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## Synthesis of new pyridoquinoxalines, thienopyridoquinoxalines and pyrimidothienopyridoquinoxalines

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Synthesis of 3-chloro-2-cyanoquinoxaline (**1**) and its reactions with sodium azide, guanidine hydrochloride, semicarbazide and thioheterocycles have been investigated (**2**–**7**). Also, the reaction of the chloro compound **1** with cyanoacetamide or cyanothioacetamide gave the pyrido[2,3-*b*]quinoxaline derivatives **8**, **9**. Compound **9** was used as a key intermediate to produce the more polyheterocyclic systems **10**–**18**.

### 1. Introduction

We have previously reported the synthesis and reactions of some new quinoxaline derivatives. Compounds with a quinoxaline nucleus significant have biological activities [1–4]. For example, pyridazinoquinoxaline and ditriazoloquinoxaline derivatives show excellent bactericidal and fungicidal activity [5, 6]. Also, 3,6,7-substituted-2-quinoxalinone and 6,7-difluoro-3-alkyl(aryl)-substituted-2-quinoxalinone have been used for their anti-microbial, anticancer, and anti HIV activities, and as interleukin receptor antagonists and can be used in the treatment of a chemokine-mediated disease, inflammatory bowel disease, Crohn's disease, Alzheimer's disease and allergic disease [7–9]. The present investigation which continues our work on the quinoxaline moiety [10–15] is concerned with the use of 3-chloro-2-cyanoquinoxaline [16] for the synthesis of many fused quinoxaline heterocycles of a new type and it therefore appears likely that these compounds will exhibit interesting biological properties.

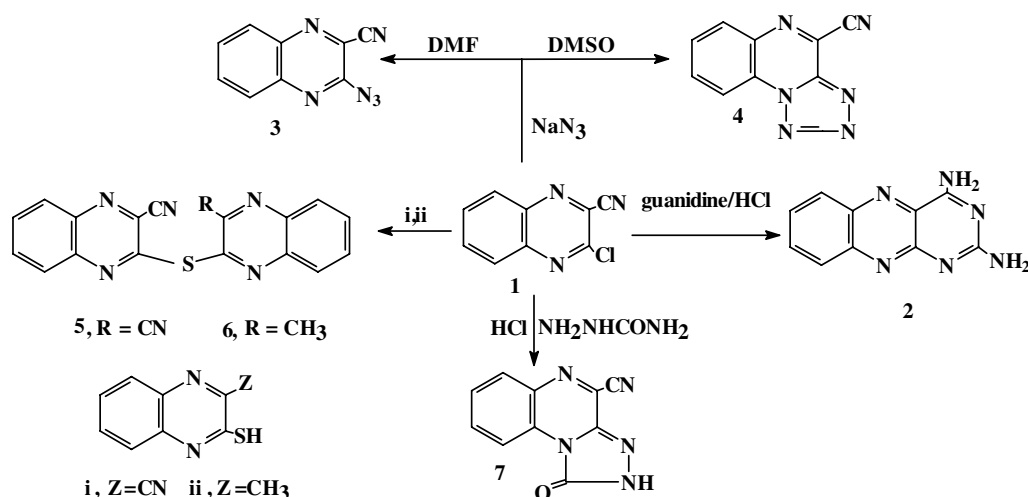
### 2. Investigations, results and discussion

3-Chloro-2-cyanoquinoxaline (**1**) [16] with a vicinal chloro-cyano group was envisaged as a potential starting material for the synthesis of fused heterocycle systems. Thus, treatment of **1** with guanidine hydrochloride in sodium

