

THE BIOTRANSFORMATION OF 8-EPICEDROL AND SOME RELATIVES BY
CEPHALOSPORIUM APHIDICOLA

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(Received 19 October 1994)

Key Word Index—*Cephalosporium aphidicola*; microbiological transformation; sesquiterpenoid; 8-epicedrol; hydroxylation.**Abstract**—The microbiological hydroxylation of 8-epicedrol and relatives bearing hydroxyl groups at C-9 and C-15 by *Cephalosporium aphidicola* has been shown to take place predominantly at C-3.

INTRODUCTION

In studies aimed at examining the scope of biosynthetically directed biotransformations by *Cephalosporium aphidicola*, we showed that the sesquiterpenoid cedrol (**1**) was hydroxylated primarily at C-3 [1]. Previous studies on the microbiological hydroxylation of cedrol by *Aspergillus niger* [2] and *Beauveria sulfurescens* [3] had shown that hydroxylation occurred predominantly at C-3, although other organisms [4] were less discriminating in their site of attack. We have now undertaken a further study of the structural scope of the biotransformation by *C. aphidicola* with some relatives of cedrol to assess the directing role of the C-8 hydroxyl group.

RESULTS AND DISCUSSION

The compounds selected for study were 8-epicedrol (**2**), cedran-9 α -ol (**13**), cedran-8 α ,9 α -diol (**15**), cedran-8 α ,15-diol (**20**) and isocedran-15-ol (**22**) which have hydroxyl groups at C-8 or on an adjacent carbon. The substrates were prepared from α - and β -cedrene by literature methods [5].

Incubation of 8-epicedrol (**2**) with *C. aphidicola* for 10 days gave a series of metabolites which were separated by chromatography. The metabolites fell into two series: those retaining the C-8 α hydroxyl group and those in which dehydration had taken place to afford the 8(9)-alkene. The location of the additional oxygen functions at C-3 was established from the changes in the ^{13}C NMR spectra (see Table 1) [6, 7]. The multiplicity of the C-3 proton resonances, when compared to previous work [1] enabled the stereochemistry of the alcohols to be assigned (3 α -ol: δ_{H} 3.70, triplet ($J = 10$ Hz) of doublets ($J = 5.4$ Hz); 3 β -ol: δ_{H} 4.32, doublet ($J = 0.9$ Hz) of doublets ($J = 5.2$ Hz)). In each case the 3 β -alcohol was the

