1, and 22.4 (CH₂ each, C-5-C-7), 13.8 (CH₃, C8), 15.3 (OCH₂*C*H₃). IR (*n*-hexane; cm⁻¹ (%)): 2064.2 (30), 1933.2 (100) (ν (C=O)), 1700.3 (75), 1684.4 (50). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 519 (5) [M⁺ - CO], 491 (5) [M⁺ - 2 CO], 463 (5) [M⁺ - 3 CO], 407 (5) [M⁺ - 5 CO], 57 (100). Anal. Calcd for C₁₈H₂₁NO₇W (547.2): C, 39.51; H, 3.87; N, 2.56. Found: C, 39.52; H, 3.76; N, 2.40.

2-(Dimethylamino)-2-(2-ethoxy-4-methyl-3-phenyl-1cyclobutenyl)-1-(pentacarbonyltungsta)ethenes ($3S^*$, $4R^*$)-10g and ($3R^*$, $4R^*$)-10g and (Z)-1-Ethoxy-2-phenylethene (11). But-2-enoic acid dimethylamide (2a; 226 mg, 2.00 mmol), phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl-(1-ethoxy-(2-phenylethanylidene))tungsten (1g; 472 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) were reacted as described above to give compound 11 (57 mg, 39%, $R_f = 0.3$ in 1/1 *n*-pentane/dichloromethane, colorless oil), compound ($3S^*$, $4R^*$)-10g (31 mg, 6%, $R_f = 0.6$ in 1/1 *n*-pentane/dichloromethane, yellow oil), and compound ($3R^*$, $4R^*$)-10g (39 mg, 7%, $R_f = 0.5$ in 1/1 *n*-pentane/dichloromethane, yellow oil).

(3.5*,4R*)-10g. ¹H NMR (CDCl₃, 25 °C): δ 7.33, 7.27, and 7.13 (2:1:2 H, m each, Ph), 4.45 (1 H, d, ${}^{3}J = 4.8$ Hz, 3'-H), 3.84 (2 H, q, OCH₂), 3.81 and 3.55 (3 H each, s each, 2 NCH₃), 3.66 (1 H, qd, ${}^{3}J = 6.9$ Hz and 4.8, 4'-H), 1.23 (3 H, t, OCH_2CH_3), 0.67 (3 H, d, ${}^{3}J = 6.9$ Hz, 4'-CH₃). ${}^{13}C$ NMR (CDCl₃): δ 241.6 (C_q, W=C), 203.2 and 198.9 (C_q each, 1:4, trans- and cis-CO W(CO)₅), 138.8 (Cq, C2'), 138.5 (Cq, i-C Ph), 130.2 (Cq, C1'); 128.4, 128.2, and 127.0 (2:1:2, CH each, Ph), 65.4 (OCH₂), 53.6 and 45.2 (2 NCH₃), 49.2 (CH, C3'), 40.5 (CH, C4'), 16.5 (4'-CH₃), 15.6 (OCH₂CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2059.9 (16), 1983.9 (3), 1966.9 (2), 1927.7 (100) (v(C≡O)). MS (70 eV; 184 W, m/e (%)): 567 (20) [M⁺], 539 (35) [M⁺ - CO], 511 (20) $[M^+ - 2 \text{ CO}]$, 483 (20) $[M^+ - 3 \text{ CO}]$, 455 (30) $[M^+ - 4$ CO], 427 (100) [M⁺ - 5 CO], 243 (5) [M⁺ -W(CO)₅). Anal. Calcd for C₂₁H₂₁NO₆W (567.3): C, 44.47; H, 3.73; N, 2.47. Found: C, 44.63; H, 3.75; N, 2.40.

