

GERMACRANES AND FLAVONOIDS FROM *ACHILLEA AGERATUM*

LUIS M. VIEIRA,* ANAKE KIJJOA,* JOSÉ A. PEREIRA,* THOMAS E. GEDRIS† and WERNER HERZ†‡

*Centro de Estudos de Química Orgânica, Fitoquímica e Farmacologia de Universidade do Porto, Instituto de Ciências Biomédicas Abel Salazar, 4000 Porto, Portugal; †Department of Chemistry, The Florida State University, Tallahassee, FL 32306-3006, U.S.A.

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Key Word Index—*Achillea ageratum*; Anthemideae; Compositae; ageratriol derivatives; germacranes; flavonoids.

Abstract—Aerial parts of *Achillea ageratum* yielded, in addition to ageratriol 9-*O*-acetylageratriol, 1-dehydroageratriol, a 1(10)-epoxygermacra-5,9-diol, and its monoacetate, as well as 5,4'-dihydroxy-3,7-dimethoxyflavone, chrysosplenetin, penduletin, hispidulin and cirsiolol. © 1997 Elsevier Science Ltd. All rights reserved

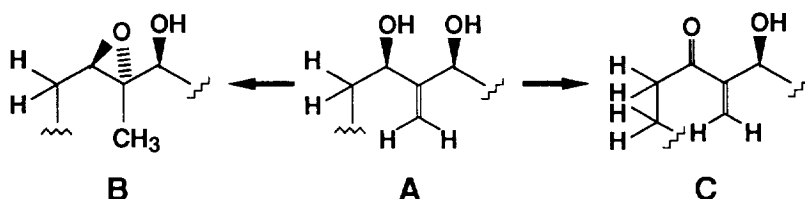
INTRODUCTION

More than 20 years ago, Trave and co-workers reported the isolation of the germacranes ageratriol (**1a**) [1–3] from the whole plant and agerol (**2**) [4, 5] from the flowers of Sardinian *Achillea ageratum* L. Biosynthesis of **1a** apparently proceeds via agerol diepoxide rather than by a process mimicking the reaction of **2** with singlet oxygen followed by reduction of the intermediate 1,5-hydroperoxide [5, 6]. The essential oil produced by steam distillation contained approximately 60% 1,8-cineole and some **3**, presumably produced by a Cope rearrangement of **2** [7]. Other workers have reported identification of *E*-dehydromatricaria ester in the root extract [8] and penduletin and vicenin in the leaf [9] of *A. ageratum* collections, the provenance of which was not specified.

RESULTS AND DISCUSSION

Examination of a collection of *A. ageratum* from Mafra, Portugal, has now yielded, in addition to ageratriol, a monoacetate **1b**, an epoxyacetate **4a**, a 2:1 mixture of 1-dehydroageratriol (**5**) and epoxyalcohol **4c** and the flavonoids **6a**, **6b** (chrysosplenetin), **6c** (penduletin), **7a** (hispidulin) and **7b** (cirsiolol). Previously reported ¹H NMR data for ageratriol (**1a**), which is

insoluble in CDCl₃, and its triacetate **1c** are sparse [1] and not useful for comparison with the new compounds. Significant signals in the ¹H NMR spectra of **1a** in DMSO-*d*₆ and **1c** in DMSO-*d*₆ and CDCl₃ are, therefore, included in Table 1, together with those of the new compounds, while the ¹³C NMR spectra are listed in Table 2. The ¹H NMR spectra of **1a–1c** and **5** exhibit three pairs of singlets, characteristic of *exo*-methylenes. Thus, the pair at highest field near δ 4.6, allylically coupled to a vinyl methyl near δ 1.65, is attributable to H-13a, b, while a second pair, which although slightly affected by acetylation remains essentially constant near δ 5.2 throughout the series, can be assigned to H-15a, b. The pair at lowest field (H-14a, b) is replaced in the spectra of monoacetate **4a**, synthetic diacetate **4b** and epoxyalcohol **4c** by a three-proton singlet in the range δ 1.33–1.42, characteristic of methyl on carbon under oxygen. Simultaneously, a *dd* at δ 4.21 (–CHOH) in the spectrum of monoacetate **1b** and δ 4.98 (or 5.00) in the spectrum of triacetate **1c** (–CHOAc) has moved upfield to δ 3.11 (from δ 3.17), thus indicating the conversion of partial structure A to B where the CH₂– group could represent C-2 or C-8, but not C-6 (see below). In the spectrum of dehydroageratriol (**5**) the chemical shifts of H-15a, b are not altered, while the paramagnetic shifts of



‡Author to whom correspondence should be addressed.

