



# Phytochemistry, Vol. 39, No. 3, pp. 735-736, 1995 Copyright © 1995 Elsevier Science Ltd Printed in Great Britain. All rights reserved 0031-9422/95 \$9.50 + 0.00

## ISOCRYPTOLEPINE FROM CRYPTOLEPIS SANGUINOLENTA

JEAN-LOUIS POUSSET, MARIE-THERESE MARTIN, AKINO JOSSANG and BERNARD BODO\*

Laboratoire de Chimie, URA 401 CNRS, Museum National d'Histoire Naturelle, 63 rue Buffon, 75005 Paris, France

(Received in revised form 27 October 1994)

**Key Word Index**—Cryptolepis sanguinolenta; Asclepiadiaceae; indoloquinoline; alkaloid.

Abstract—A new alkaloid, 5-methyl, 5H-indolo-[3,2-c]-quinoline named isocryptolepine, has been isolated from the roots of Cryptolepis sanguinolenta and its structure determined from spectroscopic data, including IR, MS and NMR.

#### INTRODUCTION

Cryptolepis sanguinolenta (Lindl) Schlechter (Asclepiadiaceae), a shrub indigenous to tropical West Africa, has long been employed in the dyeing of textiles and leather [1]; it is also used in folk medicine as an antimalarial agent [2] and such bioactivities as antibacterial activity or action on vasodilation have been reported [3].

The first alkaloid isolated from C. sanguinolenta in 1951 by Gellert et al. [4] was cryptolepine (5-methyl, 5Hindolo-[3,2-b]-quinoline), previously extracted from C. triangularis. Later, the same alkaloid was re-isolated from C. sanguinolenta by Dwuma-Badu et al. [5], along with quindoline (norcryptolepine) and a non-characterized alkaloid (CSA-3). Recently, the spirononacyclic alkaloid cryptospirolepine was isolated from the same plant [3].

This paper reports the isolation of a new alkaloid 1 (5methyl, 5H-indolo-[3,2-c]-quinoline) which is an isomer of cryptolepine 2, and thus termed isocryptolepine.

## RESULTS AND DISCUSSION

The EtOH extract of the defatted powdered roots of C. sanguinolenta was treated with aqueous HOAc, filtered, alkalinized with NH<sub>4</sub>OH, and extracted repeatedly with CH<sub>2</sub>Cl<sub>2</sub>. The combined CH<sub>2</sub>Cl<sub>2</sub> extracts were chromatographed over Al<sub>2</sub>O<sub>3</sub> eluted with a CH<sub>2</sub>Cl<sub>2</sub>-MeOH gradient. Elution with CH<sub>2</sub>Cl<sub>2</sub> gave a mixture of quindoline and isocryptolepine, and elution with CH2Cl2-MeOH (9:1) afforded cryptolepine. Quindoline and isocryptolepine (1) were separated by preparative silica gel thin-layer chromatography.

The EI mass spectrum of 1 showed the molecular ion at m/z 232 in agreement with the molecular formula  $C_{16}H_{12}N_2$ . The 16 carbon atoms were depicted in the <sup>13</sup>C NMR spectrum and distributed into 15 aromatic carbon atoms (9 CH and 6 quaternary sp<sup>2</sup> carbons) and one methyl group (Table 1). The <sup>1</sup>H NMR spectrum

\*Author to whom correspondence should be addressed.

displayed resonances for 12 protons forming two fourspin systems assigned to two ortho-disubstituted benzene rings and a singlet at  $\delta$ 9.29. These spin systems were the same as those previously described for cryptolepine (2), but the chemical shifts were different, especially the aromatic singlet at  $\delta$ 8.91 in the spectrum of cryptolepine was shifted to lower field.

Analysis of the <sup>1</sup>H-<sup>1</sup>HCOSY, <sup>13</sup>C-<sup>1</sup>HCOSY and HMBC spectra allowed the assignment of one benzene ring in an indole substructure and the second in a quinoline substructure, as in the structure of cryptolepine. However, the assembling of the substructures was different for isocryptolepine, and was suggested to be as shown in structure 1, due to the following long-range correlations observed in the HMBC. The quaternary carbon at  $\delta$ 152.1 (11a) was <sup>3</sup>J correlated to the proton at  $\delta$ 8.77 (H-1); the singlet proton at  $\delta 9.29$  was <sup>3</sup>J correlated to carbon atoms at  $\delta$ 152.1 (C-11a), 137.5 (C-4a), 43.5 (N-Me) and  $^2J$ to that of  $\delta$ 117.6 (C-6a), whereas the carbon atoms at 140.5 (C-6) and 137.5 (C-4a), gave strong correlations with the N-methyl protons. These data agreed with the position of the singlet methine  $\alpha$  to the N-methyl group. The results led to structure 1 for isocryptolepine, which has thus an angular structure isomeric with cryptolepine for which the four rings are linear. Isocryptolepine could be the partially characterized alkaloid CSA-3 [5].

#### EXPERIMENTAL

General. 1H 300.13 MHz and 13C 75.47 MHz NMR spectra were performed on an AC 300 Bruker spectrometer. The HMBC spectrum was optimized for  $J_{C-H}$ 10 Hz. EIMS: obtained with a Nermag Sidar V 3.0 mass spectrometer.