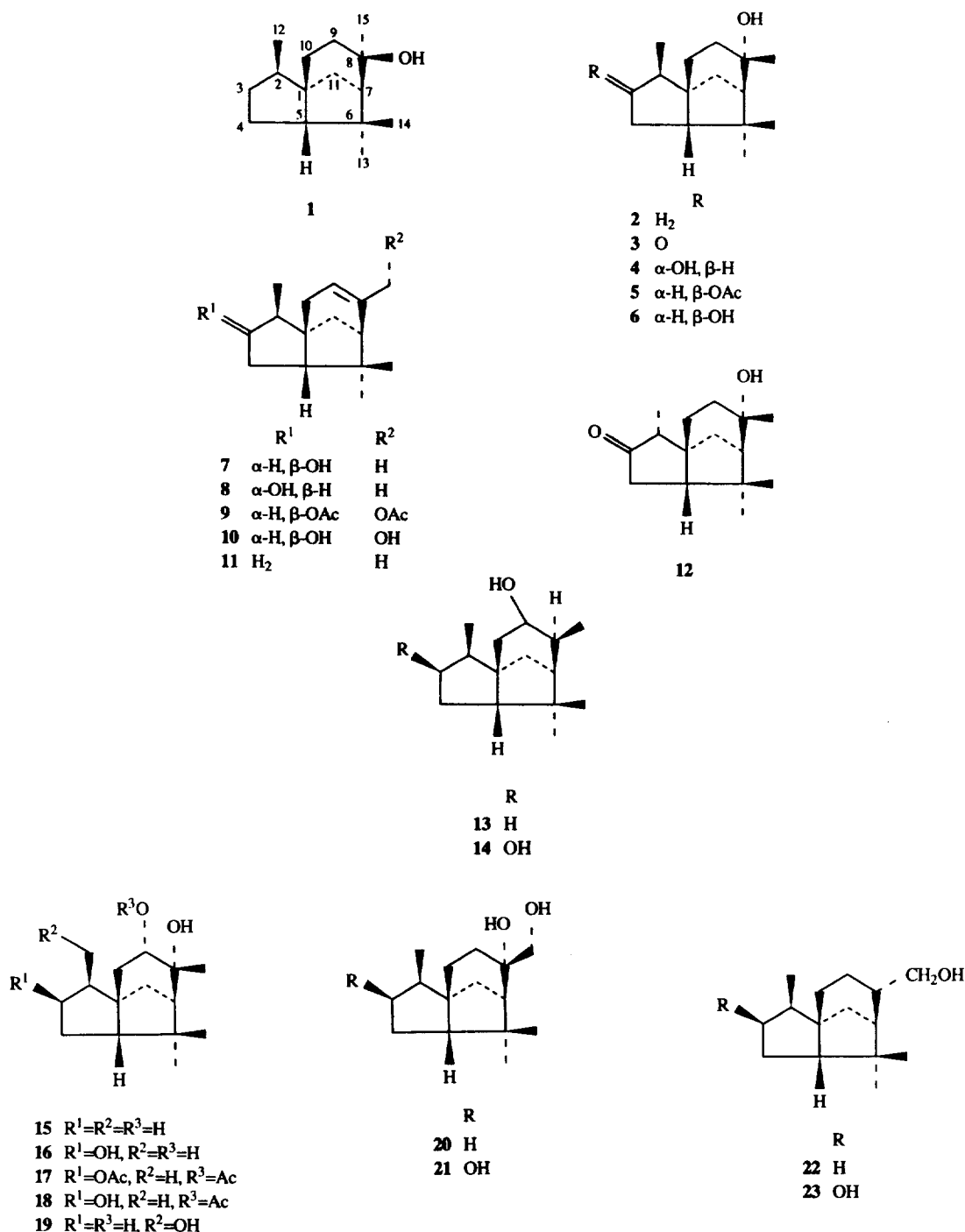
major metabolite. A mixture of cedr-8-en-3 β , 15-diol (**10**) and cedran-3 β ,8 α -diol (**6**) was separated after acetylation and the corresponding alcohols were recovered by hydrolysis. The C-2 methyl group in the 3-ketone underwent epimerization. The epimers were distinguished by the long-range coupling ($J = 1.9$ Hz) shown between the 2 α -proton (δ_{H} 2.05) in **3** and the 4 α -proton (δ_{H} 2.38, *ddd*, $J = 1.9, 12.1$ and 19.5 Hz). The epimer **12** (2 β -H, δ_{H} 2.12) showed no long-range coupling.

In several experiments, cedran-9 α -ol (**13**) was poorly transformed and most of the starting material was recovered. The minor metabolite was identified as cedran-3 β ,9 α -diol (**14**) from the ^{13}C NMR spectrum and the position of the CH(OH) resonance (δ_{H} 4.32, *dd*, $J = 4.3, 8.0$ Hz).

Incubation of cedran-8 α ,9 α -diol (**15**) gave cedran-3 β ,8 α ,9 α -triol (**16**) as the major metabolite accompanied by a minor amount of cedran-8 α ,9 α ,12-triol (**19**). Crude **16** contained some of the 3 α -isomer, as shown by its ^1H NMR spectrum. The crude triol was purified by acetylation to give the 9 α -monoacetate (**18**) and the 3 β ,9 α -diacetate (**17**). Incubation of both cedran-8 α ,15-diol (**20**) and isocedran-15-ol (**22**) with the fungus also gave the 3 β -alcohols (**21** and **23**) as metabolites (6.7% and 3% respectively). These were identified by their ^1H and ^{13}C NMR spectra (see Table 1).

As noted previously [1] the organism does not hydroxylate cedrene (**11**), but requires an oxygen function. In spite of the variations in the position of the directing hydroxyl group, oxidation by *C. aphidicola* had taken place predominantly at the C-3 β position with minor amounts of transformation at the C-3 α position and at the C-12 methyl group. The best yields occurred with a tertiary C-8 alcohol, although these were surprisingly independent of the stereochemistry of the alcohol. A possible explanation, which may accommodate this flexibility, is that the directing interaction between the hydroxyl group in the substrate and the enzyme involves not just

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a simple bonding directly to the enzyme surface but a hydrogen bonding interaction via water molecules held in a lattice adjacent to the peptide chain of the hydroxylase. This could provide the flexibility of a hydrophilic region rather than a more rigid hydrophilic centre.

