



## JASPOLYSIDE, A SECOIRIDOID GLYCOSIDE FROM *JASMINUM POLYANTHUM*

YA-CHING SHEN,\* SAHO-LING LIN and CHYH-CHUNG CHEIN

Institute of Marine Resources, National Sun Yat-sen University, 70 Lien-Hai Rd, Kaohsiung, Taiwan, R.O.C.

(Received in revised form 5 February 1996)

**Key Word Index**—*Jasminum polyanthum*; Oleaceae; secoiridoid glycoside; jaspolside.

**Abstract**—A new secoiridoid glucoside, jaspolside was isolated from the leaves of *Jasminum polyanthum* along with nine known secoiridoid glycosides: oleoside dimethyl ester, 10-hydroxyoleoside dimethyl ester, oleoside 11-methyl ester, methyl-glucoside, 8-epi-kingiside, ligustroside, angustifolioside B, oleuropein, and oleoacetoside. The structure of the new compound has been established on the basis of spectral and chemical methods.

### INTRODUCTION

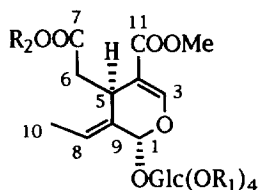
The evergreen *Jasminum* has been extensively investigated in the past few years [1–6]. As part of a study on the secoiridoid glycosides of oleaceous plants, we have undertaken the investigation of the chemical constituents of *Jasminum polyanthum* Franch. The species with fragrant, white flowers is indigenous to the western area of China and is now cultivated in Taiwan. In a previous paper, we reported on the structural elucidation of secoiridoid glycosides from the leaves of *J. polyanthum* [7]. We now report on the isolation of a new secoiridoid glycoside, jaspolside (**1**), in addition to nine known secoiridoids: oleoside dimethyl ester, 10-hydroxyoleoside dimethyl ester, oleoside 11-methyl ester (**4**), methyl-glucoside, 8-epi-kingiside, ligustroside, angustifolioside B, oleuropein and oleoacetoside, from the ethanolic extract of this plant.

### RESULTS AND DISCUSSION

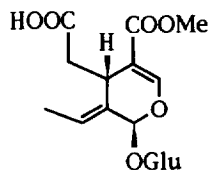
A combination of solvent partition and extensive

column chromatography allowed the isolation of jaspolside (**1**) and nine known secoiridoid glycosides: oleoside dimethyl ester, 10-hydroxyoleoside dimethyl ester, oleoside 11-methyl ester (**4**), methyl-glucoside, 8-epi-kingiside, ligustroside, angustifolioside B, oleuropein, and oleoacetoside. The structures of these compounds were identified on the basis of spectral data and comparison with authentic samples.

Jaspolside (**1**),  $[\alpha]_D^{25} -53^\circ$  (MeOH), was obtained as an amorphous solid. Its UV absorption at 234 nm and IR bands at 1720, 1690 and  $1640\text{ cm}^{-1}$  suggested the presence of an enol ether system conjugated with a carbonyl chromophore. The  $^1\text{H}$  NMR spectrum of compound **1** (Table 1) exhibited signals typical of a secoiridoid nucleus [8], including a hemiacetalic proton H-1 ( $\delta$  6.27), two vinylic protons, H-3 and H-8 ( $\delta$  7.46, 5.70), an ABX spin system comprising H-6 $\alpha,\beta$  and H-5 ( $\delta$  2.90, 2.52 and 3.70). The relationships between the protons in compound **1** was established by a COSY experiment. In the  $^{13}\text{C}$  NMR spectrum signals at  $\delta$  154.1 (C-3), 125.9 (C-8) and 132.6 (C-9), two carbonyl resonances at  $\delta$  176.6 and 168.8, a methoxy



- 1**,  $R_1 = R_2 = \text{H}$   
**2**,  $R_1 = \text{Ac}$ ,  $R_2 = \text{H}$   
**3**,  $R_1 = \text{H}$ ,  $R_2 = \text{Me}$



**4**

\*Author to whom correspondence should be addressed.