Isolation. The roots of Cryptolepis sanguinolenta (Lind1) Schlechter (60 g) were air-dried, and ground to provide a fine powder. The powder was defatted with hexane for 12 hr and then extracted by percolation with EtOH (500 ml). After evapn of the solvent, the residue was 736 Short Reports

Table 1. <sup>13</sup>C and <sup>1</sup>H NMR data for isocryptolepine (1) (CD<sub>3</sub>OD)

	1	1	НМВС
С	$\delta_{\mathrm{C}}$	$\delta_{\mathbf{H}} \ m \ J(\mathbf{Hz})$	C correlated to H
1	125.4	8.77 dd 8.1, 1.4	3
2	127.5	7.78 ddd 8.4, 8.1, 0.9	4
3	131.7	7.92 ddd 8.4, 8.1, 1.4	1
4	118.8	8.12 brd 8.1	2
4a	137.5	_	1, 3, 6, N-Me
6	140.5	9.29 s	N-Me
6a	117.6	_	6,7
6b	125.7	_	8, 10
7	120.9	8.15 dd 8.1, 1.0	9
8	122.5	7.35 ddd 8.2, 8.2, 0.8.	10
9	127.8	7.53 ddd 8.1, 8.2, 1.0	7
10	117.6	7.81 dd 8.1, 0.8	8
10a	152.0	<del></del>	7, 9
11a	152.1	_	1, 6
11b	121.6	_	2, 4
N-Me	43.5	4.36 s	6

mixed with aq. HOAc (10%, 200 ml), and allowed to stand overnight. The mixt. was filtered, alkalinized with  $NH_4OH$  to pH 10, and extracted by  $CH_2Cl_2$  (2 × 200 ml). The combined CH<sub>2</sub>Cl<sub>2</sub> extracts were washed with H<sub>2</sub>O, dried over Na2SO4, filtered and evapd to afford a dark alkaloidal residue (450 mg). This residue was dissolved in MeOH (2 ml) and chromatographed over Al<sub>2</sub>O<sub>3</sub> (30 g). Elution with hexane, and then CH<sub>2</sub>Cl<sub>2</sub> afforded a mixt. of quindoline and isocryptolepine (40 mg). Elution with CH<sub>2</sub>Cl<sub>2</sub>-MeOH (9:1) gave a violet residue of cryptolepine (400 mg). The mixt. of quindoline and isocryptolepine was further purified by prep. TLC (Kieselgel Merck 60) using toluene-Me<sub>2</sub>CO-NH<sub>4</sub>OH (50:50:2) as eluent yielding isocryptolepine (1, 5 mg, 0.08%, R, 0.45) and quindoline (5 mg,  $R_f$  0.80), which was identified by comparison with an authentic sample.

Isocryptolepine (1).  $C_{16}H_{12}N_2$ : (not crystallized); UV (EtOH)  $\lambda_{max}$  nm (log  $\varepsilon$ ): 238 (4.18), 284 (4.26), 352 sh (3.64); IR (KBr)  $\nu_{max}$  cm<sup>-1</sup>: 1644, 1617, 1460, 1354, 1242, 1123, 762, 715, 617; EIMS (m/z) 232 [M]  $^+$  (100%), 217 (16), 204 (4), 190 (10), 116 (12), 101 (5), 89 (6).

Cryptolepine (2).  $C_{16}H_{12}N_2$ ; mp 167–168°, lit. 166–169° [4]; IR (KBr)  $v_{\text{max}}$  cm<sup>-1</sup>: 1644, 1615, 1466, 1354, 1308, 1242, 1031, 775, 715, 597; EIMS (m/z) 232 [M]<sup>+</sup> (100%), 217 (32), 190 (15), 116 (16), 102 (11), 94 (15), 89 (35). <sup>1</sup>H NMR (CD<sub>3</sub>OD) δ ppm: 8.91 (s, H-11), 8.54 (d, 8.5, H-4), 8.50 (d, 9.1, H-6), 8.34 (dd, 7.8, 1.2, H-1), 8.05 (ddd, 8.5, 7.3, 1.2, H-3), 7.80 (dd, 7.8, 7.3, H-2), 7.74 (dd, 8.4, 1.3, H-9), 7.69 (dd, 8.4, 7.6, H-8), 7.34 (ddd, 9.1, 7.6, 1.3, H-7), 4.95 (s, N-Me).

Acknowledgements—We thank Dr Boye at the Centre of Scientific Research into Plant Medicine, Manpong-Akwapin, Ghana for collecting and identifying the plant material, and Dr J. P. Brouard for the mass spectra.

### REFERENCES

- 1. Saxton, J. E. (1965) in *The Alkaloids, Chemistry and Physiology* (Manske, R. H. F. and Holmes, H. L., eds), Vol. 8, p. 19. Academic Press, New York.
- Boye, G. L. and Ampofo, O. (1983) Proceedings at the First International Symposium on Cryptolepine, Kumasi, Ghana, University of Science and Technology
- Tackie, A. N., Boye, G. L., Sharaf, M. H. M., Schiff Jr, P. L., Crouch, R. C., Spitzer, T. D., Johnson, R. L., Dunn, J., Minick, D. and Martin, G. E. (1993) J. Nat. Prod. 56, 653.
- 4. Gellert, E., Raymond-Hamet and Schlittler, E. (1951) *Helv. Chim. Acta* 34, 642.
- Dwuma-Badu, D., Ayim, J. S. K., Fiagbe, N. I. Y., Knapp, J. E., Schiff Jr, P. L. and Slatkin, D. J. (1978) J. Pharm. Sci. 67, 433.