



OLEANEN AND STIGMASTEROL DERIVATIVES FROM AMBROMA AUGUSTA

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Key Word Index—Ambroma augusta; Sterculiaceae; augustic acid; olean-12-en- 2β , 3β -diol-28-oic acid; augustoside; stigmast-5,22-dien-3-O- α -D-glucopyranoside.

Abstract—Augustic acid, a new oleanane derivative, and a stigmasterol glycoside were isolated from the roots of Ambroma augusta. Their structures were established as olean-12-en-2 β ,3 β -diol-28-oic acid and stigmast-5,22-dien-3-O- α -D-glucopyranoside respectively, on the basis of full spectral analyses and chemical methods.

INTRODUCTION

Ambroma augusta Linn.f. (Sterculiaceae), commonly known as 'Ulatkambal', is a large, spreading, quickgrowing, hairy shrub found both wild and cultivated throughout the hot and moister parts of India from Uttar Pradesh to Sikkim, Khasia Hills, Assam and Bengal. The plant drug possesses uterine tonic and emmenagogue properties and is used for the treatment of dysmenor-rhoea, amenorrhoea sterility and other menstrual disorders [1]. The roots of A. augusta act as an abortifacient and antifertility agent [2, 3]. We describe herein the isolation and structural elucidation of augustic acid (1) and stigmasterol glucoside (2) from the roots of A. augusta.

RESULTS AND DISCUSSION

Compound 1 responded positively to the Liebermann-Burchard test and showed characteristic IR absorption bands at 3436 (OH), 1696 (CO₂H) and 1512 (C=C) cm⁻¹. It had a molecular ion peak at m/z 472 corresponding to the molecular formula $C_{30}H_{48}O_4$ and thereby indicated seven double bond equivalents. The mass spectrum of 1 exhibited a pair of diagnostically important mass peaks at m/z 248 and 203 [248-CO₂H] typical of the retro-Diels-Alder fragmentation in ring C of an olean-12-en derivative containing a carboxyl function in either ring D or E [4]. Acetylation did not alter the values of these two fragments whereas conversion to the methyl ester caused a mass increment of 14 amu in one of the fragments (m/z 262). Other important ions were observed at 454 $[M - H_2O]^+$, 428 $[M - CO_2]^+$, 426 $[M - HCO_2H]^+$, 408 [426 - H_2O]⁺ and 189 [204 – Me]⁺.

The ¹H NMR spectrum of 1 displayed a one-proton broad singlet at δ 5.166 assigned to H-12 and another two-protons broad singlet at δ 3.417 ascribable to hydroxymethines, one placed at C-3 on the basis of biogenetic analogy and another at C-2. The seven methyl signals, all located on saturated carbons, resonated at δ 0.885 (Me-23), 0.689 (Me-24 and Me-26), 0.905 (Me-25) 1.075 (Me-27) and 0.858 (Me-29 and Me-30).

More compelling evidence for the structure was provided by a study of its ¹³C NMR spectrum (Table 1), which showed the presence of 30 carbon atoms. The assignments of the carbon chemical shifts were made by comparison with the δ values of the corresponding carbon atoms in structurally similar compounds [5-9], namely, methyl oleanolate, hederagenin methyl ester, gypsogenin methyl ester, methyl maslinate and imberbic acid. The difference in melting points of compound 1 and its methyl ester derivative from that of maslinic acid and its methyl ester derivative suggested that they were different compounds. An upfield shift of C-2 (δ 67.15) from that of methyl maslinate (δ 68.8) [8] also suggested a difference in the stereo structure at C-2. Furthermore, apart from the resonances of the carbons of rings of A, B and C, the resonances of the ring D and E carbons in 1 agreed with the corresponding assignments in methyl maslinate, which restricted the positioning of two hydroxyl functions to ring A and a carboxyl group at C-17. The degree of protonation of each carbon atom was determined by DEPT experiments and by the ¹H-¹³C COSY spectrum. The quaternary carbon atoms were determined by substracting DEPT from a broad band ¹³C NMR spectrum. The connectivities of structure 1 were confirmed by ¹H-¹³C long range experiments. The stereochemistry of 1198 M. S. Alam et al.

