SHORT COMMUNICATIONS

Department of Organic Chemistry, Faculty of Pharmacy, Cairo University, Egypt

Fentanyl analogues with a modified propanamido group as potential affinity labels: Synthesis and *in vivo* activity

M. Y. H. ESSAWI

Ligands which form covalent bonds with a particular type or subtype of opioid receptors have provided useful tools for investigating opioid receptors and activities [1, 2]. This report describes the synthesis and analgesic activity (hotplate test) of potential affinity labels of the µ-opioid fentanyl (1) [3]. In these derivatives (2-5), the propanamido group of fentanyl was modified to include an electrophilic center which may interact with a receptor nucleophile. Reaction of 4-anilino-1-(2-phenethyl)piperidine (6) [4] with 3-chloropropionyl chloride and TEA in CH2Cl2 for 48 h at RT afforded propenanilide 2, apparently via the chloro derivative 7. Shorter reaction time in absence of TEA provided 7 · HCl (m.p. 231-235 °C, acetone) Reaction of 6 with crotonyl chloride, (E)-4-chloro-4-oxo-2-butenic acid methyl ester [5] or chloroacetyl chloride gave 3, 4 and 5, respectively. Compounds 2-5 were isolated and purified as bases or HCl salts (Table). Mouse 55 °C hot-plate test was performed as reported [6]. Solutions of compounds 2-5 were prepared by adding equimolar amounts of citric acid to their base forms in water. These and reference solutions [fentanyl citrate (Janssen) and morphine sulfate (El-Kahira)] were administered i.p. (10 mL/kg). Naloxone HCl (Dupont) was used s.c.

Compound 2 showed a remarkable dose-related analgesic effect. When compared with fentanyl (at doses of less than 1 mg/kg), it was more potent and had a longer duration of action (Fig.). The analgesic effect of fentanyl (estimated at 0.1, 0.2 and 0.5 mg/kg) dropped significantly 60-70 min and became insignificant 90-100 min after administration. At comparable doses, 2 maintained considerable analgesia 90 and 120 min after administration. In its duration, the time-response profile of 2 resembled more closely that of morphine (20 mg/kg) than that of fentanyl. At higher doses, 2 showed a maximum possible effect (MPE; animals not responding to nociception within 60 s) that was retained for several hours without signs of opioid toxicity. Thus, at a dose of 6.8 mg/kg, 50% of the animals maintained a MPE unaltered for 4.5 h. At 17 mg/kg, 100% of the animals maintained a MPE for 3 h and 50% of these animals continued as such for 6 h. Motor activity was inhibited after a dose of 25 mg/kg, however, animals were not cataleptic and returned to contineous circuling behaviour 3.5 h after treatment. Convulsions developed 1 h after a dose of 50 mg/kg and 60% lethality was observed from apparent respiratory depression.

Naloxone (2 mg/kg) administered 30 min before 2 (0.85 mg/kg) blocked the effect of 2 for about 40 min, then the tested animals showed analgesia and other mor-

Table: Physical a	and spectral	properties of	f fentanyl	derivatives 2-5
-------------------	--------------	---------------	------------	-----------------

R $1 CH_2 CH_3 (Fentanyl)$ $2 CH = CH_2$ $3 C = \overset{H}{C}CH_3$ $4 C = CCOOCH_3$ $5 CH_2Cl$ $7 CH_2CH_2Cl$						
Compd.	Yield (%)	m.p. ^a (°C) (solvent)	Formula ^b	Spectral data ^c		
2 ^d	67	101–103 (EtOAc)	C ₂₂ H ₂₆ N ₂ O	IR (KBr) 1615, 1655 cm ⁻¹ ; ¹ H NMR (CDCl ₃) δ 7.6–7.0 (m, 10H, ArH), 6.35 (d, $J = 16.7$ Hz, 1H, olefinic H cis to C=O), 5.81 (dd, $J = 16.7$ and 10.2 Hz, 1H, olefinic H geminal to C=O), 5.45 (d, $J = 10.2$ Hz, 1H, olefinic H trans to C=O), 4.74 (t, $J = 12.4$ Hz, $W_{\rm H} = 27.8$ Hz, 1H, H-4), 3.25–1.0 (m, 12H); EIMS m/z 334 (M) ⁺ (0.1%), 243 (M-CH ₂ Ph) ⁺ (100%).		
3 · HCl	21	202–204 (Me ₂ CO)	$C_{23}H_{28}N_2O\cdot HCl$	IR (KBr) 1625–1660 cm ⁻¹ ; ¹ H NMR (CD ₃ CN) δ 12.2 (br, 1H, N ⁺ H), 7.5–6.8 (m, 10H, ArH), 5.85 (m, 1H), 5.1–4.6 (m, 2H), 3.6–3.5 (m, 2H), 3.3–3.0 (m, 4H), 2.9–2.5 (m, 3H), 2.3–1.8 (m, 3H), 1.7 (d, 3H, CH ₃); EIMS <i>m</i> / <i>z</i> 346 (M-2) ⁺ (0.9%), 257 (M-CH ₂ Ph) ⁺ (4.5%).		
4	54	139–144 (CH ₂ Cl ₂)	$C_{24}H_{28}N_2O_3\\$	IR (KBr) 1630, 1660, 1725 cm ⁻¹ ; ¹ H NMR (CDCl ₃) δ 7.7–7.0 (m, 10H, Ar <i>H</i>), 6.83 (d, <i>J</i> = 15.2 Hz, 1H, C <i>H</i> COOMe), 6.57 (d, <i>J</i> = 15.2 Hz, 1H, C <i>H</i> CONPh), 4.71 (t, <i>J</i> \approx 12 Hz, <i>W</i> _H = 30.5 Hz, 1H, H-4), 3.75 (s, 3H, OCH ₃), 3.2–0.8 (m, 12 H); EIMS <i>m</i> /z 392 (M) ⁺ (0.6%), 301 (M-CH ₂ Ph) ⁺ (100%).		
5 · HCl ^e	57	250–253 (dec) (Me ₂ CO)	$C_{21}H_{25}CIN_2O\cdot HCl^f$	IR (KBr) 1665 cm ⁻¹ ; ¹ H NMR (CDCl ₃) δ 12.6 (N ⁺ <i>H</i>), 7.6–7.0 (m, 10H, Ar <i>H</i>), 4.76 (t, <i>J</i> = 11.8 Hz, <i>W</i> _H = 27.9 Hz, 1H, H-4), 3.8–2.6 (m, 9H), 2.4–1.0 (m, 5H); EIMS <i>m</i> /z 265 (M-CH ₂ Ph) ⁺ (6.2%), 91 (PhCH ₂) ⁺ (100%).		

