PII: S0031-9422(97)00350-6

LAHADININES A AND B, NEW CYANO-SUBSTITUTED INDOLE ALKALOIDS FROM KOPSIA PAUCIFLORA

TOH-SEOK KAM* and K. YOGANATHAN

Department of Chemistry University of Malaya, 59100 Kuala Lumpur, Malaysia

(Received in revised form 10 March 1997)

Key Word Index—Kopsia pauciflora; Apocynaceae; leaves; cyano-substituted indole alkaloids.

Abstract—Two new alkaloids of the aspidofractinine-type bearing cyano substituents at carbon-21 were obtained from *Kopsia pauciflora*, together with two other new alkaloids with 21-hydroxy-substitution. The structures of these alkaloids were deduced from their spectral data. © 1997 Elsevier Science Ltd

INTRODUCTION

The genus Kopsia includes some 30 species found mostly in tropical Asia [1, 2]. There are ca 17 species which occur in Malaysia. The phytochemistry of this genus has been quite well investigated. Our own studies of this genus have resulted in several novel structures and some important bioactivities [3–6]. Kopsia pauciflora Hook f. is endemic to Sabah in North Borneo and in continuation of our studies of this species [7], we now report four new alkaloids from the leaf extract.

RESULTS AND DISCUSSION

Lahadinine A (1) was obtained as a colourless oil. The UV spectrum showed absorption maxima at 227, 248, 255 (sh), 285 and 290 nm (log ε 4.36, 3.89, 3.84, 2.97 and 2.92, respectively) typical of a dihydroindole chromophore and is similar to that of other aspidofractinine-type compounds [8, 9]. The EI mass spectrum of 1 showed a $[M]^+$ at m/z 481, which was also the base peak, the odd mass indicating the presence of a third nitrogen (C₂₅H₂₇N₃O₇). This was further supported by the 25 separate carbon resonances in ¹³C NMR, as well as the observation of the fragments at m/z 455 and 454 attributable to loss of CN and HCN, respectively, in addition to peaks at m/z 463 [M-H₂O]⁺, 422 [M-CO₂Me]⁺ and 394 [M-CO₂Me- $CH_2 = CH_2$]⁺. The IR spectrum showed a weak band at 2252 cm⁻¹. The ¹H and ¹³C NMR spectral data (Tables 1 and 2) showed the presence of a methylenedioxy substituent at C-11 and C-12 (a pair of aromatic AB doublets at $\delta_{\rm H}$ 6.57 and 6.99 and another

1 R¹, R² = -O-CH₂-O-, R³ = CN
2 R¹ = R² = OMe, R³ = CN
3 R¹, R² = -O-CH₂-O-, R³ = OH
4 R¹, R² = -O-CH₂-O-, R³ = OH, N(4)
$$\rightarrow$$
O
5 R¹, R² = -O-CH₂-O-, R³ = H

pair at $\delta_{\rm H}$ 5.87 and 5.93, $\delta_{\rm C}$ 100.5), a urethane function at $N_{\rm I}$ ($\delta_{\rm C}$ 155.7), one hydroxyl group (16-OH, $\delta_{\rm H}$ 6.94) and an ester group ($\delta_{\rm C}$ 172.3). The spectra can be assigned with the aid of COSY and HETCOR and showed that 1 is an aspidofractinine-type alkaloid [8]. The NMR spectral data resembles those of the aspidofractinine derivative N-methoxycarbonyl-11,12-methylenedioxykopsinaline 5 [10, 11], which was also isolated from the stem extract of the same plant [7], except for the conspicuous absence of the H-21 signal and the appearance instead of an additional resonance in the $^{13}{\rm C}$ NMR spectrum at $\delta_{\rm C}$ 117.9. This resonance

R¹ R² CO₂Me CO₂Me OH

^{*} Author to whom correspondence should be addressed.