Table 1. Partial ^1H NMR data for compounds **1a–1c**, **4a–4c** and **5** (500 MHz, CDCl_3)

H	1a *	1b	1c *	1c †	4a	4b †	4c ‡	5 ‡
1	3.87 <i>dt</i> * (10, 5.5)	4.21 <i>dd</i> (11, 5)	4.85 <i>dd</i> (10, 5)	4.98 <i>dd</i> (11, 5.5)	3.11 <i>dd</i> (10, 4.5)	3.17 <i>dd</i> (10, 4.5)	obsc.	—
5	3.64 <i>dt</i> ‡ (8.5, 3.5)	3.90 <i>dd</i> (12, 3.5)	4.88 <i>dd</i> (11, 4)	5.00 <i>dd</i> (11, 4.5)	3.96 <i>dd</i> (12, 3)	5.01 <i>dd</i> (12, 3)	3.94 <i>dd</i> (12, 2)	3.98 <i>dd</i> (10.5, 4.5)
9	3.87 <i>dt</i> * (10, 5)	4.70 <i>dd</i> (12, 5.5)	5.47 <i>dd</i> (11, 3)	5.58 <i>dd</i> (11.5, 3)	4.24 <i>dd</i> (12, 5)	4.31 <i>dd</i> (12, 5)	Obsc.	4.78 <i>dd</i> (10.5, 5.5)
12§	1.62 <i>brs</i>	1.68 <i>brs</i>	1.65 <i>brs</i>	1.68 <i>brs</i>	1.68 <i>brs</i>	1.69 <i>brs</i>	1.65 <i>brs</i>	1.64 <i>brs</i>
13a	4.64 <i>brs</i>	4.67 <i>brs</i>	4.72 <i>brs</i>	4.77 <i>brs</i>	4.68 <i>brs</i>	4.75 <i>brs</i>	4.66 <i>brs</i>	4.63 <i>brs</i>
13b	4.64 <i>brs</i>	4.65 <i>brs</i>	4.72 <i>brs</i>	4.74 <i>t</i> (1)	4.65 <i>brs</i>	4.74 <i>brs</i>	4.62 <i>brs</i>	4.59 <i>brs</i>
14a	5.29 <i>s</i>	5.42 <i>brs</i>	5.57 <i>s</i>	5.60 <i>s</i>	1.37 <i>s§</i>	1.42 <i>s§</i>	1.33 <i>s§</i>	6.06 <i>s</i>
14b	5.16 <i>s</i>	5.36 <i>brs</i>	5.57 <i>s</i>	5.50 <i>s</i>				5.94 <i>s</i>
15a	4.96 <i>s</i>	5.16 <i>brs</i>	5.25 <i>s</i>	5.34 <i>s</i>	5.17 <i>brs</i>	5.32 <i>brs</i>	5.09 <i>brs</i>	5.16 <i>brs</i>
15b	4.94 <i>s</i>	5.12 <i>br</i>	5.24 <i>s</i>	5.32 <i>s</i>	5.16 <i>brs</i>	5.31 <i>brs</i>	5.09 <i>brs</i>	5.09 <i>brs</i>
Ac§	—	2.01 <i>s</i>	1.98 <i>s</i> , 1.96 <i>s</i> 1.93 <i>s</i>	2.09 <i>s</i> , 2.01 <i>s</i> 1.98 <i>s</i>	2.06 <i>s</i>	2.07 <i>s</i> 2.01 <i>s</i>		

*At 300 MHz, $\text{DMSO}-d_6$.†At 300 MHz, CDCl_3 .‡From an approximately 1 : 2 mixture of **4c** and **5**.

§Intensity three protons.

