invariance $x \to ax$, $E \to E$, $\sigma \to \sigma$, $\sigma' \to a^{-2}\sigma'$. This scale invariance is destroyed when further details of the transports involved are brought in.

Clearly the external field strength E_0 must lie in a rather well-defined range for such solutions. If E_0 is too small, it cannot be raised above the breakdown field by the presence of a more conducting sphere (which even if $\sigma_1/\sigma_0 \rightarrow \infty$ can at most triple the external field, it is well known). If E_0 is too large there will be a breakdown

between the distant sources of the field. If the field should fall during the existence of such a ball, it would disappear quietly, because of the small energies involved; if the field should rise, however, the ball may give way to a high-power discharge.

Thus the nonlinear conduction model is helpful in understanding some of the reported circumstances of ball lightning, and in suggesting further theoretical and experimental work.

PHYSICAL REVIEW

VOLUME 135, NUMBER 2A

20 JULY 1964

Effect of Applied Electric Fields on the Electron Spin Resonance of Fe^{3+} and Mn^{2+} in α -Al₂O₃

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The effect of applied electric fields in broadening the electron spin resonance lines of the isoelectronic ions Fe^{3+} and Mn^{2+} in α -Al₂O₃ has been observed. This is interpreted as an unresolved line "splitting" which is linear in the applied field. The change in the spin Hamiltonian D parameter with applied field parallel to the $c xxis (\partial D/\partial E)$ is 1.00 ± 0.09 for Fe^{3+} and 1.03 ± 0.09 for Mn^{2+} in units of 10^{-5} G cm/V. The existence of this effect implies that these ions replace Al³⁺ ions substitutionally in the α -Al₂O₃ lattice. A simplified calculation indicates that ionic displacement plays an important role in producing the effect in these S-state ions and thatthe resultant relative change in the even component of the axial crystalline field appears to be much larger for Mn^{2+} than Fe^{3+} .

A S has been pointed out by Bloembergen,¹ if one places a paramagnetic ion in a crystal site which lacks inversion symmetry, he can, in general, expect to produce a change in the electron spin resonance (ESR) spectrum upon application of an external electric field. In this paper, the broadening of ESR lines due to the isoelectronic pair of ions Fe^{3+} and Mn^{2+} ($3d^5$) in Al_2O_3



FIG. 1. A portion of the α -Al₂O₃ lattice showing the four types of Al³⁺ sites (after Geschwind and Remeika, Ref. 3).

is reported, and the effect is interpreted in terms of an unresolved splitting of the lines.

In the α -Al₂O₃ (corundum) lattice, the Al³⁺ ions are located in sites which lack inversion symmetry as is seen in Fig. 1. This representation of the corundum lattice is taken from an article by Geschwind and Remeika² as adapted by Artman and Murphy.³ The Al³⁺ ions are surrounded by a greatly distorted octahedral distribution of O²⁻ ions such that the point symmetry is reduced to C₃ and thus lacks inversion symmetry. As can be seen, however, sites *a* and *f* in Fig. 1 are related to one another by inversion symmetry, as are sites *b* and *c*, and all Al³⁺ sites in the lattice are so related in these two types of pairs.

In ruby, the linear electric field pseudosplitting of certain ESR lines arising from Cr^{3+} in such sites has been exhibited and measured.^{4,5} In actuality, the lines from one type of site merely shift their position while those from the inversion-related site shift by the same amount in the opposite direction. Since it is possible to substitute other iron group transition metal ions into this lattice, we considered it would be helpful in elucidating the atomic mechanism to examine the effect

¹ N. Bloembergen, Science 133, 1363 (1961).

² S. Geschwind and J. P. Remeika, Phys. Rev. **122**, 757 (1961). ³ J. O. Artman and J. C. Murphy, J. Chem. Phys. **38**, 1544 (1963).

 ⁴ E. B. Royce and N. Bloembergen, Phys. Rev. 131, 1912 (1963).
 ⁵ J. O. Artman and J. C. Murphy, Bull. Am. Phys. Soc. 7, 14 (1962).



