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## EFFUSIDES I–V: 9,10-DIHYDROPHENANTHRENE GLUCOSIDES FROM JUNCUS EFFUSUS

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Key Word Index—Juncus effusus; Juncaceae; 9,10-dihydrophenanthrene glucosides; effusides.

Abstract—Five 9,10-dihydrophenanthrene glucosides, named effusides I–V, have been isolated from the methanolic extract of *Juncus effusus*. Structures have been determined on spectroscopic grounds.

## INTRODUCTION

In a chemical investigation of *Juncus effusus*, connected to a study of the allelopathic interactions between freshwater macrophytes and microalgae [1], we have recently reported the isolation of some 9,10-dihydrophenanthrene derivatives [2–4]. In pursuing such a study we now describe the isolation of five glucosides named effusides I–V.

Effusides I and V were identified as the  $12-O-\beta-D$ -glucopyranoside (1) and the 2,12-di- $O-\beta-D$ -glucopyranoside (5) of 1,8-dimethyl-2-hydroxy-5-hydroxymethyl-7-methoxy-9,10-dihydrophenanthrene (6), while effusides II-IV were attributed structures 7-O-(2), 2-O-(3) and  $12-O-\beta-D$ -glucopyranosyl-1,8-dimethyl-2,7-dihydroxy-5-hydroxymethyl-9,10-dihydrophenanthrene (4).

The less polar effuside I (1) had a molecular formula, C<sub>24</sub>H<sub>30</sub>O<sub>8</sub>, according to the presence of 24-carbon signals in the <sup>13</sup>CNMR spectrum (Table 1) and a quasimolecular ion at m/z 469 in the FAB mass spectrum. The <sup>1</sup>H NMR spectrum (Table 2) showed two aromatic ortho coupled doublets at  $\delta$  6.76 and 7.38, an aromatic singlet at  $\delta$  7.14, two AB doublets at  $\delta$  4.68 and 4.91, a methoxyl methyl at  $\delta$  3.85, four benzylic protons as a multiplet at  $\delta$  2.62 and two methyl singlets at  $\delta$  2.17 and 2.18, beside an anomeric proton at  $\delta$ 4.32 and further signals of a saccharide moiety. The signal at  $\delta$  6.76, attributed to the H-3 proton and correlated to the carbon at  $\delta$  111.9 in the H-C one-bond COSY, gave cross peaks in the H-C long-range COSY with the signals at  $\delta$  121.4 and 128.5 which were attributed to the C-1 and C-4a carbons. Accordingly, both these carbons were correlated to the H-10 benzylic protons at  $\delta$  2.62 while the C-1 gave an additional cross peak with the H-11 methyl protons at  $\delta$  2.18. The signal at  $\delta$  7.38, linked to the carbon at  $\delta$  126.4 and attributed to the H-4 proton, was correlated to the Enzymatic hydrolysis of 1 gave D-glucose, identified by GC analysis [5], and aglycone 6. The coupling of the H-1 proton of glucose agreed with a  $\beta$ -configuration at the anomeric carbon and the differences in chemical shift and multiplicity of the H-12 protons in 1 and 6 justified the location of the saccharide residue at the C-12 position. The nOe interactions in a NOESY experiment of the anomeric proton with the H-12 methylene confirmed the structure.

Effuside II (2) had 23 carbon signals in the  $^{13}$ C NMR spectrum and a quasimolecular peak at m/z 455 in the FAB mass spectrum for the molecular formula  $C_{23}H_{28}O_8$ . Enzymatic hydrolysis gave aglycone 7 and D-glucose, and the coupling of the anomeric proton indicated a  $\beta$  configuration of the sugar. The NMR spectra of 2 were lacking in the methoxyl methyl signals and a comparison with those of 1 evidenced a significant upfield shift of the H-6 proton and a downfield shift of the C-6 carbon. These data agreed well with the presence of a hydroxyl group rather than a methoxyl one at C-7.

