

NEOVASIPYRIDONES A, B AND C: METABOLITES RELATED TO NEOVASININ, A PHYTOTOXIN OF THE FUNGUS, NEOCOSMOSPORA VASINFECTA

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Abstract—Three new metabolites, neovasipyridones A, B and C, were isolated from the phytopathogenic fungus *Neocosmospora vasinfecta* NHL2298, and their structures elucidated by spectroscopic methods. The biosynthetic relationship between the neovasipyridones and neovasinin, a phytotoxin of this fungus, is discussed.

INTRODUCTION

Neocosmospora vasinfecta E.F. Smith is a pathogen which causes root- and fruit-rot and seedling dampingoff in the Malvaceae, Leguminosae, Piperaceae and Cucurbitaceae [1]. Neocosmospora vasinfecta NHL2298, isolated from the soil of Johannesburg (Republic of South Africa) produces neovasinin (1) which is phytotoxic to soybean, a host plant of this fungus [2, 3]. Neovasinin is a new type of fungal metabolite that has a unique bicyclic unit, 2H,5H-pyrano[4,3-b]pyran-2-one, and probably biosynthesized from a hexaketide and five C₁ units. To examine its biosynthetic pathway, particularly formation of the bicyclic unit, we undertook a search for metabolites biogenetically related to the neovasinin, produced by this fungus. Very recently we reported the structures of neovasipyrones A (2a) and B (2b) and those of neovasifuranones A (3a) and B (3b) [4]. We here report the structures of three new metabolites, named neovasipyridones A (4a), B (4b) and C (4c), which seem to be biogenetically related to neovasinin.

RESULTS AND DISCUSSION

Neovasipyridones A (4a), B (4b) and C (4c) were isolated as oils with respective yields of 0.036, 0.023 and 0.018 mg g⁻¹ from air-dried mycelium of the fungus N. vasinfecta NHL 2298 grown on a malt extract medium supplemented with peptone and methionine [4, 5]. The ¹³C NMR data (Table 1) and HRFAB mass spectrum of neovasipyridone A (4a) showed its molecular formula to be C₂₁H₃₅NO₃ (five unsaturations). The ¹³C NMR

spectrum of 4a showed 21 resonances, seven of which were due to methyls, four to methylenes, five to methines and five to quaternary carbons from a DEPT experiment [6], indicating that one of the protons in the molecule is bound to oxygen or nitrogen. The 2,3-dihydro-4-pyridinone portion in 4a was indicated by two IR absorption bands at 1647 and 1580 cm⁻¹, three ¹³C resonances at δ 191.7, 158.8 and 104.9 and a UV absorption band at 322 nm [7, 8]. Two ¹³C resonances at δ 127.6 and 140.1 were attributed to two olefinic carbons of a trisubstituted

Table 1. 13C NMR data for 4a, 4b and 4c

C	4a	4b	4 c
2	76.7	76.1	76.3
3	71.1	70.9	71.0
4	191.7	191.6	191.7
5	104.9	104.8	104.8
6	158.8	158.7	158.1
7	127.6	127.7	127.5
8	140.1	140.0	140.1
9	34.0	34.0	34.0
10	30.1	30.1	30.1
11	11.8	11.8	11.8
12	13.1	13.0	13.1
13	20.6	20.6	20.6
14	28.8	28.6	28.9
15	197.6	197.6	197.6
16	34.9	34.8	34.9
17	8.5	8.6	8.5
18	61.2	53.6	62.7
19	33.7	37.6	27.2
20	26.4	25.7	19.8
21	10.8	22.0	20.1
22	16.8	22.5	_

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OH

Me

double bond. One 13 C resonance of δ 197.6 was characteristic of a ketonic carbonyl carbon conjugated to a double bond. The structure of **4a** was unambiguously deduced from analysis of H,H-COSY [9, 10], C,H-COSY [11, 12], long-range C,H-COSY (COLOC) [11, 12] (Table 2) and NOE data (Table 3).

The partial structure A (Fig. 1) was deduced from H.H-COSY and COLOC data. The structure of the side chain (C-7 to C-11), deduced to be 1,3-dimethyl-1-pentenyl from H,H-COSY, was confirmed by COLOC data. The bonds between C-2 and C-7, and between C-2 and C-3, were deduced from COLOC data. The allylic methyl carbon C-12 was correlated with the methine proton H-2, indicating the attachment of a 1,3-dimethyl-1-pentenyl group to the methine carbon, C-2. Furthermore, the oxygen-bearing quaternary carbon C-3, was correlated both with the methine proton H-2, and the methyl protons H-14, and the methyl carbon C-14 was in turn correlated with the methine proton H-2. These correlations indicated that the methyl carbon C-14 is bonded to the oxygen-bearing quaternary carbon C-3, which is attached to the methine carbon C-2. The stereochemistry of the double bond is E based on the NOE data, irradiation of the methine proton H-2 caused NOE enhancement of the signal of the olefinic proton H-8, and irradiation of the olefinic proton H-8 induced NOE enhancement of the signal of the methine proton H-2. The partial structure A therefore was established. The partial structure B (Fig. 1) also was deduced from H,H-COSY and COLOC data. An ethyl group (C-16 and C-17) was indicated by the ¹H NMR data and H,H-COSY. Attachment of the ethyl group to a ketonic carbonyl carbon