*Coupled to $-\text{OH}$ signals at 4.69 *d* (5) and 4.60 *d* (5).‡Coupled to $-\text{OH}$ signal at 4.51 *d* (3.5).Table 2. ^{13}C NMR spectra of compounds **1a–1c**, **4a–4c** and **5** (75 MHz)

C	1a *	1b *	1c †	1c *	4a †	4b †	4c *†	4c †‡	5 *†	5 †‡
1	76.6 <i>d</i>	74.1 <i>d</i> ^a	76.2 <i>d</i> ^a	77.1 <i>d</i>	59.9 <i>d</i>	59.3 <i>d</i>	58.8 <i>d</i>	60.5 <i>d</i>	203.3 <i>s</i>	203.4 <i>s</i>
2	32.1 <i>t</i>	31.6 <i>t</i>	29.9 <i>t</i>	28.0 <i>t</i>	25.7 <i>t</i>	25.5 <i>t</i>	25.5 <i>t</i>	26.0 <i>t</i>	38.6 <i>t</i>	39.1 <i>t</i>
3	22.5 <i>t</i>	21.7 <i>t</i>	21.8 <i>t</i>	22.6 <i>t</i>	21.1 <i>t</i>	22.9 <i>t</i>	22.0 <i>t</i>	22.0 <i>t</i>	24.8 <i>t</i>	25.7 <i>t</i>
4	150.4 <i>s</i>	150.7 <i>s</i>	150.0 <i>s</i>	143.3 <i>s</i>	147.3 <i>s</i> ^a	142.7 <i>s</i>	148.2 <i>s</i>	147.7 <i>s</i> ^a	148.1 <i>s</i> ^a	146.6 <i>s</i> ^a
5	69.3 <i>d</i>	69.4 <i>d</i>	70.7 <i>d</i>	73.2 <i>d</i>	75.6 <i>d</i>	77.6 <i>d</i>	76.0 <i>d</i>	76.0 <i>d</i>	66.7 <i>d</i>	69.1 <i>d</i>
6	36.9 <i>t</i>	36.5 <i>t</i>	36.4 <i>t</i>	33.8 <i>t</i>	35.9 <i>t</i>	33.7 <i>t</i>	36.4 <i>t</i>	35.8 <i>t</i>	37.4 <i>t</i>	37.6 <i>t</i>
7	37.7 <i>d</i>	37.6 <i>d</i>	37.9 <i>d</i>	37.1 <i>d</i>	38.4 <i>d</i>	38.1 <i>d</i>	38.1 <i>d</i>	38.9 <i>d</i>	36.8 <i>d</i>	37.8 <i>d</i>
8	44.4 <i>t</i>	39.8 <i>t</i>	40.1 <i>t</i>	39.6 <i>t</i>	36.0 <i>t</i>	36.1 <i>t</i>	39.7 <i>t</i>	38.4 <i>d</i>	44.0 <i>t</i>	42.5 <i>t</i>
9	72.0 <i>d</i>	74.5 <i>d</i> ^a	77.2 <i>d</i> ^a	70.4 <i>d</i>	80.9 <i>d</i>	80.7 <i>d</i>	78.1 <i>d</i>	79.9 <i>d</i>	73.7 <i>d</i>	75.3 <i>d</i>
10	154.4 <i>s</i>	149.5 <i>s</i>	147.3 <i>s</i> ^b	145.7 <i>s</i>	59.4 <i>s</i>	59.9 <i>s</i>	61.2 <i>s</i>	62.2 <i>s</i>	153.0 <i>s</i>	152.3 <i>d</i>
11	148.7 <i>s</i>	148.0 <i>s</i>	147.8 <i>s</i> ^b	146.9 <i>s</i>	146.4 <i>s</i> ^a	146.0 <i>s</i>	148.2 <i>s</i>	147.5 <i>s</i> ^a	147.4 <i>s</i> ^a	147.8 <i>s</i> ^a
12	18.4 <i>q</i>	18.2 <i>q</i>	18.0 <i>q</i>	18.0 <i>q</i>	18.2 <i>q</i>	18.2 <i>q</i>	18.3 <i>q</i>	18.1 <i>q</i>	18.0 <i>q</i>	18.2 <i>q</i>
13	110.6 <i>t</i>	110.1 <i>t</i>	111.0 <i>t</i>	111.0 <i>t</i>	111.2 <i>t</i>	111.7 <i>t</i>	110.0 <i>t</i>	111.0 <i>t</i>	109.7 <i>t</i>	110.8 <i>t</i>
14	109.7 <i>t</i>	113.6 <i>t</i>	114.7 <i>t</i>	117.5 <i>t</i>	11.5 <i>q</i>	11.6 <i>q</i>	10.6 <i>q</i>	10.6 <i>q</i>	122.6 <i>t</i>	123.1 <i>t</i>
15	112.8 <i>t</i>	113.2 <i>t</i>	115.4 <i>t</i>	117.5 <i>t</i>	115.9 <i>t</i>	118.4 <i>t</i>	113.9 <i>t</i>	116.1 <i>t</i>	114.9 <i>t</i>	116.9 <i>t</i>
Ac		169.5 <i>s</i> 21.0 <i>q</i>	171.7 <i>s</i> 22.0 <i>q</i>	170.1 <i>s</i> , 169.1 <i>s</i> , 168.9 <i>s</i> , 21.0 <i>q</i> , 20.9 <i>q</i> , 20.9 <i>q</i>	170.0 <i>s</i> 21.2 <i>q</i>	169.9 <i>s</i> , 169.7 <i>s</i> 21.2 <i>q</i> , 21.1 <i>q</i>				