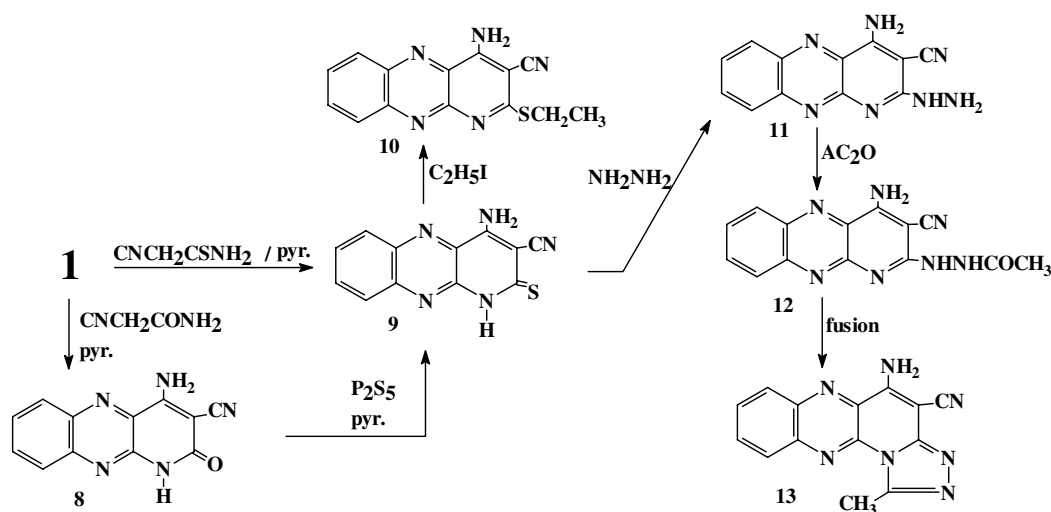
ethoxide yielded pyrimido[4,5-*b*]quinoxaline **2**. Azidoquinoxaline **3** was prepared by the reaction of chloroquinoxaline **1** with sodium azide in DMF. Like similar heterocyclic azides having the azido group attached to the cyclic carbon atom adjacent to an annular nitrogen, it may exist as a true azide or as tetrazolo[1,5-*a*]quinoxaline **4** (through reaction with sodium azide in DMSO). Treatment of **1** with 3-mercapto-2-cyano-quinoxaline and 3-mercapto-2-methyl-quinoxaline yielded the bis(quinoxalin-2-yl)sulified derivatives **5**, **6**. 5-Cyano-*s*-triazolo[4,3-*a*]quinoxaline **7** was obtained by treatment of **1** with semicarbazide hydrochloride (Scheme 1).

Reaction of **1** with cyanoacetamide in pyridine gave 4-amino-3-cyano-pyrido[2,3-*b*]quinoxalin-2(1*H*)-one (**8**) which was thionated by P<sub>2</sub>S<sub>5</sub> in pyridine to yield 4-amino-3-cyano-pyrido[2,3-*b*]quinoxaline-2(1*H*)-thione (**9**). The latter thio compound was also produced directly by reaction of **1** with cyanothioacetamide in pyridine. Compound **9** was used as a key intermediate to produce other heterocycle rings thus, reaction of **9** with ethyl iodide gave the 3-ethylthio-pyridoquinoxaline derivative **10**, while hydrazinolysis with hydrazine hydrate yielded 4-amino-3-cyano-2-hydrazino-pyrido[2,3-*b*]quinoxaline (**11**). Acylation of **11** by boiling with acetic anhydride yielded 4-amino-3-cyano-2-acetylhydrazino-pyrido[2,3-*b*]quinoxaline (**12**), and ring closure of **12** gave 6-amino-5-cyano-2-methyl-1,2,4-triazolo[4',3':1,6]-pyrido[2,3-*b*]quinoxaline (**13**, Scheme 2).