Effuside III (3) had the same molecular formula  $C_{23}H_{28}O_8$  of 2 and by enzymatic hydrolysis gave 7 and D-glucose. It showed in the <sup>1</sup>H NMR spectrum the H-12 protons as a sharp singlet shifted upfield at  $\delta$  4.48 and the anomeric proton shifted downfield at  $\delta$  4.82. These elements suggested that the saccharide moiety was linked to

OH-bearing C-2 carbon at  $\delta$  154.0, and to the signals at  $\delta$  137.9 and 125.0. These signals, also correlated to the benzylic signal at  $\delta$  2.62, were attributed to the C-1a and C-5a carbons, respectively, owing to the heterocorrelations of the latter with the H-12 protons at  $\delta$  4.68 and 4.91 and with the H-6 proton at  $\delta$  7.14, linked to the carbon at  $\delta$  111.0. This latter proton, together with the methyl protons at  $\delta$  2.17, gave cross peaks with the C-8 carbon at  $\delta$  119.8. The chemical shifts of the C-6 and C-8 carbons agreed with the presence of the methoxyl group at C-7 and in a NOESY spectrum the methyl of this group at  $\delta$  3.85 gave nOe interaction with the H-6 proton.

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Table 1. 13C NMR chemical shifts of effusides I-V

