Kinetic Resolution of Planar-Chiral Ferrocenes by Molybdenum-Catalyzed Enantioselective Metathesis

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Summary: Kinetic resolution of planar-chiral 1,1′*-diallylferrocene deri*V*ati*V*es was realized by molybdenum-catalyzed asymmetric ring-closing metathesis with excellent enantioselectivity. This is the first example of highly enantioselective metal-catalyzed methods of preparing optically active planarchiral metallocenes.*

Planar-chiral ferrocenes are important chiral scaffolds in organic and organometallic chemistry.¹ Although such ferrocenes have been used in a variety of asymmetric reactions as ligands² or catalysts, 3 their preparation in optically active form is still a challenging problem. Most nonracemic planar-chiral ferrocenes were obtained either by diastereoselective metalation utilizing chiral *ortho*-directing groups⁴ or by optical resolution of racemic compounds. Enantioselective lithiation of prochiral ferrocenes by a stoichiometric chiral base has also shown fair success.5 On the other hand, examples of *catalytic* asymmetric synthesis of planar-chiral ferrocene derivatives are extremely

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rare.6-⁸ Apart from several reports on enzymatic resolution of racemic ferrocenes,⁶ to the best of our knowledge, only two examples of catalytic induction of ferrocenyl planar chirality was reported so far with modest enantioselectivity.^{7,8}

Richards⁹ and we¹⁰ recently reported preparation of $[4]$ ferrocenophanes by Ru- or Mo-catalyzed metathesis reaction of 1,1′-diallylferrocenes.11 Here, we report kinetic resolution of planar-chiral 1,1′-diallylferrocene derivatives by Mo-catalyzed asymmetric ring-closing metathesis (ARCM).12 The reaction

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Table 1. Molybdenum-Catalyzed Asymmetric RCM Kinetic Resolution of Planar-Chiral Ferrocenes*^a*

^a The reaction was carried out in benzene in the presence of the molybdenum catalyst (*R*)-Mo*. *^b*Initial concentration of the substrate. *^c* Isolated yields of the products (**2** and **3**) and the recovered substrate (**1**). *^d*Enantiomeric excess of the recovered **1** was determined after converting them into **2** by a reaction with the second-generation Grubbs' catalyst. *^e* Calculated based on a first-order equation (ref 14).

proceeds very efficiently to afford planar-chiral ferrocenes in high yield with excellent enantiomeric enrichment.

The ferrocene substrates for this study (*rac*-**1**, see Scheme 1) possess a trisubstituted η^5 -(C₅H₂-1-allyl-2,4-R¹₂) ligand, which constructs a planar-chiral environment in the ferrocenes, and a monosubstituted η^5 -cyclopentadienyl ligand with an allylic side chain. A readily available chiral molybdenum species (*R*)- Mo^* (Scheme $1)^{13}$ was chosen as an asymmetric metathesis catalyst.

Enantioselectivity in the ARCM kinetic resolution was strongly dependent on the structure of the allylic group in the monosubstituted cyclopentadienyl moiety. Treatment of **1a** with the Mo^{*} catalyst (5 mol %) for 15 min in benzene at 23 $^{\circ}$ C yielded the corresponding ferrocenophane **2a** in 73%, and the unreacted **1a** was recovered in 26%. To our disappointment, however, the selectivity of the reaction was very low: **2a** was nearly racemic (<1% ee) and the recovered **1a** was only 2% ee (Table 1, entry 1). The RCM reaction of the ferrocene **1b**, which was with $η⁵-(C₅H₄-crotyl)$, showed slightly better enantioselectivity and the ferrocenophane **2a** of 12% ee was obtained in 44% yield (entry 2). Introduction of an *η*5-(C5H4-cinnamyl) ligand in **1c** further improved enantioselectivity of the ARCM, although diluted conditions were required to suppress formation of metathesized homodimer **3** (as a mixture of stereoisomers). A reaction of **1c** at 80 °C (initial concentration of $1c = 0.01$ mol/L) gave **2a** (66% ee, 26% yield) and recovered **1c** (29% ee, 53%). The *k*rel value ([rate of fast-reacting enantiomer]/[rate of slow-reacting enantiomer]) for this reaction was estimated to be 6.4¹⁴ (entry 3). A reaction of **1d**, which was with η^5 - $C_5H_4CH_2CH=CMe_2$, produced the dimer 3d in 38% as a sole metathesis product and **2a** was not detected (entry 4). The recovered **1d** was only 7% ee; that is, kinetic resolution of **1d** at the dimerization was not very efficient.