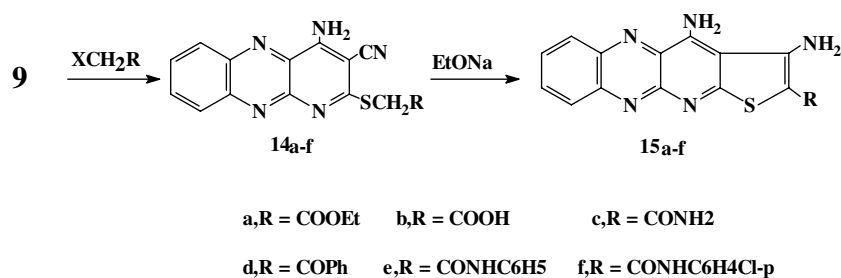
Scheme 1



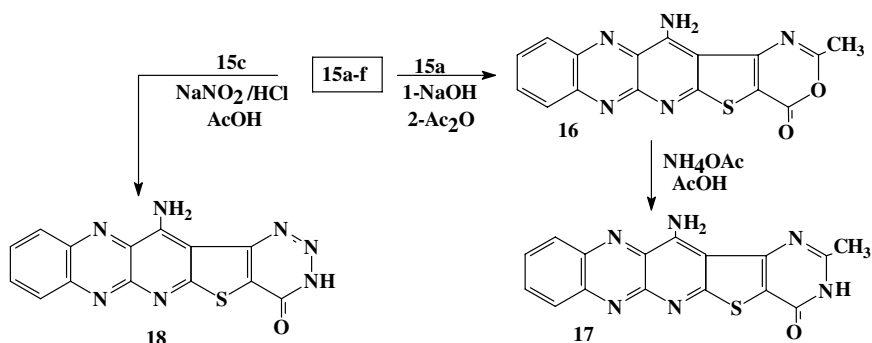
Scheme 2



Scheme 3



Scheme 4



Also, alkylation of **9** with  $\alpha$ -halo compounds (e.g. ethyl chloroacetate, chloroacetic acid, chloroacetamide, phenacyl bromide, chloroacetanilide or *p*-chloro-chloroacetanilide) in alcoholic solution of anh. sodium acetate yielded the substituted thio intermediates **14a-f**, respectively, which upon treatment with sodium ethoxide produce the thienopyridoquinoxaline derivatives **15a-f** (Scheme 3).

Some of the latter derivatives were chosen and subjected to additional reaction to build up pentacyclic heterocycles e.g. the alkaline hydrolysis of **15a** with sodium hydroxide gave the sodium salt. This on refluxing in acetic anhydride yielded the oxazino compound **16** which in turn was reacted with ammonium acetate in acetic acid to give the

pyrimidinone derivative **17**. Also, compound **15c** was reacted with concentrated hydrochloric acid and sodium nitrite in the presence of acetic acid at  $-5^\circ\text{C}$  to give the triazinothienopyridoquinoxaline derivative **18** (Scheme 4).

### 3. Experimental

Melting points were determined on a Gallenkamp apparatus and were uncorrected. IR spectra were recorded on a Pye-Unicam SP<sup>3</sup>-100 spectrophotometer using the KBr wafer technique. <sup>1</sup>H NMR spectra were measured on a Varian 390–90 MHz NMR spectrometer in a suitable deuterated solvent, using TMS as internal standard. Elemental analyses were performed on a Perkin-Elmer 240 C microanalyzer. Elemental analysis