786

Table 1. ¹H NMR spectral data for compounds (1-4) (270 MHz, CDCl₃)*

Н	1	2	3	4
3	3.05 m	3.04 m	2.81 m	3.21 m
3	3.05 m	3.04 m	3.15 m	3.51 br d (14)
5	2.94 ddd (9, 7, 2)	2.94 ddd (9, 7, 2)	3.01 m	3.31 m
5	3.07 m	3.06 m	3.15 m	3.77 m
6	1.51 m	1.60 m	1.65 m	2.13 m
6	1.93 m	$2.00 \ m$	2.05 m	2.13 m
9	6.99 d (8)	7.17 d(8)	6.75 br s	7.60 d(8)
10	6.57 d(8)	6.63 d (8)	6.54 d(8)	6.51 d(8)
14	1.43 m	1.45 m	1.25 m	1.71 m
14	1.72 m	1.74 m	1.76 m	1.71 m
15	1.60 m	1.62 m	1.46 m	1.56 br d (12)
15	1.60 m	1.62 m	1.70 m	1.65 m
17	1.49 br d (15)	1.51 br d (15)	1.56 br d (15)	1.46 br d (15)
17	3.05 m	3.04 m	2.95 br d (15)	2.87 br d (15)
18	$1.80 \ m$	1.86 m	1.67 m	1.67 m
18	2.37 m	2.40 m	2.29 m	2.26 m
19	1.51 m	1.60 m	1.65 m	1.66 m
19	1.89 m	1.94 m	1.65 m	1.66 m
16-OH	6.94 s	6.60 s	6.98 s	7.23 s
OCH ₂ O	5.93 d (1.5)		5.93 d (1.5)	5.93 d (1.5)
OCH₂O	5.87 d(1.5)	_	5.87 d(1.5)	5.85 d(1.5)
NCO ₂ Me	3.87 s	3.89 s	3.88 s	3.87 s
CO ₂ Me	3.73 s	3.75 s	3.75 s	3.76 s
11- OM e	_	3.75 s		_
12-OMe		3.82 s		

^{*} Assignments based on COSY and HETCOR.

Table 2. ¹³C NMR spectral data for compounds (1-4) (67.5 MHz, CDCl₁)*

C	1	2	3	4
2	73.7	73.9	74.6	73.6
3	44.2	44.4	43.0	62.2
5	49.0	49.2	47.7	60.2
6	39.0	39.3	36.6	33.4
7	59.8	59.8	59.7	58.8
8	130.7	133.8	131.9	129.7
9	116.8	117.9	116.3	120.1
10	103.6	107.1	103.8	104.1
11	149.0	153.7	148.4	148.4
12	134.3	138.0	134.8	133.5
13	123.2	133.8	124.1	123.9
14	19.9	20.4	17.4	18.5
15	32.4	32.5	30.3	29.7
16	74.3	75.4	75.2	74.7
17	39.1	39.5	41.8	43.1
18	23.9	24.9	23.8	22.6
19	28.4	28.4	28.9	28.9
20	36.6	36.7	36.4	36.9
21	68.4	68.5	90.6	103.2
NCO ₂ Me	53.3	53.5	53.2	53.2
CO ₂ Me	52.5	52.5	52.5	52.8
CO ₂ Me	172.3	172.6	173.0	172.8
NCO ₂ Me	155.7	157.1	155.9	156.1
$O\overline{C}H_2O$	100.5		100.3	100.2
11-OMe	_	60.2		_
12-OMe	_	56.1	_	_
CN	117.9	118.6		_

^{*} Assignments based on HETCOR and HMBC.

