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Phenylpropanosids, lignans and other constituents from *Cremanthodium* ellisii

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Together with twenty-six known compounds, a new phenylpropanosid, named cremanthodioside, was isolated from the whole plant of *Cremanthodium ellisii* Kitam. Their structures were elucidated by spectroscopic methods MS, IR, UV, NMR, including 2D NMR techniques, and by chemical methods. The anti-bacterial activity of compounds 1–6 and the anti-tumor activity of compound 1 were tested.

1. Introduction

The genus Cremanthodium (Compositae) is widely distributed in the mountains of the Himalayas and contiguous climatic regions. So far, 64 species of it are known throughout the world, all of which are distributed in China, especially in the northwest and southwest regions (most of the genus Cremanthodium grow at an elevation of 3500–5000 m) (Delectis Flora Reipublicae Popularis Sinicae Agendae Academiae Sinicae Edita 1989). Some Cremanthodium plants like the title species have been used as traditional Tibetan medicines for antiinflammation, detoxification, and relief of pain since ancient times (Northwestern Plateau Institute of Biology, Academia Sinica 1991).

Due to its medicinal value, Cremanthodium ellisii Kitam. (Chen et al. 1996, 1997) (collected in Zhang county, Gansu province) has been systematically studied by our research group. We have now also investigated C. ellisii collected in Huzhu county, Qinghai province. We found that their chemical constituents differed greatly because they were collected in different districts (Su et al. 2000). In this paper, we report the isolation and structural elucidation of a new phenylpropanosid (1) and twenty-six known compounds $(2-27)$ from the species as a continuation of our studies. Among these, compounds 1–4, 6–11, 13–16 and 19–27 were isolated from this genus for the first time. In addition, six compounds 1–6 were assayed against Staphylococcus aureus, Escherichia coli and Bacillus subtilis. The new compound 1 was also screened against human ovarian carcinoma (HO-8910), human leukemia (HL-60) and human hepatoma (SMMC-7721) cell lines.

2. Investigations, results and discussion

From the ethanol extract of the whole plant of C. ellisii, a new phenylpropanosid, named cremanthodioside (1), was isolated and elucidated, together with twenty-six known compounds: citrusin B (2) (Deyama et al. 1987), $(+)$ -pinoresinol (3) (Deyama 1983; Fonseca et al. 1979), $(+)$ -syringaresinol (4) (Abe and Yamauchi 1988; Deyama 1983), $(-)$ -4'-O-methylnyasol (5) (Su et al. 2000), syrigoylglycerol 2- O - β -D-glucopyranoside (6) (Kijima et al. 1997), syringin (7) (Niwa et al. 1988), isorhamnetin (8) (Barbear et al. 1986), isorhamnetin-3-O- β -D-glactopyranoside (9) (Barbear et al. 1986), quercetin-3- O - β -D-glactopyranoside (10) (Barbear et al. 1986), umbelliferone (11) (Razdan et al. 1987), scopoletin (12) (Razdan et al. 1987), scopolin (13) (Tsukamoto et al. 1985), scoporone (14) (Razdan et al. 1987; Yu et al. 1998), 8-hydroxy-9,10-isobutyryloxy-thymol (15) (Bohlmann and Chen 1984), sitoindoside I (16) (Chaurasia and Wichtl 1987; Luo et al. 2001), β -sitosterol (17) (Greca et al. 1990), β -daucosterol (18) (Li et al. 1994), trans-p-hydroxycinnamic acid (19) (Sadtler Research Laboratories, Inc. 1978), trans-3,4-dihydroxycinnamic acid (20) (Flamini et al. 2001), trans-4-dihydroxy-3-methoxylcinnamic acid (21) (Sadtler Research Laboratories, Inc. 1973), 3,5-dimethoxy-4-hydroxybenzoic acid (22) (Sadtler Research Laboratories, Inc. 1969, Sadtler Research Laboratories, Division of Bio-Rad Laboratories, Inc. 1979), 5-hydroxymethyl furfural (23) (Xu et al. 1995), L-chiro-inositol (24) (Dorman et al. 1970), glycerol (25) , octadecanoate glycerol (26) , (R) -(Z, Z)-1-(9,12-octadecadienoate)-glycerol (27). The structure of the new compound 1 was identified by EIMS, FABMS, IR, UV, ¹H NMR, ¹³C NMR, ¹H, ¹H COSY, HMQC, HMBC, TOCSY1D spectroscopic methods and chemical transformation. The structures of the known compounds 2–27 were elucidated by comparison of their spectral data (ORD, EIMS, FABMS, IR, ¹H NMR and $13C$ NMR) with those published in the literature or were determined on the basis of their physical properties (m.p., R_f) etc.) by comparison with those of authentic samples. Compound $\hat{1}$ was obtained as white needles, m.p. 189–190 \degree C. Its molecular formula was deduced to be

