

CHROMENES FROM *EVODIA LEPTA*

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Key Word Index—*Evodia leptae*; chromenes; leptol A; ethylleptol A; leptene A.

Abstract—Three new chromenes, leptol A, ethylleptol A and leptene A, along with the two known chromenes, isoevodionol and evodione, were isolated from the traditional Chinese herb *Evodia leptae*. The structures were elucidated by spectroscopic analysis and chemical techniques. Copyright © 1997 Elsevier Science Ltd

INTRODUCTION

Evodia leptae, a traditional Chinese herb, is widely used as an antipyretic, anti-inflammatory and analgesic; externally it is used to treat trauma, abscesses, wound infections, eczema, dermatitis and haemorrhoids [1]. Its chemical constituents have only been investigated cursorily [2]. In the present paper, we report the isolation and identification of three new chromenes (**3–5**) and the known ones, isoevodionol (**1**) [3] and evodione (**2**) [4].

RESULTS AND DISCUSSION

Compound **2** was obtained as prisms, $C_{16}H_{20}O_5$ ($[M]^+ m/z$ 292). Its IR spectrum exhibited the presence of a *gem*-dimethyl group (1380 and 1360 cm^{-1}) and a benzene ring (1590 and 1470 cm^{-1}). The electron impact (EI) mass spectrum and melting point were identical to those of evodione, a known compound the structure of which has been elucidated by total synthesis [5] and chemical decomposition [4] in a previous study. Owing to absence of ^1H NMR and IR spectra of evodione in published articles, without further work we could not be sure that **2** and evodione were same compounds. Therefore, we synthesized evodione from isoevodionol and found that the ^1H NMR and IR spectra of evodione were identical to those of **2**. Thus, the structure of **2** was determined and its ^1H NMR and IR spectra are reported for the first time.

Leptol A (**3**) ($C_{16}H_{22}O_5$ [$M]^+ m/z$ 294) was also shown to be a chromene by comparing its spectral

data with those of evodione. Its IR and mass spectra showed the presence of a hydroxyl group (broad band at 3500 cm^{-1} in IR spectrum, and fragment at $[M-15-18]^+$, 22% in the EI-mass spectrum). The ^1H NMR spectrum showed the presence of a $-\text{CH}(\text{OH})\text{CH}_3$ group. Oxidation of **3** using CrO_3 -pyridine [6] gave evodione (**2**) and led to its structure as 6-(1'-hydroxyethyl)-5,7,8-trimethoxy-2,2-dimethyl-2H-[1]-benzopyran.

Ethylleptol A (**4**) and leptene A (**5**) were identified as 6-(1'-ethoxyethyl)-5,7,8-trimethoxy-2,2-dimethyl-2H-[1]-benzopyran and 6-vinyl-5,7,8-trimethoxy-2,2-dimethyl-2H-[1]-benzopyran, respectively, by comparison of their spectral data with those of leptol A (**3**) and chemical reactions. Compound **4** was obtained by ethylation of **3** using the $\text{EtOH}-\text{HCO}_2\text{H}$ method. Dehydrolysis of **3** with formic acid in chloroform (refluxed for two hours) afforded **5**, the ^1H NMR, IR and mass spectra of which were identical to those of the natural product.

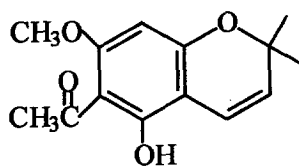
EXPERIMENTAL

General. Mps: uncorr. EI-MS: 70 eV, direct inlet. ^1H NMR: 400 MHz, CDCl_3 , chemical shifts are given in δ and refer to CDCl_3 in the residual CHCl_3 (δ 7.24). ^{13}C NMR: 100 MHz, CD_3COCD_3 . Chemical shifts are given in δ and refer to CD_3COCD_3 in the residual Me_2CO (δ 29.8).

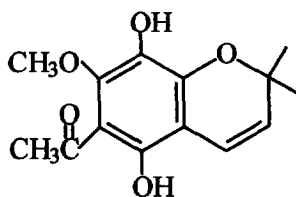
Plant material. Aerial parts of *E. leptae* (Spr.) Merr. were collected from Hainan province, P.R. China, in July 1992. A voucher sample is deposited in the Herbarium of Shanghai Institute of Materia Medica, Chinese Academy of Sciences.