(3*R**,4*R**)-10g. ¹H NMR (CDCl₃, 25 °C): δ 7.34–7.26 (5 H, m, Ph), 3.83 and 3.45 (3 H each, s each, 2 NCH₃), 3.81 (2 H, m, diastereotopic OCH₂), 3.66 (1 H, d, ³*J* = 1.2 Hz, 3'-H), 2.89 (1 H, qd, ³*J* = 6.3 and 1.2 Hz, 4'-H), 1.29 (3 H, d, ³*J* = 6.3 Hz, 4'-CH₃), 1.20 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 243.6 (C_q, W=C), 203.1 and 199.0 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 143.3 (C_q, C2'), 139.4 (C_q, *i*-C Ph), 128.8, 127.1, and 127.0 (2:1:2, CH each, Ph), 127.6 (C_q, C1'), 65.1 (OCH₂), 54.9

(CH, C3'), 53.9 and 45.5 (2 NCH₃), 45.3 (CH, C4'), 19.6 (4'-CH₃), 15.4 (OCH₂*C*H₃). IR (*n*-hexane; cm⁻¹ (%)): 2060.0 (20), 1983.7 (5), 1966.9 (2), 1926.6 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 567 (15) [M⁺], 539 (30) [M⁺ - CO], 511 (15) [M⁺ - 2 CO], 483 (15) [M⁺ - 3 CO], 455 (25) [M⁺ - 4 CO], 427 (100) [M⁺ - 5 CO].

11. ¹H NMR (CDCl₃, 25 °C): δ 7.60 (2 H, m, Ph), 7.30–7.10 (3 H, m, Ph), 6.19 and 5.2 (1 H each, d each, ${}^{3}J = 7$ Hz), 3.96 (2 H, q, OCH₂), 1.35 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): δ 146.5 and 105.5 (CH each, (O)*C*H=*C*H), 136.1, 128.1 ,and 125.6 (1:2:2, Ph), 69.0 and 15.4 (O*C*H₂*C*H₃). IR (neat; cm⁻¹ (%)): 1652.3 (100), 1261.3 (8), 1098.8 (72), 1074.0 (24), 779.4 (20), 685.1 (25). MS (70 eV), *m/e* (%): 148 (90) [M⁺], 91 (100).

2-(2-Ethoxy-4-methyl-3-phenyl-1-cyclobutenyl)-2-morpholino-1-(pentacarbonyltungsta)ethenes (3*S****,4***R****)-10h and (3***R****,4***R****)-10h. 1-(Morpholin-4-yl)but-2-en-1-one (2b; 310 mg, 2.00 mmol) was reacted with phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(1-ethoxy-(2-phenylethanylidene-))tungsten (1g; 472 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) as described above to give compound 11 (53 mg, 36%, R_f = 0.3 in 1/1** *n***-pentane/dichloromethane, colorless oil), compound (3***S****,4***R****)-10h (32 mg, 5%, R_f = 0.5 in 3/2** *n***-pentane/dichloromethane, yellow oil), and compound (3***R****,4***R****)-10h (42 mg, 7%, R_f = 0.4 in 3/2** *n***-pentane/dichloromethane, yellow oil).**

(3*S**,4*R**)-10h. ¹H NMR (CDCl₃, 25 °C): δ 7.30 (3 H, m, Ph), 7.09 (2 H, "d", Ph), 4.53, 4.29, 4.24, and 4.03 (1 H each, m each, 2 NCH₂), 4.48 (1 H, d, ${}^{3}J$ = 4.8 Hz, 3'-H), 3.96–3.83 (6 H, m, diastereotopic OC*H*₂CH₃ and 2 OC*H*₂CH₂N), 3.71 (1 H, m, ${}^{3}J$ = 6.4 and 4.8 Hz, 4'-H), 1.20 (3 H, t, OCH₂C*H*₃), 0.67 (3 H, d, ${}^{3}J$ = 6.4 Hz, 4'-CH₃). ¹³C NMR (CDCl₃): δ 239.8 (C_q, W= C), 202.9 and 198.6 (C_q each, 1:4, *trans-* and *cis-*CO W(CO)₅), 138.5 and 138.1 (C_q each, C2' and *i*-C–Ph), 129.1, 128.5, and 128.2 (1:2:2, CH each, Ph), 127.1 (C_q, C1'), 68.0 and 67.8 (2 O*C*H₂CH₂N), 65.4 (O*C*H₂CH₃), 62.8 and 54.9 (2 NCH₂), 49.2 (CH, C3'), 40.7 (CH, C4'), 16.5 (4'-CH₃), 15.4 (OCH₂*C*H₃). IR (*n*-hexane; cm⁻¹ (%)): 2060.1 (20), 1929.2 (100) (ν (C≡O)). MS (70 eV; ¹⁸⁴W, *m*/*e* (%)): 609 (5) [M⁺], 581 (15) [M⁺ – CO], 553 (5) [M⁺ – 2 CO], 525 (10) [M⁺ – 3 CO], 497 (10) [M⁺ – 4 CO], 469 (35) [M⁺ – 5 CO].

