



ABIETANE AND ICETEXANE DITERPENOIDS FROM THE ROOTS OF SALVIA ASPERA

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Abstract—From the roots of Salvia aspera, two new icetexane diterpenoids (salviasperanol and 5,6-dihidro- 6α -hydroxysalviasperanol) and one new abietane derivative (6-epi-demethylesquirolin D) were isolated in addition to the previously known diterpenoids sugiol, taxodione and demethylsalvicanol. The structures of the new compounds were established by spectroscopic means. An X-ray analysis was performed on salviasperanol.

INTRODUCTION

The phytochemical study of the aerial parts of several species of Salvia, subgenus Calosphace, led to the isolation of a number of diterpenes mainly of the neoclerodane type in more than the 80% of the species studied [1]. Some abietane and icetexane type diterpenoids have been isolated from spp. belonging to Section Erythrostachys and Tomentellae [2, 3]. On the other hand, phytochemical studies of the aerial parts and roots of European and Asiatic Salvia spp. led to the isolation of a number of diterpenoids with an abietane skeleton in almost 100% of the species studied. While the chemical composition of the roots of several European and Asiatic salvias is known and the presence of abietane quinones in the roots of these plants has been postulated as a chemotaxonomic character of the genus [4], the content of the roots of American salvias is poorly documented. As a part of our continuing systematic study of the genus, we have undertaken the analysis of the roots of Mexican Salvia spp.. In this paper we report on the diterpenoid content of the roots of Salvia aspera M. et G., a shrub placed in the section Conzattiana of the subgenus Calosphace [5].

RESULTS AND DISCUSSION

Extraction of the roots of S. aspera afforded, after extensive chromatographic purification, the previously

known abietane-type diterpenoids sugiol [6] and taxodione [7]. The icetexane diterpenoid demethylsalvicanol [8] was isolated as its diacetyl derivative 4. In addition, we isolated two new icetexane and one new abietane diterpenoids to which we assigned the structures 1-3 on the basis of the following considerations.

Compound 1 (C₂₀H₂₆O₃, MS), named salviasperanol, showed IR bands for phenolic groups (3500, 3420 cm⁻¹) and aromatic double bonds (1620 cm⁻¹). The ¹H NMR spectrum (Table 1) showed signals for two secondary

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B. ESQUIVEL et al.

Table 1. ¹H NMR spectral data for compounds 1-3 and 5, 6 (CDCl₃, 200 MHz)

H	1*	2	3†	5	6‡
6	6.04 dd	4.30 t	4.03 t	2.33 ddd	6.02 d
	(2.1, 0.8)	(6.2)	(4.4)	(11.4, 9.7, 7.7)	(2)
6α		_	_	1.44 dt	_
				(11.4, 2.5)	
7	5.07 d	4.72 d	4.63 d	4.92 dd	5.15 d
	(2.1)	(6.2)	(4.4)	(7.7, 2.5)	(2)
14	6.43 s	6.54 s	6.73 s	6.41 s	6.8 s
15	3.08 hept	3.12 hept	3.27 quint	3.08 quint	2.95 quini
	(6.8)	(6.8)	(6.8)	(7)	(6)
3H-16	1.24 d	1.24 d	1.21 d	1.23 d	1.2 d
	(6.8)	(6.8)	(6.8)	(7)	(6)
3H-17	1.30 d	1.23 d	1.20 d	1.22 d	1.15 d
	(6.8)	(6.8)	(6.8)	(7)	(6)
3H-18	1.02 s	1.03 s	1.06 s	$0.93 \ s$	1.15 s
3H-19	1.12 s	1.01 s	1.04 s	0.91 s	1.0 s
20A	2.88 d	2.76 d	5.36 s	3.16 d	2.75 d
	(16.5)	(16)		(17.6)	(16)
20B	2.65 d	2.43 d		2.69 d	2.5 d
	(16.5)	(16)		(17.6)	(16)
OAc					2.30 s§
R-OH	5.12 br s	5.2 br s	6.75 br s	5.0 br s	
	5.35 br s	5.25 br s	7.45 br s	5.2 br s	

^{*}Assignments confirmed by ¹H-¹H 2D COSY spectra.

