



ZINNIMIDINE AND 5-(3',3'-DIMETHYLALLYLOXY)-7-METHOXY-6-METHYLPHTHALIDE FROM *ALTERNARIA PORRI*

RIKISAKU SUEMITSU, KEIICHIRO OHNISHI, YOSHIYUKI MORIKAWA and SHIGEMI NAGATOMO

Doshisha University, Kamikyo-ku, Kyoto 602, Japan

(Received 24 February 1994)

Key Word Index—*Alternaria porri*; Deuteromycotina; fungus; metabolite; imidine; phthalide.

Abstract—The structures of two metabolites, isolated from the culture liquid of *Alternaria porri*, have been identified as zinnimidine and 5-(3',3'-dimethylallyloxy)-7-methoxy-6-methylphthalide. The latter is the first reported isolation as a naturally occurring product.

INTRODUCTION

The fungus *Alternaria porri* (Ellis) Ciferri, the causal fungus of black spot disease in stone-leek and onion, produces phytotoxins such as tentoxin [1], porritoxin [2], silvaticol [3] and porritoxinol [4]. In addition, we reported 6-(3',3'-dimethylallyloxy)-4-methoxy-5-methylphthalide (1) [5], although no particular inhibitory activity was apparent toward lettuce and stone-leek seedlings. This paper deals with the isolation and structural elucidation of further metabolites, zinnimidine (2) and 5-(3',3'-dimethylallyloxy)-7-methoxy-6-methylphthalide (3), from *Alternaria porri*.

RESULTS AND DISCUSSION

Silica gel chromatography of adsorbates with Amberlite XAD-7 led to the isolation of two compounds, **A** and **B**, in order of increasing R_f values. High resolution mass spectral analysis gave $C_{15}H_{19}NO_3$ for **A** and IR absorption was indicative of amides at 3456 and 1695, ether oxygen at 1115, a methoxy group at 1452, together with benzene rings at 1473 and 665 cm^{-1} . The UV spectral data showed the existence of a benzoyl group at 213 (4.73), 252 (4.04), 292 (3.67). The ^1H and ^{13}C NMR data are given in Table 1. Comparing the spectral data of **A** with those of zinnimidine (2), which was isolated from *Alternaria cichorii* [6], gave essentially the same signals. Consequently, **A** was identified as zinnimidine (2).

High resolution mass spectral analysis gave $C_{15}H_{18}O_4$ for compound **B**. IR absorption was indicative of a γ -lactone at 1758, together with benzene rings at 1610 and 695 cm^{-1} . The UV spectral data gave signals at 215.8 (4.49) and 258.5 (4.17). The ^1H and ^{13}C NMR data are shown in Table 1. In order to compare them with those of 1, spectral data of **A** are also shown in Table 1. Comparing the spectral data of **B** with those of 1, the former closely resembles the latter, except for the chemical shift of the aromatic proton at $\delta 6.62$, while the aromatic

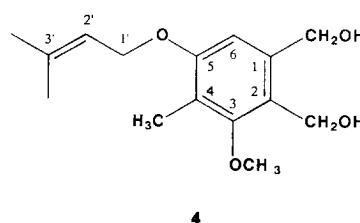
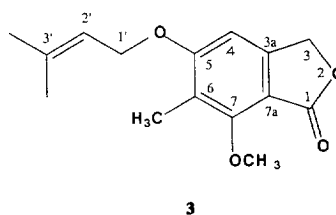
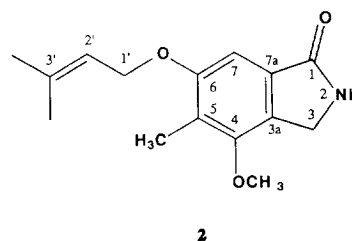
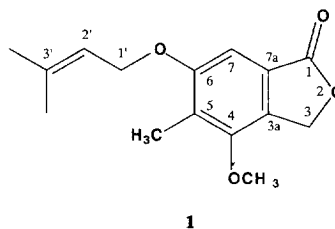


Table 1. Comparison of ^1H and ^{13}C NMR spectral data of 6-(3',3'-dimethylallyloxy)-4-methoxy-5-methylphthalide, compounds A and B.

6-(3',3'-dimethylallyloxy)-4-methoxy-5-methylphthalide (1)				A (zinnimidine)				B (3)			
H	C	H	C	H	C	H	C	H	C	H	C
3	5.83 s	171.3	1	171.2	1	171.2	1	5.18 s	3	169.0	1
		68.3	3	43.9	3	43.9	3		7a	68.7	3
		124.7	3a	126.2	3a	126.2	3a		7	148.1	7a
		159.2	4	158.7	4	158.7	4		6	163.6	7
		125.8	5	123.9	5	123.9	5		5	120.8	6
		152.9	6	153.7	6	153.7	6		4	157.8	5
7	7.08 s	101.9	7	7.11 s	7	101.2	7	6.62 s	4	99.5	4
		128.1	7a	130.8	7a	130.8	7a		3a	109.5	3a
1'	4.57 d (6.72)	65.8	1'	4.58 d (6.72)	1'	65.7	1'	4.59 d (6.71)	1'	65.8	1'
2'	5.47 t (6.72)	119.2	2'	5.18 t (6.72)	2'	119.6	2'	5.49 t (6.71)	2'	118.9	2'
OMe-4	3.89 s	138.2	3'	3.87 s	3'	137.8	3'	4.03 s	3'	138.6	3'
Me-5	2.21 s	59.3	OMe-4	2.22 s	OMe-4	59.8	OMe-4	2.15 s	OMe-7	62.1	OMe-7
Me-3'-1	1.75 s	9.7	Me-5	1.74 s	Me-5	9.6	Me-5	1.75 s	Me-6	8.7	Me-6
Me-3'-2	1.80 s	18.3	Me-3'-1	1.80 s	Me-3'-1	18.3	Me-3'-1	1.81 s	Me-3'-1	18.3	Me-3'-1
		25.8	Me-3'-2		Me-3'-2	25.8	Me-3'-2		Me-3'-2	25.8	Me-3'-2