Table 1: 1 H (400 MHz), 13 C NMR (100 MHz) and HMBC data of 1 (DMSO-d $_6$, TMS, δ , ppm)

No.	¹ H NMR	¹³ C NMR (DEPT) HMBC	
$\mathbf{1}$	3.22 (brs)	61.12 (CH ₂)	$C-3$
2	3.68 (brd, $J = 9.3$ Hz)	81.34 (CH)	$C-3$, $1''$
3	4.27 (d, $J = 9.3$ Hz)	79.25 (CH)	C-1, 2, $1', 2',$
			6', 2''
1'		128.45 (C)	
2', 6'	6.63 (d, $J = 1.8$ Hz)	106.07 (CH)	$C-3$, $1', 3', 4',$
			6''/2'
3', 5'		148.36 (C)	
4 [′]		136.17(C)	
$1^{\prime\prime}$	4.48 (d, $J = 7.5$ Hz)	98.57 (CH)	$C-2, 2'', 5''$
$2^{\prime\prime}$	2.94	80.03 (CH)	$C-3$, $1'', 3''$
	$(dd, J = 9.9, 8.1 Hz)$		
$3^{\prime\prime}$	3.35 *	73.86 (CH)	$C-2'', 4''$
$4^{\prime\prime}$	$3.15*$	71.16 (CH)	$C-3'', 5'', 6''$
$5^{\prime\prime}$	$3.29*$	79.25 (CH)	$C-1'', 4'', 6''$
$6^{\prime\prime}$	3.68 *, 3.46	61.59 (CH ₂)	$C-5''$
	$(dd, J = 12.3, 6.0 Hz)$		
OMe	3.75 (s)	56.74 (CH ₃)	$C-3', 5'$
Ar-OH	8.42 (brs)		$C-3', 4', 5'$

