



## TRITERPENES FROM TRIPTERIGIUM WILFORDII\*

Takashi Morota,† Chun-Xin Yang,‡ Hiroshi Sasaki, Wan-Zhang Qin,‡ Kô Sugama, Kang-Li Miao,‡ Takaaki Yoshino, Li-Hong Xu,‡ Masao Maruno and Bing-Hui Yang‡

Tsumura Central Laboratories, 3586 Yoshiwara, Ami-machi, Inashiki-gun, Ibaraki 300-11, Japan; ‡Zhong Shan Hospital of Shanghai Medical University, 136 Yi-Xue-Yuan Road, Shanghai 200 032, China

(Recieved in revised form 19 December 1994)

**Key Word Index**—*Tripterigium wilfordii*; Celastraceae; wilforols; oleanane; D:B-friedooleanane; triterpenes.

Abstract—The acetone extract of dried root bark of *Tripterigium wilfordii* afforded two novel oleanane and D:B-friedooleanane triterpenes: wilforol C and D. Their structures were established on the basis of chemical and spectroscopic studies.

### INTRODUCTION

In our previous paper, we reported the isolation of D:Afriedo-24-noroleanane triterpenoids from root bark of the Tripterigium wilfordii Hook f. [1]. In continuing studies on the chemical components of root bark of this plant, we have isolated two novel triterpenes, 3α,23-dihydroxyolean-12-en-28-oic acid (1) and  $3\beta$ -hydroxy-D:Bfriedoolean-5-en-29-oic acid (2), which we have named wilforols D and E, along with four known triterpenes [3 $\alpha$ , 24-dihydroxy-olean-12-en-28-oic acid (3); 3 $\beta$ , 29dihydroxy-D: B-friedoolean-5-en (4); 3-hydroxy-D: Afriedoolean-24-al-3-en-2-on-29-oic acid (cangoronine, 5) and 3-hydroxy-D: A-friedoolean-3-en-2-on-29-oic acid (6)]. Compounds 3-6 were identified by comparing their physicochemical and spectral data with those in the literature [2-5]. This paper deals with the structural investigations of the two novel triterpenes.

# RESULTS AND DISCUSSION

An acetone extract of root bark of *T. wilfordii* afforded two novel triterpenes: an oleanane triterpene (wilforol C, 1) and a D:B-friedooleanane triterpene (wilforol D, 2).

Compound 1 was obtained as needles and gave a positive response to the Liebermann-Burchard test for triterpenes. Its molecular formula,  $C_{30}H_{48}O_4$ , was determined by high-resolution (HR) mass spectrometry and was the same as that of 3.

The IR spectrum of 1 was very similar to that of 3, including the adsorption bands due to the hydroxyl (3428 cm<sup>-1</sup>) and carboxyl (1696 cm<sup>-1</sup>) groups. The EI-

mass spectrum of 1 showed the same fragmentation pattern as 3, with important fragments at m/z 472 [M<sup>+</sup>], 248, 203 (248 – COOH), which arose from retro Diels-Alder cleavage around ring C. It was a characteristic fragmentation pattern for an olean-12-ene triterpene [6, 7]. The <sup>1</sup>H and <sup>13</sup>C NMR data of 1 were also very similar to those of 3. Thus, 1 was presumed to be a structural isomer of 3.

The location of the carboxyl group was decided as follows. Among the possible positions (C-28, C-29, C-30) of the carboxyl group suggested by the mass spectral data of 1 described above, C-28 was thought to be the most reasonable because of the good agreement of the carbon signals ascribable to the ring B, C, D and E portions in the <sup>13</sup>C NMR spectra of 1 and 3 (Table 1). Furthermore, the presence of the *gem*-dimethyl group (C-29, C-30) at C-20 was confirmed by the correlation cross-peaks in the HMBC data of 1 (Fig. 1). Thus, the carboxyl group was determined to be at the C-28 position.

