# Notes

## Syntheses of 1,3-Disilaindanes, 1,4-Disilatetralines, and 1,3-Disila-1,3-dihydroisobenzofuranes Using Cobalt-Catalyzed [2+2+2] Cycloadditions

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Summary: A simple catalytic system for [2+2+2] alkyne cycloadditions consisting of cobalt(II) iodide and zinc powder in acetonitrile was applied to the direct synthesis of hydroxyalkyl-substituted 1,3-disilaindanes, 1,4-disilatetralines, and 1,3disila-1,3-dihydroisobenzofuranes starting from silicon-containing diynes and unprotected propargyl alcohols. Siliconcontaining diynes were efficiently coupled with propargyl alcohols with a terminal or internal monoyne moiety. The method reported allows efficient, one-step syntheses of the title compounds.

### Introduction

The catalytic [2+2+2] cyclotrimerization of alkynes is an attractive method for synthesizing polysubstituted benzene derivatives<sup>1,2</sup> and is demonstrated here to be an efficient method for the synthesis of hydroxyalkyl-substituted 1,3-disilaindanes, 1,4-disilatetralines, and 1,3-disila-1,3-dihydroisobenzofuranes. As we have recently demonstrated, compounds with a 1,1,3,3-

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Selected reviews: (a) Vollhardt, K. P. C. Angew. Chem. 1984, 96, 525-541; Angew. Chem., Int. Ed. Engl. 1984, 23, 539-556. (b) Schore, N. E. Chem. Rev. 1988, 88, 1081-1119. (c) Ojima, I.; Tzamarioudaki, M.; Li, Z.; Donovan, R. J. Chem. Rev. 1996, 96, 635-662. (d) Saito, S.; Yamamoto, Y. Chem. Rev. 2000, 100, 2901-2915. (e) Varela, J. A.; Saá, C. Chem. Rev. 2003, 103, 3787-3801. (f) Kotha, S.; Brahmachary, E.; Lahiri, K. Eur. J. Org. Chem. 2005, 4741-4767.

(2) Selected recent contributions: (a) Tanaka, K.; Nishida, G.; Wada, A.; Noguchi, K. Angew. Chem. 2004, 116, 6672-6674; Angew. Chem. Int. Ed. 2004, 43, 6510-6512. (b) Saino, N.; Kogure, D.; Okamoto, S. Org. Lett. 2005, 7, 3065-3067. (c) Tanaka, K.; Nishida, G.; Ogino, M.; Hirano, M.; Noguchi, K. Org. Lett. 2005, 7, 3119-3121. (d) Hilt, G.; Hess, W.; Vogler, T.; Hengst, C. J. Organomet. Chem. 2005, 690, 5170-5181. (e) Chang, H.-T.; Jeganmohan, M.; Cheng, C.-H. Chem. Commun. 2005, 4955-4957. (f) Cadierno, V.; García-Garrido, S. E.; Gimeno, J. J. Am. Chem. Soc. 2006, 128, 15094-15095. (g) Tekavec, T. N.; Zuo, G.; Simon, K.; Louie, J. J. Org. Chem. 2006, 71, 5834-5836. (3) (a) Montana, J. G.; Fleming, I.; Tacke, R.; Daiss, J. (Amedis

(3) (a) Montana, J. G.; Fleming, I.; Tacke, R.; Daiss, J. (Amedis Pharmaceuticals Ltd.), PCT Int. Pat. Appl. WO 2004/045625 A1, June 3, 2004. (b) Montana, J. G.; Showell, G. A.; Fleming, I.; Tacke, R.; Daiss, J. (Amedis Pharmaceuticals Ltd.), PCT Int. Pat. Appl. WO 2004/048390 A1, June 10, 2004. (c) Montana, J. G.; Showell, G. A.; Tacke, R. (Amedis Pharmaceuticals Ltd.), PCT Int. Pat. Appl. WO 2004/048391 A1, June 10, 2004. (d) Daiss, J. O.; Burschka, C.; Mills, J. S.; Montana, J. G.; Showell, G. A.; Fleming, I.; Gaudon, C.; Ivanova, D.; Gronemeyer, H.; Tacke, R. *Organometallics* 2005, 24, 3192–3199. (e) Miller, D. J.; Showell, G. A.; Conroy, R.; Daiss, J.; Tacke, R.; Tebbe, D. (Amedis Pharmaceuticals Ltd.), PCT Int. Pat. Appl. WO 2005/005443 A1, January 20, 2005. (f) Büttner, M. W.; Burschka, C.; Daiss, J. O.; Ivanova, D.; Rochel, N.; Kammerer, S.; Peluso-Iltis, C.; Bindler, A.; Gaudon, C.; Germain, P.; Moras, D.; Gronemeyer, H.; Tacke, R. *ChemBioChem* 2007, *8*, 1688–1699.