Table 1.  $^1\text{H}$  NMR spectral data\* (300 MHz,  $\text{CD}_3\text{OD}$ ) for compounds **1** and **4**

	<b>1</b>	COSY ( <b>1</b> )	<b>4</b>
1	6.27 s		5.94 brs
3	7.46 s		7.52 s
5	3.70 dd (8.7, 3.3)	H-6	4.01 dd (9.6, 4.2)
6 $\alpha$	2.90 dd (15.6, 3.3)	H-5	2.72 dd (14.4, 4.2)
6 $\beta$	2.52 dd (15.6, 8.7)	H-5	2.37 dd (14.4, 9.6)
8	5.70 q (7.2)	H-10	6.12 q (7.2)
10	1.79 d (7.2)	H-8	1.76 d (7.2)
11-OMe	3.69 s		3.68 s

\*TMS as internal standard.

signal at  $\delta$  51.8 and a methyl carbon signal at  $\delta$  13.5 (C-10), and also signals from a  $\beta$ -glucopyranosyl group, resembled those of oleoside 11-methyl ester and suggested that compound **1** was a stereoisomer of the compound. Acetylation of compound **1** provided a tetraacetate (**2**), which showed four acetyl singlets at  $\delta$  2.05, 2.01, 1.96 and 1.92. Upon methylation, compound **1** yielded a dimethyl ester (**3**), indicating a free carboxylic function in **1**. The  $^1\text{H}$  NMR spectrum of **3** exhibited an additional methoxyl signal ( $\delta$  3.64) arising from the C-7 ester group. A  $^1\text{H}$  NMR spectral comparison of compounds **1** and **4** revealed significant upfield shifts in compound **1** for the resonances of H-8 and H-5, while a downfield shift was observed for the signal of H-1 (Table 1). In addition, the chemical shift of Me-10 remain unchanged, suggesting that the orientation of Me-10 is identical to that of compound **4**. The specific rotation of compound **1** although negative but much smaller than that of compound **4** ( $[-176^\circ, \text{MeOH}]$ ), indicating that they are diastereoisomers [9]. Close comparison of the carbon resonances of C-4, C-5 and C-6 in compound **1** with those in **4** revealed a downfield shift for C-4 and C-5, and an upfield shift for C-6, suggesting that C-5 is the epimeric carbon atom. Moreover, the chemical shifts of C-1 and C-9 appeared at  $\delta$  93.9 and 132.6 in compound **1** relative to  $\delta$  95.5 and 130.7 in compound **4**, respectively, indicating that **1** and **4** are also epimeric at C-1. These findings pointed to the structure of compound **1**, which is a new secoiridoid glycoside bearing a carboxylic acid at C-6, an  $\alpha$ -H at C-5 and an *O*-glucosyl moiety at C-1 with  $\alpha$  orientation. Although we could not explain the reason for the large downfield shift for C-1' in some  $1\alpha$ -Glc-compounds [9, 10], both the resonances of C-1 and C-1' in compound **1** and the shift change between these carbons are similar to those reported for  $1\alpha$ -protoplumericin A [11].

This is the first example of the occurrence of a secoiridoid glucoside of this type in the Oleaceae.

#### EXPERIMENTAL

**General.** FAB-MS: VG Quattro 5022 mass spectrometer;  $^1\text{H}$   $^{13}\text{C}$  NMR and COSY Varian FT-300 spectrometer.

**Plant material.** Leaves of *J. polyanthum* were collected in March 1994, in Tai-chung county, Taiwan. A voucher specimen is kept in Institute of Marine Resources, National Sun Yat-sen University.