is due to a quaternary carbon and can be attributed to a cyano group [12] (the molecular formula of 1 shows that it differs from 5 by replacement of H by CN). The cyano substituent can a priori be placed on either of the two quaternary centres, 21 or 16. Placement of the OH at position 16 is consistent with the NMR data and is in agreement with the many aspidofractinine compounds reported, which have the same substitution pattern and stereochemistry at position 16 of the aspidofractinine skeleton. Furthermore, the assignment of the carbon-21 resonance is confirmed by HMBC; from the observed ³J interactions between C-21 and H-5, H-17 and H-19. The observed chemical shift of carbon-21 ($\delta_{\rm C}$ 68.4) firmly rules out the alternative placement of the hydroxyl substituent on C-21 and the cyano substituent on C-16, especially since the 21-OH substituted compound was also obtained (compound 3, vide infra) where the 21-carbon resonance was observed at δ 90.6. Based on the above considerations, lahadinine A is therefore the 21-cyano derivative of 5. Similarly, inspection of the spectral data and comparison with compound 6 [10, 11], which also occurs in the leaf extract, revealed lahadinine B 2 to be the 11,12-dimethoxy congenor of compound 1. In addition to the cyano-substituted compounds 1 and 2, two other related alkaloids were also obtained, viz., paucifinine (3) and paucifinine Noxide (4). Paucifinine is similar in all respects to 1 except for replacement of the 21-cyano group by a hydroxyl group. This is supported by its molecular Short Reports 787

formula (EI mass spectrum, m/z 472, $C_{24}H_{28}N_2O_8$) and the NMR data (Tables 1 and 2), in particular the presence of an oxygenated quaternary carbon attributable to C-21 ($\delta_{\rm C}$ 90.6). The identification of compound 4 as the N-oxide of 3 is based on the characteristic downfield shifts of the carbon resonances for C-3, C-5 and C-21 when compared with those of 3. Further confirmation is also provided by the ready oxidation (mCPBA) of 3 to give 4. The stereochemistry of the 21-substituent in all four compounds (1-4) is assumed to be α , in common with the stereochemistry at position 21 of the parent (unsubstituted) compounds 5 and 6 (21- α H), which also occur in the same plant [7], as well as with the many other aspidofractinine alkaloids occurring in Kopsia, assuming they share a common biogenetic origin [7-13]. The isolation of the rare cyano-substituted alkaloids 1 and 2 represents the first isolation of this class of compounds in Kopsia, although a cyano-substituted indole alkaloid has been previously reported from Alstonia angustiloba [12].

EXPERIMENTAL

Plant material. Plant material was collected in Sabah, Malaysia. Herbarium voucher specimens are deposited at the Herbarium of the Sabah Forest Department, Sandakan, Sabah, Malaysia.

Extraction and isolation. Extraction of alkaloids was carried out in the usual manner, as described in detail elsewhere [9]. Alkaloids were isolated by CC and centrifugal TLC on silica gel. Solvent systems used for CC were CHCl₃-MeOH and Et₂O. Solvent systems used for centrifugal TLC were Et₂O-hexane (1:1), Et₂O-EtOAc (10:1) and MeOH-CHCl₃ (1:9). The yields (g kg⁻¹) of the alkaloids from the leaf extract were: 1 (0.009), 2 (0.006), 3 (0.011) 4 (0.017), 5 (0.019) and 6 (0.009).

Lahadinine A (1). $[\alpha]_D - 184^\circ$ (CHCl₃, c 0.15). UV (EtOH), λ_{max} nm (log ε): 227 (4.36), 248 (3.89), 255 (3.84), 285 (2.97), 290 (2.92). IR (dry film) ν_{max} cm⁻¹ 3303, 2252, 1737, 1686. EIMS m/z (rel. int.): 481 [M⁺ C₂₅H₂₇N₃O₇] (100), 463 (15), 455 (10), 454 (10), 422 (40), 394 (20), 379 (10), 352 (75) 322 (10). 1 H and 13 C NMR: Tables 1 and 2.

Lahadinine B (2). $[\alpha]_D - 123^{\circ}C$ (CHCl₃, c 0.03). UV (EtOH), λ_{max} nm (log ε) 222 (4.39), 253 (3.79), 284 (3.06), 289 (3.03). IR (dry film) ν_{max} cm⁻¹: 3327, 2252, 1737, 1678. EIMS m/z (rel. int.): 497 [M⁺, $C_{26}H_{31}N_3O_7$] (100), 482 (50), 471 (5), 470 (5), 467 (10), 438 (25), 410 (20), 395 (15), 380 (15), 368 (40). ¹H and ¹³C NMR: Tables 1 and 2.

