

Where W_{in} is the amount of lauroyl-indapamide in liposomes and W_{total} is the total amount of lauroyl-indapamide in the liposomal preparation.

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An unusual new sulfated triterpene saponin from *Arenaria juncea*

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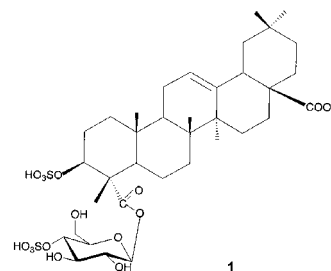
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A novel triterpene saponin, 3-O-sulfate-gypsogenic acid-23-O-β-D-4-O-sulfate-glucopyranoside (junceoside D), has been isolated from the roots of *Arenaria juncea*. The structure was characterized mainly by a combination of 2D-NMR techniques, mass spectrometry and chemical methods.

Arenaria juncea Bieb. roots endemic to China (Family: Caryophyllaceae) is well-known in the Traditional Chinese Medicine in the Shanxi, Gansu Hubei province as a substitute of the Chinese drug Yin-Chai-Hu (root of *Stellaria dichotoma* var. *lanceolata* Bge) (Cui et al. 1992) and is used to treat fever due to Yin-deficiency and fever in infant malnutrition. Our previous phytochemical studies on the methanolic extract of *Arenaria juncea* roots led to the isolation of five triterpene-saponins (Gaidi et al. 2001). A detailed further investigation of the same extract has led to the isolation of an additional triterpene-glycoside **1**. The concentrated *n*-BuOH-soluble fraction of the MeOH extract of the roots of *Arenaria juncea* Bieb. (Caryophyllaceae) was purified by precipitation with diethyl ether. The crude saponin mixture was further dialysed and subjected to multiple chromatographic steps over Sephadex LH-20 and medium pressure liquid chromatography (MPLC) over reversed-phase Si RP-18 yielding a saponin, junceoside D (**1**). Its structure was elucidated mainly by FABMS, MALDITOFMS, HRESIMS and by 600 MHz NMR spectroscopic analysis including 1D and 2D NMR experiments (¹H–¹H COSY, TOCSY, NOESY, HSQC, and HMBC).



Compound **1** was obtained as a yellow powder. The FABMS (negative-ion mode) exhibiting a quasi-molecu-

Table: ^{13}C NMR Data of the aglycon and sugar unit of saponin **1** (δ ppm, DMSO-d_6 as solvent)^a

Position	Gypsogenic acid		
	Mult. ^b	δ ^{13}C	δ ^1H
1	CH ₂	38.3	0.95, 1.52
2	CH ₂	23.0	1.36, 1.72
3	CH	80.0	4.54, dd, J = 4.1 and J = 16.1
4	C	52.8	—
5	CH	51.6	1.36
6	CH ₂	19.0	nd, nd
7	CH ₂	33.7	1.30, 1.50
8	C	39.3	—
9	CH	48.0	1.46
10	C	36.3	—
11	CH ₂	23.6	nd, nd
12	CH	121.5	5.05 (br s)
13	C	145.5	—
14	C	41.8	—
15	CH ₂	28.0	1.18, nd
16	CH ₂	23.7	nd, nd
17	C	46.7	—
18	CH	41.8	2.72
19	CH ₂	46.4	0.98, 1.50
20	C	31.0	—
21	CH ₂	34.6	0.90, nd
22	CH ₂	28.0	0.80
23	C	174.6	—
24	CH ₃	12.2	0.98 (s)
25	CH ₃	15.7	0.83 (s)
26	CH ₃	17.4	0.64 (s)
27	CH ₃	26.0	1.01 (s)
28	C	180.0	—
29	CH ₃	33.6	0.82 (s)
30	CH ₃	24.1	0.80 (s)
	Glucose		
1		94.5	5.35 (d, J = 8.2)
2		72.8	3.23 (t, J = 8.5)
3		74.3	3.49 (t, J = 9.1)
4		75.7	3.87 (br s)
5		75.8	3.02 (t, J = 7.0)
6		61.5	3.43, 3.59 (d, J = 11.4)

^a The assignments were based on the COSY, TOCSY, NOESY, HSQC, and HMBC experiments (150 MHz for ^{13}C and 600 MHz for ^1H NMR), ^b Multiplicities were assigned from DEPT spectra. nd: not determined