**Extraction and isolation.** Fresh leaves (1.1 kg) of *J. polyanthum* were extracted and fractionated as described previously [7]. After partition between *n*-BuOH and  $\text{H}_2\text{O}$ , the  $\text{H}_2\text{O}$ -soluble fraction (10 g) was chromatographed on a silica gel column (150 g) and eluted with the lower layers of mixtures of  $\text{CHCl}_3$ -MeOH- $\text{H}_2\text{O}$  of increasing polarity (540:135:120; 540:135:80; 540:135:40; 505:135:40; 450:150:40; 650:350:100, each 1 l) to yield a residue (1.7 g). Further sepn of the residue by C-18 CC (16 g) developed with a mixt. of  $\text{H}_2\text{O}$  and MeOH (gradient, 5:1, 4:1, 3:1, 2:1, 1:1, each 200 ml) yielded compound **1** (15 mg), methyl-glucoside (32 mg) and oleoside 11 methyl ester (**4**, 46 mg).

**Jaspolyside (1).**  $[\alpha]_D^{24}$ :  $-53^\circ$  (MeOH,  $c = 0.075$ ); UV  $\lambda_{\text{max}}^{\text{MeOH}}$  (log  $\epsilon$ ) nm: 234 (3.68); IR  $\nu_{\text{max}}^{\text{neat}}$   $\text{cm}^{-1}$ : 3460, 1720, 1690, 1640, 1405, 1105, 930;  $^1\text{H}$  NMR ( $\text{CD}_3\text{OD}$ ): Table 1;  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ ): Table 2; FAB-MS  $m/z$  rel. int.: 427 ( $[\text{M} + \text{Na}]^+$ , 4.0), 386 ( $[\text{M} - \text{H}_2\text{O}]^+$ , 16), 225 ( $[\text{M} - \text{Glu}]^+$ , 5.9), 207 ( $[\text{M} - \text{Glu} - \text{H}_2\text{O}]^+$ , 10).

**Jaspolyside tetraacetate (2).** Acetylation ( $\text{Ac}_2\text{O}$ -pyridine, 2:1, rt) of compound **1** (5 mg) gave after work-up **2** (4 mg) as a solid;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  6.12 (1H, s, H-1), 7.42 (1H, s, H-3), 3.60 (1H, m, H-5), 2.86 (1H, dd,  $J = 15.9, 2.7$  Hz, H-6a), 2.55 (1H, dd,  $J = 15.9, 7.8$  Hz, H-6b), 5.72 (1H, q,  $J = 6.1$  Hz, H-8), 1.73 (3H, d,  $J = 6.1$  Hz, H-10), 3.70 (3H, s, COOMe), 1.92, 1.96, 2.01, 2.05 (12H, s, OAc);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ ):  $\delta$  93.9 (*d*, C-1), 152.6 (*d*, C-3), 30.8 (*d*, C-5), 126.1 (*d*, C-8), 133.1 (*s*, C-9), 13.1 (*q*, C-10), 168.4 (*s*, C-11), 51.7 (*q*,  $\text{COOCH}_3$ ), 97.4 (*d*, C-1'), 72.3 (*d*, C-2'), 74.1 (*d*, C-3'), 69.8 (*d*, C-4'), 73.2 (*d*, C-5'), 62.9 (*d*, C-6'),

Table 2.  $^{13}\text{C}$  NMR spectral data\* (75 MHz) for compounds **1** and **4**

C		<b>1</b> $^\dagger$	<b>4</b> $^\dagger$
1	<i>d</i>	93.8	95.5
3	<i>d</i>	154.1	155.2
4	<i>s</i>	112.6	109.8
5	<i>d</i>	33.8	32.1
6	<i>t</i>	38.2	41.4
7	<i>s</i>	176.6	175.4
8	<i>d</i>	125.9	125.0
9	<i>s</i>	132.6	130.7
10	<i>q</i>	13.5	13.8
11	<i>s</i>	168.8	168.8
OMe	<i>q</i>	51.8	52.0
1'	<i>d</i>	100.0	101.1
2'	<i>d</i>	74.8	74.9
3'	<i>d</i>	78.5	78.5
4'	<i>d</i>	71.7	71.6
5'	<i>d</i>	78.1	78.1
6'	<i>t</i>	62.9	62.9

\*Multiplicities determined by DEPT (*s* = C, *d* = CH, *t* =  $\text{CH}_2$ , *q* =  $\text{CH}_3$ ).