EXPERIMENTAL

General experimental details have been described previously [8]. *Cephalosporium aphidicola* (IMI 68689) was grown in shake culture as described previously [8].

Incubation of 8-Epicedrol (2). 8-Epicedrol (2 g) in EtOH (25 ml) was evenly distributed among 50 flasks (100 ml medium per flask) of 2-day-old cultures of *C. aphidicola* and the incubation continued for a further 10 days. The metabolites were recovered in EtOAc and separated by chromatography on silica gel. Elution with EtOAc-petrol (3:7) gave sequentially cedr-8-en-3β-ol (7) (95 mg), cedr-8-en-3α-ol (8) (15 mg), 3-oxocedran-8α-ol (3) (80 mg) and 3-oxo-2-epicedran-8α-ol (12) (45 mg). Elution with EtOAc-petrol (3:2) gave a mixture followed by cedran-3α,8α-diol (4) (31 mg). The mixture was acetylated

Table 1. ^{13}C NMR spectroscopic data

Carbon atom	2	3	4	5	6	8	9	10	12	13
1	53.4	52.0	50.1	52.2	51.9	50.1	51.2	52.3	49.4	54.9
2	41.8	48.4	50.6	45.0	46.2	50.9	39.3	45.0	49.3	41.8
3	36.9	220.1	81.4	82.6	79.5	81.3	81.9	79.6	221.9	36.7
4	25.4	33.6	33.3	31.4	33.7	35.3	30.5	33.4	38.0	25.7
5	56.3	52.8	52.5	54.1	53.9	56.3	57.1	57.1	51.2	58.3
6	41.8	42.6	41.1	41.4	41.3	47.1	47.5	47.5	42.3	44.0
7	61.5	58.6	61.4	62.1	61.8	55.2	52.3	50.1	59.5	55.2
8	73.2	72.2	73.1	73.0	73.0	140.5	138.7	143.7	72.5	46.9
9	34.3	34.6	35.3	34.0	33.9	118.4	123.4	120.2	33.2	73.1
10	30.5	33.1	31.1	31.1	31.5	39.3	40.6	39.7	30.3	46.3
11	39.9	37.7	41.0	40.3	40.3	41.3	44.5	40.9	40.4	43.9
12	15.4	8.9	12.3	9.9	9.4	12.7	10.0	9.5	11.7	15.6
13	29.0	28.1	29.7	27.9	27.9	27.1	25.9	25.9	30.1	27.9
14	28.2	30.2	27.8	29.2	29.1	27.1	27.3	27.4	27.9	28.9
15	30.6	30.7	30.7	30.7	30.5	24.7	68.1	67.0	30.7	17.9
OAc				21.1			21.1, 21.2			
				170.8			170.7, 171.0			

Carbon atom	14	15	16	17	18	19	21	22	23
1	52.6	53.4	52.0	51.8	51.9	52.7	52.3	54.8	53.4
2	45.4	41.6	46.2	44.1	45.8	50.4	46.0	42.2	46.7
3	77.3	36.4	78.5	81.8	79.5	32.1	78.3	37.1	80.0
4	33.2	25.6	34.1	31.3	34.1	26.0	30.8	25.6	34.5
5	54.8	60.8	61.0	60.9	61.0	60.9	56.9	57.9	55.4
6	44.2	40.6	41.1	40.6	41.0	41.4	40.8	43.7	43.2
7	54.6	57.2	54.9	54.5	54.8	58.0	53.6	47.4	47.2
8	46.5	74.3	73.9	72.9	73.4	73.8	74.2	51.5	51.8
9	73.2	72.3	71.4	74.3	74.7	71.5	33.3	33.3	34.0
10	45.5	39.0	39.6	36.4	37.4	39.1	29.3	24.7	24.3
11	42.5	41.4	41.4	38.9	39.2	40.0	39.5	47.2	47.5
12	9.1	15.6	10.0	9.6	9.6	62.6	9.3	15.3	9.5
13	26.7	26.8	27.1	26.5	26.9	27.2	27.4	27.1	27.0
14	28.1	28.7	28.6	27.9	28.4	28.7	28.4	28.6	28.8
15	17.2	29.2	29.4	28.9	29.3	29.4	68.9	67.6	67.5
		20.8, 20.8	21.2						
		170.0, 170.4	170.2						