The locations of the two hydroxyl groups in 1 were decided as follows. One hydroxyl group was thought to be primary and the other to be secondary because of the signals in the <sup>1</sup>H NMR spectrum ( $\delta$ 3.94 (1H, t-like, J = 2.6 Hz),  $\delta 3.68 \text{ and } 3.88 \text{ (each 1H, } d, J = 10.9 \text{ Hz)}$ ). The location of the secondary hydroxyl group at the C-3 position was highly probable on a biogenetic basis and its  $\alpha$ -orientation was confirmed by the *J*-value of the H-3 methine signal (t, J = 2.6 Hz) in the <sup>1</sup>H NMR spectrum of 1. The position of the primary hydroxyl group was established to be at C-23. The signals ascribable to C-3, C-4, C-5, C-6, C-23 and C-24 in the <sup>13</sup>C NMR spectrum of 1 were different from those of 3 and in particular the hydroxymethyl signal ( $\delta$ 71.3) showed a lower shift  $(5.5 \text{ ppm}, \text{ axial} \rightarrow \text{equatorial})$  and the methyl signal  $(\delta 18.2)$  attached to the C-4 position showed a higher shift (5.4 ppm, equatorial  $\rightarrow$  axial, Table 1). The locations of the C-23 hydroxymethyl and the C-3α-hydroxyl groups

<sup>\*</sup>Part 2 in the series 'Chemical Studies on the Root Bark of T. Wilfordii'. For Part 1 see [1].

<sup>†</sup>Author to whom correspondence should be addressed.

T. Morota et al.

3  $R^1$ =OH,  $R^2$ =CH<sub>2</sub>OH,  $R^3$ =Me

$$(\delta^{1H}: 1.00, d^{13c}: 23.8)$$
  $H_3C_{30}$  29 Me  $(\delta^{1H}: 0.92, d^{13c}: 33.3)$   $(\delta^{13C}: 31.0)$ 

Fig. 1. HMBC data of compound 1.

were confirmed by the physicochemical properties of the monoacetonide (1a) prepared from 1. Thus, the structure of wilforol C was established as  $3\alpha$ -23-dihydroxy-olean-12-en-28-oic acid (1).

Wilforol D (2) was obtained as needles whose molecular formula was determined as  $C_{30}H_{48}O_3$  by HR-mass

spectrometry. Its IR spectrum showed hydroxyl  $(3400 \text{ cm}^{-1})$  and carboxyl  $(1700 \text{ cm}^{-1})$  adsorption. The <sup>1</sup>H NMR spectrum of 2 showed seven tertiary methyl, one oxygen-bearing methine ( $\delta$  3.69, t, J = 2.8 Hz) and one vinyl proton ( $\delta$  5.75, d, J = 5.8 Hz) signals. The EImass spectrum of 2 showed intense peaks at m/z 456 (M<sup>+</sup>), 304, 259, 235, 152, 134, which were consistent with the fragmentation pattern characteristic for D:Bfriedoolean-5-ene pentacyclic triterpenes (Fig. 2) [6]. The mass spectral fragmentations also suggested that the hydroxyl group was located in ring A or B and that the carboxyl group was on ring D or E. In the <sup>1</sup>H NMR spectrum of 2, a methine signal at  $\delta 3.69$  (1H, t, J = 2.8 Hz) showed the presence of the same  $3\beta$ -substituted secondary hydroxyl group in 2 as in 4. It was also supported by the good agreement of the signals due to the ring A portion of 2 and 4 in their <sup>13</sup>C NMR spectra (Table 1). The location of the carboxyl group was pre-