FIG. 2. Broadening of the low-field Fe^{3+} resonance line with the electric field applied parallel to the *c* axis. Note the change in apparent line shape with applied field.

of applied electric fields on the ESR spectra of the isoelectronic pair of ions Fe^{3+} and Mn^{2+} . These ions have a ${}^{6}S$ ground state. This requires a higher order crystal field perturbation to produce a ground-state splitting⁶ than is the case for Cr³⁺. As a result, the effect of applied electric fields is expected to be considerably smaller for them than for Cr³⁺. This expectation was confirmed experimentally.

The ESR spectra were examined with a Varian X-band spectrometer at room temperature. The Al_2O_3 crystals were grown from a PbO-PbF₂ flux doped with the appropriate impurity. They were aligned either by x ray or by making use of the natural crystal faces. Because of the growth habit (platelets with the faces perpendicular to the c axis), the samples obtained were only suitable for applying electric fields along the c axis. The samples were cut into the form of disks, and the disk faces were coated with evaporated silver electrodes. The crystals were edged with polystyrene dope to lengthen the dielectric breakdown path through the air. Beryllium-copper leaf springs were used to make electrical contact with the samples which were supported by a polystyrene rod at the center of the cylindrical cavity.

Because the observed ESR line splittings were not large enough to produce resolved lines, it was necessary to use other techniques to determine the splitting from the observed line broadening. One of these methods makes use of the increase in the second moment of the line when the line splits. If half of the original line shifts to higher magnetic fields by an amount a and half to lower fields by the same amount, the increase in the second moment S is given by

$$S - S_0 = a^2. \tag{1}$$

The line splitting 2a is thus directly determined by the

second moment increase. Equation (1) is valid for any line shape as long as the second moment can be defined. This excludes Lorentzian lines, of course. The principal problem encountered in using this technique was the fact that the second moment only increased by about 12% for the highest applied fields, and it is difficult to measure second moments to an accuracy of more than 2 or 3%. To achieve even this accuracy, it was necessary to make use of a program which made corrections for base line and line center errors.⁷

Alternatively, one may assume a line shape and calculate the increase in the derivative linewidth when the line splits by an amount 2a. The relative increases in width W for various values of $2a/W_0$ were calculated for both Lorentzian and Gaussian line shapes and were used to extract the value of the splitting from the observed linewidth increase for a given applied electric field. In practice, it was possible to measure the line widths with an accuracy of 1-2% while their relative changes were about the same as for the second moment. Since it is the change in these quantities upon application of the electric field which is of interest, this accuracy represented a worthwhile improvement over the second moment method. Furthermore, accurate linewidth measurements were much easier to make.

RESULTS

For Fe³⁺ in Al₂O₃, Fig. 2 shows the line broadening produced by applying an external electric field of 142 kV/cm. The linewidth increases from 10.5 to 13.0 G, and the line shape changes in the expected manner. These data were taken with the magnetic and electric fields parallel to the *c* axis, using the low magnetic field $-3/2 \leftrightarrow -1/2$ transition. The splitting versus applied field for this line is shown in Fig. 3, where the splittings were determined by the second moment method. As can



FIG. 3. Splitting of the $-3/2 \leftrightarrow -1/2$ Fe³⁺ resonance line as determined by the increase in second moment when the electric field is applied. The electric field is parallel to the *c* axis.

⁷ D. J. Kroon, Philips Res. Rept. 15, 501 (1960).

⁶ H. Watanabe, Progr. Theoret. Phys. (Kyoto) 18, 405 (1957).



FIG. 4. Line splitting versus applied electric field as determined from the linewidth increase (Fe^{3+} in Al₂O₃). Two assumed line shapes are shown. The electric and magnetic fields are parallel to the *c* axis. The indicated slopes of the straight lines are equal to $6R_{333}$.

be seen, the scatter of the experimental points is considerable. The behavior of the splitting with applied field for this same line is shown more clearly in the data of Fig. 4, where the splittings are derived from the linewidth increase. The linearity of the splitting with applied field is evident and is in agreement with theory.¹ The figure also illustrates the difference in the derived splitting when a Gaussian or a Lorentzian line shape is assumed. In order to obtain a more accurate value for this splitting constant, a line-shape analysis⁸ was made which showed how closely the actual lines approximated the assumed shapes, and appropriate corrections were made. The results for Mn²⁺ in Al₂O₃ were similar to those for Fe³⁺ except for a somewhat smaller change in width due to an increased initial ESR linewidth (12.3 G). For Mn²⁺, one of the best resolved lines of the electronic spin $1/2 \leftrightarrow 3/2$ hyperfine set was used for most measurements.