Table 2. COLOC data for 4a

С	COLOC
3	H-2, H-14
4	H-6, H-2, H-14,
6	H-2
7	H-2, H-12
8	H-12
9	H-11, H-13
10	H-11, H-13
12	H-2, H-8
14	H-2
15	H-17
16	H-17
19	H-21
20	H-22

Table 3. NOE observed in 4a

IRR	NOE
H-2 H-6 H-8	H-8, H-14, H-18b, H-19 H-18a* H-2

^{*}Refers to the methylene proton that resonates in the upper field.

C-15 was deduced from the COLOC correlation of the carbonyl carbon C-15 to the methyl protons H-17, thus giving the partial structure B. The partial structure C (Fig. 1) was shown to be 2-methylbutyl from the

Fig. 1. Structural units of neovasipyridone A (4a).

¹H NMR data and H,H-COSY. The partial structure D (Fig. 1) was deduced from the IR, UV and ¹³C NMR data as described above.

These four partial structures A-D were combined on the basis of chemical shift considerations and the COLOC data. The carbonyl carbon C-4 in partial structure D had COLOC correlations with the methyl protons H-14 and with the methine proton H-2, both in partial structure A, indicating a bond between C-3 and C-4. The COLOC correlation of the olefinic carbon C-6 in partial structure D to the methine proton H-2 in partial structure A, and the chemical shift of the methine carbon C-2 (δ_C 76.7), were in agreement with there being a bond between C-2 and the nitrogen atom. The finished combination of partial structures A and D gave an integrated partial structure with a 2,3-dihydro-4-pyridinone ring. The chemical shift of the ketonic carbonyl carbon C-15 (δ_C 197.6) in partial structure B indicated bonding of

the carbonyl carbon C-15 to the olefinic carbon C-5 on the 2,3-dihydro-4-pyridinone ring. The chemical shift of the methylene carbon (C-18, $\delta_{\rm C}$ 61.2) and the methylene protons (H-18a, $\delta_{\rm H}$ 3.10; H-18b, $\delta_{\rm H}$ 3.23) indicated a bond between C-18 and the nitrogen atom. This was confirmed by the NOE observations. Irradiation of H-2 caused NOE enhancement of the signal of one of the methylene protons (H-18b), and irradiation of H-6 caused NOE enhancement of one of the methylene protons (H-18a). Finally, the one remaining hydrogen atom was bonded to the oxygen atom on C-3 to give the complete structure of 4a. NOE enhancement of the signal of the methyl protons (H-14) on irradiation of the methine proton (H-2) suggested that the methyl group was pseudo-equatorial on the supposition that the H-2 proton was pseudo-axial. The stereochemistries of the other asymmetric carbons have yet to be resolved although the absolute stereochemistries of neovasinin [3] and the neovasipyrones [5] suggest that the stereochemistries of C-2, C-3 and C-9 in neovasipyridone A are (R,S,S).

Neovasipyridone B (4b) had the same molecular formula, C21H35NO3 (HRFABMS and 13C NMR data), as neovasipyridone A, and their UV spectra were very similar. Their 16 13 C resonances due to C-2 to C-17 (Table 1) were nearly superimposable, indicating that neovasipyridone B has the same structure, except for the N-alkyl group, as neovasipyridone A. The ¹H NMR (Table 4) and H,H-COSY data indicated the attachment of 3methylbutyl to the nitrogen atom on the 2,3-dihydro-4pyridinone ring. Neovasipyridone C (4c) had a molecular formula of C₂₀H₃₃NO₃ (HREIMS and ¹³C NMR data), which is equal to the loss of CH2 from neovasipyridones A and B. The 16 13C resonances due to C-2 to C-17 (Table 1) were nearly superimposable on those of neovasipyridones A and B. The ¹H NMR (Table 4) and H,H-COSY data indicated the 2-methylpropyl was a