Table 1. ¹H and ¹³C NMR chemical shifts of compound 1

	¹H NMR		
Position	α	β	¹³ CNMR
1	1.748 br s	0.775 br s	46.52
2	3.417 br s	_	67.15
3	3.417 br s		82.23
4	_	_	41.44
5	$0.748 \ dd$		
	(14.12, 12.09)		54.84
6	1.456 m	1.456 m	18.11
7	1.229 br s	1.229 br s	32.41
8	_		37.72
9	1.507 br s		47.14
10	-	_	37.72
11	1.456 m	1.456 m	22.66
12	5.166 br s	_	121.54
13	_		143.95
14	_		41.44
15	0.987 br s	0.987 br s	27.44
16	1.806 m	1.889 m	23.08
17		_	45.76
18	_	2.718 d	40.88
		(9.16)	, , , ,
19	1.585 m	1.132 br s	45.52
20	_	_	30.44
21	1.316 br s	1.212 br s	33.40
22	1.585 m	1.400 m	32.17
23	0.885 br s	_	28.85
24	0.689 br s	_	16.94
25	0.905 br s		16.34
26	0.689 br s	_	17.14
27	1.075 br s	_	25.69
28	0.905 br s	_	178.51
29	0.858 br s	_	32.89
30	0.858 br s		23.42
	0.050 57 3	_	23.72

Multiplicity for the protons and the coupling constant(s) in Hertz are given in parentheses.

1 was also elucidated on the basis of ${}^{1}H^{-1}H$ COSY analysis. Treatment of 1 with acetic anhydride and pyridine afforded a diacetyl derivative (1a), $C_{34}H_{52}O_6$, $(m/z 556, v_{max} 1735$ and $1725 \, cm^{-1}$), indicating the presence of two hydroxyl groups. Esterification of 1 with diazomethane formed a methyl ester (1b). On the basis of these findings, augustic acid (1) was identified as olean-12-en-2 β ,3 β diol-28-oic acid.

Compound 2 gave a positive Liebermann-Burchard test and a negative Ehrlich reaction. The IR spectrum exhibited strong bands at 3476 and $1062 \,\mathrm{cm}^{-1}$ characteristic of a glycoside. The FAB-mass spectrum of 2 gave a quasimolecular ion at m/z 597 [M + Na]⁺, which suggested the M_r , to be 574. The EI mass spectrum gave a fragment at m/z 396 due to the loss of glucose from the molecular ion peak.

The ¹H NMR spectrum of 2 showed the signals of two tertiary methyls at δ 0.640 (Me-18) and 1.005 (Me-19), four secondary methyls at δ 0.978 (Me-21), 0.885 (Me-26), 0.810 (Me-27) and 0.822 (Me-29), one trisubstituted ole-

finic proton at $\delta 5.317$ (H-6), two disubstituted olefinic protons at $\delta 5.149$ (H-22) and 5.006 (H-23) and one anomeric proton at δ 4.212. The ¹³C NMR spectrum (Table 2) showed the existence of 35 carbon atoms in the molecule. An anomeric signal at $\delta 100.76$ indicated the presence of a single monosaccharide moiety. The degree of protonation of each carbon atom was determined by DEPT experiments. The ¹³C DEPT NMR spectrum revealed the presence of six methyl, 10 methylene and 16 methine carbon atoms. The four methine resonances at δ 70.08, 76.74, 73.42 and 76.69 and one methylene resonance at δ 61.0 were due to C-2', C-3', C-4', C-5' and C-6', respectively, of the α-D-glucopyranoside. The olefinic resonances at δ 121.13, 137.95 and 128.79 corresponded to the C-6, C-22 and C-23 methine carbons, and a signal at δ 140.41 corresponded to the C-5 quaternary carbon of the sterol moiety.