* Overlapped signals

at m/z 394.9 $[M + Li]^{+}$ and m/z 410.9 $[M + Na]^{+}$, which was supported by 1 H NMR, 13 C NMR and DEPT data (Table 1). The IR spectrum (KBr) showed the presence of alcoholic groups $(3464, 3398, 3296, 1058, 1023 \text{ cm}^{-1})$, phenolic hydroxyl $(3568, 1225 \text{ cm}^{-1})$, typical methoxyl absorption (2852 cm^{-1}) , benzene ring $(1615, 1521,$ 1467 cm^{-1}) and C-O-C bond (1126 cm⁻¹). Its ¹H NMR spectrum gave a independent aromatic proton absorption signal at $\bar{\delta}_{H}$ 6.63 (2 H, d, J = 1.8 Hz), so that combined with the ¹³C NMR and DEPT spectra data at δ_c 148.36 $(2 \times C)$, 136.17 (C), 128.45 (C), 106.07 (2 \times CH=), compound 1 has a symmetrical four-substituted aromatic ring. Moreover, the syringyl subunit of compound 1 was deduced from signals δ_H 8.42 (1 H, brs, Ar-OH), and δ_H 3.75 (6 H, s, $2 \times$ OMe) in the ¹H NMR spectrum and δ_c 56.74 $(2 \times OMe)$ in the ¹³C NMR spectrum. In addition, the significant absorption bands at 270 and 209 nm in the UV spectrum also supported a aromatic ring with auxochromic group substitutions. Apart from the proton signals corresponding to the above mentioned groups, the ¹H NMR data displayed an anomeric proton signal of sugar at δ _H 4.48 (1 H, d, J = 7.5 Hz), and oxygen-bearing methylene and methine group signals, which were determined to be a glycerol moiety and a monosaccharide moiety by ¹H, ¹H COSY spectrum data. Furthermore, ¹³C NMR and DEPT spectral data further verified this conclusion at δ _C 98.57 (anomeric carbon of sugar) and δ _C 61.12–81.34. To determine the correlations of δ_H and δ_C , the HMQC spectrum of compound 1 was obtained (Table 1). In addition, its HMBC spectrum gave long-range correlations of δ_H 6.63 (H-2',6') with δ_C 79.25 (C-3); and of δ_H 4.27 (H-3) with δ_C 106.07 (C-2', 6') showing that the skeleton of compound 1 was 3-C-syringy-glycerol, and the coupling constant $(J_{2,3})$ of 9.3 Hz allowed its stereochemistry be assigned to the threo form (Herrera Braga et al. 1984). From the long-range correlations of δ_H 4.48 (H-1") with δ _C 81.34 (C-2); of δ _H 3.68 (H-2) with δ _C 98.57 (C-1"); δ _H 2.94 (H-2") with δ _C 79.25 (C-3); and of δ_H 4.27 (H-3) with δ_C 80.03 (C-2"), the structure of compound 1 was obtained apart from the type of the sugar. In order to confirm this point, compound 1 was acetylated to give a penta-acetate (1a). Besides absorption peaks of the $\overline{4}$ -acetyl-3,5-dimethoxyl-phenyl moiety, the ¹H NMR spectrum of compound 1a revealed oxygenbearing proton signals which were connected to saturated carbons. According to the ${}^{1}H$, ${}^{1}H$ COSY spectrum of 1a, these signals were assigned as follow: δ_H 5.33 (1 H, dd, J = 9.9, 9.6 Hz, H-3"), 5.13 (1 H, dd, J = 9.9, 9.6 Hz, H-4"), 4.68 $(1 H, d, J = 8.1 Hz, H-1'')$, 4.47 $(1 H, d, J = 9.3 Hz, H-3)$, 3.93 (2 H, m, H-2, 5"), 3.52 (1 H, dd, $J = 9.9$, 8.1 Hz, H- 2^u). The relatively large trans di-axial coupling constants (J_{aa}) between H-1ⁿ and H-2ⁿ (8.1 Hz), H-2ⁿ and H-3ⁿ (9.9 Hz) , H-3ⁿ and H-4ⁿ (9.6 Hz), H-4ⁿ and H-5ⁿ (9.9 Hz) indicated that the sugar was β -D-glucopyranose. The signals of H-2 and $H-5$ ["] were overlapped, giving the TOCSY1D spectrum of 1a (about tri-acetate sugar), then the signals at δ_H 4.28 (1 H, dd, J = 12.3, 5.1 Hz), 4.19 (1 H, dd, $J = 12.3$, 1.8 Hz) were the proton of C-6^{$\prime\prime$} and δ_H 4.05 (1 H, m) was the proton of C-1. Thus, the structure of compound 1 was elucidated as shown, and it was named cremanthodioside.

 $C_{17}H_{24}O_{10}$ from the quasi-molecular ion peaks of FABMS

Compounds 1–6 were tested for their anti-bacterial activity against Staphylococcus aureus, Escherichia coli and Bacillus subtilis by comparision with chloramphenicol as a control. The results indicated that they all exhibited comparatively strong activities against S. aureus (Table 2).

^{**} Assignments are aided by ¹H, ¹H COSY, HMQC and HMBC data *** OH of compound 1: 5.14-5.16 (C₁-OH, C₆-OH), 4.64-4.68 (C₂⁻OH, C₃⁻OH, C_{4} ⁻OH) which were assigned by HMBC data.

Zone diameter of growth inhibition: < 10 mm (-), $10-12$ mm (+), $13-15$ mm (++), and $16-20$ mm $(+ + +)$

Using the MTT method, the anti-tumor activity of compound 1 against human ovarian carcinoma (HO-8910), human leukemia (HL-60) and human hepatoma (SMMC-7721) cell lines was screened by comparision with vincristine sulphate as a control. However, they exhibited little anti-tumor activity against the cell lines (the half inhibitory concentrations (IC₅₀, ug/ml) were 286.578 ± 1.586 , 333.667 ± 2.450 and > 400 , respectively).

3. Experimental

3.1. Apparatus

Optical rotations: Perkin-Elmer 341 Polarimeter; UV: TU-1901 UV-VIS instrument; IR: Nicolet NEXUS 670 FT-IR instrument; EIMS: HP 5988A GC/MS instrument; FABMS data: VG-ZAB-HS mass spectrometer (at 70 eV); NMR: Varian Mercury Plus-300BB instrument; Silica gel $(200-300 \text{ mesh})$ for column chromatography and GF254 $(10-40 \mu)$ for TLC were supplied by the Qingdao Marine Chemical factory, Qingdao, P.R. China; Spots were detected on TLC under UV lamp and by heating after spraying with 5% H_2SO_4 in C₂H₅OH or 5% FeCl₃ in C₂H₅OH; Melting points were determined on a Kofler melting point apparatus, and are uncorrected.