*In $\text{DMSO}-d_6$.†In CDCl_3 .‡From mixture of **4c** (minor) and **5** (major).^{a,b}In the same column, interchangeable signals.

H-14a, b are indicative of conjugation with the new carbonyl group, the presence of which is also responsible for strongly deshielding a proton under a hydroxyl as in C, of necessity at C-1 if the carbonyl is at C-9, or at C-9 if the carbonyl group is at C-1. That the latter arrangement is correct is shown by the ^1H NMR spectrum, where two mutually coupled protons at relatively low field (δ 3.25 and 2.71) are each

coupled to a second pair of mutually coupled protons at δ 2.57 and 2.44; this is only possible if the carbonyl group is at C-1. Consequently, the ketodiols is identified as **5**, which is reflected in the ^{13}C NMR spectrum by pronounced paramagnetic shifts of three triplets (b) now assignable to C-2, C-3 and C-14 when compared with **1a** in the same solvent ($\text{DMSO}-d_6$).

In the case of epoxide monoacetate **4a** and the

minor constituent of the **4c**–**5** mixture, superposition of ^1H NMR signals presented considerable difficulties. However, it could be shown that the epoxidic signal of **4a** at δ 3.11 was vicinally coupled to two neighbouring protons at δ 2.41 and 1.5, the former of which exhibited three additional couplings, thus leading to formula **4a** for the monoacetate and **4c** for the diol with the epoxide closed from C-1 to C-10. Acetylation of **4a** produced the diacetate **4b**. In the ^1H NMR spectrum this was accompanied by a paramagnetic shift of the signal at δ 3.96 associated with H-5 (compare with **1b** and **5**) to δ 5.01. Thus the *dd* of **4a** at δ 4.24 (δ 4.31 in **4b**) which was unaffected by acetylation, had to be associated with H-9 under the acetate originally present in **4a**, and its high-field position can be ascribed to shielding by the epoxide (compare with H-9 at δ 3.20 of unacetylated C-9 in the NMR spectrum of synthetic agerol diepoxide [9]). The carbon shift changes accompanying the conversions **4c** $\rightarrow$ **4a** $\rightarrow$ **4b** when measured in the same solvent are also in accordance with structure **4a** of the epoxide monoacetate.

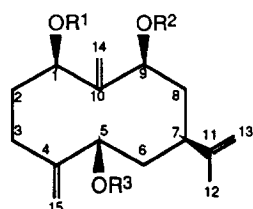
Finally, we consider the location of the acetate moiety in the naturally occurring monoacetate of ageratriol. Clearly, the acetate was not at C-5 (Tables 1 and 2) but superposition of signals in the high-field

region of the ^1H NMR spectrum again caused difficulties. However, the signal at δ 4.21 was clearly coupled to multiplets at δ 2.23 and 2.04, at least one of which was further coupled to multiplets at δ 2.17 and 2.08, thus indicating that the signal at δ 4.21 represents H-1. Hence the signal at lower field (δ 4.70) had to be ascribed to C-9, from which the inference could be drawn that the acetate was at C-9. The changes in the ^{13}C NMR spectra in the same solvent, in particular the upfield shift of the C-8 frequency accompanying the change from **1a** to **1b** and the upfield shift of the C-2 resonance in going from **1b** to **1c**, are in accordance with this formulation.

EXPERIMENTAL

Plant material. *Achillea ageratum* L. was collected in Mafra, Portugal, in July 1992. A voucher specimen was deposited in the Herbarium of the Instituto Superior de Agronomia, Lisbon.