The values obtained by the two methods for both ions are shown in Table I.⁹ In this table, the splittings are tabulated in terms of the parameter R_{333} which appears in the electric field Hamiltonian⁴

$$\Im C = \frac{1}{2} \sum_{ijk} R_{ijk} E_i (S_j S_k + S_k S_j), \quad j \le k,$$
 (2)

where E_i and S_i are the *i*th components of the applied electric field and the electronic spin operator, respectively. Equation (2) is the most general interaction which is linear in the applied field and quadratic in the spin operators. It is worthwhile to point out that the relation between 2a and R_{333} depends on the transition

TA	BLE	I.	Compari	ison	of	the	axia	al e	electri	ic fie	eld	parame	eter
R_{333}	value	es	obtained	by	the	seco	nd	mon	nent	and	the	linewi	dth
meth	ods.												

	R_{333} (10 ⁻⁵ G cm/V)			
Ion	Second moment	Linewidth		
Mn ²⁺	$0.72 {\pm} 0.06$	0.67 ± 0.06		
Fe ³⁺	$0.70 {\pm} 0.10$	0.64 ± 0.05		

considered. Making use of Eq. (2), one finds for the $1/2 \leftrightarrow 3/2$ transition $2a = 6R_{333}E_3$, while for a $3/2 \leftrightarrow 5/2$ transition $2a = 12R_{333}E_3$.

Under C_3 symmetry, the tensor R_{ijk} has only five independent parameters, and when the electric field is applied parallel to the c axis, as was necessary here, only R_{333} can be determined by experiment.⁴ A possible physical origin for such a Hamiltonian was suggested by Rado¹⁰ in his explanation of the magnetoelectric effect in Cr_2O_3 . If we compare the values of R_{333} obtained by the second moment and linewidth methods, as found in Table I, we see that for both ions the two results are in agreement within experimental error. The somewhat smaller values given by the linewidth method are thought not to reflect a systematic difference. These results give one added confidence in the use of the linewidth method when the line splitting is not resolved. This is convenient since it is considerably easier to apply in practice. As another check on this method, the broadening of the $1/2 \leftrightarrow 3/2$ low magnetic field Cr³⁺ line in Al₂O₃ was examined. The value found for R_{333} agreed within experimental error with the known value.⁴

In the case of Mn²⁺, it was possible to make measurements on ESR lines from several hyperfine sets. Although the actual amount of splitting for lines from the $3/2 \leftrightarrow 5/2$ hyperfine set was twice as large as for the $1/2 \leftrightarrow 3/2$ set, measurements were more difficult because of the increased linewidth and reduced signalto-noise ratio. The results, however, agreed with those shown in Table I. In addition, measurements were made on the $1/2 \leftrightarrow -1/2$ hyperfine set of Mn²⁺ to check for possible changes in the g tensor. No change was observed at fields up to 127 kV/cm although the broadening due to a shift of 3×10^{-4} of the initial g value would have been detectable. For the Fe³⁺ ion, no pure $1/2 \leftrightarrow -1/2$ transition exists at the X-band frequency used here (\sim 9400 Mc/sec),¹¹ and as a result no estimate could be made of the g shift although it is expected to be quite small since the experimental g value is quite close to the free spin value.

Since only electric field broadening of the ESR lines was observed for the ions considered here, it is necessary to point out certain assumptions made in the interpretation of the data. Thus it was assumed that

⁸ R. T. Weidner and C. A. Whitmer, Phys. Rev. **91**, 1279 (1953). ⁹ These values are somewhat different than those reported earlier [Bull. Am. Phys. Soc. **8**, 259 (1963)] as a result of additional measurements at higher electric fields.