**Table 1: Melting points, yields and analytical data of compounds 2–18**

Compd.	M.P °C (Yield %)	Formula Mol.Wt
<b>2</b>	310 (80)	C <sub>10</sub> H <sub>8</sub> N <sub>6</sub> , 212
<b>3</b>	180 (68)	C <sub>9</sub> H <sub>4</sub> N <sub>6</sub> , 196
<b>4</b>	220–221 (75)	C <sub>9</sub> H <sub>4</sub> N <sub>6</sub> , 196
<b>5</b>	240 (83)	C <sub>18</sub> H <sub>8</sub> N <sub>6</sub> S, 340
<b>6</b>	190 (80)	C <sub>18</sub> H <sub>11</sub> N <sub>5</sub> S, 329
<b>7</b>	290 (70)	C <sub>10</sub> H <sub>5</sub> N <sub>5</sub> O, 211
<b>8</b>	225 (77)	C <sub>12</sub> H <sub>7</sub> N <sub>5</sub> O, 237
<b>9</b>	>360 (70)	C <sub>12</sub> H <sub>7</sub> N <sub>5</sub> S, 253
<b>10</b>	115 (90)	C <sub>14</sub> H <sub>11</sub> N <sub>5</sub> S, 281
<b>11</b>	320 (81)	C <sub>12</sub> H <sub>9</sub> N <sub>7</sub> , 251
<b>12</b>	260 (82)	C <sub>14</sub> H <sub>11</sub> N <sub>7</sub> O, 293
<b>13</b>	360 (75)	C <sub>14</sub> H <sub>9</sub> N <sub>7</sub> , 275
<b>14a</b>	120 (90)	C <sub>16</sub> H <sub>13</sub> N <sub>5</sub> O <sub>2</sub> S, 339
<b>14b</b>	150 (78)	C <sub>14</sub> H <sub>9</sub> N <sub>5</sub> O <sub>2</sub> S, 311
<b>14c</b>	240 (83)	C <sub>14</sub> H <sub>10</sub> N <sub>6</sub> OS, 310
<b>14d</b>	160 (70)	C <sub>20</sub> H <sub>13</sub> N <sub>5</sub> OS, 371
<b>14e</b>	190 (80)	C <sub>20</sub> H <sub>14</sub> N <sub>6</sub> OS, 386
<b>14f*</b>	155 (85)	C <sub>20</sub> H <sub>13</sub> N <sub>6</sub> OSCl, 420.5
<b>15a</b>	255 (77)	C <sub>16</sub> H <sub>13</sub> N <sub>5</sub> O <sub>2</sub> S, 339
<b>15b</b>	240 (68)	C <sub>14</sub> H <sub>9</sub> N <sub>5</sub> O <sub>2</sub> S, 311
<b>15c</b>	330 (72)	C <sub>14</sub> H <sub>10</sub> N <sub>6</sub> OS, 310
<b>15d</b>	235 (72)	C <sub>20</sub> H <sub>13</sub> N <sub>5</sub> OS, 371
<b>15e</b>	255 (65)	C <sub>20</sub> H <sub>14</sub> N <sub>6</sub> OS, 386
<b>15f**</b>	260 (70)	C <sub>20</sub> H <sub>13</sub> N <sub>6</sub> OSCl, 420.5
<b>16</b>	225 (75)	C <sub>16</sub> H <sub>9</sub> N <sub>5</sub> O <sub>2</sub> S, 335
<b>17</b>	295 (65)	C <sub>16</sub> H <sub>10</sub> N <sub>6</sub> OS, 334
<b>18</b>	>360 (68)	C <sub>14</sub> H <sub>7</sub> N <sub>7</sub> OS, 321

\*, \*\* Cl (calc. 8.44, found 8.41, 8.38% on respectively)

gave acceptable results unless otherwise stated. Melting points, yields and spectroscopic data are listed in Tables 1 and 2.

### 3.1. 3-Chloro-2-cyanoquinoxaline (1)

Compound **1** was prepared according to the literature [16], m.p. 160 °C.

### 3.2. 2,4-Diamino-pyrimido[4,5-b]quinoxaline (2)

To a solution of sodium ethoxide [0.5 g (0.02 mol) of sodium and 50 ml abs. ethanol] guanidine hydrochloride (0.01 mol) was added and refluxed for 2 h. Compound **1** (0.01 mol) in abs. ethanol (20 ml) was added dropwise. After being well stirred the reaction mixture was refluxed for 6 h. The solid obtained upon dilution with water was filtered off and recrystallized from ethanol as yellow crystals.

### 3.3. 3-Azido-2-cyanoquinoxaline (3)

A mixture of **1** (0.01 mol) in DMF (20 ml) and sodium azide (0.01 mol) was stirred for 3 h, diluted with water and neutralized with HCl. The solid obtained upon dilution with water was filtered off and recrystallized from acetic acid as yellow crystals.

### 3.4. Tetrazolo[1,5-a]quinoxaline-4-carbonitrile (4)

A mixture of **1** (0.01 mol) in DMSO (20 ml) and sodium azide (0.01 mol) was stirred for 4 h, diluted with water and neutralized with HCl. The solid obtained upon dilution with water was filtered off and recrystallized from acetic acid as brown crystals.