Table 1. <sup>13</sup> C NMR spectral data of triterpenoids from T. wilfordii (125 MHz, in pyridine-d <sub>5</sub>
TMS as int. standard)

c	1	1a*	3	2	4	5†	6
1	33.4	32.5	33.9	18.9	19.0	32.7	34.5
2	26.5	23.0a	26.5	29.1	28.8	193.8	194.7
3	75.7	73.0	70.0	75.5	75.5	149.0	144.4
4	40.7	34.9	43.9	41.0	41.1	125.9	139.9
5	43.6	42.9	50.2	144.0	144.0	54.9	39.5a
6	18.4	17.8	19.1	119.9	119.9	33.3	38.6
7	33.0	32.5	33.8	24.1	23.9	18.9	18.2
8	39.9	39.6	40.0	45.5	48.3	49.5	50.2
9	48.1	47.5	48.2	35.0	35.2	37.1	36.9
10	37.3	36.8	37.5	50.6	50.7	55.5	55.7
11	23.7	23.3°	24.0	34.7	35.0	36.6	33.9
12	122.6	122.7	122.6	36.8	36.3	30.9	36.6
13	144.9	143.6	144.8	39.4	38.0a	39.6	39.7a
14	42.3	41.7	42.2	39.0	39.9ª	39.4	39.4ª
15	28.3	27.7	28.3	29.9	32.7	29.5	29.6
16	23.9	23.5°	23.7	30.5	30.8	30.5	30.5
17	46.7	46.6	46.7	30.6	30.8	30.5	30.5
18	42.0	41.0	42.0	45.0	42.6	44.8	44.8
19	46.5	45.9	46.5	31.0	30.2	30.9	30.9
20	31.0	30.7	31.0	40.7	33.7	40.7	40.7
21	34.2	33.9	34.3	29.4	29.3	29.4	29.4
22	33.3	32.9	33.2	37.0	39.9	37.4	37.4
23	71.3	68.4	23.6	26.2	26.3	10.7	10.6
24	18.2	17.3 <sup>b</sup>	65.8	29.5	29.6	195.8	18.9
25	15.8	15.8	15.9	17.0	16.4	17.2	$18.0^{b}$
26	17.5	17.4 <sup>b</sup>	17.4	16.2	20.4	17.9	18.1 <sup>b</sup>
27	26.2	26.0	26.1	17.6	18.2	16.2	16.4
28	180.2	184.1	180.2	31.9	32.2	32.1	32.1°
29	33.3	33.1	33.2	181.3	73.8	181.4	181.3
30	23.8	23.6	23.8	32.7	27.1	32.3	32.3°
-		0					
		Ī					
		$O - C (CH_3)_2$					
		19.3					
		29.3					
		98.1					

Assignments of 1-4 were based on the 2D NMR data (H-H, C-H, HMBC).

sumed to be at C-28, C-29 or C-30 by the EI-mass spectral data of 2. In the HMBC data of 2, the signal of the methyl at  $\delta 1.42$  showed two significant correlation peaks between that of the carboxyl at  $\delta$  181.3 and that of the quarternary carbon atom at  $\delta$  40.7 (C-20), indicating that the carboxyl group was attached to C-20. Furthermore, in the comparisons of the <sup>13</sup>C NMR spectrum of 2 with those of 5 and 6, which had the same substitution groups at the C-20 and the D/E ring structure, the signals assignable to ring D/E and their substitution groups agreed very closely with those of 5 and 6. Thus, the carboxyl group was determined to be located at C-29. From the results mentioned above, wilforol D was formulated as 3β-hydroxy-D: B-friedoolean-5-en-29-oic acid (2). The structure was further confirmed by the chemical conversion of 4 into 2 as shown in Scheme 1.

Wilforol D (2) has the same skeleton as 4. However, the <sup>1</sup>H and <sup>13</sup>C NMR data of 2 showed some disagreement with those of 4. This was caused by the difference of the C/E ring conformation which resulted from the substitution of a group at C-20 [4, 8].

Compound 3 has been isolated previously from Salvia nicolsoniana (Labiatae) [2], and 4, 5 and 6 have been isolated from various plants belonging to the Celastraceae [3-5]. This is the first report of isolation of these compounds from Tripterigium wilfordii.

