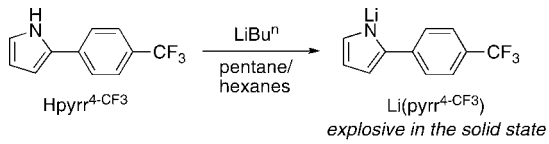
# *Additions and Corrections*

2006, Volume 25

**Douglas L. Swartz II and Aaron L. Odom*:** Synthesis, Structure, and Hydroamination Kinetics of (2,2′-Diaryldipyrrolylmethane)- and Bis(2-arylpyrrolyl)titanium Complexes

Page 6125. *Caution!* The lithium pyrrolide of 2-(4-(trifluoromethyl)phenyl)pyrrole ($Hpyrr^{4\text{-}CF3}$) was recently prepared in our laboratory by addition of *n*-butyllithium to $Hpyrr^{4\text{-}CF3}$, which allowed apparent production of the desired lithium salt. This compound, $Li(pyrr^{CF3})$, was found to be explosive in the solid state under an inert atmosphere. It is likely that *o*-$CF_3$-aryl pyrrolides can undergo a similar decomposition. *Consequently, extreme caution should be used if alkali-metal salts are produced of aryl-substituted pyrroles containing $CF_3$ groups in the ortho or para positions of the arene*.[1] It is unknown if the same decomposition can occur in an explosive manner in solution as well, but we also urge caution with solutions of such compounds. We have produced the lithium salts of pyrroles containing *m*-$CF_3$-aryl groups on numerous occasions, and, thus far, these have not undergone the same explosive decomposition.

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(1) A similar caution has been issued by Wigley and co-workers concerning Li-4-$CF_3C_6H_4$, which likely undergoes a similar decomposition: Weller, K. J.; Gray, S. D.; Briggs, P. M.; Wigley, D. E. *Organometallics* **1995**, *14*, 5588–5597.

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**Thorsten Kreickmann, Stefan Arndt, Richard R. Schrock,* and Peter Müller:** Imido Alkylidene Bispyrrolyl Complexes of Tungsten

Page 5702. In ref 4, Blanc, R. should read Blanc, F.

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