

LONG-CHAIN POLYPRENYL ACETATES IN MURRAYA EXOTICA

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Abstract—Polyprenyl acetates with an average number of isoprene residues of eight to 13 were isolated from the leaves of $Murraya\ exotica$, the content being 0.12% of the dry weight. Spectroscopic analysis revealed that all the polyprenyl acetates were long-chain homologues with the following sequence of isoprene residues: ω -trans, two-trans, four to nine cis and cis- α -terminal.

INTRODUCTION

The Rutaceae comprises many genera with important economic and medicinal uses. *Murraya* species have different folk uses especially in India, Australia and South Africa [1, 2]. Roots of *M. exotica* are used as a painkiller, whilst the leaves and bark are used in the treatment of diarrhoea and dysentry [3]. This species is also reported as one of several rutaceous plants used successfully to treat cancer [4].

In previous publications, we have reported the isolation of methoxyflavones [5, 6], coumarins, cycloartenols [7-10] and alkaloids [11, 12]. In continuation of our investigation of the constituents of M. exotica collected from the Botanic Island in Aswan (upper Egypt), the author reports the presence of polyprenyl acetates in the leaves (constituting ca 0.12% of the dry wt) with the structure 1b, where m = 5-10. The arrangement of the two trans (w-trans-trans) and cis residues indicates that the isolated polyprenyl acetates are relatively long-chain homologues of betulaprenols [13]. These polyisoprenoids have been used as starting materials for the synthesis of mammalian dolichols (2, m = 12-19, mainly 14-16) [14-16] which are now recognized as one of the most important materials in the biosynthesis of glycoproteins [17, 18].

In a search for a source of dolichols, many authors [19-22] have investigated the chain-length distribution of polyprenols in several species belonging to the Pinaceae and Ginkgoaceae. These results showed that the polyprenols of the Pinaceae are homologues having an alignment of ω -trans-trans residues followed by poly cis residues.

RESULTS AND DISCUSSION

UV spectral analysis of the isolated compounds showed no characteristic bands above 210 nm. IR spec-

tral data showed the following absorption bands: 1665 (C=C), 1005 (=CH), 840 (C=C-H) and $1715 cm^{-1}$ (CO).

FD-mass spectrometry of the isolated compounds showed a fragmentation pattern characteristic of polyisoprenoids [27], viz. fragments at m/z 135, 202, 271, 339, 407, 543, etc., indicating successive losses of isoprene units from the original compounds. The [M]⁺ of A-F appeared at m/z 604, 672, 740, 808, 876 and 944, corresponding to polyprenyl-8 to -13 acetate of the general formula 1b (m = 5-10). On saponification, the resulting polyprenols gave [M]⁺ of m/z 562, 630, 698, 766, 834 and 902 for A-F, respectively, corresponding to the general formula 1a (m = 5-10).

¹H NMR spectral data of A-F showed allylic methyl signals at δ 1.59, 1.67 and 1.75, assignable to methyl groups of the internal *trans*-isoprene residue and a terminal methyl group *cis*- to the main carbon chain [23], a methyl group of the internal *cis*-isoprene residue and a terminal, methyl group of the α-terminal *cis*-isoprene residue, respectively [23].

¹H NMR spectra of the isolated polyprenyl acetates were similar to those of the polyprenols produced by saponification, except that the =CH and -CH₂OAc protons in the α-terminal residue exhibited signals at $\delta 5.35$ (t, J = 7.3 Hz) and 4.55 (d, J = 7.3 Hz), respectively; these signals were shifted to $\delta 5.44$ and 4.08 after saponification of the isolated polyprenyl acetates. ¹H NMR spectral data can be used for chain-length determination using integration [22, 24] (Table 1). The relative intensities of signals in the isolated compounds were in good agreement with the theoretical one for structure 1b. This indicates that the principal polyprenyl acetates (A-F) were composed of two internal trans residues, two to nine internal cis residues and a cis-α-terminal residue.

The 13 C NMR spectrum is essential for the determination of the internal *cis*- and *trans*-isoprene residue alignment [23]; the α -terminal residue exhibited signals at 32.4

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Table 1. Relative intensities of ¹H NMR signals of isolated compounds

