



## NEO-CLERODANE DITERPENOIDS FROM ROOTS OF *LINARIA SAXATILIS* VAR. *SAXATILIS*

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(Received 10 February 1995)

**Key Word Index**—*Linaria saxatilis* var. *saxatilis*; Scrophulariaceae; diterpenoids; neo-clerodanes.

**Abstract**—Two new and four known neo-clerodane diterpenoids were isolated from an *n*-hexane extract of the roots of *Linaria saxatilis* var. *saxatilis*. The structures of the new compounds were established from the spectral data as 15,16-diacetoxy-15,16-epoxy-neo-cleroda-3,12Z-diene, and 15,16-diacetoxy-12,13-15,16-diepoxy-14-hydroxy-neo-clerod-3-ene.

### INTRODUCTION

The genus *Linaria* belongs to the family Scrophulariaceae (Subfam. Scrophularioideae, tribe Antirrhinae) and several species have been used in traditional medicine for the treatment of vascular disorders (*L. vulgaris*) [1], and for their tonic and antiscorbutic (*L. cymbalaria*) [2], laxative (*L. japonica*) [3] and diuretic (*L. cymbalaria* and *L. japonica*) [2, 3] effects.

For several years, our group has been studying the chemical composition of varieties of *Linaria saxatilis*. Continuing on the same line, we have addressed the chemical composition of the roots of *L. saxatilis* var. *saxatilis*, a plant native to the north and centre of the Iberian Peninsula, on which we have already published studies concerning the chemical composition of the aerial parts [4, 5], which mainly contain unsaturated neo-clerodanes at position 4(18). The chemical composition of the var. *glutinosa* [6–8] has also been studied and its roots contain six new clerodanes unsaturated at position C-3. In the present work, we report the results of study of the chemical composition of an *n*-hexane extract of the roots of the var. *saxatilis*.

### RESULTS AND DISCUSSION

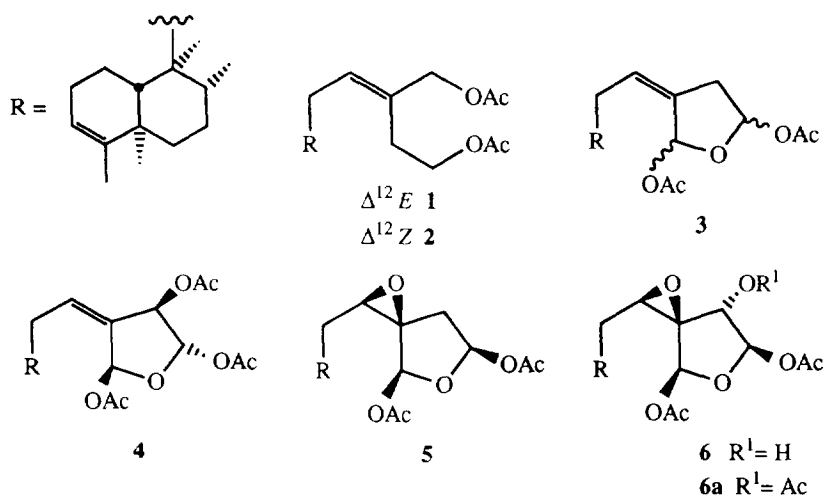
From an acetone-dewaxed *n*-hexane extract of the roots of *L. saxatilis*, six compounds of neo-clerodane structure and the triterpene friedelin were isolated and identified. The neo-clerodane nature of the isolated diterpenes and the presence of an unsaturation at C-3 were established from comparison of their spectroscopic data with those for clerodanes previously isolated from the

roots of the var. *glutinosa*. Compounds 1, 2, 4 and 5 are common to the roots of both varieties, while 3 and 6 were only isolated from the var. *saxatilis* and are described as new natural products.

The structure of 3 was established by comparison of the  $^{13}\text{C}$  and  $^1\text{H}$  NMR data with those for 15,16-diacetoxy-15,16-epoxy-neo-cleroda-4(18),12Z-diene (7) [4]. The signals assignable to the side chain almost reproduce those of the compound, whose 12Z-stereochemistry was established by NOE experiments. For the signals assignable to the bicyclic moiety, the main differences are the absence of an olefinic methylene and the presence of a methine both in  $^1\text{H}$  and  $^{13}\text{C}$  NMR ( $\delta$ 5.10 and 120.5) indicating the existence of a double bond between C-3–C-4 instead of the  $\Delta$ 4(18)-double bond of 7. The structure proposed for 3 is that of 15,16-diacetoxy-15,16-epoxy-neo-cleroda-3,12Z-diene, which is isolated with a small proportion of its epimer at C-15, similar to the case for 7.

The structure of compound 6 was assigned by comparison with that of 15,16-diacetoxy-15,16-epoxy-neo-clerod-3-ene (5), which was also obtained from the var. *glutinosa*, and its structure was unequivocally established by H C heteronuclear correlations and some NOE difference experiments [6]. The only observable difference between 5 and 6 lies in the presence in 6 of an additional secondary alcohol function (IR:  $3510\text{ cm}^{-1}$ ;  $^1\text{H}$  NMR:  $\delta$ 4.25 and  $^{13}\text{C}$  NMR:  $\delta$ 71.4); this was confirmed by acetylation which yielded the acetate 6a. The additional hydroxyl function was located at C-14 because the geminal proton to the acetate at C-15 was a doublet of 2 Hz, which also indicated the *trans*-configuration between both functions. The absolute configuration of the side chain could correspond to that shown in the figure or to its enantiomers

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## EXPERIMENTAL

**General.** Optical rotations:  $CHCl_3$ . IR:  $NaCl$ ,  $\nu_{max}$  values are expressed in  $cm^{-1}$ .  $^1H$  (200.13 MHz) and  $^{13}C$  (50.3 MHz) NMR:  $CDCl_3$  with TMS as int. standard. Chemical shifts are reported in  $\delta$  (ppm) and coupling constants are in Hz. EIMS: 70 eV. Flash chromatography: silica gel (Merck No. 9385).