## EXPERIMENTAL

General. Mps: uncorr. <sup>1</sup>H NMR: 500 MHz; <sup>13</sup>C NMR: 125 MHz with TMS as int. standard; 2D NMR: 500 MHz in common conditions; EIMS: 70 eV. Silica gel

<sup>\*</sup>Recorded at 100 MHz.

<sup>†</sup>Assignments were different from those in the literature [4].

a-c Assignments may be interchangeable in each column.

1156 T. MOROTA et al.

Fig. 2. Mass spectral fragments of compound 2.

Scheme 1. Chemical conversion of 4 into 2.

column chromatography (CC) was carried out on Kieselgel 60 (70–230 mesh); prep. HPLC: column, CIG Si-10 (silica gel,  $1.5 i.d. \times 30 cm$ ).

Plant material. See [1].

Isolation procedure. Among the frs afforded by the Sephadex LH-20 CC of fr. L3 [1], those which did not show a positive response to FeCl<sub>3</sub> were combined (1.2 g). This material was chromatographed on silica gel CC and eluted with 20% Me<sub>2</sub>CO in benzene to afford four frs (1-4). Fr. 2 was crystallized from EtOH to give 4 (210 mg). Fr. 3 was crystallized from MeOH to give wilforol C (1, 56 mg). The Me<sub>2</sub>CO in n-hexane (2:23) eluate (18.5 g) of silica gel CC of M3 [1] was further chromatographed over silica gel with an increasing amount of EtOAc in *n*-hexane to give 8 frs. Fr. 5 (2.1 g) was crystallized from MeOH to afford 3 (460 mg). Fr. 6 (5.3 g) was further purified with prep. HPLC (EtOAcn-hexane, 1:4) and the rich fraction was crystallized from MeOH to give wilforol D (2, 35 mg). The mother liquor of celastrol [1] gave mixed crystals of two compounds (860 mg) with MeOH. They were separated by silica gel CC and each of the compounds obtained was crystallized from MeOH to give 5 (170 mg) and 6 (520 mg).

Wilforol C (3α, 23-dihydroxy-olean-12-en-28-oic acid, 1). Needles (EtOH), mp > 300°.  $[\alpha]_{2}^{26}$  + 84° (pyridine, c 0.18). IR  $v_{max}^{KBr}$  cm<sup>-1</sup>: 3428, 2940, 1696. EIMS m/z (rel. int.): 472  $[M]^+$  (0.4, calc. for  $C_{30}H_{48}O_4$ ; 472.35526, found: 472.35687), 454 (0.5), 248 (100), 207 (5), 203 (77), 189 (13). <sup>1</sup>H NMR (pyridine- $d_5$ ): δ0.79 (3H, s, H-23), 0.92 (3H, s, H-29), 0.96 (3H, s, H-25), 1.00 (3H, s, H-30), 1.07 (3H, s, H-26), 1.23 (3H, s, H-27), 3.30 (1H, dd, J=13.7 and 4.6 Hz, H-18), 3.68 (1H, d, J=10.9 Hz, H-23), 3.88 (1H, d, J=10.9 Hz, H-23), 3.94 (1H, t, J=2.6 Hz, H-3), 5.60 (1H, t, J=3.7 Hz, H-12). <sup>13</sup>C NMR: see Table 1.

Acetonidation of 1. A soln of 1 (25 mg) in dry  $Me_2CO$  (3 ml) was treated with 2,2-dimethoxypropane (50  $\mu$ l) and a catalytic amount of p-toluene sulphonic acid (p-TsOH), and the mixture was stirred at 20° for 2 hr. After adding  $H_2O$  to the soln, it was neutralized with  $Na_2CO_3$ .

