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CYTOTOXIC FLAVONOIDS FROM VITEX AGNUS-CASTUS

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Key Word Index—*Vitex agnus-castus*; Verbenaceae; root bark; cytotoxic flavonoids; acylated luteolin *C* and *O*-glucosides.

Abstract—Four new flavonoids, luteolin 6-C-(4"-methyl-6"-O-trans-caffeoylglucoside), luteolin 6-C-(6"-O-trans-caffeoylglucoside), luteolin 6-C-(2"-O-trans-caffeoylglucoside), and luteolin 7-O-(6"-p-benzoylglucoside), together with four known ones 5, 4'-dihydroxy-3,6,7,3'-tetramethoxyflavone, luteolin, artemetin and isorhamnetin, were isolated from the root bark of *Vitex agnus-castus*. The structures were elucidated by spectroscopic means. © 1997 Elsevier Science Ltd

INTRODUCTION

Vitex agnus-castus (Verbenaceae), is a shrub which is widely distributed in the Middle East and southern Europe. Some iridoids and flavonoids have been isolated from the leaves or fruits [1]. During our study of the cytotoxic principles from this plant, four new flavonoids, luteolin 6-C-(4"-methyl-6"-O-trans-caffeoylglucoside) (1), luteolin 6-C-(6"-O-trans-caffeoylglucoside) (2), luteolin 6-C-(2"-O-trans-caffeoylglucoside) (3), and luteolin 7-O-(6"-p-benzoylglucoside) (4) together with four known ones, 4',5-dihydroxy-3,3',6,7-tetramethoxyflavone (5) [2], luteolin (6) [3], artemetin (7) [3-5] and isorhamnetin (8) [3] were isolated and applied to cytotoxic bioassay.

RESULT AND DISCUSSION

Luteolin 6-C-(4"-methyl-6"-O-trans-caffeoylglucoside) (1) was obtained as a pale yellow powder with a molecular formula of $C_{31}H_{28}O_{15}$ from $[M+H-CH_2+Na]^+$ at m/z 634 and $[M+H-CH_2]^+$ at m/z 611 in the FABMS. The NMR spectra showed the presence of a flavone skeleton with one methoxyl (δ_H : 3.35 s, δ_C : 49.7 q), one caffeoyl (δ_H : 6.26 d H-8", 7.53 d H-7", 6.88 dd H-6", 6.73 d H-5" and 6.99 d H-2") and one C-glucosyl group (δ_H : 4.95 d d = 9.9 Hz, H-1", δ_C : 75.4 d C-1"). The flavone skeleton was elucidated as 5,7,3',4'-tetrahydroxyflavone from the UV spectrum with various shift reagents. The C-glucose was assigned to C-6 from the cross peak between C5–OH-5, C10–OH-5, C6–OH-5, C6–H-1", C7–H-1" and

Luteolin 6-C-(6"-O-trans-caffeoylglucoside) (2) was obtained as a pale yellow powder with a molecular formula of $C_{30}H_{26}O_{15}$ from $[M+H+Na]^+$ at m/z 634 and $[M+H]^+$ ion at m/z 611 in the FABMS. The 1H and ^{13}C NMR spectra were quite similar to those of 1 except for the absence of the methoxyl signal.

Luteolin 6-C-(2"-O-trans-caffeoylglucoside) (3) was obtained as a pale yellow powder with the same molecular formula of $C_{30}H_{26}O_{15}$ as 2 from the $[M+Na]^+$ ion at m/z 633 in the FABMS. The NMR spectra were similar to those of 2, but the chemical shifts of C-2" and C-5" were shifted downfield in comparison with those of 2, and those of C-1", C-3", and C-6" were shifted upfield. Also, the chemical shifts of H-1", H-2" and H-3" in 3 were shifted downfield in comparison with those of 2, but those of H-5" and H-6" were shifted upfield. These results suggested that the caffeoyl group at C-6" in 2 is at C-2" in 3.

Luteolin 7-O-(6"-p-benzoylglucoside) (4) was obtained as a pale yellow powder with a molecular formula of $C_{28}H_{24}O_{13}$ from the 591[M+Na]⁺ ion at m/z 591 in the FAB mass spectrum. The NMR spectra showed the presence of one p-benzoyl group (δ_H : 7.84, d, H-2" and H-6"; 6.65, d, H-3" and -5"), one O- β -glucosyl (δ_H : 5.13, d, d = 7.3 Hz, H-1"; δ_C : 101.0, d, C-1"), and a luteolin skeleton as shown in Tables 1 and 2. The O-glucosyl group was confirmed to be

C5-H-1" in the HMBC spectrum measured in DMSO- d_6 . The methoxyl group was assigned to C-4" from the cross peak between C-4"-OCH₃ in the HMBC spectrum of 1. The caffeoyl group was assigned to C-6" from the cross peak between C-9"-H-6" in the HMBC spectrum and the *trans*-geometry was elucidated from the coupling constants between H-7" and H-8" (J = 15.9 Hz).