(3*R**,4*R**)-10h. ¹H NMR (CDCl₃, 25 °C): δ 7.34–7.25 (5 H, m, Ph), 4.80, 4.35, 4.09, and 4.01 (1 H each, m each, 2 NCH₂), 3.92–3.75 (6 H, m, diastereotopic OCH₂CH₃ and 2 OCH₂-CH₂N), 3.65 (1 H, d, ³*J* = 1.2 Hz, 3'-H), 2.94 (1 H, qd, ³*J* = 6.6 and 1.2 Hz, 4'-H), 1.28 (3 H, d, ³*J* = 6.6 Hz, 4'-CH₃), 1.18 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃): δ 242.6 (C_q, W=C), 202.7 and 198.6 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 141.9 (C_q, C2'), 139.2 (C_q, *i*-C Ph); 128.8, 127.2, and 127.0 (2:1:2, CH each, Ph), 126.4 (C_q, C1'), 68.0 and 67.8 (2 O*C*H₂CH₂N), 65.2 (O*C*H₂CH₃), 63.0 and 55.1 (2 NCH₂), 54.9 (CH, C3'), 45.4 (CH, C4'), 19.7 (4'-CH₃), 15.4 (OCH₂*C*H₃). IR (*n*-hexane; cm⁻¹ (%)): 2060.6 (20), 1967.2 (2), 1927.4 (100) (ν (C=O)), 1700 (7), 1653.0 (8). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 609 (15) [M⁺], 581 (30) [M⁺ - CO], 553 (15) [M⁺ - 2 CO], 525 (20) [M⁺ - 3 CO], 497 (30) [M⁺ -4 CO], 469 (100) [M⁺ - 5 CO].

2-(Dimethylamino)-2-(3-(2,4,6-cycloheptatrienyl)-2ethoxy-4-methyl-1-cyclobutenyl)-1-(pentacarbonyltungsta)ethenes ($3S^*$, $4R^*$)-10i and ($3R^*$, $4R^*$)-10i and 5-(2,4,6-Cycloheptatrienyl)-2-(dimethylamino)-3-ethylidene-1-(pentacarbonyltungsta)-1-penten-4-one (12i). But-2-enoic acid dimethylamide (2a; 226 mg, 2.00 mmol), phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(2-(2,4,6cycloheptatrienyl)-1-ethoxy-1-ethylidene)tungsten (1h; 486 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) were reacted as described above to give a 7/2 mixture of compounds ($3S^*$, $4R^*$)-10i and ($3R^*$, $4R^*$)-10i (256 mg, 44%, R_f = 0.7 in 2/1 n-pentane/dichloromethane, yellow oil) and compound 12i (13 mg, 2%, R_f = 0.4 in 1/1 n-pentane/dichloromethane). (3S*,4R*)-10i.



¹H NMR (CDCl₃, 25 °C): δ 6.66 (2 H, m, 5"-H and 4"-H), 6.21 (2 H, m, 3"-H and 6"-H), 5.37 and 5.15 (1 H each, m each, 2"-H and 7"-H), 3.80 (2 H, m, diastereotopic OCH₂CH₃), 3.77 and 3.42 (3 H each, s each, 2 NCH₃), 3.48 (2 H, m, 3'-H and 4'-H), 1.67 (1 H, m, 1"-H), 1.29 (3 H, t, OCH₂CH₃), 1.02 (3 H, d, ${}^{3}J = 6.6$ Hz, 4'-CH₃). ${}^{13}C$ NMR (CDCl₃): δ 246.7 (C_q, W=C), 203.2 and 198.6 (Cq each, 1:4, trans- and cis-CO W(CO)₅), 139.6 $(C_q, C2')$, 131.0 and 130.9 (CH each, C-4" and C-5"), 129.4 $(C_q, C2')$ C1'), 124.5 and 124.5 (CH each, C-3" and C-6"), 124.1 and 124.0 (CH each, C-2"and C-7"), 65.3 (OCH2), 53.1 and 46.8 (NCH3 each), 45.1 (CH, C3'), 39.1 (CH, C4'), 37.3 (CH, C1"), 15.6 and 15.5 (OCH₂CH₃ and 4'-CH₃). IR, MS, and CHN analytical data of a 7/2 mixture of compounds (3.S*,4R*)-10i and $(3R^*, 4R^*)$ -10i are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2060.6 (20), 1966.5 (5), 1929.3 (100) (v(C≡O)). MS (70 eV; ¹⁸⁴W, m/e (%)): 581 (15) [M⁺], 553 (25) [M⁺ - CO], 469 (25) [M⁺ -4 CO], 441 (35) [M⁺ - 5 CO], 91 (100). Anal. Calcd for C₂₂H₂₃-NO₆W (581.3): C, 45.46; H, 3.99; N, 2.41. Found: C, 45.34; H, 4.21; N, 2.38.