methyl groups at $\delta 1.24$ (d, J = 6.8 Hz) and $\delta 1.30$ (d), coupled with a proton responsible of an heptuplet at $\delta 3.08 (J = 6.8 \text{ Hz})$. These signals indicated the presence of an aromatic isopropyl group. A singlet at $\delta 6.43$ was ascribed to one aromatic proton and two broad singlets, exchangeable with D_2O , at $\delta 5.12$ and $\delta 5.35$, were assigned to two hydroxy phenol groups. These data indicated the presence of a pentasubstituted aromatic ring in 1, similar to that present in demethylsalvicanol [8]. The ¹³C NMR spectrum (Table 2) was in agreement with this conclusion and also showed signals for four methylene groups at δ 38.6 (C-1), 19.0 (C-2), 40.0 (C-3), 30.2 (C-20) and only two additional methyl groups at δ 29.9 (C-18) and 27.7 (C-19) indicating an icetexane type skeleton for salviasperanol (1) [3, 9]. That 1 possesses a trisubstituted double bond was deduced from the presence of two additional sp² carbon atoms in the ¹³C NMR spectrum; a singlet (δ 149.5) and a doublet (δ 128.3), ascribed to C-5 and C-6, respectively. A double doublet, observed in the ¹H NMR spectrum of 1 (Table 1), at δ 6.04 was assigned to the olefinic proton H-6. Homonuclear 2D COSY experiments indicated that H-6 is longe-range coupled (J = 0.8 Hz) with a complex signal at δ 2.1 ascribed to H-1 β by HETCOR experiments and with a doublet at δ 5.07 (J = 2.1 Hz) which was ascribed to the geminal proton of an ethereal function. The chemical shift of this signal indicated a benzylic and allylic nature for this proton and was assigned to H-7. The ethereal function present in 1 must be cyclic, as indicated by its molecular formula. In

the 13 C NMR spectrum of 1 (Table 2) two signals due to oxygenated sp³ carbon atoms were observed, a doublet at δ 79.9 ascribed to C-7 and a singlet at δ 83.75, assigned to C-10.

The structure proposed for 1 was confirmed by X-ray diffraction analysis. The computer generated perspective drawing of salviasperanol is shown in Fig. 1. The A ring in the structure of 1, adopts a distorted chair conformation, flattened at C-5, the five- and six-membered heterocyclic ring exhibit envelope and sofa conformations respectively, having the oxygen atom O-1 as a common flap. The essentially planar aromatic ring is almost coplanar to the mean plane formed by the C-7, C-8, C-9, C-20 and C-10 atoms of the pirane moiety (angle between planes 3.5°). The isopropyl group is tilted, so that one of the methyl groups, C-16, lies nearly perpendicular (torsion angle C-14, C-13, C-15, C-16 = 99.4°) to the aromatic ring. Phenolic H-atoms are involved in the formation of intramolecular (O-3-H-3 \rightarrow O-2:2.15 Å) and intermolecular $(O-2-H-2 \rightarrow O-1:1.97 \text{ Å})$ hydrogen bonds; the latter giving arise to formation of endless chains parallel to the (100) direction, responsible for the crystal packing. Figure 1 shows only the relative stereochemistry of salviasperanol, since attempts to obtain a suitable derivative for absolute stereochemistry determination were unsuccessful.

Catalytic hydrogenation of 1 was possible only when PtO₂ was used as catalyst. The ¹H NMR spectrum of the dihydro derivative 5 (Table 1) was in agreement with the

[†]Run in CDCl₃-DMSO-d₆ (4:1) solution.

[‡]Run at 80 MHz.

[§]Six protons intensity.

Table 2. ¹³C NMR data for compounds 1-3 (CDCl₃, 50 MHz)*

C	1	2	3
1	38.6 t	29.6 t	29.3 t
2	19.0 t	15.2 t	18.2 t
3	40.0 t	39.0 t	40.9 t
4	33.6 s	31.3 s	33.5 s
5	149.5 s	58.1 d	56.3 s
6	128.3 d	77.9 d	68.9 d
7	79.9 d	78.4 d	73.7 d
8	131.4 s	126.6 s	128.7 s
9	117.0 s	$117.3 \ s$	122.1 s
10	83.8 s	80.5 s	46.8 s
11	142.6 s	142.3 s	143.2 s
12	139.9 s	140.6 s	142.9 s
13	131.0 s	132.3 s	133.9 s
14	112.1 d	115.7 d	115.7 d
15	27.0 d	27.2 d	26.7 d
16	22.4 q	22.6 q	22.5 q
17	22.8 q	22.6 q	22.4 q
18	29.9 q	30.1 q	33.2 q
19	27.7 q	27.8 q	22.9 q
20	30.2 t	30.4 t	91.0 d

*Multiplicities were determined by the DEPT pulse sequence.