with Ac_2O -pyridine and separated by chromatography on silica gel with $\text{Me}_2\text{CO}-\text{CH}_2\text{Cl}_2$ (1:9) to give 3 β ,15-diacetoxycedr-8-ene (9) (20 mg) and 3 β -acetoxycedran-8 α -ol (5) (150 mg). These were then treated with 10% (w/v) $\text{NaOH}-\text{MeOH}$ at room temp. overnight to afford cedr-8-en-3 β ,15-diol (10) (10 mg) and cedran-3 β ,8 α -diol (6) (110 mg), respectively. The metabolites were characterized as follows.

Cedr-8-en-3 α -ol (7) was identical to material described previously [1].

Cedr-8-en-3 α -ol (8). Oil, MS m/z : 220 $[\text{M}]^+$ ($\text{C}_{15}\text{H}_{24}\text{O}$, 220). IR $\nu_{\text{max}} \text{ cm}^{-1}$: 3450, 1645. NMR δ_{H} (CDCl_3): 0.96 (3H, d , $J = 7$ Hz, H-12), 0.99 (3H, s , H-14), 1.05 (3H, s , H-13), 1.68 (3H, $br s$, H-15), 3.70 (1H, td , $J = 10$ and 5.4 Hz, H-3 β), 5.19 (1H, $br s$, H-9). Oil, 3-Oxocedran-8 α -ol

(3). MS m/z : 236 $[\text{M}]^+$ ($\text{C}_{15}\text{H}_{24}\text{O}_2$, 236). IR $\nu_{\text{max}} \text{ cm}^{-1}$: 3485, 1735. NMR δ_{H} (CDCl_3): 0.93 (3H, d , $J = 7$ Hz, H-12), 0.92 (3H, s , H-14), 1.19 (3H, s , H-13), 1.32 (3H, s , H-15), 2.05 (2H, m , H-2, H-5), 2.18 (1H, dd , $J = 5.6$ and 19.5 Hz, H-4 β), 2.38 (1H, ddd , $J = 1.9$, 12.1 and 19.5 Hz, H-4 α).

3-Oxo-2-epicedran-8 α -ol (12). Oil, MS m/z : 236 $[\text{M}]^+$ ($\text{C}_{15}\text{H}_{24}\text{O}_2$, 236). IR $\nu_{\text{max}} \text{ cm}^{-1}$: 3490, 1735. NMR δ_{H} (CDCl_3): 0.98 (3H, d , $J = 7$ Hz, H-12), 1.02 (3H, s , H-14), 1.21 (3H, s , H-13), 1.35 (3H, s , H-15), 2.12 (1H, q , $J = 7$ Hz, H-2 β), 2.25 (1H, dd , $J = 5.6$, 19.5 Hz, H-4 β), 2.36 (1H, dd , $J = 12.1$, 19.5 Hz, H-4 α).

3 β ,15-Diacetoxycedr-8-ene (9). Oil, IR $\nu_{\text{max}} \text{ cm}^{-1}$: 1745. NMR δ_{H} (CDCl_3): 0.83 (3H, d , $J = 7$ Hz, H-12), 0.98 (3H, s , H-14), 1.04 (3H, s , H-13), 2.04 and 2.07 (each

3H, s, OAc), 4.42 and 4.48 (each 1H, *d*, *J* = 12 Hz, H-15), 5.23 (1H, *dd*, *J* = 4.8 and 9.8 Hz, H-3 α), 5.56 (1H, *br s*, H-9).

3 β ,15-Dihydroxycedran-8-ene (10). Mp 141°. MS *m/z*: 236 [M]⁺ (C₁₅H₂₄O₂, 236). IR ν_{\max} cm⁻¹: 3300, 1650. NMR δ_{H} (CDCl₃): 0.91 (3H, *d*, *J* = 7 Hz, H-12), 0.99 (3H, *s*, H-14), 1.02 (3H, *s*, H-13), 3.96 and 4.05 (1H, *d*, *J* = 13.5 Hz, H-15), 4.35 (1H, *dd*, *J* = 5 and 10 Hz, H-3 α), 5.52 (1H, *br s*, H-9).