tetramethyl-1,3-disilaindane or 1,1,4,4-tetramethyl-1,4-disila-1,2,3,4-tetrahydronaphthalene skeleton are of great interest because of their potential application as pharmaceuticals<sup>3</sup> or fragrances,<sup>4</sup> and 1,3-disila-1,3-dihydroisobenzofuranes have been reported to be valuable reagents in the generation of benzynes.<sup>5</sup>

Recently, we have shown that cobalt(II) iodide in the presence of zinc powder in acetonitrile can efficiently catalyze arene syntheses by [2+2+2] cycloaddition of monoynes with diynes,<sup>6</sup> and neither additional zinc iodide nor additional ligands are necessary for this reaction (in this context, see also ref 2d). In these syntheses, an alkynylborolane or THP-protected propargyl alcohols were used as the monoynes. In the present study we demonstrate that non-protected propargyl alcohols can also be used in these [2 + 2 + 2] cycloadditions (in this context, see also ref 7), and the protection/deprotection steps<sup>6a</sup> can be omitted.

#### **Results and Discussion**

As shown in Scheme 1 and Table 1, a series of cobaltcatalyzed [2+2+2] cycloaddition reactions of the diynes 1a-cwith the alcohols 2a-d were performed to give compounds 3al. A detailed procedure for the synthesis of 3a and analytical data (<sup>1</sup>H, <sup>13</sup>C, and <sup>29</sup>Si NMR; C,H analyses) for 3a, 3e, and 3iare reported in the Experimental Section.

The reactions proceeded very fast and were complete within ca. 5 min. The products were isolated in satisfying yields (44– 66%, not yet optimized), although they are lower compared to those obtained with the respective THP-protected substrates.<sup>6a</sup> Because of their relatively high polarity, the products **3a**–**1** could be easily separated from the byproducts by column chromatography.<sup>8</sup> Both terminal (**2a**, **2d**) and internal monoynes (**2b**, **2c**) could be used in these syntheses. The yields of compounds **3a**–**d** (1,3-disilaindane skeleton; Table 1, entries 1–4) and **3i–l** 

(4) (a) Büttner, M. W.; Penka, M.; Doszczak, L.; Kraft, P.; Tacke, R. Organometallics 2007, 26, 1295–1298. (b) Büttner, M. W.; Burschka, C.; Junold, K.; Kraft, P.; Tacke, R. ChemBioChem 2007, 8, 1447–1454. (5) (a) Kitamura, T.; Meng, Z.; Fujiwara, Y. Tetrahedron Lett. 2000,

(7) (a) Fatland, A. W.; Eaton, B. E. *Org. Lett.* **2000**, *2*, 3131–3133. (b) Saino, N.; Amemiya, F.; Tanabe, E.; Kase, K.; Okamoto, S. *Org. Lett.* **2006**, *8*, 1439–1442.

(8) Diyne "dimers" and "trimers" were typical byproducts of the [2+2+2] cycloadditions described in this paper.

<sup>(5) (</sup>a) Kitamura, T.; Meng, Z.; Fujiwara, Y. *Tetrahedron Lett.* **2000**, *41*, 6611–6614. (b) Chen, Y.-L.; Zhang, H.-K.; Wong, W.-Y.; Lee, A. W. M. *Tetrahedron Lett.* **2002**, *43*, 2259–2262.

<sup>(6) (</sup>a) Doszczak, L.; Fey, P.; Tacke, R. *Synlett* **2007**, 753–756. (b) Büttner, M. W.; Nätscher, J. B.; Burschka, C.; Tacke R. *Organometallics* **2007**, *26*, 4835–4838.