### 3.5. Bis (3-cyanoquinoxaline-2-yl)sulfide (5)

A mixture of **1** (0.01 mol) in 20 ml of 25% aqueous NaOH and 2-cyanoquinoxaline-3(1H)-thione (0.01 mol) was heated for 2 h. The reaction mixture was cooled, diluted with water and neutralized with dilute acetic acid. The solid obtained was filtered off and recrystallized from ethanol as pale red crystals.

### 3.6. 3-Cyano-3'-methyl-bis(quinoxalin-2-yl)sulfide (6)

A mixture of **1** (0.01 mol) in 20 ml of 25% aqueous NaOH and 3-methylquinoxaline-2(1H)-thione (0.01 mol) was heated for 2 h. The reaction mixture was cooled, diluted with water and neutralized with dilute acetic acid. The solid obtained was filtered off and recrystallized from ethanol as red crystals.

**Table 2: Spectroscopic data of compounds 2–18**

Compd.	IR( $\nu$ cm <sup>-1</sup> )/ <sup>1</sup> H NMR $\delta$ (ppm)
<b>2</b>	3120 (NH <sub>2</sub> ), 1620 (C=N); (DMSO-d <sub>6</sub> ): $\delta$ 4.2 (s, 2 H, NH <sub>2</sub> ), $\delta$ 6.2 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.3–7.8 (m, 4 H, Ar-H).
<b>3</b>	2210 (CN), 2120 (N3); (CF <sub>3</sub> COOD): $\delta$ 7.3–8.0 (m, 4 H, Ar-H).
<b>4</b>	2200 (CN), 1620 (C=N); (CF <sub>3</sub> COOD): $\delta$ 7.4–8.1 (m, 4 H, Ar-H).
<b>5</b>	2220 (bro. 2 CN), 1620 (C=N); (CDCl <sub>3</sub> ): $\delta$ 7.5–8.0 (m, 8 H, Ar-H).
<b>6</b>	2220 (CN), 1610 (C=N); (CDCl <sub>3</sub> ): $\delta$ 2.8 (s, 3 H, CH <sub>3</sub> ), $\delta$ 7.5–8.3 (m, 8 H, Ar-H).
<b>7</b>	3230 (NH), 2220 (CN), 1680 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 7.5–7.8 (m, 4 H, Ar-H), $\delta$ 10.5 (s, 2 H, NH).
<b>8</b>	3180–3420 (NH, NH <sub>2</sub> ), 2220 (CN), 1670 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 6.2 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.4–7.9 (m, 4 H, Ar-H), 9.1 (s, 1 H, NH).
<b>9</b>	3200–3380 (NH, NH <sub>2</sub> ), 2210 (CN), 1230 (C=S); (DMSO-d <sub>6</sub> ): $\delta$ 4.9 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.4–8.1 (m, 4 H, Ar-H), $\delta$ 9.5 (s, 1 H, NH).
<b>10</b>	3400 (NH <sub>2</sub> ), 2980 (CH, aliph.), 2220 (CN); (DMSO-d <sub>6</sub> ): $\delta$ 1.3–1.6 (t, 3 H, CH <sub>3</sub> ), $\delta$ 3.2–3.5 (q, 2 H, CH <sub>2</sub> ), $\delta$ 5.8 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.6–8.2 (m, 4 H, Ar-H).
<b>11</b>	3220, 3430 (NH, NH <sub>2</sub> ), 2200 (CN); (CF <sub>3</sub> COOD): $\delta$ 7.5–8.2 (m, 4 H, Ar-H).
<b>12</b>	3220, 3500 (NH, NH <sub>2</sub> ), 2220 (CN), 1620 (C=N); (CF <sub>3</sub> COOD): $\delta$ 2.4 (s, 3 H, CH <sub>3</sub> ), $\delta$ 7.6–8.1 (m, 4 H, Ar-H).
<b>13</b>	3320 (NH <sub>2</sub> ), 2220 (CN); (CDCl <sub>3</sub> ): $\delta$ 2.4 (s, 3 H, CH <sub>3</sub> ), $\delta$ 6.0 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.6–8.2 (m, 4 H, Ar-H).
