

Department of Chemistry, Faculty of Science, Zagazig University, Zagazig, Egypt

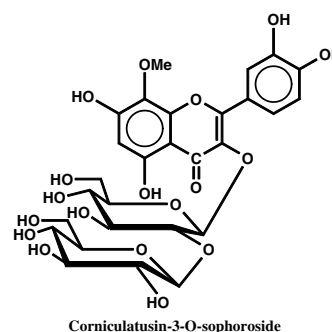
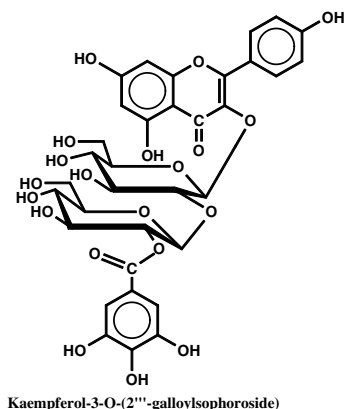
## Polyphenols of Egyptian Rosaceae plants – two new flavonoid glycosides from *Sanguisorba minor* Scop.

A. M. D. EL-MOUSALLAMY

Two new flavonol glycosides, 8-methoxyquercetin-3-*O*- $\beta$ -glucosyl-(1'''-2'')-*O*- $\beta$ -glucoside and kaempferol-3-*O*-[2'''-galloyl-*O*- $\beta$ -glucosyl-(1'''-2'')-*O*- $\beta$ -glucoside], together with five known quercetin and kaempferol 3-*O*-mono-glycosides, were isolated and identified from the aerial parts of *Sanguisorba minor* Scop. (Rosaceae). All structures were determined by routine methods of analysis and confirmed mostly by negative ESI-MS,  $^1\text{H}$ - and  $^{13}\text{C}$  NMR.

### 1. Introduction

Many members of Rosaceae predominantly produce a variety of polyphenolic conjugates which includes beside, ellagitannins [1, 2] several flavonoid glycosides. These flavonoids mainly are apigenin-*C*-glycosyls [3, 4], quercetin and kaempferol 3-*O*-glycosides and 3-*O*-galloylated glycosides [5]. They also, include 8-methoxykaempferol-3-*O*-neohesperidoside, 3-*O*-glucoside [6], 3-*O*-sophoroside [7]. Rosaceae is represented in Egypt by seven species, belonging to six different genera [8]. They provide extracts which are used in traditional medicine for treating cardiac insufficiency corresponding to stages I & II, viral infection diseases, diarrhea and enuresis [9]. In previous studies of the phenolic conjugates of Egyptian Rosaceae medicinal plants, the isolation and characterization of a di-*C*- $\beta$ -cellobiosylapigenin, together with 13 other phenolics, from the whole plant of *Cotoneaster orbicularis* Schldl. [3], and of quercetin-3-*O*-galloylgalactoside, among 10 additional phenolics from the whole *Rosa arabica* plant [5], as well as the isolation and characterization of a diacetyl-2''-*O*- $\alpha$ -rhamnosylvitexin in addition to 5 flavonoid conjugates from the leaves of *Crataegus sinaica* Boiss. [4], have been reported. In continuation of these studies isolated and identified, along with the known glycosides, kaempferol-3-*O*- $\beta$ -glucuronide (3), 3-*O*- $\beta$ -galactide (4), quercetin-3-*O*- $\beta$ -glucuronide (5), -3-*O*- $\beta$ -galactoside (6) and -3-*O*- $\beta$ -arabinoside (7) the new conjugates, 8-methoxyquercetin-3-*O*- $\beta$ -glucopyranosyl-(1'''-2'')- $\beta$ -glucopyranoside, or corniculatusin-3-*O*- $\beta$ -sophoroside (1) and kaempferol-3-*O*-[2'''-galloyl- $\beta$ -glucosyl-(1'''-2'')- $\beta$ -glucoside], or kaempferol-3-*O*-[2'''-galloyl- $\beta$ -sophoroside) (2), from the aqueous ethanolic whole plant extract of *Sanguisorba minor* Scop.