Table 1. Yields of the Arenes 3 Obtained in the Reaction ofthe Diynes 1 with the Propargyl Acohols 2 in AcetonitrileUsing a  $CoI_2/Zn$  System for Catalysis<sup>a,b</sup>

entry	diyne	Х	monoyne	R	R′	product (%) <sup>c</sup>
1	<b>1</b> a	CH <sub>2</sub>	2a	Н	Н	<b>3a</b> (65)
2	1a	$CH_2$	2b	Me	Н	<b>3b</b> (60)
3	1a	$CH_2$	2c	Et	Н	<b>3c</b> (58)
4	1a	$CH_2$	2d	Н	Me	<b>3d</b> (63)
5	1b	$CH_2CH_2$	2a	Н	Н	<b>3e</b> (44)
6	1b	$CH_2CH_2$	2b	Me	Н	<b>3f</b> (51)
7	1b	CH <sub>2</sub> CH <sub>2</sub>	2c	Et	Н	<b>3g</b> (50)
8	1b	CH <sub>2</sub> CH <sub>2</sub>	2d	Н	Me	<b>3h</b> (47)
9	1c	0	2a	Н	Н	<b>3i</b> (64)
10	1c	0	2b	Me	Н	<b>3j</b> (52)
11	1c	0	2c	Et	Н	<b>3k</b> (62)
12	1c	0	2d	Н	Me	<b>3l</b> (59)

<sup>*a*</sup> Ratio of reagents: **1** (mmol)/**2** (mmol)/CoI<sub>2</sub> (mmol)/Zn (mmol)/MeCN (mL): 1/1.4/0.025/0.1/2. <sup>*b*</sup> Reaction required cooling in a water or an ice bath (depending on the scale) to keep the temperature below 40 °C. <sup>*c*</sup> Isolated yield.

(1,3-disila-1,3-dihydroisobenzofurane skeleton; Table 1, entries 9–12) are higher than those of compounds 3e-h (1,4-disilatetralin skeleton; Table 1, entries 5–8) as in the case of the THP-protected substrates.<sup>6a</sup>

In summary, we have discovered that a simple catalytic system consisting of commercially available cobalt(II) iodide and zinc powder can efficiently catalyze [2+2+2] cycloadditions of unprotected propargyl alcohols, containing terminal or internal monoyne moieties, with diynes. The reactions can be performed under mild conditions, and a relatively low (2.5%) catalyst load gives almost complete consumption of the starting materials in a very short time. Moreover, the reaction does not need to be performed at high dilution, and the workup and isolation are straightforward.

#### **Experimental Section**

General Procedures. All syntheses were carried out under dry nitrogen. Acetonitrile was dried and purified according to standard procedures and stored under dry nitrogen. The <sup>1</sup>H, <sup>13</sup>C, and <sup>29</sup>Si NMR spectra were recorded at 22 °C on a Bruker Avance 400 NMR spectrometer (<sup>1</sup>H, 400.1 MHz; <sup>13</sup>C, 100.6 MHz; <sup>29</sup>Si, 79.5 MHz), and CDCl<sub>3</sub> was used as the solvent. Chemical shifts (ppm) were determined relative to internal CHCl<sub>3</sub> (<sup>1</sup>H,  $\delta$  7.24), internal CDCl<sub>3</sub> (<sup>13</sup>C,  $\delta$  77.0), or external TMS (<sup>29</sup>Si,  $\delta$  0). Assignment of the <sup>1</sup>H NMR data was supported by <sup>1</sup>H, <sup>1</sup>H gradient-selected COSY, <sup>13</sup>C, <sup>1</sup>H gradient-selected HMQC and gradient-selected HMBC, and <sup>29</sup>Si, <sup>1</sup>H gradient-selected HMQC experiments (optimized for <sup>2</sup>J<sub>SiH</sub> = 7 Hz). Assignment of the <sup>13</sup>C NMR data was supported by DEPT 135 and the aforementioned <sup>13</sup>C, <sup>1</sup>H correlation experiments.