Assignments were confirmed by the ${}^{1}H^{-13}CHETCOR$ spectra.

structure proposed for it. Acetylation of 1 yielded the oily derivative 6. The IR spectrum of 6 showed the absence of hydroxyl absorptions and a band at 1770 cm⁻¹ due to the aromatic acetate groups.

The mass spectrum of the second icetexane-type diterpenoid isolated from the roots of S. aspera was consistent with the molecular formula $C_{20}H_{28}O_4$. Structure 2 was proposed for it, based on spectroscopic data. Its IR

spectrum showed bands for hydroxy phenolic groups $(3693, 3604, 3551 \,\mathrm{cm}^{-1})$ and aromatic double bonds (1601, 1539 cm⁻¹). The ¹H and ¹³C NMR spectra of 2 were similar to those of 1, however, they revealed that it was devoid of the 5,6 double bond (Tables 1 and 2). A triplet observed at $\delta 4.30$ (J = 6.2 Hz) in the ¹H NMR spectrum of 2 was assigned to the geminal proton of an hydroxyl group. The coupling constant of this signal indicated an a orientation for this hydroxyl group, which was located at the C-6 position. This proton was coupled with H-7, the geminal proton of the ethereal ring, which was responsible for a doublet at $\delta 4.72$. The trans fusion depicted in 2 was established with the aid of the coupling constant (J = 6.2 Hz) found for H-6, and by comparison with related bicyclic systems [10]. On the basis of the above findings, 2 is 5,6-dihydro-6α-hydroxysalviasper-

The mass spectrum of 3 showed the molecular ion at m/z 348, consistent with a molecular formula $C_{20}H_{28}O_5$ and seven unsaturation units. The spectral data found for 3 indicated that this compound possessed an abietanetype skeleton. The ¹³C NMR spectrum of 3 (Table 2) showed five singlets between δ 122 and 143, and one doublet at δ 115.7, indicating that four of these unsaturations corresponded to a pentasubstituted aromatic ring [8]. A singlet at $\delta 6.73$ in the ¹H NMR spectrum of 3 was ascribed to the aromatic H-14. The characteristic signals of an isopropyl group bound to an aromatic ring, as well as two broad singlets, exchangeable with D2O, corresponding to the hydroxy phenolic groups at the C-11 and C-12 positions, were also observed in the ¹H NMR spectrum of 3 (Table 1). Two singlets at δ 1.04 (3H) and 1.04 (3H) were ascribed to the C-18 and C-19 methyl groups. The angular methyl group (C-20), frequently found in abietane-type diterpenes, was not observed. A singlet at δ 5.36 indicated that C-20 was part of a cyclic hemiketalic group closed to C-7. Therefore, a doublet at $\delta 4.63 (J = 4.4 \text{ Hz})$ was ascribed to the geminal proton of

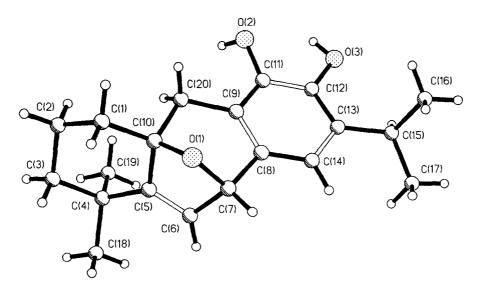


Fig. 1. Computer generated perspective drawing of salviasperanol (1), showing atom numbering.

the hemiketalic ring closure, i.e. H-7. It was coupled with a proton responsible of a triplet at $\delta 4.03$ (J = 4.4 Hz), ascribed to H-6, which must be bound to a carbon atom bearing an additional hydroxy group. The coupling constants of H-6 suggested an α-equatorial orientation for this hydroxy group and an α-axial orientation for H-5, as compared with the related compound 16-hydroxyisorosmanol [11]. The presence of three doublets at δ 91.0 (C-20), 73.7 (C-7) and 68.9 (C-6) in the ¹³C NMR spectrum of 3 was in agreement with the previous discussion (Table 2). The chemical shift of the C-19 methyl group ($\delta 1.04$) suggested a 20R configuration, i.e. the hydroxy group of the hemiketalic moiety points toward the C ring of 3, otherwise an strong deshielding must be expected for this methyl group. A careful inspection of a Dreiding model of 3 supports this conclusion.

The structure of 3 is close to that proposed for esquirolin D, an abietane-type diterpenoid isolated from Coleus eskirolii [12]. However, in esquirolin D, a β orientation for the C-6 hydroxy group was proposed. The presence of an additional methyl group in the ketalic moiety of esquirolin D was not found in 3. On the basis of the above findings 3 must be 6-epi-demethylesquirolin D.