<b>14a</b>	3400 (NH <sub>2</sub> ), 2980 (CH, aliph.), 2200 (CN), 1730 (C=O); (CDCl <sub>3</sub> ): $\delta$ 1.5–1.9 (t, 3 H, CH <sub>3</sub> ), $\delta$ 3.9–4.1 (q, 2 H, CH <sub>2</sub> ), $\delta$ 4.6 (s, 2 H, CH <sub>2</sub> ), $\delta$ 6.1 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.2–8.0 (m, 4 H, Ar-H).
<b>14b</b>	3580 (OH), 3420 (NH <sub>2</sub> ), 2220 (CN), 1690 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 4.2 (s, 2 H, CH <sub>2</sub> ), $\delta$ 5.9 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.5–7.9 (m, 4 H, Ar-H).
<b>14c</b>	3400 (NH <sub>2</sub> ), 2200 (CN), 1680 (C=O); (CF <sub>3</sub> COOD): $\delta$ 4.3 (s, 2 H, CH <sub>2</sub> ), $\delta$ 7.4–8.00 (m, 4 H, Ar-H).
<b>14d</b>	3250, 3400 (NH <sub>2</sub> ), 2200 (CN), 1740 (C=O); (CF <sub>3</sub> COOD): $\delta$ 4.1 (s, 2 H, CH <sub>2</sub> ), $\delta$ 7.4–8.2 (m, 9 H, Ar-H).
<b>14e</b>	3150, 3330 (NH, NH <sub>2</sub> ), 2200 (CN), 1700 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 4.2 (s, 2 H, CH <sub>2</sub> ), $\delta$ 6.1 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.5–8.4 (m, 9 H, Ar-H), 8.9 (s, 1 H, NH).
<b>14f</b>	3100, 3390 (NH, NH <sub>2</sub> ), 2200 (CN). 1660 (C=O); (CF <sub>3</sub> COOD): $\delta$ 4.3 (s, 2 H, CH <sub>2</sub> ), $\delta$ 7.3–8.2 (m, 8 H, Ar-H).
<b>15a</b>	3300, 3400 (NH <sub>2</sub> ), 1660 (C=O); (CDCl <sub>3</sub> ): $\delta$ 1.2–1.5 (t, 3 H, CH <sub>3</sub> ), $\delta$ 3.9–4.1 (q, 2 H, CH <sub>2</sub> ), $\delta$ 6.4 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.4–8.0 (m, 4 H, Ar-H).
<b>15b</b>	3280, 3400 (NH <sub>2</sub> ), 1700 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 5.9 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.6–7.9 (m, 4 H, Ar-H).
<b>15c</b>	3180 (NH <sub>2</sub> ), 1660 (C=O); (CF <sub>3</sub> COOD): $\delta$ 7.3–8.1 (m, 4 H, Ar-H).
<b>15d</b>	3320 (NH <sub>2</sub> ), 1680 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 6.3 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.5–8.4 (m, 9 H, Ar-H).
<b>15e</b>	3200–3420 (NH, NH <sub>2</sub> ), 1650 (C=O); (CF <sub>3</sub> COOD): $\delta$ 7.5–8.4 (m, 9 H, Ar-H).
<b>15f</b>	3300, 3480 (NH, NH <sub>2</sub> ), 1640 (C=O); (CF <sub>3</sub> COOD): $\delta$ 7.5–8.4 (m, 8 H, Ar-H).
<b>16</b>	3350 (NH <sub>2</sub> ), 1690 (C=O); (CDCl <sub>3</sub> ): $\delta$ 2.3 (s, 3 H, CH <sub>3</sub> ), $\delta$ 6.0 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.2–8.1 (m, 4 H, Ar-H).
<b>17</b>	3160–3400 (NH, NH <sub>2</sub> ), 1660 (C=O); (DMSO-d <sub>6</sub> ): $\delta$ 6.1 (s, 2 H, NH <sub>2</sub> ), $\delta$ 7.6–8.2 (m, 4 H, Ar-H), 9.8 (s, 1 H, NH).
<b>18</b>	3200, 3420 (NH, NH <sub>2</sub> ), 1650 (C=O); (CF <sub>3</sub> COOD): $\delta$ 7.5–8.3 (m, 4 H, Ar-H).