3 β -Acetoxycedran-8 α -ol (5). Oil, MS *m/z*: 280 [M]⁺ (C₁₇H₂₈O₃, 280). IR ν_{\max} cm⁻¹: 330, 1740. NMR δ_{H} (CDCl₃): 0.82 (3H, *d*, *J* = 7 Hz, H-12), 1.01 (3H, *s*, H-14), 1.15 (3H, *s*, H-13), 1.32 (3H, *s*, H-15), 2.02 (3H, *s*, OAc), 5.26 (1H, *dd*, *J* = 4 and 8.2 Hz, H-3 α).

Cedran-3 β ,8 α -diol (6). Mp 127°. (Found: C, 75.5; H, 11.1. C₁₅H₂₆O₂ requires C, 75.6; H, 10.9%.) IR ν_{\max} cm⁻¹: 3300. NMR δ_{H} (CDCl₃): 0.89 (3H, *d*, *J* = 7 Hz, H-12), 1.02 (3H, *s*, H-14), 1.16 (3H, *s*, H-13), 1.33 (3H, *s*, H-15), 4.28 (1H, *dd*, *J* = 4 and 8.3 Hz, H-3 α).

Cedran-3 α ,8 α -diol (4). Mp 148°. (Found: C, 75.4; H, 11.0. C₁₅H₂₆O₂ requires C, 75.6; H, 10.9%.) IR ν_{\max} cm⁻¹: 3300. NMR δ_{H} (CDCl₃): 0.97 (3H, *d*, *J* = 7 Hz, H-12), 1.04 (3H, *s*, H-14), 1.17 (3H, *s*, H-13), 1.33 (3H, *s*, H-15), 3.60 (1H, *td*, *J* = 10.6 and 5 Hz, H-3 β).

Incubation of cedran-9 α -ol (13). The alcohol (2 g) was incubated with *C. aphidicola* as above. On chromatography the starting material (1.76 g) was recovered. Further elution gave cedran-3 β ,9 α -diol (**14**) (20 mg), mp 148–150°. (Found: C, 75.4; H, 10.9. C₁₅H₂₆O₂ requires C, 75.6; H, 10.9%.) IR ν_{\max} cm⁻¹: 3300. NMR δ_{H} (CDCl₃): 0.95 (3H, *s*, H-14), 0.98 (3H, *d*, *J* = 7 Hz, H-12), 1.16 (3H, *s*, H-13), 1.23 (3H, *d*, *J* = 7 Hz, H-15), 3.90 (1H, *dd*, *J* = 6.5 and 10 Hz, H-9 β), 4.32 (1H, *dd*, *J* = 4.3 and 8 Hz, H-3 α).

Incubation of cedran-8 α ,9 α -diol (15). The diol (2 g) was incubated with *C. aphidicola* as above and the metabolites were separated by chromatography. Elution with EtOAc–petrol (3:2) gave crude cedran-3 β ,8 α ,9 α -triol (**16**) (1 g) and cedran-8 α ,9 α ,12-triol (**19**) (30 mg). Crude **16** was acetylated with Ac₂O–pyridine at room temp. overnight and the acetates purified by chromatography to give 3 β ,9 α -diacetoxycedran-8 α -ol (**17**) (250 mg) and 9 α -acetoxycedran-3 β ,8 α -diol (**18**) (300 mg). Hydrolysis (10% NaOH–MeOH, room temp., overnight) of the monoacetate (100 mg) gave pure cedran-3 β ,8 α ,9 α -triol (**16**) (60 mg). The metabolites were characterized as follows.

Cedran-3 β ,8 α ,9 α -triol (16). Mp 150°. (Found: C, 68.1; H, 10.6. C₁₅H₂₆O₃·0.5H₂O requires C, 68.4; H, 9.9%.) IR ν_{\max} cm⁻¹: 3400. NMR δ_{H} (CDCl₃): 0.98 (3H, *d*, *J* = 7 Hz, H-12), 1.02 (3H, *s*, H-14), 1.15 (3H, *s*, H-13), 1.45 (3H, *s*, H-15), 3.85 (1H, *dd*, *J* = 6.5, 10.2 Hz, H-9), 4.35 (1H, *dd*, *J* = 4.2 and 8.5 Hz, H-3 α).