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Table 1. ¹H NMR data of flavonoids from Vitex agnus-castus (400 MH₂, δ, CD₃OD)

	1	2	3	4
3	6.42 s	6.51 s	6.41 s	6.59 s
6				6.54 d(2.2)
8	6.39 s*	6.47 s*	6.38 s*	$6.72 \ d(2.2)$
2'	7.27 br s	7.33 br s	7.26 br s	7.35 br s
5′	6.73 d (8.8)	6.88 d (8.9)	6.83 d (7.8)	6.88 d (8.9)
6′	7.26 br d (8.8)	7.34 br d (8.9)	$7.25 \ br \ d \ (7.8)$	7.34 br $d(\hat{3}.9)$
OMe	3.35 s		, ,	(****)
1"	4.95 d (9.9)	4.94 d (9.9)	5.12 d (10.0)	5.13 d (7.3)
2"	4.25 t (9.0)	4.23 t (8.9)	5.67 br	3.55
3"	3.56 m	3.53 m	3.77 t (9.0)	3.55
4"	3.56 m	3.53 m	$3.63 \ t \ (9.4)$	3.44 t (9.1)
5"	3.70 m	3.68 m	3.53 m	3.90 dt (9.8, 2.2)
6"	4.56 d (10.5)	4.54 d (10.2)	3.95 d (10.5)	4.74 dd (12.0, 2.2)
	4.40 dd (12.0, 5.4)	4.37 dd (12.1, 5.5)	3.82 dd (12.1, 5.3)	4.31 dd (11.9, 7.8)
2‴	$6.99 \ d \ (1.7)$	7.02 d(1.9)	6.90 d(1.4)	7.84 d (8.8)
3‴			` ,	6.65 d(8.8)
5‴	6.73 d (8.2)	6.75 d (8.2)	6.69 d (8.2)	6.65 d (8.8)
6‴	6.88 dd (8.2, 1.7)	6.92 dd (8.2, 2.0)	6.80 dd (8.3, 1.7)	7.84 d (8.8)
7‴	7.53 d (15.9)	7.56 d (15.9)	7.36 d(15.9)	, ,
8‴	6.26 d (15.9)	6.28 d(15.9)	6.05 d (15.9)	

^{*} Signals reduce when immersed in the solvent (CD₃OD) for long periods.

connected to C-7 from the UV spectrum with the absence of a shift of Band II in the presence of sodium acetate. The *p*-benzoyl group was established to be connected to C-6" from the chemical shifts of C-6" (δ_c : 65.2, t) and H-6" (δ : 4.74, dd; 4.31, dd).

Compounds 5–8 were identified as 4',5-dihydroxy-3,3',6,7-tetramethoxy-flavone, luteolin, artemetin and isorhamnetin, respectively, by comparing their physi-

- 1. R₁=H, R₂=CH₃, R₃=trans caffeoyl
- 2. $R_1=R_2=H$, $R_3=trans$ caffeoyl
- 3. R₁=trans caffeoyl, R₂=R₃=H

$$R_1$$
 R_2
 R_3
 R_4
 R_6

- 4. R_1 =O-Glc(6"-p-hydroxybenzoyl), R_2 = R_4 =H, R_3 = R_5 = R_6 =OH
- 5. $R_1=R_2=R_4=R_5=OCH_3$, $R_3=R_6=OH$
- 6. R₁=R₃=R₅=R₆=OH, R₂=R₄=H
- 7. $R_1=R_2=R_4=R_5=R_6=OCH_3$, $R_3=OH$
- 8. $R_1=R_3=R_4=R_6=OH$, $R_5=OCH_3$, $R_2=H$

cal and spectral data with those reported in literature, and from chemical evidence [2–5]. The IC₅₀ values (μ g ml⁻¹) of 1–7 showing cytotoxic activity against P388 lymphocytic leukemia cells were 7.6 in 1, 14 in 2, 56 in 3, 70 in 4, 0.1 in 5, 0.31 in 6 and 8.1 in 7.