Paucifinine (3). $[\alpha]_D - 91^\circ$ (CHCl₃, c 0.08). UV (EtOH), λ_{max} nm (log ε): 227 (4.51), 248 (4.08), 255 (4.03), 285 (3.13), 290 (3.09). IR (dry film) ν_{max} cm⁻¹:

3302, 1733, 1681. EIMS m/z (rel. int.): 472 [M⁺, $C_{24}H_{28}N_2O_8$] (35), 454 (100), 446 (10), 413 (10), 395 (10), 364 (15), 351 (10), 323 (8), 262 (10), 246 (6), 124 (4), 109 (5). ¹H and ¹³C NMR: Tables 1 and 2.

Paucifinine-N-oxide (4). [α]_D -36° (CHCl₃, c 0.30). UV (EtOH), $\lambda_{\rm max}$ nm (log ε): 227 (4.50), 248 (4.06), 255 (3.97), 285 (3.21), 290 (3.17). IR (dry film) $\nu_{\rm max}$ cm⁻¹ 3291, 1733, 1681. EIMS m/z (rel. int.): 488 [M⁺, C₂₄H₂₈N₂O₉] (25), 470 (100), 454 (70), 428 (10), 411 (35), 395 (15), 369 (10), 341 (10), 246 (15), 124 (5) 109 (5). 1 H and 13 C NMR: Tables 1 and 2.

Oxidation of 3 to 4. mCPBA (2 mg, 0.011 mmol) was added to a stirred soln of 3 (5 mg, 0.01 mmol) in CH_2Cl_2 (10 ml) at 4°. After ca 10 min, satd Na_2CO_3 soln (10 ml) was added and the mixt. extracted with CH_2Cl_2 (3×10 ml). The extract was then dried (Na_2SO_4) and the solvent evapd off. Centrifugal chromatography over silica gel (MeOH–CHCl₃) afforded 3 mg (58%) of 4.

Acknowledgements—We thank the University of Malaya and IRPA for financial support of this work and Dr K. M. Wong, Leopold Madani and Julius Kulip of the Sabah Forest Department for assistance during the collection of plant material.

REFERENCES

- 1. Markgraf, F., Blumea, 1972, 20, 416.
- Sevenet, T., Allorge, L., David, B., Awang, K., Hadi, A. H. A., Kan-Fan, C., Quirion, J. C., Remy, F., Schaller, H. and Teo, L. E., *Journal of Ethnopharmacology*, 1994, 41, 147.
- Kam, T. S., Yoganathan, K., Koyano, T. and Komiyama, K., Tetrahedron Letters, 1996, 37, 5765.
- 4. Kam, T. S., Yoganathan, K. and Chen Wei, Tetrahedron Letters, 1996, 37, 3603.
- 5. Kam, T. S., Yoganathan, K. and Chuah, C. H., Tetrahedron Letters, 1995, 36, 759.
- Kam, T. S., Yoganathan, K. and Chuah, C. H., Tetrahedron Letters, 1994, 35, 4457.
- Kam, T. S., Arasu, L. and Yoganathan, K., *Phytochemistry*, 1996, 43, 1385.
- 8. Kam, T. S., Yoganathan, K., Chuah, C. H. and Chen Wei, *Phytochemistry*, 1993, 32, 1343.
- Kam, T. S. and Tan, P. S., Phytochemistry, 1990, 29, 2321.
- Feng, X. Z., Kan, C., Potier, P., Kan, S. K. and Lounasmaa, M., *Planta Medica*, 1983, 48, 280.
- 11. Zheng, J. J., Zhou, Y. L. and Huang, Z. H., Acta Chimica Sinica, English Edition, 1989, 2, 168.
- Zeches, M., Ravao, T., Richard, B., Massiot, G., Le Men-Olivier, L. and Verpoorte, R., *Journal of Natural Products*, 1987, 50, 714.
- 13. Kam, T. S. and Yoganathan, K., *Phytochemistry*, 1996, **42**, 539.