(3*R**,4*R**)-10i. ¹H NMR (CDCl₃, 25 °C): δ 6.64 (2 H, m, 5"-H and 4"-H), 6.21 (2 H, m, 3"-H and 6"-H), 5.32 and 5.23 (1 H each, m, 2"-H and 7"-H), 3.80 (2 H, m, diastereotopic OCH₂), 3.79 and 3.42 (3 H each, s each, 2 NCH₃), 2.88 (1 H, qd, ³*J* = 6.6 and 1.6 Hz, 4'-H), 2.86 (1 H, dd, ³*J* = 8.4 and 1.6 Hz, 3'-H), 1.57 (1 H, m, 1"-H), 1.29 (3 H, t, OCH₂CH₃), 1.22 (3 H, d, ³*J* = 6.6 Hz, 4'-CH₃). ¹³C NMR (CDCl₃): δ 246.7 (C_q, W=C), 203.1 and 198.8 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 142.3 (C_q, C2'), 129.4 and 129.4 (CH, C-4"and C-5"), 127.4 (C_q, C1'), 125.3 and 125.3 (CH, C-3"and C-6"), 125.0 and 124.7 (CH, C-2"and C-7"), 53.5 and 45.3 (NCH₃ each), 51.4 (CH, C3'), 40.6 (CH, C4'), 38.8 (CH, C1"), 15.6 (OCH₂*C*H₃), 15.5 (4'-CH₃).





¹H NMR (CDCl₃ 600 MHz, 25 °C): δ 6.65 (2 H, m, 5'-H and 4'-H), 6.19 (2 H, m, 3'-H and 6'-H), 6.13 (1 H, q, ${}^{3}J = 7$ Hz, 1"'-H), 5.28 and 5.21 (2 H each, dd each, ${}^{3}J = 6.0$ and 9.6 Hz, 2'-H and 7'-H), 3.87 and 3.15 (3 H each, s each, 2 NCH₃), 3.22 and 2.97 (1 H each, ddd each, ${}^{2}J = 17.4$ Hz, ${}^{3}J = 7$ Hz, COCH₂), 2.39 (1 H, m, 1"-H), 1.73 (3 H, d, ${}^{3}J = 7$ Hz, CH₃). ¹³C NMR (CDCl₃): δ 253.4 (C_q, W=C), 202.9 and 198.1 (C_q each, 1:4, *trans*- and *cis*-CO W(CO)₅), 153.5 (C_q, C3), 131.2 and 130.7 (CH, C4' and C5'), 126.0 (CH, C1''), 125.1 and 125.1 (CH, C3'and C6'), 125.1 and 124.9 (CH, C2'and C-7'), 53.3 and 43.5 (NCH₃ each), 40.5 (CO*C*H₂), 34.7 (CH, C1'), 15.3 (CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2064.0 (20), 1983.8 (20), 1966.5 (5), 1931.2 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m/e* (%)): 525 (10) [M⁺ - CO], 469 (5) [M⁺ - 3 CO], 441 (5) [M⁺ - 4 CO], 413 (10) [M⁺ - 5 CO], 91 (100).