It was found that the enantioselectivity was dramatically improved by introducing a η^5 -(C₅H₄-methallyl) moiety in the ferrocene substrates. The RCM reaction of **1e** at 23 °C (initial concentration of $1e = 0.1$ mol/L) proceeded with excellent enantioselectivity. The bridged ferrocene **2e** was obtained in

Figure 1. ORTEP drawing of $(+)$ - (R) -2e with 40% thermal ellipsoids. All hydrogen atoms are omitted for clarity.

nearly enantiomerically pure form (>99.5% ee) in 23% yield, and unreacted **1e** of 78% ee was recovered in 30% yield. The k_{rel} for this reaction was $> 500^{14}$ (entry 5). Unfortunately, however, 47% of the substrate was converted to dimer **3e**. The dimer **3e**, which could be converted to the ferrocenophane **2e** by treatment with the second-generation Grubbs catalyst, was only 7% ee; that is, the dimer formation occurred with negligible enantioselectivity. As before, dimerization was minimized in the reaction under diluted conditions. A reaction of **1e** with initial concentration of 0.005 mol/L affords **2e** of 96% ee in 47% yield, and unreacted **1e** of 95% ee was recovered in 46% (entry 6). Although somewhat higher temperature (50 °C) was required to gain a reasonable reaction rate under the highdilution conditions, enantioselectivity of the reaction was still excellent (k_{rel} = 183) and nearly perfect kinetic resolution of *rac*-**1e** was achieved. As shown in entries 7 and 8, similarly high levels of enantioselectivity and reaction efficiency were observed for kinetic resolution of **1f** ($k_{\text{rel}} = 26$) and **1g** ($k_{\text{rel}} =$ 165) also under the optimized conditions.

Single-crystal X-ray crystallography revealed that absolute configuration of the ARCM product **2e** (Table 1, entry 5; $>99.5\%$ ee), which was dextrorotatory ($\left[\alpha\right]^{30}$ _D +22 (*c* 0.50 in CHCl3)), was *R*, as shown in Figure 1. The configurations of other ARCM products were assigned by analogy.

For highly enantioselective kinetic resolution of *rac*-**1**, an allyl group in a trisubstituted cyclopentadienyl needs to be more reactive than an allylic substituent in a monosubstituted cyclopentadienyl. When the (*R*)-Mo* species approaches *rac*-**1a**, initial metathesis occurs with the less crowded C_5H_4 -allyl moiety preferentially to form a diastereomeric mixture of (R_a, R_p) - and (R_a, S_p) -4 (" R_a " represents absolute configuration of the axially chiral biaryl moiety in Mo*) and formation of **5** is unfavorable (Scheme 2, top). Whereas the allyl group in C_5H_4 -allyl is remote from the planar-chiral η^5 -C₅H₂'Bu₂(allyl) moiety, initial me-

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tathesis with (R) -Mo* takes place with negligible stereoselectivity. Following the RCM step in **4** is an intramolecular process, and thus both (R_a, R_p) - and (R_a, S_p) -4 can be smoothly transformed into the corresponding ferrocenoephanes to give **2a** in nearly

racemic form. On the other hand, with a less reactive methallyl group in a monosubstituted Cp in place of the allyl group, the reaction between (*R*)-Mo* and *rac*-**1e** takes place at the allyl group in the planar-chiral trisubstituted Cp preferentially and formation of **7** is minor. While (R_a, R_p) -6 is transformed into (*R*)-2e smoothly via RCM, the epimeric (R_a , S_p)-6 is forced to take an unfavorable conformation due to the steric repulsion between a 'Bu group in η^5 -C₅H₂'Bu₂(allyl) and a 'Bu group in the chiral biaryl ligand in (*R*)-Mo* to liberate (*S*)-**1e** (Scheme 2, bottom).

In summary, we have developed an effective method for kinetic resolution of racemic planar-chiral ferrocene derivatives via molybdenum-catalyzed asymmetric ring-closing metathesis. This is the first example of highly enantioselective metalcatalyzed induction of planar chirality in metallocene substrates.

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Supporting Information Available: Detailed experimental procedures, compound characterization data, and crystallographic data (CIF file). This material is available free of charge via the Internet at http://pubs.acs.org.

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