### 2. Investigations, results and discussion

Compounds 1–7 were isolated from an aqueous ethanolic (3:1) extract of the ground aerial parts of *S. minor* by applying repeated Sephadex LH-20 column and preparative paper chromatography (PPC). The known compounds 3–7 showed chromatographic, UV absorption, hydrolytic, negative ESI-MS,  $^1\text{H}$ - and  $^{13}\text{C}$  NMR results identical with those of kaempferol-3-*O*- $\beta$ -galactopyranoside [10], kaempferol-3-*O*- $\beta$ -glucopyranuronide [11], quercetin-3-*O*- $\beta$ -galactopyranoside [12], quercetin-3-*O*- $\beta$ -glucopyranuronide [11] and quercetin-3-*O*- $\beta$ -arabinopyranoside [13], respectively.

Compound 1 was obtained as a brownish yellow amorphous powder. The molecular weight ( $M_r$ ) was determined as 656 (negative ESI-MS,  $[\text{M}-1]^-$  ion at  $m/z$  655). Direct measurements of the ESI-MS at collision induced dissociation voltage (CID) [14] led to formation of fragment ions at  $m/z$  640, 493, 475, 331.07, 315.07, 303.07 and 166, a pattern of fragmentation which when incorporated with the above given analytical data would be best interpreted in terms of a 3,8-di-*O*-substituted gossypetin structure in which the substituent at position 8 is a methyl group and that at position 3 is a di-*O*-hexoside. This view was supported by acid hydrolysis of 1, (aqueous 2 N HCl, 100 °C, 2 h) to yield 8-methoxyquercetin, corniculatusin [15] ( $R_f$ s, UV spectral, -ve ESI-MS,  $^1\text{H}$  and  $^{13}\text{C}$  NMR data) and glucose (Co-PC). Partial  $\beta$ -glucosidase hydrolysis of 1 (0.5 ml of the enzyme in 0.05 M acetate buffer, pH 5.1, 37 °C) gave an intermediate which was isolated from an ethyl acetate extract of the hydrolysate by PPC and was then identified as corniculatusin-3-*O*- $\beta$ -glucoside [16]. Partial acid hydrolysis of 1 (0.1 N aqueous HCl, 100 °C, 5 min) gave the same intermediate, corniculatusin 3-*O*-glucoside (Co-PC). Consequently, compound 1 must be corniculatusin 3-*O*-glucosylglucoside.  $^1\text{H}$  NMR of 1 confirmed this conclusion and showed, in the recorded

spectrum (DMSO- $d_6$ ), two anomeric sugar proton resonances at  $\delta$  5.68 ( $d$ ,  $J = 8$  Hz) assignable to the  $1''$ - $\beta$ -anomeric glucosyl proton, and at  $\delta$  4.61 ( $d$ ,  $J = 8$  Hz) attributable to the  $1'''$ - $\beta$ -anomeric glucoside proton. It also showed a singlet at  $\delta$  3.81, assignable to a methoxyl protons and in the aromatic region, a singlet at  $\delta$  6.22, a doublet ( $J = 7.5$  Hz) at 6.90, a doublet ( $J = 2.5$  Hz) at 7.60 and a double doublet ( $J = 7.5$  and 2.5 Hz) at 7.64, assignable to the protons of the 8-OMe, 6-H, 5'-H, 2'-H and 6'-H, respectively in the corniculatusin moiety of **1**. Further confirmation of the achieved structure of **1** was received through  $^{13}\text{C}$  NMR analysis. The spectrum exhibited in the aromatic region a pattern of carbon resonances consistent with 3- $O$ -substituted corniculatusin (see Experimental). It also showed a group of 12 carbon resonances in the sugar region, thus indicating a di- $O$ -glycosylation. The anomeric sugar carbon resonances were recognized at  $\delta$  98.6 (C-1''), and at 103.9 (C-1'''). The carbon resonance localized in this spectrum, at  $\delta$  83.2 was assigned to the glucosylated C-2'' glucoside carbon, being shifted downfield due to glucosylation of its geminal hydroxyl group ( $\alpha$  effect), (in comparison with the corresponding resonances in the spectra of flavonol-3- $O$ - $\beta$ -glucoside [17]). That  $O$ -glycosylation in **1** is present at C-3 of the flavonol moiety which was confirmed by the upfield shift of this carbon resonance and the accompanying large downfield shift of the C-2 resonance of the same moiety [all in comparison with the corresponding resonances in the spectrum of the aglycone corniculatusin (see Experimental)]. Other glucosyl and glucoside carbon resonances, in the received spectrum possessed chemical shift values which were in close agreement with structure of **1** as corniculatusin-3- $O$ - $\beta$ -glucopyranosyl-(1'''2'')- $\beta$ -glucopyranoside, or corniculatusin-3- $O$ - $\beta$ -sophoroside.