### 3.7. 5-Cyano-1,2,4-triazolo[4,3-a]quinoxaline (7)

A mixture of **1** (0.01 mol) and semicarbazide hydrochloride (0.012 mol) in abs. ethanol (25 ml) was treated with a few drops of conc HCl and refluxed for 7 h. The solid obtained was filtered off and recrystallized from ethanol as yellow crystals.

**3.8. 4-Amino-3-cyano-pyrido[2,3-b]quinoxalin-2(1H)-one (8)**

A mixture of **1** (0.01 mol) and cyanoacetamide (0.01 mol) in pyridine (30 ml) was refluxed for 4 h, poured onto cold water and neutralized with dilute acetic acid. The solid obtained was filtered off and recrystallized from ethanol as brown crystals.

**3.9. 4-Amino-3-cyano-pyrido[2,3-b]quinoxaline-2(1H)-thione (9)**

A mixture of **1** (0.01 mol) and cyanothioacetamide (0.01 mol) in pyridine (35 ml) was refluxed for 4 h, poured onto cold water and neutralized with dilute acetic acid. The solid obtained was filtered off and recrystallized from acetic acid as bright deep red crystals.

**3.10. 4-Amino-3-cyano-2-ethylthio-pyrido[2,3-b]quinoxaline (10)**

A mixture of **9** (0.01 mol), ethyl iodide (0.01 mol) and anhydrous sodium acetate (5 g) in ethanol (40 ml) was refluxed for 2 h, poured onto cold water. The solid obtained was filtered off and recrystallized from ethanol as pale yellow crystals.

**3.11. 4-Amino-3-cyano-2-hydrazino-pyrido[2,3-b]quinoxaline (11)**

A mixture of **9** (0.01 mol) and hydrazine hydrate (6 ml) was refluxed in ethanol (35 ml) for 4 h or until evolution of H<sub>2</sub>S ceased then cooled, and the yellow precipitate was filtered off and recrystallized from ethanol.

**3.12. 4-Amino-3-cyano-2-acetylhydrazino-pyrido[2,3-b]quinoxaline (12)**

A solution of **11** (0.01 mol) in acetic anhydride (25 ml) was refluxed for 3 h, then cooled and poured onto ice/water. The precipitate thus formed was collected and recrystallized from ethanol as pale yellow crystals.

**3.13. 6-Amino-5-cyano-2-methyl-1,2,4-triazolo[4',3':1,6]pyrido[2,3-b]quinoxaline (13)**

Acetylhydrazino **12** (0.5 g) was heated to melting and refluxed for 15 min, then cooled. The solid thus formed was recrystallized from acetic acid as pale brown crystals.

**3.14. 4-Amino-3-cyano-2-substitutedthio-pyrido[2,3-b]quinoxaline (14a–f)**

A mixture of **9** (0.1 mol) and  $\alpha$ -halo carbonyl compound (0.1 mol) in ethanol (30 ml) in the presence of anh. sodium acetate (5 g) was refluxed for 2 h, and poured onto cold water. The solid obtained was filtered off and recrystallized from ethanol. The physical constants and spectral data of compounds **14a–f** are summarized in Table 1.

**3.15. 3,4-Diamino-2(substituted)thieno[2',3':5,6]pyrido[2,3-b]quinoxaline (15)**

A sample of compounds **14a–f** (0.5 g) in (25 ml) ethanolic ethoxide solution was refluxed for 1 h. The solid product separated from the hot mixture was filtered off and recrystallized from the proper solvent. The physical constants and spectral data of compounds **15a–f** are summarized in Tables 1, 2.