Compounds 1, 5 and 3 showed no lethal effects in the brine shrimp (*Artemia salina*). On the other hand 4 was active at 55 ppm.

It is of interest to note that from the aerial parts of S. aspera only tri-nor-dammarane triterpenoids and neoclerodane diterpenoids were isolated [13]. The presence of abietane and icetexane diterpenoids in the roots of S. aspera (subgenus Calosphace) could be of phytogeographical importance, since the same chemical profile have been found in the roots of S. coulterii [14], placed in the same subgenus, and from the roots of S. apiana [15], an American species belonging to the subgenus Audibertia and also from the roots of the European S. canariensis [16], placed in the subgenus Salvia. Abietane-derived diterpenoids have been also isolated from the roots of S. lavanduloides (personal communication from Dr Emma Maldonado, Instituto de Química UNAM), S. fruticulosa [17] and S. melissodora (Esquivel et al., unpublished results), species also belonging to subgenus Calosphace [5].

EXPERIMENTAL

Mps: Uncorr.; EI-MS: 70 eV, direct inlet; UV, MeOH; ¹H NMR: 200 MHz, CDCl₃, unless noted otherwise; ¹³C NMR: 50 MHz, CDCl₃, TMS as int. standard. Plant material was collected in the state of Puebla (México) and a voucher specimen (MEXU544950) is deposited in the herbarium of the Instituto de Biología de la UNAM.

Extraction, fractionation and isolation of the diterpenoids from the roots of Salvia aspera. Dried and powdered roots of S. aspera (1.506 kg) were extracted with Me_2CO for 4 days at room temp. The solvent was removed in vacuo to yield 62.7 g of a gummy residue which was partitioned between $MeOH-H_2O$ (4:1) and C_6H_6 -petrol (1:1). The aq. methanolic fraction was concd in vacuo, H_2O was added and the mixture was extracted with

EtOAc. The organic phase was dried with Na₂SO₄ and the solvent removed to yield 25.9 g of a gum, which was subjected to vacuum chromatography over silica gel. Mixtures of petrol-EtOAc and EtOAc-MeOH of increasing polarity were used as eluents. From the fractions eluted with petrol-EtOAc (9:1) salviasperanol (1) (260 mg) was isolated. Some fractions eluted with petrol-EtOAc (9:1) were combined and purified by flash chromatography (CH₂Cl₂-Me₂CO, 24:1) to yield 2 (12 mg) as an amorphous powder. Some fractions eluted with petrol-EtOAc (6:4) were rechromatographed over silica gel, using mixtures of CH₂Cl₂ of increasing polarity as eluents, to yield 13 mg of clovandiol and 24 mg of 3 as a crystalline solid. From the fractions eluted with petrol-EtOAc (17:3) of the original column, 6 mg of taxodione, 2 mg of sugiol and 13 mg of clovandiol were isolated after extensive chromatorgaphic purifications.

The C_6H_6 -petrol phase (25.9 g) of the original partition was subjected to vacuum chromatography in a similar manner as the polar phase, to yield an additional crop of taxodione (56 mg) and β -sitosterol (13 mg). Some fractions eluted with petrol-EtOAc (9:1) (494 mg) wee combined and treated with Ac₂O-pyridine in the usual manner to yield 613 mg of a crude mixture which was subjected to flash chromatography using petrol-Me₂CO (17:3), after crystallization from petrol-EtOAc, 4 (20 mg) was isolated.

Salviasperanol (1). Amorphous solid, mp 203–205°; $[\alpha]_D - 31.5$ (CHCl₃; c 0.2); UV λ_{max} nm (log ε): 205 (3.13), 275 (4.5), 280 (4.6); IR ν_{max}^{nujol} cm⁻¹: 3500, 3420, 3230, 1620, 1260, 1160, 855; ¹H NMR: Table 1; ¹³C NMR: Table 2; MS m/z (rel. int.): 314 (16.9), 301 (1.2), 299 (30), 285 (100), 267 (30), 257 (40), 243 (17), 215 (97.9), 173 (45), 115 (30), 91 (30), 83 (30), 55 (50), 43 (50), 41 (99). $C_{20}H_{26}O_2$ requires [M] ⁺ at m/z 314.