The COSY and relayed COSY experiments reflected that the anomeric doublet at $\delta 4.212$ belonged to one glucose residue. In the COSY spectra all coupling constants were large and corresponded to the equatorial proton of one α -D-glucopyranose residue. The $^{13}C^{-1}H$ COSY-90° spectrum allowed complete correlation of the protonated carbon resonances with those of the ^{1}H spectrum. The $^{1}H^{-1}H$ COSY measurements also confirmed the assignments. The COSY-45° spectrum showed strong coupling interaction between the H-3 methine and H-4 methylene.

Acetylation of 2 with acetic anhydride-pyridine afforded a tetraacetyl derivative (2a) ($v_{\rm max}^{\rm KBr}$ 1760 cm⁻¹). The positive FAB-mass spectrum of 2a gave a peak at m/z 765 [M + Na]⁺. Acetylation deshielded all the methine protons alpha to acetates in the $\delta 5.035$ -3.455 region. The homonuclear ¹H-¹H COSY study of 2a not only correlated each proton in the spin system, but also revealed several interesting correlations via long range coupling. The assignment of carbon resonances of 2a was achieved by the heteronuclear ¹H-¹³C COSY spectrum including both one-bond and long-range coupling.

Acid hydrolysis of 2 yielded D-glucose and an aglycone (2b) which was identified as stigmasterol by spectral data as well as by direct comparison with the authentic samples (co-TLC, mmp). On the basis of these findings, 2 was identified as stigmast-5,22-dien-3-O- α -D-glucopyranoside.

EXPERIMENTAL

Mps: uncorr. IR: JASCO FT/IR 5000, KBr; UV: Beckmann DU-64, MeOH; ¹H (600 MHz) and ¹³C (150 MHz) NMR: JEOL A-600, DMSO- d_6 with TMS as int. standard; MS: Hitachi M-80; CC: silica gel (Qualigen, 60–120 mesh); TLC: Silica gel 60 F₂₅₄ pre-coated plates. The spots were visualized by exposure to I₂vapour and by spraying with vanillin–H₂SO₄.

Plant material. The roots of A. augusta were purchased from Herba Indica, Chandigarh, and identified in our Botany Department by Dr M. P. Sharma.

Isolation of constituents. Air dried and crushed roots of A. augusta (3 kg) were exhaustively extracted (Soxhlet)

Table 2. ¹H and ¹³C NMR chemical shifts of compounds 2 and 2a

	2		2a	
Position	¹H NMR	¹³ CNMR	¹H NMR	¹³ C NMR
1	2.344 m	38.27	0.991 m	37.20
	2.093 m		1.817 m	
2	1.306 m	33.30	$0.902 \ br \ s$	33.94
			1.270 m	
3	3.469	76.89	3.455 br m	72.92
	$(W_{1/2} = 22.35)$		$(W_{1/4} = 22.71)$	
4	1.775 d (10.63) 36.79		1.887 br s	39.73
	1.142 (6.59)		1.135 m	
5	_	140.41	_	140.36
6	5.317 br s	121.13	5.331 br s	122.16
7	1.481 m	31.31	1.459 m	31.94
8	1.495 br s	31.38	1.270 br s	31.87
9	0.916 br s	49.56	0.818 br s	50.17
10	_	36.16	_	36.13
11	1.178 m	22.57	1.998 br s	21.20
			1.981br s	
12	1.138 m	39.63	0.964 br s	40.45
	1.957 m		1.459 m	
13	_	40.03		42.33
14	1.070 m	56.22	0.835 br s	56.75
15	1.033 m	24.79	1.034 m	24.29
16	1.792 br s	29.21	1.637 br s	28.22
	1.637 br s			
17	1.043 m	56.12	1.043 m	56.05
18	0.640 s	11.80	0.653 s	11.98
19	1.005 s	19.03	1.002 s	19.34
20	1.306 m	35.42	1.270 m	36.72
21	0.978 d (6.23)	18.78	0.991 d (6.60)	18.77
22	5.149 m	137.95	5.164 m	138.27
23	5.006 m	128.79	5.052 m	129.31
24	0.917 br s	45.11	1.550 br s	29.16
25	1.438 m	31.26	1.459 m	31.87
26	0.885 d (6.59)	19.64	0.891 d (6.59)	19.80
27	0.810 d (6.96)	18.88	0.818 d (6.96)	19.03
28	1.054 br s	23.80	1.196 br s	23.07
29	0.822 d (7.33)	11.62	0.835 d (7.32)	11.85
1'	4.212 d (9.69)	100.76	4.576 d (8.06)	99.64
2'	3.054 m	70.08	3.455 m	80.07
3'	3.403 m	76.74	3.482 m	71.51
4' 	3.113 m	73.42	5.035 m	68.55
5'	3.070 m	76.69	4.919 m	71.55
5a'	4.406 d (5.5)	61.0	4.817 d (10.10)	62.10
6b'	4.212 d (7.69)		4.081 d (10.10)	
COCH ₃	_		2.054 s	170.69, 21.20
			2.025 s	170.35, 21.04
			1.998 s	169.40, 20.76
			1.981 s	169.30, 20.71