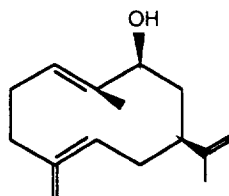
Extraction and isolation. Powdered dry aerial parts (2 kg) were percolated with MeOH at room temp. to exhaustion. Evaporation of the solution at red. pres. gave 500 g of crude material, which was dissolved in CHCl_3 , filtered and evaporated at red. pressure. The



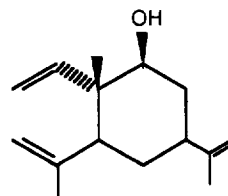
1 a $\text{R}^1, \text{R}^2, \text{R}^3 = \text{H}$

b $\text{R}^1, \text{R}^3 = \text{H}, \text{R}^2 = \text{Ac}$

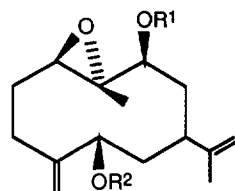
c $\text{R}^1, \text{R}^2, \text{R}^3 = \text{Ac}$



2



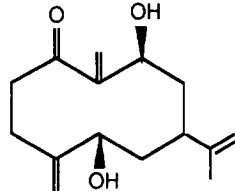
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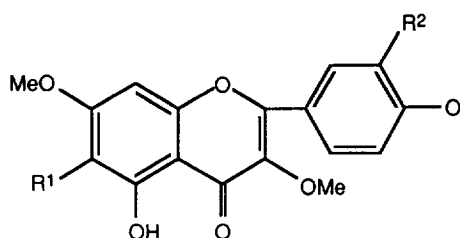
4 a $\text{R}^1 = \text{Ac}, \text{R}^2 = \text{H}$