						Chemical shif-	Chemical shift (δ -values) and assignment	nment	
Compound	No. of isoprene units	$[M]^+$	Molecular formula	1.59 Me trans, \to (trans)	1.67 Me cis, æ(cis)	1.75 Me ¤(cis)	4.52 4.58 <u>CH</u> 20H	\$.12 =CH-	5.29 5.33 5.40 =CHCH2OH
A	∞	96	C4,H6gO,	3.13*(3)	4.88*(5)	1.03*(1)	1.98*(2)	7.01*(7)	0.97*(1)
В	6	672	$C_{47}H_{76}O_{2}$	2.94*(3)	6.12*(6)	0.96*(1)	2.02*(2)	7.80*(8)	1.20*(1)
C	10	740	C, H, O,	3.19*(3)	6.79*(7)	0.97*(1)	2.04*(2)	8.73*(9)	1.29*(1)
D	11	808	C4,H42O,	3.24*(3)	7.75*(8)	1.07*(1)	1.98*(2)	9.82*(10)	1.27*(1)
E	12	876	C ₆₂ H ₁₀₀ O ₂	3.17*(3)	8.85*(9)	1.06*(1)	2.07*(2)	10.90*(11)	1.00*(1)
ч	13	944	$C_{67}H_{108}O_2$	2.98*(3)	10.11*(10)	0.95*(1)	2.04*(2)	11.82*(12)	1.20*(1)

Theoretical values in parentheses.

Observed and theoretical values for Me protons are the number of methyl groups

(C-1), 142.4 (C-2), 119.3 (C-3), 61.0 (C-4), 170.6 (Me-CO) and 20.9 (Me-CO). The C-1 methylene carbons exhibited signals around δ 32-40 reflecting the linkage of the cisand trans-isoprene residues where carbon atoms are designed as follows [24]: the signal at δ 39.7 is assigned to C-1 methylene carbons of the trans-isoprene residue in the trans-trans and trans linkage. The signals at δ 32.2 are assigned to C-1 methylene carbons of the cis-isoprene residue in cis-cis and trans-cis linkages, respectively. The absence of signals around δ 40, which is characteristic for a cis-trans linkage, indicates that the trans-isoprene units are incorporated in a ω -trans-trans linkage. The presence of the ω -trans linkage is also confirmed by close inspection of the characteristic C-2 signal of the terminal unit at δ 131.1 (the ω -trans linkage in model compounds showed a signal around δ 131.0-131.3 while the ω -cis-linkage showed a signal around δ 131.5–131.6) [25]. The relative intensities of the C-1 methylene signals of the isolated compounds were in good agreement with the theoretical values obtained according to the general structure 1b (Table 2). This indicates that the ω -terminal, two trans residues, four to nine cis residues and cis-α-terminal are aligned in that order.

Reviewing the current literature, it was found that all the polyprenols isolated from the needles or leaves of Gymnosperms investigated so far are long-chain homologues of the betulaprenol-type (1a, average $m \ge 12$). On the other hand, no such polyprenols have been found in the leaves of Angiosperms in which the dominant polyprenols are relatively short-chain homologues (average $m \le 8$) of the ficaprenol-type in which the internal three trans-residues are aligned ω -trans-trans [22, 26].

The present study is the first report of polyprenyl acetates of the betulaprenol-type, having 8-13 isoprene units in their skeletons, not only in the Rutaceae but also in Angiosperms.

EXPERIMENTAL

General. UV were recorded in MeOH, IR as KBr pellets. ¹H NMR spectra were determined in CDCl₃ at 400 MHz using TMS as int. standard. ¹³C NMR spectra were recorded at 100 MHz (CDCl₃, TMS). MS were recorded at 70 eV. Reverse phase HPLC: UV detector, Lichrosorp RP-18. CC: alumina (Merck). TLC: silica gel G (Merck), cyclohexane–EtOAc (9:1).

Isolation and purification. Air-dried powdered leaves of M. exotica (5 kg) collected in October from the Botanic Island of Aswan (upper Egypt) were extracted with cyclohexane. After concn under red. pres., 10 g of this extract was chromatographed on an alumina column $(4.5 \times 75 \text{ cm} \times 1 \text{ kg})$ using cyclohexane–EtOAc (9:1). The oily fr. $(R_f \ 0.51)$ was chromatographed by HPLC after unsuccessful trials for isolating its components by normal chromatographic methods. Reversed-phase prep. RP-18 column, MeOH–EtOH (3:1) flow rate 3 ml min⁻¹, UV detector at 215 nm.

Table 2. Relative intensities of C-1 methylene carbon signals in the isolated con	apounds.
Chemical shift [δ-values] and assignmen	ıt

Compound	No. of isoprene residues	Chemical shift [δ -values] and assignment			
		32.0	32.20	32.40	39.70
<u> </u>	8	1.02(1)	3.01(3)	0.91(1)	2.10(2)
В	9	1.07(1)	4.40(4)	1.26(1)	2.10(2)
C	10	1.11(1)	5.03(5)	0.93(1)	1.98(2)
D	11	0.94(1)	5.90(6)	1.05(1)	2.08(2)
E	12	1.20(1)	7.01(7)	0.97(1)	1.95(2)
F	13	0.89(1)	7.89(8)	0.95(1)	2.13(2)

Theoretical values in parentheses.

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