EXPERIMENTAL

General. Mps were uncorr. 1 H and 13 C NMR: Bruker AM 400 and 500 MHz at 303K. NOESY experiments were made with mixing time of 0.6 sec and precessed on a Bruker data station with an Aspect 3000 computer. Silica gel CC was carried out on Merck Kieselgel 60 (70–230 mesh) at amounts equivalent to $100 \times$ the sample amount. MPLC was performed with a column (22 mm i.d. \times 300 mm) packed with 40 μ m silica gel or 20 μ m ODS. HPLC was

: HMBC correlations of compound 1 measured in DMSO

Table 2. ¹³C NMR data of flavonoids from *Vitex agnus*castus (100 MHz, δ, CD₃OD)

	1	2	3	4
2	166.1 s	166.3 s	166.2 s	166.9 s
3	103.8 d	104.0 d	103.9 d	103.2 d
4	183.9 s	184.0 s	183.8 s	184.1 s
5	162.1 s	162.2 s	161.4 s†	158.9 s
6	108.8 s	108.9 s	107.9 s	101.2 d
7	165.3 s	164.9 s	164.8 s	164.6 s
8	95.5 d†	95.3 d†	95.3 d†	96.1 d
9	158.7 s	158.8 s	158.8 s	163.1 s
0	105.1 s	105.3 s	$105.0 \ s$	107.1 s
1'	123.4 s	123.6 s	123.5 s	122.0 s
2'	114.1 d	114.2 d	114.2 d	114.4 d
3′	147.0 s*	$147.0 \ s$	146.9 s*	147.1 s
4'	151.1 s	151.0 s	151.0 s	151.3 s
5′	116.8 d	116.8 d	116.8 d	116.8 d
6′	120.4 d	120.4 d	120.4 d	120.6 d
ЭМе	49.7 q			
1"	75.4 d	75.5 d	73.3 d	$101.0 \ d$
2"	72.5 d	72.6 d	73.9 d	74.6 d
3"	79.9 d	$80.0 \ d$	78.0 d	77.8 d
4"	71.9 d	71.9 d	71.8 d	72.2 d
5"	79.9 d	80.0 d	82.8 d	75.7 d
6"	65.1 <i>t</i>	65.0 t	62.8 1	65.2 t
1‴	127.7 s	127.8 s	127.7 s	123.5 s
2""	115.1 d	115.2 d	115.1 d	132.9 d
3‴	146.7 s*	146.8 s	146.7 s*	116.2 d
4‴	149.6 s	149.6 s	149.5 s	163.6 s
5‴	116.5 d	116.5 d	116.5 d	116.2 a
6‴	123.1 d	123.1 d	122.9 d	132.9 a
7‴	147.2 d	147.2 d	$147.0 \ d$	168.1 s
8‴	114.9 d	115.0 d	114.9 d	
9‴	169.3 s	169.3 s	168.3 s	

^{*} Signals may be interchangable in each column.

performed with a Hibar RT RP-18 column (20 mm i.d. \times 250 mm) packed with 7 μ m ODS. The NMR coupling constants (*J*) are given in Hz.

Plant material. The fruits of Vitex agnus-castus L. (Verbenaceae) were collected at Israel in 1995. The plant was identified by Professor Dan Palevitch (Action Director, The Northern Research Center, Agricultural Research Organization, Israel). A voucher specimen has been deposited at the Department of Natural Medicines, Tokyo University of Pharmacy and Life Science.

Extraction and isolation. The fruits of V. agnuscastus L. (2 kg) were powdered and extracted × 3 with MeOH (3 l). The concd extract (207 g) was partitioned between n-hexane and H₂O, between CHCl₃ and H₂O, as well as n-BuOH and H₂O, successively. The n-BuOH soluble fr. (47 g) was subjected to HP-20 CC using H₂O, 40% MeOH, 60% MeOH, 80% MeOH and MeOH as eluting solvent systems to give five frs (Fr. A–E). Fr. D (16.9 g) was further chromatographed on a Sephadex LH-20 column, and then

a silica gel column eluted with CHCl₃-MeOH (1:1) and CHCl₃-MeOH-H₂O (8.9:1:0.1-6.7:3:0.3) successively and finally purified by means of ODS MPLC and HPLC with MeOH-H₂O or CH₃CN-H₂O as eluants to give 1 (46 mg), 2 (13 mg), 3 (32 mg), 4 (2.5 mg), 6 (11 mg), and 8 (1.7 mg). Also, the *n*-hexane extract (55 g) was subjected to silica gel CC and eluted with *n*-hexane-EtOAc (10:0-0:10), followed with EtOAc-MeOH (1:1). The EtOAc-MeOH (1:1) eluate was further isolated with ODS MPLC and finally purified with ODS HPLC using 80% MeOH to give 5 (2.4 mg) and 7 (25 mg).