Extraction and isolation. Dried and powdered plant material (10 kg) was extracted $\times 2$ with 95% EtOH at

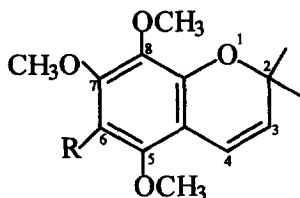
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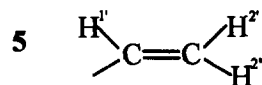
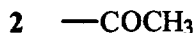
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room temp. over 2 weeks. The combined extracts were evapd to dryness under red. pres. (35°) and the residue (250 g) obtained subjected to CC over silica gel, eluting with petrol-EtOAc (10:1) to give an orange oil (80 g). Part of this oil (20 g) was fractionated by silica gel CC eluting with a petrol-EtOAc gradient. The frs obtained were repeatedly chromatographed by silica gel CC using petrol-EtOAc mixts to give the following chromenes (in increasing order of chromatographic polarity): **5** (37 mg), **1** (1504 mg), **4** (327 mg), **2** (1390 mg), **3** (2517 mg).

Evodione (2). C₁₆H₂₀O₅. Prisms, mp 57° (in EtOAc). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3040, 2970, 2940, 2840, 1710, 1590, 1470, 1378, 1364, 1240, 1060. ¹H NMR: δ 6.47 (1H, *d*, *J* = 10.0 Hz, H-4), 5.59 (1H, *d*, *J* = 10.0 Hz, H-3), 3.86 (3H, *s*, -OCH₃), 3.82 (3H, *s*, -OCH₃), 3.70 (3H, *s*, -OCH₃), 2.48 (1H, *s*, -COCH₃), 1.46 (6H, *s*, *gem*-dimethyl). EI-MS *m/z* (rel. int.): 292 [M]⁺ (17), 277 (100), 247 (18).

Leptol A (3). C₁₆H₂₂O₅. Orange oil. [α]_D²⁶ = +1.18° (CHCl₃; *c* 0.398). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3500, 3040, 2960, 2930, 2830, 1740, 1590, 1465, 1374, 1360, 1230. ¹H NMR: δ 6.46 (1H, *d*, *J* = 10.0 Hz, H-4), 5.57 (1H, *d*, *J* = 10.0 Hz, H-3), 5.08 (1H, *q*, *J* = 6.6 Hz, H-1'), 3.97 (3H, *s*, -OCH₃), 3.81 (3H, *s*, -OCH₃), 3.73 (3H, *s*, -OCH₃), 1.52 (3H, *d*, *J* = 6.6 Hz, H-2'), 1.48 (3H, *s*, CH₃-2), 1.42 (3H, *s*, CH₃-2). ¹³C NMR: δ 152.9, 150.2, 146.9, 139.6, 130.0 (C-3), 124.6, 117.9 (C-4), 112.5, 76.9 (C-2), 63.9 (C-1'), 63.5 (-OCH₃), 62.0 (-OCH₃), 61.0 (-OCH₃), 27.9 (*gem*-dimethyl), 24.7 (C-2'). EI-MS *m/z* (rel. int.): 294 [M]⁺ (16), 279 (100), 261 (21), 249 (28), 231 (9).

Ethylleptol A (4). C₁₈H₂₆O₅. Oil. [α]_D²⁶ = -1.21° (CHCl₃; *c* 0.347). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3040, 2970, 2930, 2840, 1590, 1465, 1375, 1360, 1232, 1030. ¹H NMR: δ 6.47 (1H, *d*, *J* = 10.0 Hz, H-4), 5.54 (1H, *d*, *J* = 10.0

Hz, H-3), 4.81 (H, *q*, *J* = 6.6 Hz, H-1'); 3.82 (3H, *s*, -OCH₃), 3.81 (3H, *s*, -OCH₃), 3.68 (3H, *s*, -OCH₃), 3.36 (2H, *m*, H-1''), 1.57 (3H, *d*, *J* = 6.6 Hz, H-2'), 1.45 (3H, *s*, CH₃-2), 1.42 (3H, *s*, CH₃-2), 1.14 (3H, *t*, *J* = 7.0 Hz, H-2''). EI-MS *m/z* (rel. int.): 322 [M]⁺ (18), 307 (100), 277 (16), 263 (16), 247 (10), 217 (6).