¹⁰ G. T. Rado, Phys. Rev. Letters 6, 609 (1961).

¹¹ L. S. Kornienko and A. M. Prokhorov, Zh. Eksperim. i Teor. Fiz. 40, 1594 (1961) [English transl.: Soviet Phys.—JETP 13, 1120 (1961)].

TABLE II. Crystal radii in octahedral coordination.ª

Ion	Radius (Å)		
Gd ³⁺	1.02		
Mn ²⁺	0.80		
Cr ³⁺	0.69		
Fe ³⁺	0.64		
Al ³⁺	0.50		

^a L. Pauling, Nature of the Chemical Bond (Cornell University Press, Ithaca, New York, 1960), 3rd ed., p. 518.

upon application of an electric field the ESR line splits into two parts of equal intensity which retain the same shape and width as the original line. In the case of Cr³⁺ in Al₂O₃, it has indeed been shown⁴ that the linewidth is not altered appreciably when the line is completely split by the electric field. On the other hand, the equal intensity of the two parts would imply that sites related by inversion symmetry are equally populated. In this regard, the observed lack of equality of the intensity of the Gd³⁺ resonance from two physically equivalent sites in Al₂O₃ (such as sites a and b of Fig. 1) should be noted. This has been attributed to a nonequivalence of the two sites during the growth process and is associated with the large size of the Gd³⁺ ion² (Table II). However, the Fe³⁺ spectrum is known to be normal,² and while the complexity of the Mn²⁺ spectrum when the angle between the magnetic field and the c axis is not near 0 or 90° precludes a similar statement concerning this ion, we note that its crystal radius is not greatly different from those of Cr³⁺ and Fe³⁺ (Table II). It is therefore expected that the assumption of equality of intensity would be valid for both Fe³⁺ and Mn²⁺. Experimentally, it is possible to resolve this question by measuring the shift in line centroid when the electric field is applied.⁴ By this means it was found that for both Mn²⁺ and Fe³⁺ the inversion related sites were equally populated to within one part in fourteen.

It should be pointed out that the nonvanishing value of R_{333} found for these two ions shows conclusively that they are situated in sites which lack inversion symmetry. This finding strongly suggests that the ions enter substitutionally into the Al³⁺ sites rather than into sites such as d and e (Fig. 1) which possess inversion symmetry. The axial symmetry of the Mn²⁺ spectrum indicates that any local charge compensation must lie on the c axis.¹² In the case of the crystals used here, no additional impurities have been added to achieve charge compensation. Furthermore, charge compensation by Mn⁴⁺ is ruled out since no spectrum due to this ion is seen. While it is possible that a monovalent interstitial or quadrivalent substitutional impurity may be located adjacent to the Mn²⁺ ion, the ESR spectrum would then indicate that almost every Mn²⁺ ion is so compensated.

A summary of the "best" electric field effect parameters for Fe^{3+} and Mn^{2+} is presented in Table III along with the corresponding Cr^{3+} values for comparison. These values represent an average of those obtained by the second moment and linewidth methods. For Fe³⁺ they include determination on crystals from two different batches, one of which contained Fe³⁺ as an unwanted impurity, while for Mn²⁺, results from lines measured in four hyperfine sets are included.

It is of some interest to consider the relative size of the electric field effect for two isoelectronic ions such as we have here. Table III illustrates the fact that within experimental error, the parameter R_{333} has the same value for both ions and is almost an order of magnitude smaller than for Cr^{3+} . We can consider R_{333} as expressing the change in the axial crystal field parameter D with applied field E. We then find the relation

$$\frac{3}{2}R_{333} = \partial D/\partial E. \tag{3}$$

The third column in Table III lists the relative changes $D^{-1}(\partial D/\partial E)$ with applied electric field. In these terms, we see that Mn²⁺ and Cr³⁺ are quite similar in their relative D changes while the effect for Fe³⁺ is much smaller. This results, of course, from the large D value for Fe³⁺ (+1796.4 G¹¹) as compared to isoelectronic Mn²⁺ (+208.3 G¹²). This difference in D is characteristic of all crystals in which both ions have been examined in sites of axial symmetry. Finally, in the last column of Table III, the relative change in g value with applied field is listed.