Table 4. ¹H NMR data for 4a, 4b and 4c

Н	4a	4b	4c
2	3.82 s	3.80 s	3.80 s
6	8.21 s	8.25 s	8.20 s
8	5.12 dq (9.5, 0.9)	5.12 dq (10.8, 1.2)	5.13 dq (9.5, 1.0)
9	2.28 m	2.25 m	2.28 m
10	1.10-1.40 m	1.101.42 m	1.07-1.40 m
11	0.83 t (7.6)	$0.82\ t\ (7.5)$	$0.81 \ t \ (7.5)$
12	1.43 d (0.9)	1.41 d (1.2)	$1.41 \ d(1.0)$
13	0.93 d (6.8)	0.91 d (6.8)	0.91 d (6.8)
14	1.38 s	1.34 s	1.37 s
16	2.86 dq (17.1, 7.2)	2.84 dq (16.5, 7.5)	2.84 dq (17.0, 7.0)
	2.98 dq (17.1, 7.2)	2.97 dq (16.5, 7.5)	2.97 dq (17.0, 7.0)
17	1.10 t (7.2)	1.09 t (7.5)	1.08 t (7.0)
18	3.10 dd (13.5, 8.6)	3.29 dd (13.5, 7.5)	2.94 dd (14.0, 9.0)
	3.23 dd (13.5, 5.8)	3.39 dd (13.5, 7.0)	3.28 dd (14.0, 5.2)
19	1.80 m	1.60 m	2.03 m
20	1.25 m	1.60 m	0.97 d (6.8)
21	0.95 t (7.2)	0.95 d (6.5)	$0.98 \ d \ (6.8)$
22	0.98 d (6.3)	$0.96 \ d \ (6.5)$	` '

Scheme 1. Proposed biosynthetic relationships between neovasinin (1) and the neovasipyridones (4), neovasipyrones (2) and neovasifuranones (3).

N-alkyl instead of the 2-methylbutyl in neovasipyridone A or the 3-methylbutyl in neovasipyridone B.

From their chemical structures, we propose the biosynthetic relationships between neovasinin (1), neovasipyridones (4), neovasipyrones (2) and neovasifuranones (3) shown in Scheme 1. The relationships between neovasinin, neovasipyrones and neovasifuranones has been discussed elsewhere [4]. Here, we discuss the relationship between neovasipyridones and neovasifuranones. Neovasipyridones and neovasifuranones are derived from a hexaketide chain plus five C1 units via the hypothetical common intermediate 6. Neovasipyridones (4) are formed from 6 through oxidation into aldehyde 7, Schiff base formation with alkyl amines which seem to be from isoleucine, leucine and valine in the respective case of neovasipyridones A (4a), B (4b) and C (4c) then dehydration/cyclization. In contrast, neovasifuranones (3) are formed from 6 through dehydration/cyclization. A leaf spot bioassay indicated that the neovasipyridones like the neovasipyrones were not phytotoxic to soybean plants although neovasinin and neovasifuranones are phytotoxic.

EXPERIMENTAL

General. NMR spectra were recorded in CDCl₃ on a JEOL JNM GX-270 FT NMR spectrometer. NMR chemical shifts were referenced against CDCl₃ ($\delta_{\rm H}$ 7.26, $\delta_{\rm C}$ 77.00). MS were obtained with a JEOL DX-300

spectrometer. Glycerol was the matrix used for FABMS. In EIMS the ionization voltage was 70 eV. Fourier-transform IR spectra were obtained with a JASCO FT/IR-7000 spectrometer. Optical rotations were measured with a Horiba SEPA-200 high sensitive polarimeter. Daisogel IR-60 was the silica gel used for column chromatography. Prep. TLC was done on a Merck Kieselgel 60 HF₂₅₄ glass plate $(20 \times 20 \times 0.05 \text{ cm})$.

Fungus. The strain NHL2298 of Neocosmospora vasin-fecta E.F. Smith var. africana (von Arx) Cannon et Hawksworth used in this study was a gift from Dr S. Udagawa of the National Institute of Hygienic Sciences (present address: Nodai Research Institute, Tokyo University of Agriculture, Setagaya-ku, Tokyo 156, Japan) in 1983 and has been maintained on potato dextrose agar.