3.2. Plant material

Cremanthodium ellisii Kitam. was collected in Huzhu county, Qinghai province, P.R. China, in August 1999. It was identified by Prof. Guo-liang Zhang, School of Life Science, Lanzhou University. A voucher specimen (No. Ce-0802) was deposited in the herbarium of our institute.

3.3. Extraction and isolation

Air-dried and pulverized whole plant of C. ellisii Kitam. (7.5 kg) was extracted with methanol three times at room temperature (each for one week). The extract was concentrated under reduced pressure to yield a residue (340 g) , from which we obtained fifteen compounds (Su et al. 2000). This was then extracted with ethanol three times under the same conditions, and the resultant extract (170 g) was suspended in hot water (60 \degree C, 400 ml). The suspension was extracted, successively, with petroleum ether (60-90 °C) and EtOAc, then was concentrated under reduced pressure to yield residues of 20 g (part 1) and 35 g (part 2), respectively, and the remaining residue yielded 100 g (part 3). The combined part 1 and part 2 residues (55 g) were chromatographed over a silica gel column (700 g) eluted with a gradient of petroleum ether (60–90 °C)-acetone $(50:1$ to $1:1$ and CH_3COCH_3 , 500 ml each eluent). Combination of the appropriate fractions (monitored by TLC analysis) led to eight crude fractions (A-H). Fraction A (petroleum ether-acetone $50:1$ to $20:1$, 15 g) was obtained as white wax and yellow oil which were apparently mainly volatile oils and fatty hydrocarbons so have not been studied in detail; fraction B (petroleum ether-acetone 15 : 1, 5 g) was mainly volatile oil and colorless needles, which were recrystallized in acetone to obtain compound 17; fraction C (petroleum ether-acetone $10:1$, 1 g) was repeatedly chromatographed over a silica gel column to give compound 15 (25 mg). Fraction D (petroleum ether-acetone $6:1, 1.5$ g) was chromatographed over a silica gel column eluted with a gradient of petroleum ether $(60-90 °C)$ -acetone $(8:1, 1)$ 5 : 1, 3 : 1, 100 ml each eluent) to obtain fractions D_1-D_3 . Compound 14 (5 mg) and compound 26 (10 mg) were obtained by chromatography over a silica gel column from fraction D_1 . Fraction D_2 was repeatedly chromatographed over a silica gel column to give compound 5 (60 mg). Fraction D_3 was repeatedly recrystallized in EtOAc to obtain compound 11 (12 mg); fraction E (petroleum ether-acetone 3 : 1, 5.5 g) was repeatedly recrystallized in methanol to give compound 12 (200 mg), the mother liquid was chromatographed over a silica gel column eluted with a gradient of petroleum ether (60-90 °C)-acetone (8:1, 6:1, 6:1, 2:1 and 1:1, 100 ml each eluent) to obtain fractions E_1-E_2 . Fraction E_1 (400 mg) was rechromatographed over a silica gel column eluted with benzene-acetone (10 : 1) yielding compound 27 (30 mg). Fraction E_2 (400 mg) was repeated chromatographed over a silica gel column eluted with CHCl₃ to give compound $\overrightarrow{4}$ (15 mg). Fraction F (petroleum ether-acetone 1:1, 4.3 g) separated out as a white solids and was purified by prep. TLC on silica gel to obtain compound 16 (24 mg), the mother liquid was chromatographed over a silica gel column eluted with a gradient of petroleum ether $(60-90 °C)$ acetone (8:1 to 1:1, 100 ml each eluent) to obtain fractions F_1-F_3 after combination according to TLC analysis. Fraction F_1 (310 mg) was rechromatographed over a silica gel column eluted with benzene-acetone (8 : 1) to give compound 3 (17 mg); Compound 21 (30 mg) was obtained by prep. TLC on silica gel from fraction F_2 (50 mg). Fraction F_3 (300 mg) was rechromatographed over a polyamide column eluted with methanolwater (5:1), and was then purified by prep. TLC on silica gel to yield compound 20 (50 mg); fraction G (acetone, 4 g) was recrystallized in CH3OH to give compound 18, while the mother liquid was chromatographed over a silica gel column eluted with $CH_2Cl_2-CH_3OH$ (10 : 1), and was then recrystallized in CH₃OH to obtain compound 8 (50 mg). Fraction H (acetone, 10 g) was recrystallized in CH₃OH to yield a mixture of compound 9 and compound 10 (50 mg), and compound 9 (25 mg) and compound 10 (5 mg) were then isolated by prep. TLC on silica gel. The mother liquid was repeatedly chromatographed over a silica gel column to yield compounds 22 (2 mg) and 19 (20 mg).