2-(3-(2,4,6-Cycloheptatrienyl)-2-ethoxy-4-methyl-1-cyclobutenyl)-2-morpholino-1-(pentacarbonyltungsta)ethenes $(3S^*,4R^*)$ -10j and $(3R^*,4R^*)$ -10j and 5-(2,4,6**Cycloheptatrienyl)-3-ethylidene-2-morpholino-1-(pentacarbonyltungsta)-1-penten-4-one (12j).** 1-(Morpholin-4-yl)but-2-en-1-one (**2b**; 310 mg, 2.00 mmol), phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(2-(2,4,6-cycloheptatrienyl)-1-ethoxy-1-ethylidene)tungsten (**1h**; 486 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) were reacted as described above to give a 12/1 mixture of compounds ($3S^*, 4R^*$)-**10j** and ($3R^*, 4R^*$)-**10j** (286 mg, 46%, $R_f = 0.6$ in 2/1 *n*-pentane/ dichloromethane, yellow oil) and compound **12j** (14 mg, 2%, $R_f = 0.4$ in 1/1 *n*-pentane/dichloromethane).

(**3***S**,**4***R**)-**10j**. ¹H NMR (CDCl₃, 25 °C): δ 6.67 (2 H, m, 5"-H and 4"-H), 6.21 (2 H, m, 3"-H and 6"-H), 5.28 and 5.16 (1 H each, m each, 2"-H and 7"-H), 4.45, 4.24, 4.12, and 3.94 (1 H each, m each, 2 NCH₂), 3.93-3.64 (6 H, m, diastereotopic OCH2CH3 and 2 OCH2CH2N), 3.54 (2 H each, m, 3'-H and 4'-H), 1.66 (1 H, m, 1"-H), 1.28 (1 H, t, OCH₂CH₃), 1.01 (3 H, d, ${}^{3}J = 6.6$ Hz, 4'-CH₃). ${}^{13}C$ NMR (CDCl₃): δ 244.3 (C_q, W=C), 202.9 and 198.5 (1:4, trans- and cis-CO W(CO)₅), 139.4 (Cq, C2'), 131.0 and 130.9 (CH, C4" and C5"), 128.6 (Cq, C1'), 124.7 and 124.6 (CH, C3" and C6"), 123.7 and 123.6 (CH, C2" and C7"), 67.8 and 67.6 (2 OCH2CH2N), 65.5 (OCH2CH3), 62.4 and 54.8 (2 NCH₂), 46.7 (CH, C3'), 39.0 (CH, C1"), 37.4 (CH, C4'), 15.5 and 15.4 (4'-CH₃ and OCH₂CH₃). IR and MS analytical data of a 12/1 mixture of compounds $(3S^*, 4R^*)$ -10j and $(3R^*, 4R^*)$ -10j are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2060.8 (21), 1983.8 (11), 1967.4 (3), 1930.9 (100) (v(C≡O)). MS (70 eV; ¹⁸⁴W, m/e (%)): 623 (10) [M⁺], 595 (15) [M⁺ - CO], 539 (5) $[M^+ - 3 CO]$, 511 (20) $[M^+ - 4 CO]$, 483 (25) $[M^+ - 5 CO]$, 91 (100).

(3*R**,4*R**)-10j. Some typical signals which have been collected from the reaction mixture are given. ¹H NMR (CDCl₃, 25 °C): δ 6.64 (2 H, m, 5"-H and 4"-H), 5.36 and 5.17 (1 H each, m each, 2"-H and 7"-H), 2.99 (1 H, dq, ³*J* = 1.2 and 6.8 Hz, 4'-H), 2.77 (1 H, dd, ³*J* = 1.2 and 7 Hz, 3'-H), 2.05 (1 H, m, 1"-H), 1.22 (1 H, t, ³*J* = 7 Hz, OCH₂*C*H₃). ¹³C NMR (CDCl₃, 25 °C): δ 198.5, 137.8 (C_q, C2'), 131.1 and 130.9 (CH each, C-5" and C-4"), 129.0 (C_q, C1'), 125.5 and 125.2 (CH each, C-3" and C-6"), 124.5 and 124.3 (CH each, C-2"and C-7"), 67.9 and 67.7 (OCH₂*C*H₂N each), 65.7 (O*C*H₂CH₃), 62.8 and 55.0 (NCH₂ each), 51.4 (CH, C3'), 40.6 (CH, C1"), 38.8 (CH, C4'), 15.2 (4'-CH₃ or OCH₂*C*H₃).