Compound **2** was isolated as an amorphous light yellow powder, which appeared to be a kaempferol derivative with a substituted 3-hydroxyl group as could be concluded from its chromatographic behavior and from its UV spectra in methanol and with diagnostic reagents [14, 15]. Acid hydrolysis of **2** gave kaempferol (Co-PC), gallic acid (isolated by PPC from an ethyl acetate extract of the hydrolysate, Co-PC, UV spectra,  $^1\text{H}$  and  $^{13}\text{C}$  NMR) and glucose (Co-PC). The compound was recovered unchanged on enzymic hydrolysis with  $\beta$ -glucosidase and exhibited on negative ESI-MS analysis a molecular ion peak at  $[\text{M}-\text{H}]^-$   $m/z$  761 corresponding to a molecular weight ( $M_r$ ) of 762. These data suggested a kaempferol-3-monogalloylated-diglucoside in which the galloyl moiety is esterifying one of the hydroxyl groups of the terminal glucose moiety. The disaccharide was then obtained from **2**, by alkaline hydrogen peroxide oxidation and was identified to be sophoroside (Co-PC) [7]. Structure elucidation of **2** was achieved by  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopic analysis (DMSO- $d_6$ ). The received spectra unambiguously identified the compound as kaempferol derivative bearing a 3- $O$ -substituent. The number and characteristic chemical shifts of the  $^{13}\text{C}$  glycosidic signals indicated the presence of two glucopyranose moieties in which the hydroxyl groups of the C-6 and C-4 carbons were unsubstituted [17]. The relatively downfield resonance at  $\delta$  ppm 83.16 is unequivocally attributable to the glucosylated C-2'', while that at 74.8 is assignable to the galloylated C-2''' in the terminal glucose moiety. Both moieties had  $\beta$ -glycosidic linkages. This followed from the magnitude of the vicinal proton couplings ( $J = 8$  Hz) of the anomeric protons in the  $^1\text{H}$  spectrum and from the chemical shift values of the resonances of the anomeric carbons in the  $^{13}\text{C}$  spec-

trum [ $\delta$  ppm 101.94 (C-1'''), 98.75 (C-1'')]. The recognizable upfield shift ( $\Delta\delta = 3.38$  ppm) of the resonance of C-1''', in comparison with the corresponding resonance in the spectrum of kaempferol 3- $O$ - $\beta$ -sophoroside [17], is obviously due to the galloylation sited at its vicinal carbon C-2'''. Galloylation at that position also resulted in a downfield shift of the resonances, in the  $^1\text{H}$  spectrum, of the H-2''' to  $\delta$  ppm 5.12 ( $t$ ,  $J = 8$  Hz). The symmetrical galloyl moiety itself revealed its presence in the  $^{13}\text{C}$  NMR spectrum by its characteristic pattern of resonances (see Experimental). That the sophorose moiety is linked to the kaempferol hydroxyl group number 3 followed from the lowfield location ( $\delta$  ppm 5.62) of the anomeric glucoside proton resonance and unambiguously confirmed by the upfield of the aglycone C-3 resonance and the accompanying large downfield shift of the resonance of the *ortho*-carbon C-2 (all in comparison with the corresponding chemical shifts in the spectrum of kaempferol itself) [11]. Hence, The structure of **2** was determined as kaempferol-3- $O$ -[2'''-galloyl- $\beta$ -glucopyranosyl-(1''' 2'')- $O$ - $\beta$ -glucopyranoside].