Example: Synthesis of 5-(Hydroxymethyl)-1,1,3,3-tetramethyl-1,3-disilaindane (3a). Bis(ethynyldimethylsilyl)methane (1a) (902 mg, 5.00 mmol), prop-2-yn-1-ol (2a) (392 mg, 6.99 mmol), and cobalt(II) iodide (39 mg, 125 µmol) in acetonitrile (630  $\mu$ L) were added one after another to a stirred suspension of zinc (33 mg, 505  $\mu$ mol) in acetonitrile (10 mL). When the reaction mixture started to warm up (after ca. 1 min), it was cooled in an ice bath. After 5 min, the cooling bath was removed, and triethylamine (100  $\mu$ L) was added. The solvent was removed under reduced pressure, and the residue was purified by column chromatography on silica gel (63–200  $\mu$ m; eluent, *n*-pentane/diethyl ether (1/1 (v/v)) to afford **3a** (769 mg, 65%) as a white solid. <sup>1</sup>H NMR:  $\delta - 0.04$  (s, 2 H, SiCH<sub>2</sub>Si), 0.275 (s, 6 H, SiCH<sub>3</sub>), 0.282 (s, 6 H, SiCH<sub>3</sub>), 1.64 (t,  ${}^{3}J_{HH} = 6.0$  Hz, 1 H, OH), 4.70 (d,  ${}^{3}J_{HH} = 6.0$ Hz, 2 H, CH<sub>2</sub>OH), 7.35–7.38 (m, 1 H, HC-6), 7.54–7.58 (m, 2 H, *H*C-4, *H*C-7). <sup>13</sup>C NMR:  $\delta$  –2.4 (Si*C*H<sub>2</sub>Si), 0.53 (2 C, Si*C*H<sub>3</sub>), 0.54 (2 C, SiCH<sub>3</sub>), 65.7 (CH<sub>2</sub>OH), 127.5 (HC-6), 130.2 (HC-4), 132.0 (HC-7), 140.9 (C-5), 150.0 and 151.1 (C-3a, C-7a). <sup>29</sup>Si NMR: δ 8.7, 8.9. Anal. Calcd for C<sub>12</sub>H<sub>20</sub>O<sub>2</sub>Si<sub>2</sub>: C, 60.95; H, 8.53. Found: C 60.5; H 8.4.

Analytical Data for 6-(Hydroxymethyl)-1,1,4,4-tetramethyl-1,4-disila-1,2,3,4-tetrahydronaphthalene (3e). <sup>1</sup>H NMR:  $\delta$  0.21 (s, 6 H, SiCH<sub>3</sub>), 0.22 (s, 6 H, SiCH<sub>3</sub>), 0.99 (s, 4 H, SiCH<sub>2</sub>C), 1.60 (t, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 1 H, OH), 4.67 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2 H, CH<sub>2</sub>-OH), 7.32-7.35 (m, 1 H, HC-7), 7.46-7.48 (m, 1 H, HC-5), 7.48-7.51 (m, 1 H, HC-8). <sup>13</sup>C NMR:  $\delta$  -1.52 (2 C, SiCH<sub>3</sub>), -1.49 (2 C, SiCH<sub>3</sub>), 7.49 (SiCH<sub>2</sub>C), 7.53 (SiCH<sub>2</sub>C), 65.6 (CH<sub>2</sub>OH), 126.8 (HC-7), 131.9 (HC-5), 133.8 (HC-8), 140.3 (C-6), 145.3 and 146.3 (C-4a, C-8a). <sup>29</sup>Si NMR:  $\delta$  -7.0, -6.9. Anal. Calcd for C<sub>13</sub>H<sub>22</sub>-OSi<sub>2</sub>: C, 62.34; H, 8.85. Found: C, 62.0; H, 8.7.

Analytical Data for 5-(Hydroxymethyl)-1,1,3,3-tetramethyl-1,3-disila-2-oxaindane (3i). <sup>1</sup>H NMR:  $\delta$  0.34 (s, 6 H, SiCH<sub>3</sub>), 0.35 (s, 6 H, SiCH<sub>3</sub>), 1.74 (t, <sup>3</sup>J<sub>HH</sub> = 5.7 Hz, 1 H, OH), 4.72 (d, <sup>3</sup>J<sub>HH</sub> = 5.7 Hz, 2 H, CH<sub>2</sub>OH), 7.38–7.43 (m, 1 H, HC-6), 7.54–7.59 (m, 2 H, HC-4, HC-7). <sup>13</sup>C NMR:  $\delta$  0.99 (4 C, SiCH<sub>3</sub>), 65.5 (CH<sub>2</sub>-OH), 127.9 (HC-6), 129.4 (HC-4), 131.2 (HC-7), 141.4 (C-5), 147.5 and 148.7 (C-3a, C-7a). <sup>29</sup>Si NMR:  $\delta$  15.4, 15.5. Anal. Calcd for C<sub>11</sub>H<sub>18</sub>O<sub>2</sub>Si<sub>2</sub>: C, 55.41; H, 7.61. Found: C, 55.1; H, 7.3.

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