12j. ¹H NMR (CDCl₃, 600 M, 25 °C): δ 6.65 (2 H, m, 5"-H and 4"-H), 6.18 (2 H, m, 3"-H and 6"-H), 6.15 (1 H, q, ${}^{3}J = 7$ Hz, C=CH), 5.26 and 5.20 (1 H each, m each, 2"-H and 7"-H), 4.60, 4.23, 3.79, and 3.52 (1 H each, m each, 2 NCH₂), 3.71-3.66 (4 H, m, 2 OCH₂CH₂N), 3.25 and 3.00 (1 H each, dd each, $^{2}J = -17.3$ Hz, $^{3}J = 6.9$ and 7 Hz, diastereotopic CH₂CO), 2.37 (1 H, m, CH, 1"-H), 1.75 (3 H, t, ${}^{3}J = 7$ Hz, CH₃). ${}^{13}C$ NMR (CDCl₃): δ 250.8 (C_q, W=C), 202.6 and 197.9 (C_q each, 1:4, trans- and cis-CO W(CO)₅), 196.14 (Cq, C=O), 152.6 (Cq, C= CH), 131.3 and 130.7 (CH each, C4" and C5"), 126.2 (CH, C= CH), 125.2 and 125.2 (CH each, C3" and C6"), 125.1 and 124.7 (CH each, C2" and C7"), 67.7 and 67.3 (OCH2CH2N each), 62.5 and 53.9 (NCH2CH2O each), 40.7 (CH2, CH2CO), 34.6 (CH, C1"), 15.6 (CH₃). IR (*n*-hexane; cm⁻¹ (%)): 2064.0 (22), 1972.2 (2), 1932.7 (100) (v(C≡O)), 1700 (13) (v(C=O)). 1653.0 (15). MS (70 eV; ^{184}W , m/e (%)): 567 (10) [M⁺ - CO], 511 (5) [M⁺ - 4 CO], 483 (5) $[M^+ - 5 CO]$, 91 (100).

2-(Dimethylamino)-3-ethylidene-7-hydroxy-1-(pentacarbonyltungsta)-1-nonen-4-one (12k). But-2-enoic acid dimethylamide (**2a**; 113 mg, 1.00 mmol), phosphorus oxychloride (153 mg, 1.00 mmol), and pentacarbonyl(3-ethyl-2-oxacyclopentanylidene)tungsten (**1i**; 211 mg, 0.50 mmol) were treated with triethylamine (202 mg, 2.00 mmol) as described above for 12 h to give a 4/3 mixture of diastereomeric compounds **12k** (108 mg, 40%, $R_f = 0.5$ in dichloromethane/ diethyl ether 4:1, yellow oil).