### 3. Experimental

#### 3.1. Instruments and materials

NMR: Jeol YH 300 spectrometer, 300 MHz ( $^1\text{H}$  NMR) and 75 MHz ( $^{13}\text{C}$  NMR), respectively.  $^1\text{H}$  resonances were measured relative to TMS and  $^{13}\text{C}$  NMR resonances to DMSO- $d_6$  and converted to TMS scale by adding 39.5. ESI-MS: Micromass Quattro-LC triple quadrupole mass spectrometer equipped with a "Z-Spray" electrospray ion source. UV: UV/V Shimadzu spectrometer. PC (descending): Whatman no. 1 paper, using solvent systems: (1)  $\text{H}_2\text{O}$ ; (2) 15% HOAc; (3) BAW (*n*-BuOH–HOAc– $\text{H}_2\text{O}$ , 4:1:5, upper layer); (4)  $\text{C}_6\text{H}_6$ –*n*-BuOH– $\text{H}_2\text{O}$ -pyridine (1:5:3:3, upper layer). Solvents 3 and 4 were used for sugar analysis and solvent 3 for preparative paper chromatographic (PPC) isolation, on Whatman no. 3MM paper. Fresh herbs of *Sanguisorba minor* growing wild near El-Ariesh, north of Sinai proper, Egypt were collected in November, 2000 and authenticated by Dr. M. El Gibali, National Research Centre (NRC), Dokki, Cairo, Egypt. A voucher specimen has been deposited at the herbarium of the NRC.

#### 3.2. Extraction, isolation and purification

The collected plant specimen was exhaustively extracted with EtOH– $\text{H}_2\text{O}$  (3:1). The concentrated extract was separately applied to polyamide (6S Riedel De Haen AG-Seelze-Hannover) columns and eluted with  $\text{H}_2\text{O}$ , followed by  $\text{H}_2\text{O}$ –MeOH mixtures of decreasing polarities, to yield eleven fractions (I–XI). Compound **1** was isolated from fraction IV (eluted by 30% MeOH) by PPC, using BAW, Compound **2**, **3** and **5** from fraction V (eluted by 40% MeOH) by repeated Sephadex LH-20 column fractionation using *n*-BuOH saturated with  $\text{H}_2\text{O}$  for elution. Compounds (**4** & **6**) from fraction VI (eluted by 50% MeOH) by Sepadex LH-20 (Pharmacia) column using and  $\text{H}_2\text{O}$ –MeOH mixture (7:3) as an eluent, (**7**) from VI (eluted by 50%) by sephadex LH-20, using  $\text{H}_2\text{O}$ –MeOH mixture (1:1) for elution.

#### 3.3. New natural products

##### 3.3.1. Corniculatusin-3- $O$ - $\beta$ -glucopyranosyl-(1'''-2'')- $\beta$ -glucopyranoside (**1**)

$M_r$  656, -ve ESI-MS  $[\text{M}-\text{H}]^-$ : 655;  $R_f$  values: 0.51 ( $\text{H}_2\text{O}$ ), 0.75 (HOAc), 0.50 (BAW). UV ( $\lambda_{\text{max}}$  nm): MeOH 257, 272, 328, 358 (inflection); NaOAc 279, 320, 382; NaOAc +  $\text{H}_3\text{BO}_3$  265, 300 (inflection), 374;  $\text{AlCl}_3$  277, 305 (inflection), 360, 438; NaOMe 277, 330, 415. Complete acid hydrolysis of **1** gave glucose (Co-PC), and corniculatusin [ $R_f$ s: 02 ( $\text{H}_2\text{O}$ ), 04 (HOAc), 48 (BAW); UV ( $\lambda_{\text{max}}$  nm): MeOH;  $M_r$  332, -ve ESI-MS,  $[\text{M}-\text{H}]^-$ : 331;  $^1\text{H}$  NMR:  $\delta$  ppm 3.83 ( $s$ , 8-OMe), 6.24 ( $s$ , H-6), 6.89 ( $d$ ,  $J = 8$  Hz, H-5'), 7.52 ( $d,d$ ,  $J = 8$  and 2.5 Hz, H-6'), 7.64 ( $d$ ,  $J = 2.5$  Hz, H-2');  $^{13}\text{C}$  NMR:  $\delta$  ppm 60.49 (OMe-8), 146.66 (C-2), 135.77 (C-3), 176.9 (C-4), 156.1 (C-5), 98.7 (C-6), 156.22 (C-7), 127.38 (C-8), 148.55 (C-9), 103.76 (C-10), 124.35 (C-1'), 115.48 (C-2'), 145.07 (C-3'), 147.38 (C-4'), 115.5 119.92 (C-6'). Partial acid or  $\beta$ -glucosidase hydrolysis of **1** gave corniculatusin-3- $O$ - $\beta$ -glucopyranoside [ $R_f$ s: 0.32 ( $\text{H}_2\text{O}$ ), 0.45 (HOAc), 0.52 (BAW); UV ( $\lambda_{\text{max}}$  nm): MeOH 258, 269, 357;  $M_r$  494, -ve ESI-MS,  $[\text{M}-1]^-$  493; Normal acid hydrolysis yielded glucose and corniculatusin];  $^1\text{H}$  NMR of (**1**):  $\beta$ -glucosyl moiety:  $\delta$  ppm 4.61 ( $d$ ,  $J = 8$  Hz, H-1'''), 3.0–3.80 ( $m$ , other glucosyl protons),  $\beta$ -glucoside moiety: 5.68 ( $d$ ,  $J = 8$  Hz, H-1''), 3.0–3.80 ( $m$ , other glucoside protons), corniculatusin