The product was then dissolved in  $H_2O$  and extracted with  $CH_2Cl_2$ . Work-up of the  $CH_2Cl_2$  extract in the usual manner yielded the product, which was purified by silica gel CC with  $Me_2CO$  in n-hexane (2:23) to furnish 1a (13.4 mg):amorphous powder. EIMS m/z (rel. int.): 512  $[M]^+$  (1), 454 (2), 248 (100), 207 (11), 203 (73), 189 (16).  $^1H$  NMR (pyridine- $d_5$ ):  $\delta$ 0.69 (3H, s), 0.76 (3H, s), 0.90 (3H, s), 0.93 (3H, s), 0.94 (3H, s), 1.17 (3H, s), 1.39 (3H, s), 1.40 (3H, s), 2.82 (1H, dd, J = 13.6 and 3.9 Hz), 3.25 (1H, d, J = 12.4 Hz), 5.28 (1H, t, J = 3.4 Hz).  $^{13}C$  NMR: see Table 1.

Wilforol D (3 $\beta$ -hydroxy-D:B-friedoolean-5-en-29-oic acid, **2**). Needles (EtOH), mp 280–282°. [ $\alpha$ ]<sub>D</sub><sup>26</sup> + 47° (pyridine; c 0.53). IR v<sub>max</sub><sup>KBr</sup> cm<sup>-1</sup>: 3400, 3036, 2952, 1700. EIMS m/z (rel. int.): 456 [M]<sup>+</sup> (10, calc. for  $C_{30}H_{48}O_{3}$ : 456.36035; found: 456.36152), 438 (5), 304 (100), 289 (94), 275 (15), 259 (39), 235 (44), 152 (45), 134 (96). <sup>1</sup>H NMR (pyridine- $d_5$ ):  $\delta$  0.98 (3H, s, H-26), 1.00 (3H, s, H-25), 1.07 (3H, s, H-24), 1.11 (3H, s, H-28), 1.25 (3H, s, H-27), 1.39 (3H, s, H-23), 1.42 (3H, s, H-30), 2.37 (1H, dt, J = 14.0 and 3.9 Hz, H-22), 2.54 (1H, d, J = 14.0 Hz, H-21), 2.70 (1H, d, J = 15.2 Hz, H-19), 3.69 (1H, t, J = 2.8 Hz, H-3), 5.75 (1H, d, J = 5.8 Hz, H-6). <sup>13</sup>C NMR: see Table 1.

3β, 29-Dihydroxy-D: B-friedoolean-5-en diacetate (4a). An amorphous powder. IR  $v_{max}^{KBr}$  cm<sup>-1</sup>: 2932, 1736. EIMS m/z (rel. int.): 526 [M]<sup>+</sup> (3), 466 (21), 332 (82), 317 (35), 259 (70). 134 (78). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 0.85 (3H, s), 0.98 (3H, s), 1.04 (3H, s), 1.05 (3H, s), 1.07 (3H, s), 1.13 (3H, s), 1.20 (3H, s), 2.01 (3H, s, Ac), 2.07 (3H, s, Ac), 3.76 (2H, s), 4.69 (1H, dd, J = 3.5 and 2.3 Hz), 5.89 (1H, d, J = 5.9 Hz).

Selective deacetylation of **4a**. Compound **4a** (174 mg) was dissolved in 0.5 M KOH in MeOH and left for 1.5 hr at 22°. Usual work-up gave a residue (172 mg) which was purified with prep. HPLC (Me<sub>2</sub>CO–n-hexane, 1:9) to give 120 mg 29-hydroxy-D:B-friedoolean-5-en-3 $\beta$ -yl (**4b**), needles (MeOH), mp 221–222°. IR  $\nu_{max}^{KBr}$  cm<sup>-1</sup>: 3500, 2932, 1712. EIMS m/z (rel. int.): 484 [M]<sup>+</sup> (3), 424 (11), 290 (82), 275 (43), 259 (100), 134 (56). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 0.85 (3H, s), 1.00 (3H, s), 1.02 (3H, s), 1.04 (3H, s), 1.07 (3H, s), 1.09 (3H, s), 1.21 (3H, s), 2.01 (3H, s, Ac), 3.23 (1H, d, d) = 10.3 Hz), 3.29 (1H, d, d) = 10.3 Hz), 4.69 (1H, dd, d) = 3.6 and 2.4 Hz), 5.56 (1H, d, d) = 6.0 Hz).