The 100 g remainder (part 3) was chromatographed over a silica gel column (700 g) eluted with a gradient of $CHCl₃-CH₃OH$ (10:1 to 1:1 and CH3OH, 500 ml each fluent). Combination of the appropriate fractions (monitored by TLC analysis) led to seven fractions $(A'-G')$, and compound 24 was obtained as a white powder with the chromatograph. Fraction A' (CHCl₃–CH₃OH 10 : 1, 450 mg, 2000 ml) was chromatographed over a silica gel column eluted with $CHCl₃$ -acetone (8:1), and was then purified by prep. TLC on silica gel to obtain compound 23 (5 mg). Fraction B' (CHCl₃–CH₃OH 10 : 1, 1.5 g, 10000 ml) was repeatedly chromatographed over a silica gel column to give compound 13 (25 mg); fraction C' (CHCl₃–CH₃OH 5 : 1, 1 g, 2500 ml) was rechromatographed over a silica gel column to obtain compound 7 (14 mg); fraction \overline{D} (CHCl₃–CH₃OH 5 : 1, 1 g, 1000 ml) was chromatographed over a silica gel column eluted with EtOAc-CH₃OH (20:1, 10:1 and 5:1) to yield fraction D_1 ['] (EtOAc-CH₃OH 5:1, 500 mg), which was repeatedly recrystallizated in methanol to obtain compound 1 (30 mg); fraction E' (CHCl₃–CH₃OH 5 : 1, 1.6 g, 4000 ml) was chromatographed over a silica gel column eluted with a gradient of EtOAc-CH₃OH (20:1, 10:1 and 5:1), and was then purified by prep. TLC on silica gel to yield compound 2 (30 mg); fraction \hat{F}' (CHCl₃– CH₃OH 5:1, 1.5 g, 4000 ml) was repeatedly chromatographed over a silica gel column after being chromatographed over a silica gel column eluted with EtOAc-CH₃OH (20:1, 10:1 and $5:1$) to obtain compound 25 (100 mg). Fraction G' (CHCl₃–CH₃OH 5 : 1, 1.5 g, 2500 ml) was chromatographed over a silica gel column eluted with EtOAc-CH₃OH (20:1, $10:1$ and $5:1$) to obtain crude compound 6 (50 mg), which was then repeatly recrystallized in methanol to yield compound 6 (24 mg).

3.3.1. Cremanthodioside

White needles (CH₃OH), m.p. 189–190 °C; molecular formula: C₁₇H₂₄O₁₀; $[\alpha]_D^{20}$ + 75.0 (c, 0.2, CH₃OH); FABMS m/z (sgly): 394.9 [M+Li]⁺, 410.9 [M+Na]⁺ IR (v_{max}^{KBF} , cm⁻¹): 3464, 3398, 3296, 1058, 1023 (C–O–H), 3568, 1225 (Ar-OH), 2852 (OMe); 1615, 1521, 1467 (benzene ring) and 1126 (C-O–C); UV $\lambda_{\text{max}}^{\text{M6OH}}$ (nm): 270, 209; ¹H NMR, ¹³C NMR, DEPT and HMBC data: see Table 1.