DISCUSSION

Since the theory of the D parameter as arising from the even components of the crystalline field potential is not fully understood in the case of d^5 ions,¹³ it does not seem profitable at this time to make any detailed calculations based on the odd crystal field components. Such calculations would be required, of course, to make a proper estimate of the change in D with applied field which arises via the mixing of odd excited states into the even ground state by the odd components of the crystalline field potential. We therefore content ourselves with some qualitative remarks. Clogston,¹⁴ in explaining the large optical Faraday rotation of the

TABLE III. Summary of the electric field parameters for $\mathrm{Mn^{2+}}$ and $\mathrm{Fe^{3+}}$ in $\mathrm{Al_2O_3}.$

Ion	R_{333} (10 ⁻⁵ G cm/V)	$D^{-1}(\partial D/\partial E)$ (10 ⁻⁹ cm/V)	$g^{-1}(\partial g/\partial E)$ (10 ⁻⁹ cm/V)
Mn ²⁺ Fe ³⁺ Cr ³⁺	0.69 ± 0.06 0.67 ± 0.06 6.43 ± 0.11^{a}	50 ± 4 5.6 ± 0.5 46.6 ± 0.8^{a}	≤2.4

^a Taken from Ref. 4 for comparison.

¹³ For recent discussions see: W. J. Nicholson and G. Burns, Phys. Rev. **129**, 2490 (1963); and A. M. Germanier, D. Gainon, and R. Lacroix, Phys. Letters **2**, 105 (1962).
¹⁴ A. M. Clogston, J. Appl. Phys. **31**, 198S (1960).

¹² W. Low and J. T. Suss, Phys. Rev. 119, 132 (1960).

ferrimagnetic garnets, postulated that the ${}^{6}P(3d^{4}p)$ state of Fe³⁺ is depressed from the free-ion value of 190 000 cm⁻¹ into the neighborhood of 50 000 cm⁻¹, based on optical absorption measurements on Fe⁸⁺ in Al₂O₃. Another possibility considered was that the strong absorption at 50 000 cm⁻¹ may arise instead from charge transfer states associated with the oxygen ligands. In any case, there are states of predominantly odd parity near 39 000 and 51 000 cm⁻¹. Such states could contribute to the microwave electric field effect by being mixed into the ground state, as noted above.

It should be pointed out, however, that in the case of Cr^{3+} in Al₂O₃, detailed calculations⁴ indicate that the predominant effect of the applied field is connected with the motion of the Cr^{3+} ion relative to the surrounding O^{2-} ions rather than due to the distortion of the Cr^{3+} electronic wave function. A molecular orbital calculation for the complex containing the paramagnetic ion¹⁵ was found necessary to predict the proper effect of the ionic motion. In this calculation, it was shown that the sign of D and its proper magnitude can be explained by an anisotropic spin-orbit coupling. The change in this anisotropy produced by ionic motion then gives rise to the observed effect.

In the present case, an attempt is made to get some idea of the size of relative change in D with electric field within the framework of a crystal field model. This can be done by considering the motion of the ion as producing a change in the even axial component $V_{\rm ax}$ of the crystalline potential. The D parameter may be expressed in the following manner,^{6,13}

$$D = D_1 V_{\rm ax} + D_2 V_{\rm ax}^2. \tag{4}$$

The work of Nicholson and Burns¹² suggests that the simple crystal field calculations of Watanabe⁶ give the wrong sign for D_2 . In the following, the form of (4) is taken as valid as suggested by theory, but the signs of D_1 and D_2 are those determined by experiment. This assumption is not arbitrary since it is clear that D must vanish in the absence of an axial field, and can in general be expressed in the form of a power series in $V_{\rm ax}$. For the small relative ionic displacement x produced by available applied electric fields (about 10^{-12} cm for 10^5 V/cm),

$$V_{\rm ax} = V_{\rm ax}^{0} + (\partial V_{\rm ax}/\partial x)_{0}x.$$
 (5)