1   121.4   120.9   122.5   123.0   123.1     2   154.0   153.7   153.9   154.0   154.3     3   111.9   111.9   111.7   111.9   112.0     4   126.4   126.2   126.3   125.9   126.5     4a   128.5   128.0   129.1   128.5   127.8     1a   137.9   138.2   137.7   138.2   138.3     5   130.9   131.1   135.7   135.8   131.6     6   111.0   115.5   115.1   114.3   111.0     7   155.0   156.2   153.6   153.5   155.4     8   119.8   120.2   120.0   120.8   121.5     8a   138.3   138.2   138.5   138.0   138.3     5a   125.0   127.1   125.4   126.9   127.9     9   25.0   24.9   25.0   25.1   25.0     10   25.9   26.0   26.0   25.9   25.9     11   11.5   11.6 <th>С</th> <th>1</th> <th>2</th> <th>3</th> <th>4</th> <th>5</th>	С	1	2	3	4	5
3   111.9   111.9   111.7   111.9   112.0     4   126.4   126.2   126.3   125.9   126.5     4a   128.5   128.0   129.1   128.5   127.8     1a   137.9   138.2   137.7   138.2   138.3     5   130.9   131.1   135.7   135.8   131.6     6   111.0   115.5   115.1   114.3   111.0     7   155.0   156.2   153.6   153.5   155.4     8   119.8   120.2   120.0   120.8   121.5     8a   138.3   138.2   138.5   138.0   138.3     5a   125.0   127.1   125.4   126.9   127.9     9   25.0   24.9   25.0   25.1   25.0     10   25.9   26.0   26.0   25.9   25.9     11   11.5   11.6   11.6   11.5   11.8     12   68.9   69.4   62.2   62.0   68.7     Me   11.5   11.6	1	121.4	120.9	122.5	123.0	123.1
4   126.4   126.2   126.3   125.9   126.5     4a   128.5   128.0   129.1   128.5   127.8     1a   137.9   138.2   137.7   138.2   138.3     5   130.9   131.1   135.7   135.8   131.6     6   111.0   115.5   115.1   114.3   111.0     7   155.0   156.2   153.6   153.5   155.4     8   119.8   120.2   120.0   120.8   121.5     8a   138.3   138.2   138.5   138.0   138.3     5a   125.0   127.1   125.4   126.9   127.9     9   25.0   24.9   25.0   25.1   25.0     10   25.9   26.0   26.0   25.9   25.9     11   11.5   11.6   11.6   11.5   11.8     12   68.9   69.4   62.2   62.0   68.7     Me   11.5   11.6   11.8   11.8   11.5     OMe   55.4   55.5		154.0	153.7	153.9	154.0	154.3
4a     128.5     128.0     129.1     128.5     127.8       1a     137.9     138.2     137.7     138.2     138.3       5     130.9     131.1     135.7     135.8     131.6       6     111.0     115.5     115.1     114.3     111.0       7     155.0     156.2     153.6     153.5     155.4       8     119.8     120.2     120.0     120.8     121.5       8a     138.3     138.2     138.5     138.0     138.3       5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5 <td>3</td> <td>111.9</td> <td>111.9</td> <td>111.7</td> <td>111.9</td> <td>112.0</td>	3	111.9	111.9	111.7	111.9	112.0
1a     137.9     138.2     137.7     138.2     138.3       5     130.9     131.1     135.7     135.8     131.6       6     111.0     115.5     115.1     114.3     111.0       7     155.0     156.2     153.6     153.5     155.4       8     119.8     120.2     120.0     120.8     121.5       8a     138.3     138.2     138.5     138.0     138.3       5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5     55.5       Glc-1     101.6     101.1     101.6 <td>4</td> <td>126.4</td> <td>126.2</td> <td>126.3</td> <td>125.9</td> <td>126.5</td>	4	126.4	126.2	126.3	125.9	126.5
5     130.9     131.1     135.7     135.8     131.6       6     111.0     115.5     115.1     114.3     111.0       7     155.0     156.2     153.6     153.5     155.4       8     119.8     120.2     120.0     120.8     121.5       8a     138.3     138.2     138.5     138.0     138.3       5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4	4a	128.5	128.0	129.1	128.5	127.8
6     111.0     115.5     115.1     114.3     111.0       7     155.0     156.2     153.6     153.5     155.4       8     119.8     120.2     120.0     120.8     121.5       8a     138.3     138.2     138.5     138.0     138.3       5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     76.7	1a	137.9	138.2	137.7	138.2	138.3
7     155.0     156.2     153.6     153.5     155.4       8     119.8     120.2     120.0     120.8     121.5       8a     138.3     138.2     138.5     138.0     138.3       5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8 <td< td=""><td>5</td><td>130.9</td><td>131.1</td><td>135.7</td><td>135.8</td><td>131.6</td></td<>	5	130.9	131.1	135.7	135.8	131.6
8   119.8   120.2   120.0   120.8   121.5     8a   138.3   138.2   138.5   138.0   138.3     5a   125.0   127.1   125.4   126.9   127.9     9   25.0   24.9   25.0   25.1   25.0     10   25.9   26.0   26.0   25.9   25.9     11   11.5   11.6   11.6   11.5   11.8     12   68.9   69.4   62.2   62.0   68.7     Me   11.5   11.6   11.8   11.8   11.5     OMe   55.4   55.5     Glc-1   101.6   101.1   101.6   101.3   101.6     2   73.5   73.5   73.4   73.4   73.5     3   76.8   76.8   77.0   77.0   76.9     4   70.1   70.1   69.7   69.8   69.7     5   76.6   76.8   76.7   76.6   76.6     6   61.1   61.1   60.8   60.8   60.7     Gle	6	111.0	115.5	115.1	114.3	111.0