^a Uncorrected, Griffin m.p. apparatus. ^b Analysed for C, H, N (±0.4% of theoretical values). ^c IR: Shimadzu 435; ¹H NMR: Bruker 200 (ppm from internal TMS); MS: Hewlett Packerd 5988. ^d HCl salt: m.p. 191–194 °C. ^e Base: m.p. 90–94 °C. ^f H: calculated 6.66%; Found 5.9%.

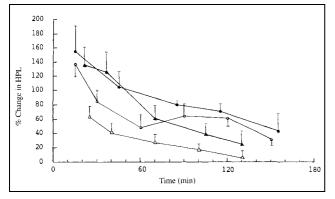


Fig.: % Changes in hot-plate latency (HPL) at several time intervals after i.p. administration at zero time of compound 2 (\bullet 0.25, \bigcirc 0.045 mg/kg) and of fentanyl (\blacktriangle 0.5, \triangle 0.1 mg/kg). Values are the mean \pm the standard error of the mean. A cut off latency time of 30 s was used.

phine-like effects for about 50 min. Also, naloxone (2 mg/kg) administered 40 min after treatment of 2 (0.85 mg/kg) reversed its effects for 70 min, then the effects of 2 returned nearly to the same level (80% of MPE) as before naloxone-administration.

Compound **3** was clearly less potent than **2** or fentanyl, however, it also showed an extented duration of analgesia that was transiently reversed by naloxone. Compounds **2** and **3** were not tested for antagonistic activity. Methyl fumarate ester **4** produced no analgesia at any time after injection. A lethality of 50% (preceded by convulsions) was observed at a dose of 80 mg/kg. Also, **4** (at 20, 40 or 80 mg/kg) exhibited no inhibition of analgesia induced by morphine (20 mg/kg) as tested 2 and 24 h after administration of **4**. In addition, **4** (40 mg/kg) produced no inhibition of fentanyl-induced analgesia (0.25 mg/kg) as tested 2 h after its administration, Chloroacetanilide (**5**) showed only a weak and transient analgesia at 17 mg/kg.

Interpretation of the above results can be difficult because of pharmacokinetic and dispotional factors *in vivo*. It is conceivable, however, that the temporary reversal of the effect of **2** by naloxone was due to noncompetitive binding of naloxone to a regulatory site which allosterically inhibited the pharmacological response of covalently bound **2** until naloxone was eliminated [7, 8]. Because naloxone could not reverse the sustained effect of the opiate-derived irreversible agonist β -chloroymorphamine (β -COA) [9], it is likely that competetive antagonism in that case was essential. This implies that **2** has a mode of interaction with μ -receptors different from morphine [10]. Binding and *in vitro* studies should delineate this and other possibilities.

Acknowledgements: I acknowledge the help and technical facilities provided by Dr. Moustafa El-Sayed, Professor of Pharmacology, Faculty of Pharmacy, Cairo University.

References

- 1 Takemori, A. E.; Portoghese, P. S.: Annu. Rev. Pharmacol. Toxicol. 25, 193 (1985)
- 2 Korlipara, V. L.; Takemori, A. E.; Portoghese, P. S.: J. Med. Chem. 38, 1337 (1995)
- 3 Zhu, J.; Yin, J.; Law, P.-Y.; Claude, P. A.; Rice, K. C.; Evans, C. J.; Chen, C.; Yu, L.; Liu-Chen, L.-Y.; J. Biol. Chem. 271, 1430 (1996)
- 4 Janssen, P. A. J.: Br. J. Anaesth. 34, 260 (1962)
- 5 Acheson, R. M.; Feinberg, R. S.; Hands, A. R.: J. Chem. Soc. 526 (1964)
- 6 Eddy, N. B.; Leimbach, D.: J. Pharmacol. Exp. Ther. 107, 385 (1953)
 7 Cowi, A. L.; Kosterlitz, H. W.; Watt, A. J.; Nature (London) 220, 1040
- (1968)8 Sayer, L. M.; Larson, D. L.; Takemori, A. E.; Portoghese, P. S.: J. Med. Chem. 27, 1325 (1984)

9 Caruso, T. P.; Takemori, A. E.; Larson, D. L., Portoghese, P. S.; Science 204, 316 (1979)

10 Portoghese, P. S.: J. Med. Chem. 8, 609 (1965)

Received August 19, 1998 Accepted November 5, 1998 Dr. M. Y. H. Essawi Department of Organic Chemistry Faculty of Pharmacy Cairo University Kasr-El-Ainy, Cairo 11562 Egypt.