proton in **1** is at $\delta 7.08$ (Table 1). The strikingly lower shift of the aromatic proton in **1** can be accommodated by the anisotropy of the carbonyl group. Starratt has reported that two phthalides, **1** (mp 85–86°) and **3** (mp 106–108°), were obtained on chromic oxidation of zinniol (**4**) [7]. Comparing the UV, IR and ^1H NMR spectral data of **B** with those of **3**, the former closely resembles the latter. Consequently, based on those spectral data and the coincidence of melting point, **B** was identified as **3**. Compound **3** is the first example found naturally occurring in fungi.

EXPERIMENTAL

Fungus. The strain of *A. porri* used in this experiment was purchased from IFO (Institute for Fermentation, Osaka), strain no. 9762.

Extraction and isolation of zinnimidine and 5-(3',3'-dimethylallyloxy)-7-methoxy-6-methylphthalide. The culture conditions using Richards medium were the same as those previously reported [5] and the isolation method was the same as that previously reported [3]. The Me_2CO extract of adsorbents obtained from Amberlite XAD-7 was subjected to prep. TLC (Merck Silica gel 60) in C_6H_6 – Me_2CO – HOAc (60:40:1) to obtain fraction A (R_f 0.82–0.95) and B (R_f 0.33–0.60). Fraction A was further subjected to TLC in C_6H_{14} – EtOAc (2:1) and finally the fr. at R_f 0.58–0.64 was chromatographed in CHCl_3 – MeOH (97:3). The fr. at R_f 0.52 gave needles of mp 107–109° in a yield of 5 mg from 300 l of culture medium. Fraction B was subjected to prep. TLC in CHCl_3 – MeOH (9:1) and the fr. at R_f 0.55–0.73 further chromatographed in C_6H_{14} – EtOAc (5:1). Finally, the fr. at R_f 0.38–0.53 was purified by HPLC using YMC S-343

(Yamamura Chemical Labs) with 40% MeCN and gave needles of mp 136–138° in a yield of 6 mg from 300 l of the culture medium.

5-(3',3'-dimethylallyloxy)-7-methoxy-6-methylphthalide (3). Needles, mp 107–109° (lit. mp 106–108° [7]); IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 1758 (γ -lactone, lit. 1743 [6]), 1610, 695 (phenyl); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 215.8 (4.49), 258.5 (4.17) (lit. 259 (4.18) [7]). EI-MS m/z : 262.1221 $[\text{M}]^+$ corresponding to $\text{C}_{15}\text{H}_{18}\text{O}_4$ (262.1204), 194 $[\text{M} - \text{C}_5\text{H}_8]^+$, 165 $[\text{M} - \text{CHO}]^+$, 69 $[\text{C}_5\text{H}_9]^+$. ^1H (400 MHz, CDCl_3) and ^{13}C (100 MHz, CDCl_3) NMR spectra: Table 1.

Zinnimidine (2). Needles, mp 136–138°; IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3456 (NH), 1695 (CONH), 1473, 665 (phenyl), 1115 (ether); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 213 (4.73), 252 (4.04), 292 (3.67). EI-MS m/z : 261.1329 $[\text{M}]^+$ corresponding to $\text{C}_{15}\text{H}_{19}\text{NO}_3$ (261.1364), 193 $[\text{M} - \text{C}_5\text{H}_8]^+$. ^1H (400 MHz, CDCl_3) and ^{13}C (100 MHz, CDCl_3) NMR spectra: Table 1.

REFERENCES

1. Suemitsu, R., Ohnishi, K., Nobuhara, T., Horiuchi, M. and Horiuchi, K. (1990) *Agric. Biol. Chem.* **54**, 2449.
2. Suemitsu, R., Ohnishi, K., Horiuchi, M., Kitaguchi, A. and Odamura, K. (1992) *Phytochemistry* **31**, 2325.
3. Suemitsu, R., Ohnishi, K., Horiuchi, M., Morikawa, Y., Sakaki, Y. and Matsumoto, Y. (1993) *Biosci. Biotech. Biochem.* **57**, 334.
4. Suemitsu, R., Ohnishi, K., Morikawa, Y., Ideguchi, I. and Uno, H. (1994) *Phytochemistry* **35**, 603.
5. Suemitsu, R., Ohnishi, K., Horiuchi, M. and Morikawa, Y. (1992) *Biosci. Biotech. Biochem.* **56**, 986.
6. Stierle, A., Hershenhorn, J. and Strobel, G. (1993) *Phytochemistry* **32**, 1145.
7. Starratt, A. (1968) *Can. J. Chem.* **46**, 767.