$^\dagger$ Measured in  $\text{CD}_3\text{OD}$ .

20.4, 20.5 ( $\times 2$ ), 20.6 (*q*, OAc), 170.7, 171.2, 171.6, 172.2 (*s*, OAc).

*Jaspolyside methyl ester* (**3**). Compound **1** (5 mg) was treated with an excess of  $\text{CH}_2\text{N}_2$  overnight at  $0-5^\circ$ . The reaction mixt. was red. under vacuum to give **3** (3.5 mg);  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  6.25 (1H, *s*, H-1), 7.47 (1H, *s*, H-3), 3.65 (1H, *m*, H-5), 2.89 (1H, *dd*,  $J = 16$ , 3 Hz, H-6a), 2.66 (1H, *dd*,  $J = 16$ , 7.5 Hz, H-6b), 5.63 (1H, *q*,  $J = 6.5$  Hz, H-8), 1.78 (3H, *d*,  $J = 6.5$  Hz, H-10), 3.69 (3H, *s*, 11-COOMe), 3.64 (3H, *s*, 7-COOMe);  $^{13}\text{C}$  NMR ( $\text{CD}_3\text{OD}$ ):  $\delta$  93.7 (*d*, C-1), 154.2 (*d*, C-3), 106.2 (*s*, C-4), 33.7 (*d*, C-5), 37.7 (*t*, C-6), 174.1 (*s*, C-7), 126.1 (*d*, C-8), 132.3 (*s*, C-9), 13.4 (*q*, C-10), 168.7 (*s*, C-11), 51.8 (*q*,  $\text{COOCH}_3$ ), 52.1 (*q*, COOMe), 101.1 (*d*, C-1'), 74.7 (*d*, C-2'), 78.4 (*d*, C-3'), 71.6 (*d*, C-4'), 78.1 (*d*, C-5'), 62.8 (*d*, C-6').

**Acknowledgements**—The authors thank the National Science Council of the Republic of China for financial support (NSC 84-2732-B-110-001 and NSC 85-2113-M-110-006), and Ms Ho Chao Lein and Yu Shiu Ching, NSC southern Instrument Center, National Sun Yat-sen University, for measurement of NMR and mass spectral data.

## REFERENCES

1. Shen, Y. C. and Chen, C. H. (1989) *J. Nat. Prod.* **52**, 1060.
2. Shen, Y. C., Lin, C. Y. and Chen, C. H. (1990) *Phytochemistry* **29**, 2905.
3. Chen, H. C., Shen, Y. C. and Chen, C. H. (1991) *J. Nat. Prod.* **54**, 1087.
4. Shen, Y. C. and Chen, C. H. (1994) *J. Chin. Chem. Soc.* **41**, 473.
5. Zhang, Y. J., Liu, Y. Q., Pu, X. Y. and Yang, C. R. (1995) *Phytochemistry*, **38**, 899.
6. Tanahashi, T., Shimada, A., Nagakura, N., Inoue, K., Ono, M., Fujita, T. and Chen, C. C. (1995) *Chem. Pharm. Bull.* **43**, 729.
7. Shen, Y. C., Lin, S. L. and Chein, C. C. (1996) *J. Chin. Chem. Soc.* **43**, 171.
8. Boros, C. A. and Stermitz, F. R. (1991) *J. Nat. Prod.* **54**, 1173.
9. Takeda, Y., Morimoto, Y., Matsumoto, T., Honda, G., Tabata, M., Fujita, T., Otsuka, H., Sezik, E. and Yesilada, E. (1995) *J. Nat. Prod.* **58**, 1217.
10. Murai, F., Tagawa, M., Damtoft, S., Jensen, S. R. and Nielsen, B. J. (1984) *Chem. Pharm. Bull.* **32**, 2809.
11. Abe, F., Chen, R.-F. and Yamauchi, T. (1988) *Chem. Pharm. Bull.* **36**, 2784.