8a     138.3     138.2     138.5     138.0     138.3       5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8     69.7       5     76.6     76.8     76.7     76.6     76.6       6     61.1     61.1     60.8     60.8     60.7	7	155.0	156.2	153.6	153.5	155.4
5a     125.0     127.1     125.4     126.9     127.9       9     25.0     24.9     25.0     25.1     25.0       10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8     69.7       5     76.6     76.8     76.7     76.6     76.6       6     61.1     61.1     60.8     60.8     60.7       Gle-1'     2'     73.4     73.4     73.4	8	119.8	120.2	120.0	120.8	121.5
9 25.0 24.9 25.0 25.1 25.0 10 25.9 26.0 26.0 25.9 25.9 11 11.5 11.6 11.6 11.5 11.8 12 68.9 69.4 62.2 62.0 68.7 Me 11.5 11.6 11.8 11.8 11.5 OMe 55.4 55.5 Glc-1 101.6 101.1 101.6 101.3 101.6 2 73.5 73.5 73.4 73.4 73.5 3 76.8 76.8 76.8 77.0 77.0 76.9 4 70.1 70.1 69.7 69.8 69.7 5 76.6 76.6 6 61.1 61.1 60.8 60.8 60.7 Glc-1' 2' 73.4 3' 76.9 4' 70.0 5' 76.6	8a	138.3	138.2	138.5	138.0	138.3
10     25.9     26.0     26.0     25.9     25.9       11     11.5     11.6     11.6     11.5     11.8       12     68.9     69.4     62.2     62.0     68.7       Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8     69.7       5     76.6     76.8     76.7     76.6     76.6       6     61.1     61.1     60.8     60.8     60.7       Gle-1'     2'     73.4     73.4     73.4       3'     76.9     76.9     76.9     76.9     76.9       4'     70.0     76.9     76.9     70.0     76.9       <	5a	125.0	127.1	125.4	126.9	127.9
11   11.5   11.6   11.6   11.5   11.8     12   68.9   69.4   62.2   62.0   68.7     Me   11.5   11.6   11.8   11.8   11.5     OMe   55.4   55.5   55.5     Glc-1   101.6   101.1   101.6   101.3   101.6     2   73.5   73.5   73.4   73.4   73.5     3   76.8   76.8   77.0   77.0   76.9     4   70.1   70.1   69.7   69.8   69.7     5   76.6   76.8   76.7   76.6   76.6     6   61.1   61.1   60.8   60.8   60.7     Glc-1'   2'   73.4   73.4   73.4     3'   76.9   76.9   76.9   70.0   70.0     5'   76.6   76.6   76.6   76.6   76.6	9	25.0	24.9	25.0	25.1	25.0
12   68.9   69.4   62.2   62.0   68.7     Me   11.5   11.6   11.8   11.8   11.5     OMe   55.4   55.5     Glc-1   101.6   101.1   101.6   101.3   101.6     2   73.5   73.5   73.4   73.4   73.5     3   76.8   76.8   77.0   77.0   76.9     4   70.1   70.1   69.7   69.8   69.7     5   76.6   76.8   76.7   76.6   76.6     6   61.1   61.1   60.8   60.8   60.7     Glc-1'   2'   73.4     3'   76.9   70.0     4'   70.0   76.6     5'   76.6   76.6	10	25.9	26.0	26.0	25.9	25.9
Me     11.5     11.6     11.8     11.8     11.5       OMe     55.4     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8     69.7       5     76.6     76.8     76.7     76.6     76.6       6     61.1     61.1     60.8     60.8     60.7       Gle-1'     2'     73.4       3'     76.9     76.9       4'     70.0     76.6       5'     76.6     76.6	11	11.5	11.6	11.6	11.5	11.8
OMe     55.4     55.5       Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8     69.7       5     76.6     76.8     76.7     76.6     76.6       6     61.1     61.1     60.8     60.8     60.7       Gle-1'     2'     73.4       3'     76.9     76.9       4'     70.0     76.6       5'     76.6     76.6	12	68.9	69.4	62.2	62.0	68.7
Glc-1     101.6     101.1     101.6     101.3     101.6       2     73.5     73.5     73.4     73.4     73.5       3     76.8     76.8     77.0     77.0     76.9       4     70.1     70.1     69.7     69.8     69.7       5     76.6     76.8     76.7     76.6     76.6       6     61.1     61.1     60.8     60.8     60.7       Glc-1'     101.2     73.4     76.9     76.9       4'     70.0     76.6     76.6     76.6	Me	11.5	11.6	11.8	11.8	11.5
2 73.5 73.5 73.4 73.4 73.5 3 76.8 76.8 77.0 77.0 76.9 4 70.1 70.1 69.7 69.8 69.7 5 76.6 76.8 76.7 76.6 76.6 6 61.1 61.1 60.8 60.8 60.7 Glc-1' 101.2 2' 73.4 3' 76.9 4' 70.0 5' 76.6	OMe	55.4				55.5
3 76.8 76.8 77.0 77.0 76.9 4 70.1 70.1 69.7 69.8 69.7 5 76.6 76.8 76.7 76.6 76.6 6 61.1 61.1 60.8 60.8 60.7 Glc-1' 101.2 2' 73.4 3' 76.9 4' 70.0 5' 76.8 77.0 77.0 76.9 77.0 76.9	Glc-1	101.6	101.1	101.6	101.3	101.6
4 70.1 70.1 69.7 69.8 69.7 5 76.6 76.8 76.8 76.7 76.6 76.6 6 61.1 61.1 60.8 60.8 60.7 Glc-1' 101.2 2' 73.4 3' 76.9 4' 70.0 5' 76.6		73.5	73.5	73.4	73.4	73.5
5 76.6 76.8 76.7 76.6 76.6   6 61.1 61.1 60.8 60.8 60.7   Glc-1' 101.2   2' 73.4   3' 76.9   4' 70.0   5' 76.6	3	76.8	76.8	77.0	77.0	76.9
6 61.1 61.1 60.8 60.8 60.7 Glc-1' 101.2 2 73.4 3' 76.9 4' 70.0 5' 76.6		70.1	70.1	69.7	69.8	69.7
Glc-1' 101.2   2' 73.4   3' 76.9   4' 70.0   5' 76.6	5	76.6	76.8	76.7	76.6	76.6
2' 73.4 3' 76.9 4' 70.0 5' 76.6	6	61.1	61.1	60.8	60.8	60.7
3' 76.9 4' 70.0 5' 76.6	Glc-1'					101.2
4' 70.0 5' 76.6	2'					73.4
5' 76.6						76.9
						70.0
6′ 61.0						76.6
	6′					61.0