The quantity $(\partial V_{ax}/\partial x)_0$ is nonvanishing because of the lack of inversion symmetry at the cation site. Otherwise the change would be quadratic in x. From (4) and (5) we obtain

$$\frac{1}{D} \frac{\partial D}{\partial E} = \frac{1}{V_{\rm ax}^0} \left(\frac{\partial V_{\rm ax}}{\partial x} \right)_0 \left[\frac{D_1 + 2D_2 V_{\rm ax}^0}{D_1 + D_2 V_{\rm ax}^0} \right] k, \qquad (6)$$

where $k \equiv \partial x / \partial E$ is the relative ionic motion per unit applied field. If we consider the experimental data for

the relationship between D and V_{ax} as found for Fe³⁺ in various crystals by Nicholson and Burns,13 we conclude that D_1 and $D_2 V_{ax}^0$ have the same sign in Al_2O_3 . It follows that the value of the expression in brackets in (6) lies between one and two. It is assumed that the same sign relationship is true for Mn²⁺. Further, we expect k to have roughly the same value for Fe^{3+} and Mn^{2+} . (Although the force on the Mn^{2+} ion will be roughly two-thirds that on the Fe³⁺ ion, the restoring forces will be correspondingly reduced because of the relaxation of the lattice about the site of the divalent ion.) As was noted in Table III, however, the value of $D^{-1}(\partial D/\partial E)$ is about 9 times larger for Mn²⁺. It is concluded, therefore, that the quantity $(V_{ax}^{0})^{-1}$ $\times (\partial V_{\rm ax}/\partial x)_0$ is much larger in the case of Mn²⁺ if ionic motion is important in giving rise to the electric field effect. Experimentally, it should be possible to obtain the relative sizes of V_{ax}^{0} for the two ions from Mossbauer¹³ and electron nuclear double resonance measurements of the quadrupole splitting. By thus considering the relative change in D rather than the absolute change, the difference in the sensitivity of the two ions to axial crystalline fields (reflected in D_1 and D_2) is eliminated for the most part. This difference shows up in the considerably different D values of the two ions as noted above.

This technique permits an order of magnitude estimate of the value of $D^{-1}(\partial D/\partial E)$ due to ionic motion to be made. The work of Laurance and Lambe¹⁶ allows the quantity $(V_{ax}^{0})^{-1}(\partial V_{ax}/\partial x)_{0}$ to be evaluated at various positions of the ion along the c axis of the crystal since they have performed a point-charge calculation of the electric field gradient. In the absence of a molecular orbital calculation, this appears to be the only reasonable way to estimate the relative change in V_{ax} with motion. The electric field gradient is the leading term in the axial field and is assumed here to be dominant. If we consider the normal Al³⁺ ion position, we find that $(V_{ax}^{0})^{-1}(\partial V_{ax}/\partial x)_{0} = 6.7$ Å⁻¹. Furthermore, for an applied field of 10^5 V/cm, $kE \simeq 2.2 \times 10^{-4}$ Å for a trivalent ion in Al₂O₃. Then using (6) we obtain $D^{-1}(\partial D/\partial E) = (15-30) \times 10^{-4}/10^5$ V/cm. This is of the same order of magnitude as the experimentally determined values for Fe³⁺, especially, and Mn²⁺. If we consider the case of Cr3+, Kamimura17 has indicated that the relationship between the trigonal field and Dis linear, even including anisotropic spin-orbit effects. The calculations of Lohr and Lipscomb, however, suggest that this is not valid, perhaps because of the importance of the $Cr^{3+2}T_2(t_2^3)$ state which Kamimura neglected. For a linear dependence, the quantity in brackets in (6) is unity while for the molecular orbital calculation, the corresponding value is 2.5. Then $D^{-1}(\partial D/\partial E) = (15-38) \times 10^{-9}$ cm/V, which is comparable in magnitude to the measured value. Although

¹⁵ L. L. Lohr, Jr., and W. N. Lipscomb, J. Chem. Phys. 38, 1607 (1963).

¹⁶ N. Laurance and J. Lambe, Phys. Rev. 132, 1029 (1963).