3 β ,9 α -Diacetoxycedran-8 α -ol (17). Mp 120–124°. (Found: C, 67.1; H, 8.6. C₁₉H₃₀O₅ requires C, 67.4; H, 8.9%.) IR ν_{\max} cm⁻¹: 3300, 1745. NMR δ_{H} (CDCl₃): 0.70 (3H, *d*, *J* = 7 Hz, H-12), 0.91 (3H, *s*, H-14), 1.09 (3H, *s*, H-13), 1.16 (3H, *s*, H-15), 1.92 and 1.99 (each 3H, *s*, OAc), 4.97 (1H, *dd*, *J* = 6.4 and 10.5 Hz, H-9), 5.13 (1H, *dd*, *J* = 3.8 and 7.8 Hz, H-3).

9 α -Acetoxycedran-3 β ,8 α -diol (18). Mp 132–134°. (Found: C, 68.1; H, 9.3. C₁₇H₂₈O₄ requires C, 68.9; H, 9.5%.) IR ν_{\max} cm⁻¹: 3340, 1745. NMR δ_{H} (CDCl₃): 0.91 (3H, *d*, *J* = 7 Hz, H-12), 1.04 (3H, *s*, H-14), 1.21 (3H, *s*, H-13), 1.29 (3H, *s*, H-15), 2.10 (3H, *s*, OAc), 4.29 (1H, *dd*, *J* = 4.1, 8.2 Hz, H-3), 5.10 (1H, *dd*, *J* = 6.5 and 10.6 Hz, H-9).

Cedran-8 α ,9 α ,12-triol (19). Mp 152°. (Found: C, 69.1; H, 10.7. C₁₅H₂₆O₃·0.5 H₂O requires C, 68.4; H, 9.9%.) IR ν_{\max} cm⁻¹: 3350. NMR δ_{H} (CDCl₃–pyridine): 1.12 (3H, *s*, H-14), 1.24 (3H, *s*, H-13), 1.60 (3H, *s*, H-15), 3.68 (1H, *dd*, *J* = 6.1, 10.3 Hz, H-9), 3.91 and 3.95 (each 1H, *d*, *J* = 7 Hz, H-12).

Incubation of cedran-8 α ,15-diol (20). The diol (1.5 g) was incubated with *C. aphidicola* as above. The metabolites were separated by chromatography to give the starting material and cedran-3 β ,8 α ,15-triol (100 mg), mp 123–126°. (Found: C, 68.7; H, 10.2. C₁₅H₂₆O₃·0.5 H₂O requires C, 68.4; H, 9.9%.) IR ν_{\max} cm⁻¹: 3300. NMR δ_{H} (CDCl₃): 0.77 (3H, *d*, *J* = 7 Hz, H-12), 0.80 (3H, *s*, H-14), 0.94 (3H, *s*, H-13), 3.45 and 3.54 (each 1H, *d*, *J* = 10.8 Hz, H-15), 4.16 (1H, *dd*, *J* = 4 and 8.2 Hz, H-3 α).

Incubation of iso cedran-15-ol (22). The alcohol (1 g) was incubated with *C. aphidicola* as above. The metabolites were separated by chromatography to give cedran-3 β ,15-diol (**23**) (30 mg), mp 111–114°. (Found: C, 75.4; H, 10.9. C₁₅H₂₆O₂ requires C, 75.6; H, 10.9%.) IR ν_{\max} cm⁻¹: 3667. NMR δ_{H} (CDCl₃): 0.92 (3H, *d*, *J* = 7 Hz, H-12), 0.96 (3H, *s*, H-14), 1.11 (3H, *s*, H-13), 3.66 (2H, *d*, *J* = 7 Hz, H-15), 4.29 (1H, *dd*, *J* = 4 and 8 Hz, H-3 α).

Acknowledgements—Part of this work was carried out under the HEJ Institute, University of Karachi: University of Sussex Link Scheme funded by the British Council. We thank Professor Atta-ur-Rahman and Dr D. R. M. Walton for establishing this scheme. We also thank the ERASMUS scheme for providing support for E.G.

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