b $\text{R}^1, \text{R}^2 = \text{Ac}$

c $\text{R}^1, \text{R}^2 = \text{H}$



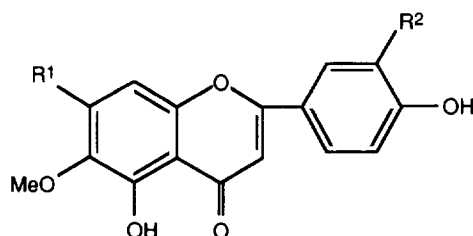
5



6 a $\text{R}^1, \text{R}^2 = \text{H}$

b $\text{R}^1 = \text{OMe}, \text{R}^2 = \text{OH}$

c $\text{R}^1 = \text{OMe}, \text{R}^2 = \text{H}$



7 a $\text{R}^1 = \text{OH}, \text{R}^2 = \text{H}$

b $\text{R}^1 = \text{OMe}, \text{R}^2 = \text{OH}$

residue (115 g) was taken up in hot EtOH, mixed with hot H₂O containing Pb(OAc)₂, and HOAc, allowed to stand overnight, filtered, concentrated at red. pressure to remove EtOH, and extracted with CHCl₃. Concentration of the dried extract in vacuo furnished 59 g of residue which was chromatographed on silica gel 60 (200 g), eluents petrol-CHCl₃ and CHCl₃-Me₂CO (1000 ml frs) as follows: Frs 1-35 (petrol-CHCl₃, 4:1), frs 36-84 (petrol-CHCl₃, 3:2), frs 85-108 (petrol-CHCl₃, 2:3), frs 109-129 (petrol-CHCl₃, 1:4), frs 130-155 (CHCl₃), frs 156-186 (CHCl₃-Me₂CO, 9:1), frs 187-207 (CHCl₃-Me₂CO, 4:1), frs 208-231 (CHCl₃-Me₂CO, 7:3), frs 232-248 (CHCl₃-Me₂CO, 3:2), frs 249-271 (CHCl₃-Me₂CO, 1:1), frs 272-281 (CHCl₃-Me₂CO, 2:3), and frs 282-294 (CHCl₃-Me₂CO, 1:4). Frs 1-47 contained non-polar material which was not studied further. Frs 48-53 were combined (1.2 g) and subjected to CC (silica gel 60, 129), 50-ml subfrs being collected as follows: subfrs 1-30 (petrol-CHCl₃, 4:1), subfrs 31-40 (petrol-CHCl₃, 3:2), subfrs 41-60 (petrol-CHCl₃, 1:4), subfrs 61-66 (CHCl₃-Me₂CO, 9:1), subfr 67 (Me₂CO), subfr 68 (MeOH). Subfrs 17-30 (560 mg) were combined and purified by prep. TLC (silica gel, petrol-EtOAc-HCO₂H, 70:30:1) to give 296 mg of **4a**. Frs 54-59 (1 g) were combined, but were not purified further. Frs 60-63 (0.6 g) were combined; recryst. from petrol-CHCl₃ afforded 0.4 g of chrysopenetin (**6b**) identified by MS, UV and NMR spectrometry and conversion of 30 mg of **6b** to 12 mg of 4'-acetoxy-5-hydroxy-3,6,7,3'-tetramethoxyflavone, ¹H NMR (DMSO-*d*₆) δ 12.45 (*brs*, 5-OH), 7.77 (*d*, *J* = 2 Hz, H-2'), 7.69 (*dd*, *J* = 8.5 and 2 Hz, H-6'), 7.33 (*d*, *J* = 8.5 Hz, H-5'), 6.97 (*s*, H-8), 3.93, 3.89, 3.87, 3.74 (each *s* and 3p, -OMe), 2.40 and 2.31 (each *s* and 3p, 4'-OAc), and 16 mg of 5,4'-diacetoxy-3,6,7,3'-tetramethoxyflavone, ¹H NMR (DMSO-*d*₆) δ 7.76 (*d*, *J* = 2 Hz, H-2'), 7.67 (*dd*, *J* = 8.5, 2 Hz, H-6'), 7.38 (*s*, H-8), 7.32 (*d*, *J* = 8.5 Hz, H-5), 3.99, 3.89, 3.78, 3.74 (each *s* and 3p, -OMe), 2.40, 2.31 (each *s* and 3p, 4'-OAc). Purification of the mother liquor of **6b** by reverse phase PTLC (RP-18, MeOH-H₂O, 7:3) afforded an additional 20 mg of **4a**. Frs 64-84 were combined, but contained a mixture of components with similar *R*_f values and were not examined further.

Frs 85-87 (0.8 g) were rechromatographed (silica gel 60), 50-ml subfrs being collected as follows: subfrs 1-13 (petrol-CHCl₃, 4:1), subfrs 14-50 (petrol-CHCl₃, 3:2), subfrs 51-72 (petrol-CHCl₃, 2:3). Purification of subfrs 14-46 by TLC (silica gel, toluene-EtOAc-Me₂CO-HCO₂H, 60:35:5:1) gave 0.32 g of **1b**. Recryst. of fr. 88 gave 27 mg of flavone **6a** identified by MS, UV and NMR spectrometry. Frs 100-111 (2.3 g) were combined. Purification by TLC (silica gel, CHCl₃-Me₂CO-HCO₂H, 90:10:1) afforded 56 mg of penduletin (**6c**) identified by UV and NMR spectrometry and conversion of 24 mg of the flavone to 4'-acetoxy-5-hydroxy-3,6,7-trimethoxyflavone (12 mg), ¹H NMR (DMSO-*d*₆) δ 12.45 (*brs*, 5-OH), 8.13 (*d*, *J* = 8.5 Hz, H-3' and H-5'), 7.83

(*d*, *J* = 8.5 Hz, H-2' and H-6'), 6.96 (*s*, H-8) 3.93 (*s*, 3p, 7-OMe), 3.85 (*s*, 3p, 3-OMe), 3.74 (*s*, 3p, 6-OMe), 2.33 (*s*, 3p, OAc), and 5,4'-diacetoxy-2,6,7-trimethoxyflavone (8 mg), ¹H NMR (DMSO-*d*₆) δ 8.11 (*d*, 2p, *J* = 9 Hz, H-3' and H-5'), 7.37 (*s*, H-8), 7.37 (*d*, 2p, *J* = 9 Hz, H-2' and H-6'), 3.98 (*s*, 3p, 7-OMe) 3.76 (*s*, 3p, 3-OMe), 3.73 (*s*, 3p, 7-OMe), 2.39 and 2.32 (both *s* and 3p, -OMe). Frs 112-126 contained a mixture of components with similar *R*_f values and were not studied further.