Multiplicity for the protons and the coupling constant(s) in Hertz are given in parentheses.

with EtOH (95%) and the combined extracts concd to dryness under red. pres. to yield a brown semi-solid residue (100 g). The residue was fractionated between EtOAc and MeOH, and EtOAc solubles (25 gm) were subjected to CC on silica gel. Elution was carried out with petrol and petrol containing increasing amounts of EtOAc.

Augustic acid (1). Frs eluted with petrol–EtOAc (7:3) yielded a residue which was crystallized from petrol–Me₂CO as white crystals (60 mg), mp 256–258°. TLC R_f 0.33 (toluene–EtOAc, 1:1). IR $\nu_{\rm max}$ cm⁻¹: 3436, 2948, 1696, 1512, 1464, 1390, 1282, 1180, 1110, 1050, 1034, 994, 962, 822. UV $\lambda_{\rm max}$ 216 (log ε 4.2). FAB-MS (rel. int. m/z: 495 [M + Na]⁺ (14.8); EIMS (rel. int.) m/z: 472

1 R₁ = H, R₂=H, R₃=H 1a R₁ = Ac, R₂= Ac, R₃=H 1b R₁ = H, R₂=H, R₃=Me

$$R = \frac{\text{OAc}}{\text{CH}_2\text{OAc}}$$

26 R = H

[M] ⁺ (C₃₀H₄₈O₄) (4.1), 454 (1.0), 436 (1.0), 428 (3.0), 426 (6.1), 408 (2.4), 393 (1.0), 248 (100), 233 (2.1), 219 (2.1), 206 (6.1), 204 (25.8), 203 (80.0), 191 (3.8), 189 (22.2), 175 (21.2), 161 (13.9), 147 (13.6), 133 (31.2), 119 (32.1), 105 (32.2), 91 (30.0), 80 (32.6), 69 (54.1), 55 (75.2), ¹H and ¹³C NMR: Table 1.

Acetylation of compound 1. Compound 1 (15 mg), dissolved in a 1: 1 mixt. of pyridine– Ac_2O , was warmed slightly. On standing overnight and on usual work-up it afforded a diacetyl derivative (1a) (12 mg), mp 148–150°, IR v_{max} cm⁻¹: 3310, 1735, 1725, 1685, 1525, 1465, 1385, 1270, 1185, 1055, 1030, 995, 955, 825. EIMS (rel. int.) m/z:

556 [M]⁺ (1.0), 496 (21.0), 451 (5.1), 436 (2.5), 248 (85.1), 233 (15.3), 187 (33), 43 (100).