**3.16. 13-Amino-2-methyl-oxazino[4'',5'':4',5']thieno[2',3':2,3]pyrido[2,3-b]quinoxalin-4-one (16)**

A sample of **15a** (1 g) was refluxed in 30 ml alcoholic NaOH 10% for 2 h. The red sodium salt was separated and was then filtered off, and washed several times with ethanol. The latter sod. salt was refluxed in acetic anhydride (25 ml) for 2 h. The solid product which was produced on heating was filtered off and recrystallized from ethanol as yellow crystals.

**3.17. 13-Amino-2-methyl-pyrimido[4'',5'':4',5']thieno[2',3':2,3]pyrido[2,3-b]quinoxalin-4(3H)-one (17)**

A mixture of the oxazino compound **16** (0.5 g) and ammonium acetate (4 g) in acetic acid (20 ml) was refluxed for 2 h, and the solid product separated from the hot mixture was filtered off, washed with water and recrystallized from acetic acid as yellow crystals.

**3.18. 13-Amino-1,2,3-triazino[4'',5'':4',5']thieno[2',3':2,3]pyrido[2,3-b]quinoxalin-4(3H)-one (18)**

The title compound was prepared by treatment of compound **15c** (0.01 mol) with hydrochloric acid while adding dropwise sodium nitrite solution (20 ml) at  $-5^{\circ}\text{C}$  in presence of acetic acid (10 ml) and stirring for 2 h. The solid separated was filtered off and recrystallized from acetic acid as yellow crystals.

**References**

- Bamby, R. E.; in: Wolf, M. E. (ed.): *Burgers Medicinal Chemistry*, part II, 69 John Wiley – NewYork 1979
- Sarges, R.; Howard, H. R.; Lebel, L. A.; Seymour, P. A.; Koe, B. K.: *J. Med. Chem.* **33**, 2240 (1990)
- El-Kerdawy, M.; Tantawy, A.; Gad, L. M.; Rady, E.: *Chin. Pharm. J.* **43**, 335 (1991); *C.A.* **116**, 59325b (1992)
- El-Deen, I. M.; Abd El-Fattah, M. E.: *Indian J. Heterocyclic. Chem.* **8** (4), 319 (1999)
- Wermuth, C. G.: *Actural Chim. Ther.* **12**, 3 (1985)
- Reddysastry, C. V.; Jogibhukta, M.; Verma, R. K.; Kaushal, R.: *Indian J. Chem.* **27B**, 1110 (1988)
- Sanna, P.; Carta, A.; Loriga, M.; Zanetti, S.; Sechi, L.: *Farmaco* **54** 161 (1999)
- Sanna, P.; Carta, A.; Loriga, M.; Zanetti, S.; Sechi, L.: *Farmaco* **54** 169 (1999)
- Kenneth, G. C.; Connor, T. D.; Jack, L. J.; Miller, R. S.: (Warner-Lambert Company, USA) 26 Aug. 1999, USAppl (patent), *C. A.* **131**, 170359q (1999)
- Moustafa, O. S.: *Phosphorus, Sulfur, Silicon*, **131**, 49 (1997)
- Moustafa, O. S.; Bachiet, E. A.; Badr, M. Z. A.: *Afinidad* **476**, 285 (1998)
- Moustafa, O. S.: *Phosphorus, Sulfur, Silicon*, in press
- Moustafa, O. S.; Badr, M. Z. A.: *Phosphorus, Sulfur, Silicon*. **119**, 127 (1996)
- Moustafa, O. S.: *J. Chinese Chem. Soc.* **47** (2), 351 (2000)
- Geies, A. A.; El-Deen, A. M. K.; Moustafa, O. S.; *Pharmazie* **52**, 436 (1996)
- Yoshida, K.; Otomasu, H.: *Chem. Pharm. Bull.* **32**, 3361 (1999)

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