3.3.2. Acetylation of cremanthodiumol

Compound 1 (2 mg) was acetylated with a mixture of $(Ac)₂O$ and pyridine $(500 \mu l$ each) at room temperature for 72 h and was concentrated under reduced pressure to yield the penta-acetate 1a (3 mg). Colorless gum; molecular formula: $\overline{C}_{27}H_{34}O_{15}$; $[\alpha]_{D}^{20} + 54.0$ (c, 0.1, CHCl₃); EIMS m/z
(rel int): 556 [M-Ac]⁺ (1.3), 252 (0.7), 213 (1.1), 196 (0.3), 182 (1.4), 167 (1.0) , 153 (2.8) , 97 (9.1) , 43 (100) ; IR (v_{max}^{KBr}, cm^{-1}) : 2850 (OMe) ; 1747 (C=O), 1606, 1464 (benzene ring) and 1230, 1130, 1042 (C-O–C); UV
 $\lambda_{\text{max}}^{C=O}$ (nm): 272, 240; ¹H NMR δ ppm (CDCl₃, 300 Hz): 6.54 (2H, d, $J = 1.8$ Hz, H-2', 6'), 5.33 (1 H, dd, $J = 9.9$, 9.6 Hz, H-3"), 5.13 (1 H, dd, $J = 9.9, 9.6$ Hz, H-4"), 4.68 (1 H, d, $J = 8.1$ Hz, H-1"), 4.47 (1 H, d, $J = 9.3$ Hz, H-3), 4.28 (1 H, dd, $J = 12.3$, 5.1 Hz, H-6ⁿ), 4.19 (1 H, dd, $J = 12.3, 1.8$ Hz, H-6"), 4.05 (2 H, m, H-1), 3.93 (2 H, m, H-2, 5"), 3.82 $(3 \text{ H}, \text{ s}, \text{ OMe})$, 3.52 $(1 \text{ H}, \text{ dd}, \text{ J} = 9.9, \text{ 8.1 Hz}, \text{ H-2}^{\prime\prime})$, $2.33, 2.02, 2.03$, 2.04, 2.08 (3 H each, s, OAc).

3.3.3. Citrusin B

Colorless gum; molecular formula: $C_{27}H_{36}O_{13}$; $[\alpha]_D^{20} - 43.6$ (c, 1.7, CH₃OH); FABMS m/z (sgly): 575 [M+Li]⁺, 591 [M + Na]^{+ 1}H NMR δ ppm (CD₃OD, 300 MH_Z): 7.12 (1 H, d, J = 8.1 Hz, H-5), 7.07 (1 H, d, J = 2.0 Hz, H-2), 6.91 (1 H, dd, J = 8.1, 2.0 Hz, H-6), 6.72 (2 H, d, J = 2.0 Hz, H-2', 6'), 6.54 (1 H, d, J = 15.9 Hz, H-7'), 6.31 (1 H, dt, $J = 15.9, 5.4$ Hz, H-8'), 4.94 (1 H, d, $J = 5.4$ Hz, H-7), 4.91 (2 H, m, H-8,

 $1'$), 4.57 (brs, OH), 4.22 (2H, d, J = 5.4 Hz, H-9'), 3.84 (3H, s, C_3 -OMe), 3.81 (6 H, s, $C_{3'}$, $C_{5'}$ -OMe), 3.68 (2 H, dd, J = 12.3, 3.6 Hz, H-9); ¹³C NMR δ ppm (CD₃OD, 75 MH_Z): 153.37 (C-3', 5'), 149.28 (C-3), 146.01 (C-4), 136.32 (C-4'), 135.12 (C-1'), 133.59 (C-1), 130.20 (C-7'), 128.72 (C-8'), 119.62 (C-6), 116.45 (C-5), 111.20 (CH-2), 103.69 (C-2', 6'), 101.78 (C-1"), 86.09 (C-8), 76.98 (C-3"), 76.61 (C-5"), 73.73 (C-2"), 72.60 (C-7), 70.16 (C-4"), 62.40 (C-9'), 61.32 (C-6"), 60.26 (C-9), 55.49 (OCH3).