¹⁷ H. Kamimura, Phys. Rev. 128, 1077 (1962).

it has been suggested³ that the odd components of the axial crystalline field contribute importantly to the value of D, the arguments given are no longer considered to be valid.¹⁸ While it is clear that such an oversimplified calculation, based as it is on a point charge model, cannot be expected to give accurate results; it nevertheless indicates that ionic motion probably plays an important role in the microwave electric field effect for Fe³⁺ and Mn²⁺ just as for Cr³⁺. However, the above discussion does not rule out possible significant contributions from the effect of distortion of the electronic wave function of these S-state ions.

It would be valuable to examine ions which are isoelectronic to the more tractable Cr^{3+} ion. These include V^{2+} and Mn^{4+} which have D parameters in Al_2O_3 closely similar¹⁶ to that for Cr^{3+} . In addition, the effect of

¹⁸ J. O. Artman and J. C. Murphy (to be published).

electric fields on the optical spectra of these ions has been studied and interpreted in terms of electronic wave function distortion and ionic displacement.¹⁹ Both of these ions are presently under study in this Laboratory.

ACKNOWLEDGMENTS

The author would like to express his thanks to V. J. Folen for many illuminating discussions and for his help in x-ray crystal orientation, to R. A. Becker for growing the crystals used, and to the Electron Tubes Shop at the Naval Research Laboratory for their help in fabricating and silvering the samples. I also profited by discussions with J. O. Artman and J. C. Murphy and from the receipt of a copy of M. D. Sturge's manuscript prior to publication.

¹⁹ M. D. Sturge, Phys. Rev. 133, A795 (1964).

PHYSICAL REVIEW

VOLUME 135, NUMBER 2A

20 JULY 1964

Coercive Force of Iron Resulting from the Interaction of Domain Boundaries with Large Nonmagnetic Inclusions

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The interaction of large inclusions with slowly moving domain boundaries has been quantitatively examined and the contribution made by such inclusions to the coercive force predicted. Several physical models of the interaction between large inclusions and moving Bloch walls were investigated by making calculations of the energy of the closure domain configuration about the inclusions for various positions of the moving domain boundary. By statistically treating a random distribution of inclusions, the coercive force was calculated as a function of the inclusion distribution parameters. Several features of the interactions between spike-shaped closure domains and moving domain boundaries have been elucidated.

I. INTRODUCTION

I T has been shown by many investigators that dislocations, nonmagnetic inclusions, and other chemical and physical inhomogeneities influence the properties of bulk ferromagnetic materials. One manner in which imperfections affect the magnetic properties of a material is by acting as impediments or obstacles to the motion of domain boundaries. An analysis of the interaction between domain boundaries and structural imperfections is important in the understanding of irreversible ferromagnetic properties at low and intermediate frequencies.

The particular problem of the determination of the contribution to the coercive force which results from the interaction of domain boundaries with large nonmagnetic inclusions is treated in this paper. It consists of an analysis of the interaction of a moving boundary with the subsidiary domain structure about such inclusions.

One of the first attempts to evaluate the effects of

nonmagnetic inclusions on coercive force was that by Kersten.¹ He supposed that the binding energy between a Bloch wall and a nonmagnetic inclusion is given by the reduction in interfacial energy caused by the intersection of the particle by the domain boundary. It was shown by Néel² that when a Bloch wall bisects an inclusion, the reduction in the magnetostatic energy is much greater than the change in Bloch wall energy and is therefore more important in determining the binding energy. Néel³ also later showed that it is necessary to adopt a realistic statistical model of the particle distribution to be able to compute the coercive force which arises from small inclusions. Dijkstra and Wert,⁴ using a simplified form of Néel's statistical model, calculated the coercive force due to inclusions with diameters up to the thickness of a Bloch wall.

- ³ L. Néel, Ann. Univ. Grenoble 22, 299 (1946).
- ⁴L. J. Dijkstra and C. Wert, Phys. Rev. 79, 979 (1950).

¹ M. Kersten, Physik Z. 44, 63 (1943).

² L. Néel, Cahiers Phys. 25, 21 (1944).