**Plant material.** *Linaria saxatilis* var. *saxatilis* was collected in July 1991 at Campo de Ledesma (Salamanca, Spain) and identified by Prof. M. Ladero. A voucher specimen is deposited at the Botany Dept. of Pharmacy, Salamanca (Register No. SALAF 21386).

**Extraction and isolation.** Once triturated, air-dried roots (750 g) were extracted by the Soxhlet procedure with *n*-hexane for 15 hr, and resulting extract was cooled overnight at  $-20^\circ$ . The soluble fr. was defatted with  $Me_2CO$ , yielding 6.4 g of a viscous material from which, after repeated CC and/or prep. TLC and/or crystallization, the following compounds were isolated: 1 and 2 (22 mg), 3 (81 mg), 4 (208 mg), 5 (380 mg), friedelin (17 mg) and 6 (23 mg).

15,16-Diacetoxy-15,16-epoxy-neo-cleroda-3,12Z-diene (3). Eluted with *n*-hexane-EtOAc (8:2);  $[\alpha]_D^{25}$ :  $-41.9$  (c

1.25). IR  $\nu_{max}$   $cm^{-1}$ : 1750, 1670, 1200, 1050, 880.  $^1H$  NMR: Table 1, and  $^{13}C$  NMR: Table 2.

15,16-Diacetoxy-12,13-15,16-diepoxy-14-hydroxy-neo-clerod-3-ene (6). Eluted with *n*-hexane-EtOAc (8:2);  $[\alpha]_D^{25}$ :  $-0.4$  (c 1.4). IR: 3500, 1750, 1640, 1220, 1080, 1030, 1000, 900, 850.  $^1H$  NMR: Table 1, and  $^{13}C$  NMR: Table 2.

14,15,16-Triacetoxy-12,13-15,16-diepoxy-neo-clerod-3-ene (6a). By acetylation of 10 mg 6 with  $Ac_2O$ -pyridine,

Table 1.  $^1H$  NMR data for compounds 3 and 6 (200 MHz,  $CDCl_3$ , TMS as int. standard;  $J$  (Hz) in parentheses)

H	3	6
3	5.10 <i>m</i>	5.20 <i>m</i>
12	5.70 <i>m</i>	3.40 <i>t</i> (6.0)
14	2.60–3.00 <i>m</i>	4.25 <i>m</i>
15	6.30 <i>d</i> (5.4)	6.30 <i>d</i> (2.0)
16	6.70 <i>s</i>	6.05 <i>s</i>
Me-17	0.85 <i>d</i> (6.1)	0.88 <i>d</i> (6.1)
Me-18	1.50 <i>br s</i>	1.57 <i>br s</i>
Me-19	1.00 <i>s</i>	1.01 <i>s</i>
Me-20	0.70 <i>s</i>	0.77 <i>s</i>
MeCO <sub>2</sub>	2.00 <i>s</i>	2.06 <i>s</i>
	2.05 <i>s</i>	2.11 <i>s</i>

Table 2.  $^{13}C$  NMR data for compounds 3 and 6 (50.3 MHz,  $CDCl_3$ , TMS as int. standard)

C	3	6
1	26.7	26.8
2	19.0	21.0
3	120.5	120.8
4	144.3	143.8
5	39.9	39.8
6	37.6	36.7
7	27.5	27.3
8	36.6	37.6
9	38.3	37.8
10	47.7	48.2
11	36.6	32.3
12	125.6	57.2
13	133.1	68.9
14	36.8	71.4
15	97.0	103.6
16	93.9	99.6
17	16.1	16.2
18	17.9	17.9
19	19.8	19.1
20	17.7	17.5
MeCO <sub>2</sub>	169.7	169.1
		169.3
MeCO <sub>2</sub>	20.6	21.0
		19.9

9 mg **6a** was obtained. Mp 117 (*n*-hexane). IR  $\nu_{\max}$   $\text{cm}^{-1}$ : 1760, 1640, 100, 890. EIMS  $m/z$  (rel. int.): 418 ( $[\text{M}]^+ - \text{AcOH}$ , 6), 300 (4), 282 (6), 246 (7), 215 (8), 189 (48), 175 (77), 133 (42), 119 (71), 107 (86), 95 (100).  $^1\text{H}$ NMR:  $\delta$  0.78 (3H, s, H-20), 0.89 (3H, *d*,  $J = 6.0$  Hz, H-17), 1.03 (3H, s, H-19), 1.58 (3H, *br s*, H-17), 2.04 (3H, s, OAc), 2.08 (3H, s, OAc), 2.10 (3H, s, OAc), 3.40 (1H, *t*,  $J = 6$  Hz, H-12), 5.12 (1H, *m*, H-3), 5.21 (1H, *br s*, H-14), 6.17 (1H, s, H-16), 6.25 (1H, s, H-15).

**Acknowledgements**—Financial support for this work came from the Spanish DGICYT (Grant No. PB 89-394) and Junta de Castilla y León (Consejería de Cultura y Turismo, SA-64/12/92).

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