Extraction and isolation of neovasipyridones A (4a), B (4B) and C (4c). The fungus was grown in a 500 ml conical flask containing liquid medium (200 ml × 200) composed of sucrose (50 gl⁻¹), peptone (3 gl⁻¹), the extract from 100 gl⁻¹ on malt, L-methionine (300 mgl⁻¹) and H_2O , without shaking at 24° for 21 days in the dark. The mycelial mats obtained after filtering the culture broth were rinsed with H_2O and dried under a hood. The dried mats (620 g) were soaked in Me_2CO overnight, after which the Me_2CO extract was obtained by filtration. This treatment was repeated $3 \times$. The combined Me_2CO extracts were evaporated to dryness. The residue was dissolved in 1 M NaHCO₃ (500 ml) and treated with EtOAc (500 ml). The EtOAc soln obtained was placed

over Na₂SO₄ and evaporated to dryness. The residue was partitioned with MeOH-H₂O (9:1, 250 ml) and n-hexane (250 ml). The n-hexane layer obtained was extracted with MeOH-H₂O (9:1, 2×250 ml). The combined MeOH soln was evaporated to dryness. The residue (13.5 g) was subjected to silica gel column chromatography (340 g, 32×4.6 cm). The column was developed successively with 2000 ml (5 \times 400 ml) each of 0%, 2%, 5%, 10% and 20% Me₂CO in *n*-hexane. Frs 1-3 eluted with 10% Me₂CO in n-hexane contained the neovasipyridones. The first fraction (455 mg) was purified by Sephadex LH-20 column (80 × 2.7 cm) chromatography with MeOH as the solvent. Each 7 ml of eluate constituted one fr., and frs 41-44 were combined and evaporated to dryness. The residue (30 mg) further was purified by prep. TLC [Me₂CO-CHCl₃ (5:95)] to afford neovasipyridone A (15.5 mg). The second and third fractions were combined and evaporated to dryness. The residue (1.29 g) was subjected to Sephadex LH-20 column chromatography (122 × 3.2 cm) with MeOH as the solvent. Each 10 ml of eluate constituted one fr., and frs 56-61 were combined and evaporated to dryness. The residue (73 mg) contained neovasipyridones A, B and C. Removal of neovasipyridone A from the mixture was achieved by prep. TLC [EtOAc-C₆H₆ (7:93), quadruple development], giving neovasipyridone A (6.7 mg) as a constituent of a higher R_f band on TLC. The lower R_f band contained neovasipyridones B and C. Neovasipyridone B was separated from C by repeated HPLC (column: DAISOPAK SP-120-5-ODS-A $(250 \times 10 \text{ mm})$, solvent: MeOH-H₂O (9:1), flow rate: 1.0 ml min⁻¹, detector: 220 nm), giving neovasipyridones B (14.0 mg, R_t 28.0 min) and C (11.1 mg, R, 33.0 min).

Neovasipyridone A (4a). Oil. $[\alpha]_{\rm D}^{20}$ + 366° (EtOH; c 1.0). UV $\lambda_{\rm max}^{\rm EtOH}$ nm (log ε): 202 (3.79), 262 (4.10), 322 (4.14). IR $\nu_{\rm max}^{\rm Film}$ cm $^{-1}$: 3398, 2964, 2930, 2874, 2858, 1647, 1580, 1462, 1375, 1350, 1315, 1267, 1241, 1172, 1114. 1 H and 13 C NMR spectral data: Tables 1 and 4. FABMS m/z (rel. int.): 350 [M + H] $^{+}$ (100), 332 (10), 306 (14), 236 (30), 196 (12); exact mass calcd for C₂₁H₃₆NO₃ 350.2695, found 350.2655.

Neovasipyridone B (4b). Oil. $[\alpha]_{\rm D}^{20}$ + 392° (EtOH; c 1.0). UV $\lambda_{\rm max}^{\rm EtOH}$ nm (log ε): 200 (3.71), 263 (4.15), 322 (4.19). IR $\nu_{\rm max}^{\rm Film}$ cm⁻¹: 3362, 2960, 2874, 1638, 1574, 1460, 1392, 1377, 1334, 1292, 1257, 1201, 1156, 1116, 1083, 1029. ¹H and ¹³C NMR spectral data: Tables 1 and 4. FABMS m/z (rel. int.): 350 [M + H]⁺ (100), 306 (15),

196 (10), 154 (39), 149 (27), 136 (33), 107 (15); exact mass calcd for C₂₁H₃₆NO₃ 350.2695, found 350.2666.

Neovasipyridone C (4c). Oil. $[\alpha]_D^{20} + 383^\circ$ (EtOH; c 1.0). UV λ_{max}^{EtOH} nm (log ε): 203 (3.98), 262 (4.11), 323 (4.14). IR ν_{max}^{Film} cm⁻¹: 3406, 2967, 2926, 2874, 1639, 1579, 1460, 1369, 1315, 1261, 1157, 1113. ¹H and ¹³C NMR spectral data: Tables 1 and 4. EIMS m/z (rel. int.): 335 [M] + (55), 292 (100), 278 (23), 222 (15), 182 (41), 125 (19); exact mass calcd for $C_{20}H_{33}NO_3$ 335.2461, found 335.2488.

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