3.3.4. $(+)$ -Pinoresinol

Colorless gum; molecular formula: $C_{20}H_{22}O_6$; $[\alpha]_D^{20}$ + 30.6 (c, 1.1, CHCl₃);
EIMS m/z (rel int): 358 [M];⁺ (28), 327 [M-OMe]⁺ (4), 205 [M-ArCHO– H]⁺ (15), 163 [ArCH=C=O–H]⁺ (27), 151 [ArCHO–H]⁺ (100), 137 (52) , 131 (35), 124 [Ar-H]⁺ (18), 123 [Ar]⁺ (9); ¹H NMR δ ppm (CDCl₃, 300 MHz): 6.90 (2 H, d, J = 1.8 Hz, H-2, 2'), 6.89 (2 H, d, J = 8.4 Hz, H-5, 5'), 6.82 (2 H, dd, J = 8.4, 1.8 Hz, H-6, 6'), 5.66 (2 H, brs, Ar-OH), 4.74 (2 H, d, J = 3.6 Hz, H-7, 7'), 4.24 (2 H, dd, J = 7.2, 6.9 Hz, H-9e, $9e'$), 3.90 (6 H, s, OMe), 3.86 (2 H, dd, J = 7.2, 3.3 Hz, H-9a, 9a'), 3.10 (2 H, m, H-8, 8'); ¹H NMR δ ppm (CD₃COCD₃, 300 MH_Z): 6.97 (2 H, d, $J = 1.8$ Hz, H-2, 2'), 6.80 (4 H, m, H-5, 5', 6, 6'), 4.66 (2 H, d, $J = 3.6$ Hz, H-7, 7'), 4.19 (2H, H-9e, 9e'), 3.83 (6H, s, OMe), 3.77 (2H, dd, $J = 8.7$, 6.6 Hz, H-9a, 9a'), 3.08 (2 H, m, H-8, 8'); ¹³C NMR δ ppm (CDCl₃, 75 MHz): 146.93 (C-3, 3'), 145.45 (C-4, 4'), 133.10 (C-1, 1'), 119.36
(C-6, 6'), 114.68 (C-5, 5'), 108.97 (C-2, 2'), 86.29 (C-7, 7'), 71.89 (C-9, 9'), 56.34 (C-OCH₃), 54.19 (C-8, 8'); ¹³C NMR δ ppm (CD₃COCD₃, 75 (C-6, 6'), 114.80 (C-5, 5'), 109.96 (C-2, 2'), 85.97 (C-7, 7'), 71.54 (C-9, 9'), 55.59 (OCH₃), 54.57 (C-8, 8').

$3.3.5.$ (+)-Syringaresinol

Colorless gum; molecular formula: $C_{22}H_{26}O_8$; $[\alpha]_D^{20} + 14.6$ (c, 0.8, CHCl₃); ¹H NMR δ ppm (CDCl₃, 300 MHz): 6.58 (4 H, d, J = 2.0 Hz, H-2, 2', 6, 6'), 5.53 (2 H, brs, Ar-OH), 4.74 (2 H, d, J = 3.6 Hz, H-7, 7'), 4.28 (2 H, dd, J = 8.7, 6.6 Hz, H-9e, 9e'), 3.90 (14 H, brs, 4OMe, H-9a, 9a'), 3.10 (2 H, m, H-8, 8'); ¹³C NMR δ ppm (CDCl₃, 75 MHz): 147.40 (C-3, 3', 5, 5'), 134.55 (C-4, 4'), 132.32 (C-1, 1'), 102.96 (C-2, 2', 6, 6'), 86.30 (C-7, 7'), 72.02 (C-9, 9'), 56.60 (OCH₃), 54.57 (C-8, 8').

3.4. Anti-bacterial activity assays

The anti-bacterial activity assay was carried out according to the cup-plate method. Chloramphenicol was used as a positive control. Three strains of bacteria: Staphylococcus aureus, Escherichia coli and Bacillus subtilis, were cultured in beef broth and incubated at 37° C for 24 h. After dilution of the beef broth, the three bacteria were cultured individually in agar medium dishes, six cups $(8 \times 10 \text{ mm})$ were put onto the dishes, and each compound being tested $(0.2 \text{ ml of } 100 \text{ µg/ml})$ was added to the cups separately under aseptic conditions. The dishes were then cultured at 37 °C for 24 h. The zone of inhibition of the growth of bacteria, produced by diffusion of the compounds from the cup into the surrounding medium, was measured to evaluate the anti-bacterial activity. Each test was performed in duplicate.

3.5. Anti-tumor activity assays

Anti-tumor activity assays were carried out according to the sulforhodamine B (MTT) colorimetric assay. Human ovarian carcinoma (HO-8910), human leukemia (HL-60) and human hepatoma (SMMC-7721) cell lines were cultured in RPMI-1640 medium supplemented with 10% fetal calf serum, at 37 °C under a humidified atmosphere of 5% CO₂, and distributed in replicate 96-well plates with 4×103 cells/well for 24 h. Then, using vincristine sulfate as a positive control, compound 1 was added. After 48 h exposure to the toxins, cell viability was determined by measuring the absorbance at 515 nm with an ELISA reader. Each test was performed in 5 replicates.

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