Frs 127-136 were combined and recryst. from MeOH to give 20 mg of hispidulin (**7a**) identified by UV and NMR spectrometry. Purification of the mother liquor by TLC (silica gel, toluene-EtOAc-Me₂CO-HCO₂H, 60:35:5:1) gave a 2:1 mixture of **5** and **4c**. Frs 137-147 contained a mixture of components with similar *R*_f values and were not studied further. Frs 142-155 (2.5 g) were combined and recryst. from CHCl₃ to give cirsiolol (**7b**, 150 mg) identified by MS, UV and NMR spectrometry and conversion of 20 mg to 17 mg of 3',4'-diacetoxy-5-hydroxy-3,6,7-trimethoxyflavone, ¹H NMR (DMSO-*d*₆) δ 12.41 (*brs*, 5-OH), 8.06 (*d*, *J* = 8.5 Hz, H-6'), 7.98 (*s*, H-2'), 7.52 (*d*, *J* = 8.5 Hz, H-5'), 3.92, 3.87, 3.74 (each *s* and 3p, -OMe), 2.34 (*s*, 6p, 2-OAc). TLC (silica gel, toluene-EtOAc-Me₂CO-HCO₂H, 60:35:5:1) of the mother liquor afforded an additional 70 mg of **7b**. Frs 156-167 contained mixtures of compounds with similar *R*_f. Frs 168-186 (2 g) were combined. Recryst. from Me₂CO afforded 168 mg of pure ageratriol (**1a**).

Ageratriol (1a). Crystals. Found: mp 191-193°; [α]_D²⁰ + 30° (*c* 0.005 g ml⁻¹, MeOH). Lit.[1]: mp 195°, [α]_D²⁰ + 30.5° (*c* 2 MeOH). MS PCI *m/z* (rel. int.) 253 ([M + H]⁺ 8), 235 (50), 217 (100), 199 (89), 189 (58), 157 (37), 147 (38), 133 (44), 121 (49), 107 (12). ¹H NMR in Table 1; ¹³C NMR in Table 2.

Acetylation of **1a** (22 mg) with Ac₂O-Py and work-up in the usual manner afforded the triacetate **1c** (16 mg) as a viscous liquid. ¹H NMR spectrum in Table 1; ¹³C NMR in Table 2.

(1R*,5R*,7S*,9S*)-9-Acetoxy-1,5-dihydroxygermacra-4(15),-10(14),11(13)-triene (ageratriol 9-acetate) (**1b**). Viscous liquid. [α]_D²⁰ -18° (*c* 0.0006 g ml⁻¹; CHCl₃). MS PCI *m/z* (rel. int.) 295 ([M + H]⁺ 8.7), 277 (23.5), 235 (53.4), 233 (27.3), 217 (100), 199 (25.3), 197 (68.0). ¹H NMR (500 MHz, CDCl₃) in Table 1; ¹³C NMR in Table 2.

(1R*,5R*,7S*,9S*,10S*)-9-Acetoxy-1(10)-epoxy-5-hydroxygermacra-4(15),11(13)-diene (**4a**). Viscous liquid: [α]_D²⁰ -46° (*c* 0.0041 g ml⁻¹; CHCl₃). MS PCI (rel. int.) 295 ([M + H]⁺ 18.9), 277 (100), 253 (32.8), 235 (66.1), 217 (42.8). IR (film), ν_{max} , cm⁻¹: 3600-3200, 3070, 2940, 2870, 1740, 1645, 1460, 1375, 1240, 1080, 1030, 995, 905, 855. ¹H NMR in Table 1; ¹³C NMR in Table 2.

Acetylation of **4a** (20 mg) with Ac₂O-pyridine followed by the usual work-up furnished 13 mg of **4b**. ¹H NMR in Table 1; ¹³C NMR in Table 2.

Mixture of (5R*,7S*,9S*)-5,9-dihydroxy-1-oxogermacra-4(15),10(14),11(13)-triene and (1R*,5R*,7S*,

9S*,10S*)-5,9-dihydroxy-1(10)-epoxygermacra-4(15), 11(13)-diene (**5** and **4c**). MS (PCI) m/z (rel. int.) 251 ($[M + H^+ \text{ of } \mathbf{5}]$ 1), 233 (16), 215(8), 173(5), 121(S), 85(11), 83(23), 57(77), 43(100). ^1H NMR (500 MHz, CDCl_3) in Table 1; ^{13}C NMR of mixture in Table 2.

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