Oxidation of 4b. Several drops of 25% CrO<sub>3</sub> (in H<sub>2</sub>SO<sub>4</sub>-H<sub>2</sub>O, 1:9) was added to the Me<sub>2</sub>CO soln of 4b (50 mg) and the mixture stirred for 3 hr at room temp., then iso-PrOH was added. The reaction mixture was partitioned between CHCl<sub>3</sub> and 1% HCl. Usual post-treatment of the CHCl<sub>3</sub> layer gave a residue (65 mg)

which was purified with prep. HPLC (Me<sub>2</sub>CO-*n*-hexane, 1:9) to give 55 mg of D:B-friedoolean-5-en-3-β-yl-29-oic acid (**4c**), needles (MeOH), mp 241–243°. IR  $v_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3256, 2932, 1726. EIMS m/z (rel. int.): 498 [M]<sup>+</sup> (2), 438 (20), 304 (78), 289 (73), 259 (30), 235 (38), 134 (90). <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ0.86 (3H, s), 0.93 (3H, s), 0.97 (3H, s), 1.036 (3H, s), 1.042 (3H, s), 1.08 (3H, s), 1.25 (3H, s), 2.01 (3H, s, Ac), 4.68 (1H, t, J = 3.1 Hz), 5.53 (1H, d, J = 6.0 Hz).

Hydrolysis of 4c. Compound 4c (55 mg) was hydrolysed with 4% KOH (3 ml, in 95% EtOH) under reflux for 1 hr. Usual post-treatment gave a residue (49 mg), which was purified with prep. HPLC (Me<sub>2</sub>CO–n-hexane: 3:17) to give 39 mg 3-β-hydroxy-D:B-friedoolean-5-en-29-oic acid (4d). Its physicochemical and spectral data (mp, mmp, TLC, IR, MS, <sup>1</sup>H and <sup>13</sup>C NMR) were identical to those of 2.

Acknowledgements—The authors are grateful to Mrs Y. Tanaka, Miss Y. Imamura and Miss N. Kanda for 1D and 2D NMR spectra; Mr K. Kano and Mr T. Katsuhara for mass spectra. We also thank the people in Zhon Shan Hospital who helped to make the plant extract. We are indebted to Dr T. Tamaki, Hachiouji Medical Center of Tokyo Medical University, and Dr Y. Komatsu, Mr M. Ando, Dr H. Kawamura, Dr H. Maruyama and other members of Tsumura & Co. for supporting this work.

### REFERENCES

- 1. Morota, T., Yang, C. X., Ogino, T., Qin, W. Z., Katsuhara, T., Xu, L. H., Komatsu, Y., Miao, K. L., Maruno, M. and Yang, B. H. *Phytochemistry* (in press).
- 2. Rogelio, P. M. (1986) J. Nat. Prod. 49, 225.
- 3. Gamini, W. and Kumar, V. (1985) *Phytochemistry* 24, 2369.
- 4. Itokawa, H., Shirota, O., Ikuta, H., Morita, H., Takeya, K. and Iitake, Y. (1991) *Phytochemistry* 30, 3713.
- Sousa, J. R., Jannotti, N. K., Silvia, G. D. F. and Pinheiro, J. A. (1988) Gazz. Chim. Ital. 118, 821.
- Shiojima, K., Arai, Y., Masuda, K., Takase, Y., Ageta, T. and Ageta, H. (1992) Chem. Pharm. Bull 40, 1683.
- Budzikiewics, H., Wilson, J. M. and Djerassi, C. (1963)
  J Am. Chem. Soc. 85, 3688.
- 8. Net, F. R. A. and Sanders, J. K. M. (1983) J. Chem. Soc. Perkin Trans I 181.