Leptene A (5). C₁₆H₂₀O₄. Pale yellow oil. IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3010, 2960, 2920, 2860, 1585, 1465, 1385, 1366, 1230, 1050, 980, 910. ¹H NMR: δ 6.72 (1H, *dd*, *J*₁ = 18.1 Hz, *J*₂ = 11.8 Hz, H-1'), 6.54 (1H, *d*, *J* = 10.1 Hz, H-4), 6.00 (1H, *dd*, *J*₁ = 18.1 Hz, *J*₂ = 2.3 Hz, H-2''), 5.57 (1H, *d*, *J* = 10.1 Hz, H-3), 5.35 (1H, *dd*, *J*₁ = 11.8 Hz, *J*₂ = 2.3 Hz, H-2'), 3.83 (3H, *s*, -OCH₃), 3.82 (3H, *s*, -OCH₃), 3.66 (3H, *s*, -OCH₃), 1.45 (6H, *s*, *gem*-dimethyl). EI-MS *m/z* (rel. int.): 276 [M]⁺ (83), 261 (100), 245 (23), 231 (43), 213 (9).

5,8-Dihydroxy-7-methoxy-6-acetyl-2,2-dimethyl-2H-[1]-benzopyran (6). Emerald prisms, mp 112° (petrol-EtOAc). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400, 2970, 2920, 1599, 1460, 1380, 1365, 1243, 1090. ¹H NMR: δ 13.5 (1H, *s*, OH-5), 6.67 (1H, *d*, *J* = 10.0 Hz, H-4), 5.50 (1H, *d*, *J* = 10.0 Hz, H-3), 3.96 (3H, *s*, -OCH₃), 2.64 (3H, *s*, -COCH₃), 1.47 (6H, *s*, *gem*-dimethyl). HR-MS: [M]⁺ *m/z* 264.0980 (calcd for C₁₄H₁₆O₅ 264.0997). EI-MS *m/z* (rel. int.): 264 [M]⁺ (28), 249 (100), 234 (22), 216 (9), 140 (9).

Synthesis of evodione from isoevodionol. 1. **Preparation of 5,8-dihydroxy-7-methoxy-6-acetyl-2,2-dimethyl-2H-[1]-benzopyran (6)** [7]. Isoevodionol (140 mg, 0.56 mmol) was added to 2 ml aq. soln of KOH (158 mg, 2.82 mmol) and a small quantity of pyridine added to the soln until the material dissolved completely. Then, 3 ml aq. soln of K₂S₂O₈ (305 mg, 1.13 mmol) was slowly added over 2 hr at 0°. The mixt. was stirred continuously overnight at room temp., Et₂O (3 ml) added the next day and the mixt. acidified with excess

3 N HCl with stirring for 10 min. The aq. layer was extracted $\times 2$ with Et₂O. The combined Et₂O layers were washed with 5% NaHCO₃ soln, dried (Na₂SO₄), evapd *in vacuo* and subjected to CC over silica gel to obtain **6**. 2. *Methylation of 6 to give evodione* [8]. Compound **6** (12 mg) in dry Me₂CO (2 ml) was refluxed with Me₂SO₄ (50 μ l) in the presence of dry K₂CO₃ (200 mg) for 5 hr, then filtered, inorganic salts washed out with hot Me₂CO and combined, filtered and concd. The residue was subjected to CC eluting with petrol–EtOAc (10:1) to give evodione (**2**) (5 mg). HR-MS: [M]⁺ *m/z* 292.1313 (calcd for C₁₆H₂₀O₅, 292.1311). IR, MS and ¹H NMR data identical to natural product.

Synthesis of 4 from 3. Compound **3** (30 mg) was dissolved in EtOH (3 ml) and HCO₂H (0.5 ml) added to the soln. The mixt. was stirred continuously for 5 hr at room temp., neutralized with satd NaHCO₃ soln, extracted with Et₂O, the organic layer washed with brine and dried (Na₂SO₄). Removal of solvent gave a residue which was subjected to CC using petrol–

EtOAc (8:1) to furnish **4**, whose IR, ¹H NMR and EI-MS were identical to those of natural product.

REFERENCES

1. Jiangsu New Medical College, *The Dictionary of Chinese Herb*, Vol. 1. China, 1986, p. 68.
2. Gunawardana, Y. A. G. P. and Cordell, G. A., *Journal of the Science Society of Thailand*, 1987, **13**, 107.
3. Allan, R. D., Correll, R. L. and Well, R. J., *Tetrahedron Letters*, 1969, **53**, 4673.
4. Wright, S. E., *Journal of the Chemical Society*, 1948, 2005.
5. Huls, R. and Brunelle, S., *Bulletin de la Societe Chimique, Belgique*, 1959, **68**, 325.
6. Ratcliffe, R. and Rodehorst, R., *Journal of Organic Chemistry*, 1970, **35**, 4000.
7. Sethna, S. M., *Chemical Reviews*, 1951, **49**, 91.
8. Ahluwalia, V. K., Arora, K. K. and Manchanda, S., *Indian Journal of Chemistry*, 1982, **21B**, 1093.