This compound was a 4/3 mixture of diastereomers; chemical shifts of minor isomer are given in braces. ¹H NMR (CDCl₃, 600 MHz, 25 °C): δ 6.17 {6.17} (1 H, q, ${}^{3}J = 7$ Hz, 1'-H), 3.88 and 3.18 {3.88 and 3.19} (3 H each, s each, 2 NCH₃), 3.56 {3.56} (1 H, m, 7-H), 3.02 and 2.83 {2.95 and 2.87} (1 H each, m each, diastereotopic 5-H₂), 1.84 and 1.73 {1.94 and 1.65} (1 H each, m each, diastereotopic 6-H₂), 1.73 {1.73} (3 H, d, ${}^{3}J$ = 7 Hz, 2'-H₃), 1.50 {1.50} (2 H, m, 8-H₂), 0.95 {0.95} (3 H, t, ${}^{3}J$ = 7 Hz, 9-H₃). ¹³C NMR (CDCl₃): δ 253.2 {253.1} (C_q, W=C), 202.9 and 198.1 {202.9 and 198.1} (Cq each, 1:4, trans- and *cis*-CO W(CO)₅), 198.0 {197.9} (C_q, C1'), 153.2 {153.2} (C_q, C3), 125.7 {125.7} (CH, C1'), 72.9 {72.5} (CH, C7), 53.3 and 43.5 $\{53.3 \text{ and } 43.5\}$ (NCH₃ each), 34.7 $\{34.7\}$ (CH₂, C5), 30.7 $\{30.6\}$ (CH₂, C6), 30.4 {30.3} (CH₂, C8), 15.2 {15.2} (CH₃, C2'), 9.9 {9.8} (CH₃, C9). IR (*n*-hexane; cm⁻¹ (%)): 2063.5 (18), 1970.5 (2), 1931.2 (100) (ν (C=O)). IR (neat; cm⁻¹ (%)): 3413.9 (16) $(\nu(O-H)), 2060.1 (13), 1967.5 (7), 1916.8 (37), 1891.1 (40),$ 1675.9 (60) (v(C=O)), 1341.1 (32), 1221.1 (30), 1213.6 (29), 1092.7 (24). MS-EI (70 eV; 184W, m/e (%)): 517 (25) [M+ - H₂O], 489 (10) $[M^+ - CO - H_2O)]$, 461 (25) $[M^+ - 2 CO - H_2O]$, 405 (15) $[M^+ - 4 CO - H_2O]$, 377 (70) $[M^+ - 5 CO - H_2O]$, 197 (100). ESI-MS (m/e (%)): 572 (100), 570 (100), 568 (70) [M⁺ + Cl], 536 (25), 534 (30), 532 (20) [M⁺]; isotope pattern in good agreement with expectation. Anal. Calcd for C₁₇H₂₁NO₇W (535.2): C, 38.15; H, 3.95; N, 2.62. Found: C, 38.20; H, 4.07; N, 2.58.

2-((Dimethylamino)-4-ethoxy-6-phenyl)-1-(pentacarbonyltungsta)-1,3-hexadien-5-ynes (3*E***)-16a and (3***Z***)-16a. Acetic acid dimethylamide (15a; 174 mg, 2.00 mmol), phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl(3phenyl-1-ethoxy-2-propyn-1-ylidene)tungsten (1b; 486 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) were reacted as described above to give an 8/1 mixture of compounds (3***E***)-16a and (3***Z***)-16a (212 mg, 39%, R_f = 0.4 in 2/1** *n***-pentane/ dichloromethane, yellow crystals from 10/1** *n***-pentane/dichloromethane at -20 °C, mp 81-82 °C).**

(3*E*)-16a.



¹H NMR (CDCl₃, 25 °C): δ 7.40 and 7.35 (2:3, m each, Ph), 6.37 (1 H, s, 3-H), 3.99 (2 H, q, OCH₂), 3.82 and 3.48 (3 H each, s each, 2 NCH₃), 1.38 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): 247.0 (C_q, W=C), 203.7 and 198.3 (C_q, 1:4, *trans*- and *cis*-CO W(CO)₅), 131.5, 129.3, and 128.5 (2:1:2, CH each, Ph), 129.3 (CH, C3), 125.1 (C_q, HC=C, C4), 121.5 (C_q, *i*-C Ph), 98.7 and 81.6 (C_q each, *C*=*C*), 64.9 (OCH₂), 53.3 and 44.2 (NCH₃ each), 14.8 (OCH₂CH₃). IR, MS, and CHN analytical data obtained from the 8/1 mixture of compounds (3*E*)-**16a** and (3*Z*)-**16a** are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2061.4 (15), 1983.9 (10), 1927.3 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, *m*/e (%)): 551 (20) [M⁺], 523 (10) [M⁺ - CO], 495 (20) [M⁺ - 2 CO], 467 (100) [M⁺ - 3 CO], 411 (45) [M⁺ - 5 CO].

Anal. Calcd for $C_{20}H_{17}NO_6W$ (551.2): C, 43.58; H, 3.11; N, 2.54. Found: C, 43.69; H, 2.90; N, 2.44. (3Z)-16a.