Esterification of compound 1. Compound 1 (15 mg) was treated with excess of ethereal CH_2N_2 at room temp. On evapn of the solvent, the Me ester (1b) was obtained, mp 175–176°; EIMS (rel. int.) m/z: 486 [M]⁺ (3.1), 440 (23.1), 262 (100), 216 (3.5), 203 (8.5), 187 (5.1).

Compound 2 (stigmasterol glucoside). Frs eluted with petrol–EtOAc (1:1) yielded a residue which was crystallized from CHCl₃–MeOH as white crystals (60 mg), mp 268–70°. TLC R_f 0.5 (CHCl₃–MeOH, 9:1). IR $v_{\rm max}$ cm $^{-1}$: 3476, 2944, 1646, 1556, 1370, 1340, 1214, 1168, 1114, 1062, 1026. UV $\lambda_{\rm max}$ 209 (log ε 2.9). FAB-MS (rel. int.) m/z: 597 [M + Na] + (9.2) 574 (14.2). EIMS (rel. int.) m/z 412 (82.5), 396 (91.8), 394 (34.0), 381 (24.4), 329 (3.2), 303 (5.3), 255 (6.6), 213 (5.7), 189 (4.0), 173 (5.8), 157 (10.0), 145 (13.2),131 (13.1), 117 (11.6), 105 (25.8), 93 (19.0), 81 (28.8), 69 (73.6), 55 (100). 1 H and 13 C NMR: Table 2.

Acetylation of compound **2**. Compound **2** (20 mg) was acetylated in the same manner as described above to give a tetraacetyl product (**2a**), mp 146–147°. IR v_{max} cm⁻¹: 2990, 1760, 1641, 1458, 1385, 1258, 1220, 1176, 1110, 1080, 1042, 910. FAB-MS (rel. int.) m/z: 765 [M + Na] + (8.2), 743 [M + H] + (7.2). EIMS (rel. int.) m/z: 396 (10.3), 382 (3.2), 353 (1.1), 331 (5.0), 255 (1.1), 213 (5.1), 201 (2.2), 185 (3.0), 169 (25.1), 157 (8.9), 145 (15.2), 133 (7.2), 119 (9.5), 105 (18.6), 93 (23.8), 81 (30.3), 69 (65.3), 55 (100). ¹H and ¹³C NMR: Table 2.

Acid hydrolysis of compound 2. Compound 2 (10 mg) was refluxed with 2 N HCl in 80% MeOH (1:1, 10 ml) for 4 hr. After cooling, the reaction mixt. was poured into crushed ice, and the hydrolysate was then extracted with EtOAc to give the aglycone. The aglycone, mp 169° , was identified as stigmasterol by spectral and chromatographic comparison (co-TLC, mmp). The neutralized (Ag₂CO₃) and concd aq. hydrolysate showed the presence of glucose on comparison with authentic sugars on silica gel TLC, R_f 0.4 (EtOAc-HOAc-H₂O-MeOH, 6:1:1:2).

REFERENCES

- 1. The Wealth of India: Indian Raw Materials (1985) Vol. IA, p. 222. CSIR, New Delhi.
- Vohora, S. B., Garg, S. K. and Chaudhury, R. R. (1969) Indian J. Med. Res. 57, 893.
- Pakrashi, A., Basak, B. and Mookerji, N. (1975) Indian J. Med. Res. 63, 378.
- Budzikiewicz, H., Wilson, J. M. and Djerassi, C. (1963)
 J. Am. Chem. Soc. 85, 3688.
- Tori, K., Seo, S., Shimaoka, A. and Tamita, Y. (1974) Tetrahedron Letters 48, 4227.
- Seo, S., Tomita, Y. and Tori, K. (1975) Tetrahedron Letters 1, 7.
- Tori, K., Yoshimua, Y., Seo, S., Sakurawi, K., Tomita, Y. and Ishii, H. (1976) Tetrahedron Letters 46, 4163.
- 8. Knight, S. A. (1974) Org. Magn. Res. 6, 603.
- 9. Rogers, C.B. and Subramony, G. (1988) Phytochemistry 27, 531.