5,6-Dihydro-6α-hydroxysalviasperanol (2). Amorphous solid, mp 184–190°; $[\alpha]_D$ – 20 (CHCl₃; c0.1); UV λ_{max} nm (log ε): 205 (4.5), 270 (3.8); IR $\nu_{max}^{CHCl_3}$ cm⁻¹: 3693, 3604, 3551, 1601, 1539, 1451, 1259, 1006, 929; ¹H NMR: Table 1; ¹³C NMR: Table 2; MS m/z (rel. int.): 332 (13.7), 314 (5.9), 301 (2.9), 285 (3), 206 (20), 194 (20), 179 (50), 91 (50). 69 (50), 57 (30), 55 (30), 43 (82), 41 (100). C₂₀H₂₈O₄ requires [M]⁺ at m/z 332.

6-epi-Demethylesquirolin D (3). Mp 203–205°; $[\alpha]_D$ – 56 (CHCl₃; c0.2); $[A]_D$ max $[A]_D$ cm⁻¹: 3616, 3534, 1451, 1392, 1033, 1011; $[A]_D$ H NMR: Table 1; $[A]_D$ CNMR: Table 2; MS m/z (rel. int.): 348 (5.5), 331 (12.5), 330 (55), 303 (20), 302 (60), 301 (40), 273 (50), 232 (30), 231 (100), 115 (3)), 91 (20), 69 (30), 55 (30), 43 (50), 41 (50). $[C]_D$ requires $[M]_D^+$ at $[M]_D^$

Acetylation of salviasperanol (1). Compound 1 (60 mg) in pyridine (0.5 ml) was treated with Ac₂O (0.5 ml) for 10 hr at room temp. After usual work-up and flash chromatography purification, 40 mg of **6** was obtained as an oil. IR $\nu_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 1770, 1630, 1170, 1050, 1090, 880.; ¹H NMR: Table 1; MS m/z (rel. int.): 398 (3), 383 (4), 370 (5), 369 (10), 356 (10), 341 (30), 327 (20), 309 (20), 299 (10), 268 (20), 267 (100). C₂₄H₃₀O₅ required [M]⁺ at m/z 398.

Catalytic hydrogenation of salviasperanol (1). Compound 1 (30 mg) in EtOAc (5 ml) was treated with H_2 in

the presence of PtO₂ (12 mg) as catalyst at room temp. for 6 days. After usual work-up and flash chromatography purification, **5** was isolated as a crystalline solid, mp 196° dec; IR $v_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3604, 3543, 1633, 1596, 1500, 1268, 1253; ¹H NMR: Table 1; MS m/z (rel. int.): 318 (2), 317 (20), 316 (100), 301 (20), 298 (20), 283 (15), 273 (10), 255 (30), 245 (20), 233 (15), 231 (20), 229 (20), 219 (20), 205 (20), 179 (20), 175 (20), 128 (30), 115 (30), 91 (30), 55 (40), 43 (50), 41 (70). $C_{20}H_{28}O_3$ requires [M]⁺ at m/z 316.

X-Ray structure determination of salviasperanol (1). A single crystal of dimension of $0.40 \times 0.17 \times 0.14$ mm, was obtained from slow evaporation of petrol-EtOAc and mounted on a Nicolet P3/F diffractometer with Nifiltered Cu- K_{α} radiation ($\lambda = 1.54178 \text{ Å}$). Orthorhombic system, space group $P2_12_12_1$, with dimensions cell a = 5.777 (1) Å, b = 11.841 (2) Å, c = 24.868 (5) Å, Z = 4, $D_x = 1.228 \text{ g cm}^{-3}, m = 0.64 \text{ mm}^{-1}, F(000) = 680, M_r$ = 314.4. Refined unit-cell parameters were determined by least-squares treatment of the diffractometer setting angles of 25 reflections (8.27 $< 2\theta < 22.31^{\circ}$), range of 0 < h < 6, 0 < k < 12, 0 < l < 25, with variable scan speed (min 4; max 29.9° min⁻¹). From a total of 1056 independent measurements, those 1005 reflections with $[F > 3.0\sigma(F_0)]$ were retained for the structure analysis and the usual Lorentz and polarization corrections were applied.

The structure was solved by direct methods [18, 19]. 214 parameters were refined by full-matrix least-squares with positional and anisotropic thermal parameters for non-H atoms. Hydrogen atoms were placed in calculated positions. Hydrogens on O-11 and O-12 were located in a difference Fourier map and then refined. All hydrogens with isotropic thermal factor U = 0.06 A. Final R = 7.25 ($R_{\rm w} = 7.69$).

Non-hydrogen atom positional parameters, bonds lengths and angles, anisotropic thermal parameters, hydrogen atom parameters and a list of observed and calculated structure amplitudes have been deposited with the Cambridge Crystallographic Data Centre. Neutral atom scattering factors used in the structure–factor calculations were taken from ref. [20].

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