moiety:  $\delta$  ppm 3.81 (OMe-8), 6.22 (*s*, H-6), 7.90 (*d*,  $J = 7.5$  Hz, H-5), 7.6 (*d*,  $J = 2.5$  Hz, H-2'), 7.64 (*d,d*,  $J = 7.5$  and  $2.5$  Hz, H-6').  $^{13}\text{C}$  NMR of **1**: corniculatusin moiety:  $\delta$  ppm 156.5 (C-2), 133.7 (C-3), 177.9 (C-4), 155.7 (C-5), 98.6 (C-6), 157.2 (C-7), 128.3 (C-8), 148.9 (C-9), 103.8 (C-10), 122.4 (C-1'), 115.9 (C-2'), 145.7 (C-3'), 148.9 (C-4'), 116.4 (C-5'), 121.7 (C-6');  $\beta$ -glucosyl moiety:  $\delta$  ppm 103.9 (C-1'''), 74.9 (C-2'''), 77.0 (C-3'''), 70.0 (C-4'''), 77.3 (C-5'''), 61.4 (C-6''');  $\beta$ -glucoside moiety:  $\delta$  ppm 98.6 (C-1''), 83.2 (C-2''), 78.0 (C-3''), 70.0 (C-4''), 77.4 (C-5''), 61.38 (C-6'').

### 3.3.2. Kaempferol-3-O-[2''-galloyl- $\beta$ -glucopyranosyl-(1'''-2''')-O- $\beta$ -glucopyranoside] (**2**)

$M_r$  762, -ve ESI-MS [M-H] $^-$ : 761;  $R_f$  values: 0.57 (H<sub>2</sub>O), 0.75 (HOAc), 0.59 (BAW). UV ( $\lambda_{\text{max}}$  nm): MeOH 267, 295 (inflection), 354; NaOAc 270, 302 (inflection); NaOAc + H<sub>3</sub>BO<sub>3</sub> 266, 297, 352; AlCl<sub>3</sub> 268, 303 (inflection), 355; NaOMe 274, 325m 403. Normal acid hydrolysis of **2** gave glucose (Co-PC), gallic acid [ $R_f$ : 0.53 (H<sub>2</sub>O), 0.59 (HOAc), 0.77 (BAW)]; UV ( $\lambda_{\text{max}}$  nm): MeOH 272;  $^1\text{H}$  NMR:  $\delta$  ppm 7.0 (*s*, H-3''' and H-5'''),  $^{13}\text{C}$  NMR: 120.6 (C-1'''), 108.8 (C-2''') and C-6'''), 145.5 (C-3''' and C-5'''), 138.1 (C-4''') and kaempferol [ $R_f$ : 0.02 (H<sub>2</sub>O), 0.04 (HOAc), 0.88 (BAW)]. Hydrogen peroxide oxidation: The released sugar obtained by H<sub>2</sub>O<sub>2</sub> treatment, for 12 h at room temperature, of an alkaline solution of **2** was co-chromatographed against sophorose [ $R_f$ : 0.57(BAW)].  $^1\text{H}$  NMR of **2**: kaempferol moiety:  $\delta$  ppm 6.22 (*d*,  $J = 2.5$  Hz, H-6), 6.44 (*d*,  $J = 2.5$  Hz, H-8), 6.98 (*d*,  $J = 7.5$  Hz, H-3 & H-5), 8.05 (*d*,  $J = 7.5$ , H-2 & H-6);  $\beta$ -glucosyl moiety:  $\delta$  ppm 4.80 (*d*,  $J = 8$  Hz, H-1'''), 5.12 (*t*,  $J = 8$  Hz, H-2'''), 3.2-4.0 (*m*, H-3''', H-4''', H-5''' and 2 H-6''');  $\beta$ -glucoside moiety:  $\delta$  ppm 5.62 (*d*,  $J = 8$  Hz, H-1''), 3.2-4 (*m*, remaining glucoside protons); galloyl moiety:  $\delta$  ppm 6.92 (*s*, H-2''' and H-6''').  $^{13}\text{C}$  NMR of **7**: Kaempferol moiety:  $\delta$  ppm 156.5 (C-2), 132.4 (C-3), 176.1 (C-4), 156.9 (C-5), 98.3 (C-6), 160.8 (C-7), 129.1 (C-8), 147.9 (C-9), 103.0 (C-10), 122.3 (C-1'), 130.5 (C-2' & C-6'), 115.8 (C-3' & C-5'), 160.1 (C-4'), 60.3 (OMe-8);  $\beta$ -glucosyl moiety:  $\delta$  ppm 101.9 (C-1'''), 74.8 (C-2'''), 74.6 (C-3'''), 69.9 (C-4'''), 78.0 (C-5'''), 61.2 (C-6''');  $\beta$ -glucoside moiety:  $\delta$  ppm 98.7 (C-1''), 83.16 (C-2''), 76.6 (C-3''), 69.7 (C-4''), 76.6 (C-5''), 61.3 (C-6''); galloyl moiety:  $\delta$  ppm 119.4 (C-1'''), 10807 (C-2''') and C-6'''), 145.48 (C-3''') and C-5'''), 138.6 (C-4'''), 165.8 (C-7''').