an aromatic hydroxyl group and the nOe interaction between the anomeric proton and the aromatic singlet at  $\delta$  7.18 indicated the C-7 position.

Also effuside IV (4) had the same molecular formula and afforded by hydrolysis 7 and D-glucose. The H-3 proton was shifted downfield at  $\delta$ 6.97 and gave nOe interaction with the anomeric proton according to the presence of the saccharide moiety at C-2.

Effuside V (5), the most polar compound, had a molecular formula C<sub>30</sub>H<sub>40</sub>O<sub>13</sub>, on the basis of the FAB-MS and <sup>13</sup>C NMR data, and gave by hydrolysis 6 and D-glucose. It showed in the <sup>1</sup>H NMR spectrum the H-12

protons as two AB doublets at  $\delta$  4.70 and 4.88, the H-3 doublet at  $\delta$  7.00 and the H-6 singlet at  $\delta$  7.16. The comparison of these chemical shifts with those of the other effusides suggested that D-glucose was at C-2 and C-12 while the methoxyl group was at C-7. Accordingly in the NOESY experiment the anomeric proton at  $\delta$  4.29 was correlated to the H-12 protons, the anomeric proton at  $\delta$  4.84 gave interaction with the H-3 proton and the methyl at  $\delta$  3.82 was correlated to the H-6 proton.

## **EXPERIMENTAL**

NMR spectra were recorded at 400 MHz for  $^1H$  and 100 MHz for  $^{13}C$  on a Bruker AC 400 spectrometer in DMSO- $d_6$  (aglycones in acetone- $d_6$ ). One bond and long-range H-C COSY experiments were performed with the XHCORR microprogramme using delays corresponding to  $J_{C,H}160$  Hz and 8 Hz, respectively. EI mass spectra were obtained with a Kratos MS 50 apparatus and FAB mass spectra with a VG ZAB 2SE apparatus. DCCC was run with a mixture CHCl<sub>3</sub>-MeOH- $H_2O$  (13:7:4 for 1-4, 26:14:5 for 5) using the more polar upper layer as the mobile phase. Reverse-phase HPLC was performed using LiChrosorb RP8 column (MeOH- $H_2O$ , 1:1) for 1-4 and LiChrosorb NH<sub>2</sub> (CH<sub>3</sub>CN- $H_2O$ , 4:1) for 5.

Isolation of effusides. Juncus effusus, collected in the summer near Naples was air dried and extracted with Et<sub>2</sub>O and then with MeOH. The MeOH extract (350 g) after removal of the solvent was distributed between EtoAc and H<sub>2</sub>O.