lar ion peak at m/z 807 $[\text{M}-\text{H}]^-$ and the high-resolution ESI mass spectrum, HRESIMS (positive ion mode) showing a pseudomolecular ion peak at m/z 831.2568 $[\text{M} + \text{Na}]^+$ (Calcd for 831.2544, $\text{C}_{35}\text{H}_{52}\text{O}_{17}\text{NaS}_2$) indicated a molecular weight of 808, compatible with the molecular formula $\text{C}_{35}\text{H}_{52}\text{O}_{17}\text{S}_2$. Other significant ion peaks visible in the FABMS at m/z 727 $[(\text{M}-\text{H})-80]^-$, 565 $[(\text{M}-\text{H})-80-162]^-$, 485 $[(\text{M}-\text{H})-80-162-80]^-$ corresponded to the successive losses of two sulfate groups $[\text{SO}_3]^-$ and one hexosyl moiety. The fragment ions at m/z 485 corresponded to the pseudomolecular ion of the aglycone. The presence of a sulfate group was concluded by the presence of a peak at m/z 97 $[\text{OSO}_3\text{H}]^-$ in the negative FABMS. The MALDITOFMS of **1** showing a $[\text{M}-\text{H}]^-$ ion peak at m/z 807 and $[(\text{M} + \text{K}-\text{H})-\text{H}]^-$ ion peak at m/z 845 followed by a fragment ion peak at m/z 727 $[\text{M}-\text{H}-80]^-$ confirmed the proposed molecular weight. Mineral acid hydrolysis of **1** with 2 N TFA afforded an artifactual aglycone and glucose (co-TLC). The aglycone was identified as gypsogenic acid from 2D NMR spectra of **1** (Table) (Koike

et al. 1999). The downfield shift observed in the HSQC spectrum for the Agly H-3/Agly C-3 resonance at δ_{H} 4.54 (dd, J = 4.1 and J = 16.1)/ δ_{C} 80.5 suggested that the aglycone residue was esterified by the SO_3H group at the position C-3, that was in good agreement with literature data for similarly sulfated saponins (Acebes et al. 1998). Compound **1** was shown to contain one sugar residue from the HSQC spectrum which showed the anomeric ^1H NMR signal at δ 5.35 (d, J = 8.2 Hz) giving a correlation with the ^{13}C NMR signal at δ 94.5. The ring protons of the monosaccharide residue were assigned starting from the readily identifiable anomeric proton by means of the COSY, TOCSY, HSQC and HMBC spectra. Evaluation of spin-spin couplings and chemical shifts allowed the identification of one β -glucopyranosyl moiety (Glc). The common D-configuration was assumed, according to those most often encountered among the plant glycosides. In the HMBC spectrum, the carbonyl C-28 observed at δ 180.0 was typical of a unsubstituted group but the upfield shift of C-23 at δ_{C} 174.6 indicated that C-23 was substituted by the glucosidic ester chain. Unequivocal demonstration of the linkage at this position was obtained from the HMBC experiment, giving a cross peak between $\delta_{\text{H}-24}$ 0.98 (s) and $\delta_{\text{C}-23}$ 174.6 and between the anomeric ^1H NMR signal at $\delta_{\text{H}-1}$ 5.35 (d, J = 8.2 Hz, Glc H-1) and the ^{13}C NMR signal at δ_{C} 174.6 (Agly C-23). The downfield shift observed in the HSQC spectrum for the Glc H-4/Glc C-4 resonance at δ_{H} 3.87 (br s)/ δ_{C} 75.7 suggested that the Glc residue was substituted at the position C-4 but the substituent does not contain carbon atoms. It was suggested that the substituent is a second SO_3H group. This was confirmed by FABMS which presented a fragment ion at m/z 565 $[(\text{M}-\text{H})-80-162]^-$.

Based on the above results, the structure of compound **1** was determined as 3-O-sulfate-gypsogenic acid-23-O- β -D-4-O-sulfate-glucopyranoside, a new natural compound. Sulfated triterpene saponins have been already found in Caryophyllaceae (Acebes et al. 1998) but according to an updated literature search and to previous reviews on sulfated-saponins (Amimoto et al. 1993; Higuchi et al. 1984; Kostova et al. 2002; Sánchez-Contreras et al. 1998, 2000; Sasmakov et al. 2003; Shaker et al. 2000; Sung et al. 1991), **1** is a first triterpene-saponin containing two sulfate groups.

Experimental

1. General procedures

All physical data of the isolated compound were obtained on the same instruments as those used in a previous paper (Gaidi et al. 2001). All NMR chemical shifts (δ) are given in ppm and the samples were solubilized in DMSO-d_6 (δ 39.5). Furthermore, HRESIMS spectra were carried out on a Q-TOF1 micromass spectrometer. MALDITOFMS was conducted using a PerSeptive Biosystems Voyager DE-STR mass spectrometer. TLC and HPTLC employed precoated Si gel plates 60 F₂₅₄ (Merck). The following TLC solvent systems were used: for saponins (a) CHCl_3 -MeOH-AcOH-H₂O (15:8:3:2); for sapogenins (b) toluene-Me₂CO (4:1); for monosaccharides (c) CHCl_3 -MeOH-H₂O (8:5:1). Spray reagents for the saponin was: Komarowsky reagent, a mixture (5:1) of *p*-hydroxybenzaldehyde (2% in MeOH) and H₂SO₄ 50%; for the sugars: diphenylaminephosphoric acid reagent. Isolations were carried out using a Gilson medium-pressure liquid chromatography (MPLC) system (Gaidi et al. 2001).