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## References

- 1 Nonaka, G.; Tanaka, T.; Nishioko, I.: J. Chem. Soc. Perkin I 1067 (1982)
- 2 Yoshida, T.; Feng, W.; Okuda, T.: Chem. Pharm. Bull. **40**, 1997 (1992)
- 3 El-Mousallamy, M. D.; Hussein, S. A. M.; Merfort, I.; Nawwar, M. A. M.: Phytochemistry **53**, 699 (2000)
- 4 El-Mousallamy, M. D.: J. Nat. Prod. Sci. **3**, 53 (1998)
- 5 Souleman, A. M. A.; El-Mousallamy, M. D.: J. Nat. Prod. Sci. **6**, 82 (2000)
- 6 Dauguet, J.; Bert, M.; Dolley, J.; Beakaert, A.; Lewin, G.: Phytochemistry **33**, 1503 (1993)
- 7 Ferreres, F.; Barberan, T.; Tomas-Lorente, F.; Nieto, J.; Rumbero, A.; Olias, J. M.: Phytochemistry **28**, (1991)
- 8 Tackholm, V.: Student flora of Egypt, p. 215 Cairo University Press, Egypt, 1974
- 9 Boulos, L.: Medicinal Plants of North Africa, Reference Publications, Inc., Michigan, U.S.A. 1983
- 10 Nawwar, M. A. M.; Ishak, M. S.; Michael, H. N.; Buddrus, J.: Phytochemistry **23**, 2110 (1984)
- 11 Nawwar, M. A. M.; Souleman, A. M.; Buddrus, J.; Linscheid, M.: Phytochemistry **23**, 2347 (1984)
- 12 Nawwar, M. A. M.; El-Mousallamy, M. D.; Barakat, Buddrus, J.; Linscheid, M.: Phytochemistry **28**, 3201 (1989)
- 13 Pachaly, P.; Klein, M.: Planta Med. **53**, 442 (1987)
- 14 Barakat, H. H.; Hussein, S. A. M.; Merfort, I.; Linscheid, M.; Nawwar, M. A. M.; Phytochemistry **46**, 935 (1997)
- 15 Harborne, J. B.; Saleh, N. A. M.; Smith, D. M.: Phytochemistry **17**, 589 (1978)
- 16 Agnese, A. M.; Chiale, C. A.; Cabrera, J. L.; Juliani, H. R.: J. Nat. Prod. **49**, 528 (1986)
- 17 Agrawal, P. K.: Carbon 13-NMR of the Flavonoids, Elsevier, New York 1989

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A. M. D. El-Mousallamy  
c/o Prof. M. Nawwar  
National Research Centre  
Dokki, Cairo  
Egypt