The organic layer was chromatographed on silica gel and the fractions eluted with CHCl<sub>3</sub>-MeOH (9:1) were rechromatographed on Sephadex LH-20 eluting with MeOH-H<sub>2</sub>O (3:1). DCCC chromatography and HPLC chromatography of fractions 130-143 gave pure 1 (10 mg), 2 (8 mg), 3 (11 mg) and 4 (6 mg). The aq. layer was chromatographed on Amberlite and the MeOH fraction was distributed between *n*-BuOH and H<sub>2</sub>O. DCCC and HPLC processes on the aq. layer gave pure 5 (11 mg).

Enzymatic hydrolysis of glucosides. Pure effuside (3 mg) in  $H_2O$  (0.5 ml) was treated with  $\beta$ -glucosidase (1 mg, Sigma) at 37° for 12 hr. The reaction mixture was extracted with EtOAc:the organic layer gave aglycone 6 (7) while HPLC chromatography (CH<sub>3</sub>CN-H<sub>2</sub>O, 4:1) of the

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Table 2. <sup>1</sup>H NMR chemical shifts of effusides I-V

Н	1	2	3	4	5
3	6.76 d	6.68 d	6.71 d	6.97 d	7.00 d
	(8.4)	(8.3)	(8.3)	(8.8)	(8.7)
4	7.38 d	7.31 d	7.33 d	7.37 d	7.43 d
	(8.4)	(8.3)	(8.3)	(8.8)	(8.7)
6	7.14 s	6.88 s	7.18 s	6.94 s	7.16 s
					2.54 m
9	2.62 m	2.62 m	2.62 m	2.62 m	2.63 m
					2.54 m
10	2.62 m	2.62 m	2.62 m	2.62 m	2.63 m
11	2.18 s	2.14 s	2.19 s	2.21 s	2.23 s
	4.68 d	4.45 d			4.70 d
	(11.0)	(10.7)	4.48 s	4.47 s	(11.0)
12	4.91 d	4.80 d			4.88 d
	(11.0)	(10.7)			(11.0)
Me	2.17 s	2.12 s	2.13 s	2.11 s	2.15 s
OMe	3.85 s				3.82 s
Glc-1	4.32 d	4.30 d	4.82 d	4.79 d	4.29 d
	(7.7)	(7.8)	(7.3)	(7.3)	(7.6)
2	3.12*	3.11*	3.26*	3.28*	3.05*
3	3.19*	3.20*	3.25*	3.25*	3.19*
4	3.18*	3.18*	3.10*	3.12*	3.15*
5	3.20*	3.20*	3.16*	3.15*	3.28*
	3.52 dd	3.51 dd	3.46 dd	3.44 dd	
	(5.7 11.3)	(5.6 11.2)	(5.5 11.2)	(5.6 11.1)	3.45*
6	3.63 dd	3.71 dd	3.69 dd	3.71 dd	3.69*
	(1.6 11.3)	(1.7 11.2)	(1.5 11.2)	(1.5 11.1)	5.07
Glc-1'					4.84 d
					(7.1)
2'					3.28*
3′					3.26*
4′					3.08*
5′					3.18*
					3.45*
6′					3.69*

<sup>\*</sup>Overlapping signals.

aq. layer gave D-glucose. The D-configuration was determined by treatment with L-cysteine methylester hydrochloride and TMS-imidazole [5].

Aglycone 6. <sup>1</sup>H NMR  $\delta$  6.75 (d, 1H, J = 8.8 Hz, H-3), 7.43 (d, 1H, J = 8.8 Hz, H-4), 7.07 (s, 1H, H-6), 2.64 (s, 4H, H-9 and H-10), 2.21 (s, 3H, H-11), 4.69 (s, 2H, H-12), 3.83 (s, 3H, OMe), 2.16 (s, 3H, Me). Aglycone 7: <sup>1</sup>H NMR  $\delta$  6.75 (d, 1H, J = 8.7 Hz, H-3), 7.39 (d, 1H, J = 8.7 Hz, H-4), 6.87 (s, 1H, H-6), 2.62 (s, 4H, H-9 and H-10), 2.18 (s, 3H, H-11), 4.52 (s, 2H, H-12), 2.15 (s, 3H, Me).

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