¹H NMR (CDCl₃, 25 °C): δ 7.42 and 7.34 (2:3 H, m each, Ph), 6.17 (1 H, s, 3-H), 4.13 (2 H, q, ³*J* = 7 Hz, OCH₂), 3.77 and 3.37 (3 H each, s each, 2 NCH₃), 1.29 (3 H, t, ³*J* = 7 Hz, OCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): 246.9 (C_q, W=C), 203.8 and 198.4 (C_q, 1:4, *trans-* and *cis*-CO W(CO)₅), 131.7, 129.0, and 128.4 (2:1:2, CH each, Ph), 129.4 (CH, C3), 125.1 (C_q, HC=C, C4), 121.7 (C_q, *i*-C Ph), 92.1 and 82.4 (C_q each, C=C), 65.4 (OCH₂), 53.0 and 44.8 (NCH₃ each), 15.5 (OCH₂CH₃).

2-((Dimethylamino)-3-ethyl-4-ethoxy-6-phenyl)-1-(pentacarbonyltungsta)-1,3-hexadien-5-ynes (3*E***)-16b and (3***Z***)-16b. Butanoic acid dimethylamide (15b; 230 mg, 2.00 mmol), phosphorus oxychloride (306 mg, 2.00 mmol), pentacarbonyl-(3-phenyl-1-ethoxy-2-propyn-1-ylidene)tungsten (1b; 486 mg, 1.00 mmol), and triethylamine (404 mg, 4.00 mmol) were reacted as described above to give a 14/1 mixture of compounds (3***E***)-16b and (3***Z***)-16b (429 mg, 74%, R_f = 0.5 in 2/1** *n***-pentane/ dichloromethane, yellow crystals from 10/1** *n***-pentane/dichloromethane at -20 °C; mp 86-87 °C).**

(3E)-16b. ¹H NMR (CDCl₃, 25 °C): δ 7.39 and 7.31 (2:3, m each, Ph), 4.09 and 3.98 (1 H each, m each, diastereotopic OC-H₂), 3.81 and 3.40 (3 H each, s each, 2 NCH₃), 2.83 and 2.47 (1 H each, m each, CCH₂CH₃), 1.32 (3 H, t, OCH₂CH₃), 1.06 (3 H, t, CCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): 253.5 (C_q, W=C), 202.9 and 198.4 (Cq, 1:4, trans- and cis-CO, W(CO)_5), 142.7 (Cq, C3), 131.7, 128.7, and 128.4 (2:1:2, CH each, Ph), 124.5 (C_q, C4), 122.0 (C_q, *i*-C Ph), 97.5 and 82.0 (C_q each, *C*≡*C*), 65.2 (OCH₂), 53.2 and 44.2 (NCH₃ each), 22.6 (CCH₂CH₃), 15.3 (OCH₂CH₃), 11.7 (CCH₂CH₃). IR, MS, and CHN analytical data obtained for a 14/1 mixture of compounds (3E)-16b and (3Z)-**16b** are as follows. IR (*n*-hexane; cm⁻¹ (%)): 2061.4 (20), 1931.5 (100) (ν (C=O)). MS (70 eV; ¹⁸⁴W, m/e (%)): 579 (20) [M⁺], 523 (10) $[M^+ - 2 CO]$, 495 (100) $[M^+ - 3 CO]$, 467 (20) $[M^+ - 4]$ CO], 439 (20) [M⁺ - 5 CO]. Anal. Calcd for C₂₂H₂₁NO₆W (579.3): C, 45.62; H, 3.65; N, 2.42. Found: C, 45.44; H, 3.42; N. 2.36.

(3Z)-16b. Only a partial assignment is given, due to overlapping signals. ¹H NMR (CDCl₃, 25 °C): δ 3.75 and 3.31 (3 H each, s each, 2 NCH₃), 2.70 (2 H, m, CCH₂CH₃), 1.26 (3 H, t, OCH₂CH₃). ¹³C NMR (CDCl₃, 25 °C): 202.2 and 198.6 (C_q, 1:4, *trans-* and *cis*-CO W(CO)₅), 137.9 (C_q, C3), 131.3, 128.6, and 128.4 (2:1:2, CH each, Ph), 122.6 and 122.2 (C_q each, C4 and *i*-C Ph), 95.3 and 81.3 (C_q each, *C*=*C*), 64.5 (OCH₂), 52.8 and 44.6 (NCH₃ each).

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Supporting Information Available: Tables giving details of the X-ray crystal structure analyses and figures giving additional views of the structures. This material is available free of charge via the Internet at http://pubs.acs.org.

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