Luteolin 6-C-(4"-methyl-6"-O-trans-caffeoylgluco-side) (1). Yellow powder, mp 224–226° (from MeOH), $[\alpha]_D$ –44.1° (MeOH; c 0.29), UV λ_{max}^{MeOH} nm: 245, 272 (sh), 295 (sh), 336; NaOMe: 269, 389; AlCl₃: 273, 302 (sh), 371 (sh), 425; AlCl₃–HCl: 281, 297 (sh), 340, 386 (sh); NaOAc: 274, 289 (sh), 332; NaOAc-H₃BO₃: 260, 305 (sh), 358; FABMS m/z: 634 [M+H-CH₂+Na]⁺, 611 [M+H-CH₂]⁺, 575, 517, 445, 391, 329, 163, 135 [100]; ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

Luteolin 6-C-(6"-O-trans-caffeoylglucoside) (2). Yellow powder, mp 224–226° (from MeOH), [a]_D – 51.0° (MeOH; c 0.49), UV λ_{max}^{MeOH} nm: 245, 272 (sh), 295 (sh), 336; NaOMe: 267, 389; AlCl₃: 273, 302 (sh) 370 (sh), 424; AlCl₃–HCl: 281, 298 (sh), 339, 386 (sh); NaOAc: 274, 333; NaOAc–H₃BO₃: 261, 297 (sh), 357; FABMS m/z: 634 [M+H+Na]⁺, 611 [M+H]⁺, 531, 443, 391, 329, 299, 177, 149 [100]; ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

Luteolin 6-C-(2"-O-trans-caffeoylglucoside) (3). Yellow powder, mp 213–215° (from MeOH), $[o]_D$ –239.3° (MeOH; c 0.57), UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 246, 272 (sh), 295 (sh), 335; NaOMe: 235 (sh), 278, 337 (sh), 389; AlCl₃: 273, 304 (sh), 380, 420; AlCl₃–HCl: 261 (sh), 278, 298 (sh), 342, 386 (sh); NaOAc: 279, 330; NaOAc-H₃BO₃: 260, 305 (sh), 357; FABMS m/z: 633 [M+Na]⁺, 353, 329, 193, 163 [100], 135; ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

Luteolin 7-O-(6"-p-benzoylglucoside) (4). Yellow powder, mp > 300° (from MeOH), $[\alpha]_D$ -78.6° (MeOH; c 0.028), UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 255, 349; NaOMe: 269 (sh), 296, 402; AlCl₃: 268, 295 (sh), 335 (sh), 427; AlCl₃-HCl: 260, 294 (sh), 355 (sh), 386; NaOAc: 258, 365; NaOAc-H₃BO₃: 258, 373; FABMS m/z: 591 [M+Na]⁺, 531, 429, 385, 287, 147 [100]; ¹H and ¹³C NMR spectra are listed in Tables 1 and 2.

Bioassay of cytotoxic activity against P388 cells. MTT (3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide) colorimetric assay was performed in a 96-well plate [6]. The assay is based on the reduction of MTT by the mitochondrial dehydrogenase of viable cells to give a blue formazan product which can be measured spectrophotometrically. Mouse P388 leukemia cells $(2 \times 10^4 \text{ cell ml}^{-1})$ were inoculated in each well $100 \ \mu \text{l ml}^{-1}$ of RPMI-1640 medium (Nissui Pharm. Co., Ltd) supplemented with 5% fetal calf serum. (Mitsubishi Chemical Industry Co., Ltd) and kanamycin $(100 \ \mu \text{g ml}^{-1})$ at 37° in a humidified atmo-

 $[\]dagger$ Signals reduce when immersed in the solvent (CD3OD) for long periods.

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sphere of 5% CO₂. Various drug concs (10 μ l) were added to the cultures at day 1 after transplantation. At day 3, 20 μ l of the MTT soln (5 mg ml⁻¹) per well was added to each cultured medium. After a further 4 hr of incubation 100 μ l of 10% sodium dodecyl sulphate-0.01 N HCl soln was added to each well and the formazan crystals in each well were dissolved by stirring with a pipette. The optical density measurements were made using a microplate reader (Tohso MPR-A4i) at two wavelengths (550 and 700 nm). In all these experiments 3 replicate wells were used to determine each data point.

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