2. Plant material

The roots of *Arenaria juncea* were collected in July 1990 in Hubei Province, People Republic of China. A voucher specimen (No 48-11) is deposited in the Herbarium of the Laboratory of Pharmacognosy, Faculty of Pharmacy, University of Burgundy (Gaidi et al. 2001).

3. Extraction

Dried powdered roots of *Arenaria juncea* (639 g) were defatted with 5 l *n*-hexane and extracted successively with 5 l CHCl₃ and 5 l MeOH. After removal of the solvent by evaporation under reduced pressure, the MeOH extract (30 g) was suspended in H₂O (400 ml) and submitted to partition between H₂O saturated *n*-BuOH (3 × 200 ml). After evaporation of the solvent under reduced pressure, 23 g of the *n*-BuOH fraction was obtained. The *n*-BuOH extract was solubilized in MeOH (10 ml) and precipitated in Et₂O (3 × 250 ml) yielding 7 g of a crude saponin fraction of which 4 g was dialysed for 3 days and submitted to column chromatography on Sephadex LH-20 eluted by MeOH yielding four fractions (F1–F4).

4. Isolation and characterization of 1

Five saponins were obtained from F2 (810 mg) (Gaidi et al. 2001). F3 (226 mg) was first fractionated by column chromatography on Sephadex LH-20 eluted by MeOH, then by MPLC on reversed phase material, Li-chroprep RP-18, Merck (40–63 µm) eluted with MeOH–H₂O (linear gradient 50–60%) to give compound 1. Final purification was carried out on Sephadex LH-20 eluted by MeOH to obtain compound 1 (10 mg).

Junceoside D (1): Yellow powder. $[\alpha]_{20}^{25} + 16^\circ$ (c 0.10, MeOH); IR ν_{\max} KBr cm⁻¹ 3398, 2928, 1735, 1718, 1615, 1386, 1220; negative FABMS glycerol matrix: m/z 807 [M–H]⁻, 727 [(M–H)–80]⁻, 565 [(M–H)–80–162]⁻, 485 [(M–H)–80–162–80]⁻; positive HRESIMS: m/z 831.2568 [M + Na]⁺ (Calcd for C₃₅H₅₂O₁₇NaS₂). MALDITOFMS: m/z 807 [M–H]⁻, 845 [(M + K–H)–H]⁻, 727 [M–H–80]⁻. ¹H and ¹³C NMR: Table.

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Clausine Z, a new carbazole alkaloid from *Clausena excavata* with inhibitory activity on CDK5

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A new carbazole alkaloid, named clausine Z, has been isolated from the stems and leaves of *Clausena excavata* Burm. (Rutaceae). Its structure was established by spectroscopic methods. The compound exhibits inhibitory activity against cyclin-dependent kinase 5 (CDK5) and shows protective effects on cerebellar granule neurons *in vitro*.

CDK5 is a member of the family of cyclin dependent kinases (CDKs). Contrarily to the other members of this family, CDK5 is not involved in the cell cycle control but plays a crucial role in the development of the central nervous system and the regulation of neuronal signal transduction. Experimental studies have shown that over-activation of CDK5 might contribute to the pathology of neurodegenerative diseases such as Alzheimer's disease, amyotrophic lateral sclerosis and cerebral ischemia (Dhavan and Tsai 2001; Shelton and Johnson 2004).

In the course of our search for new CDK5 inhibitors, we noticed that an EtOAc extract of the stems and leaves of *Clausena excavata* Burm. (Rutaceae) showed *in vitro* CDK5 inhibitory activity. *C. excavata* is a shrub growing in Southeast Asia known to contain, among others, coumarins and carbazole alkaloids (Ito et al. 1996, 1997, 2000; Wu et al. 1999). Semipreparative fractionation of the bioactive extract by HPLC-UV and subsequent testing of the fractions enabled the activity to be attributed to a main peak (Rt 8.9 min) with a UV spectrum corresponding to a 9H-carbazole ring system. The compound (**1**) was isolated from an ethanolic extract of the stems and leaves by a combination of liquid/liquid partition and CC on silica-gel and Sephadex LH20. The ESI TOF MS (positive ion mode, [M + H]⁺ m/z 228.0630) allowed the molecular formula to be established as C₁₃H₉NO₃. 1D and 2D NMR data revealed the presence of a formyl group ($\delta^1\text{H}$ 9.91, $\delta^{13}\text{C}$ 191.6) and two hydroxy groups ($\delta^1\text{H}$ 10.24, $\delta^1\text{H}$ 9.07). The ¹H and ¹³C resonances of the formyl group, derived from a HC-HSQC experiment, were used as starting point for the assignment of the regiochemistry of compound 1. The close neighbourhood of the formyl proton with two aromatic protons (H-2 d, J = 1.3 Hz and H-4 d, J = 1.3 Hz) was confirmed in the HH-ROESY spectrum by the two corresponding intense cross peaks, as well as in the HC-HMBC spectrum, which showed HC